

1 **Organic geochemistry of the early Toarcian oceanic anoxic event in Hawsker**
2 **Bottoms, Yorkshire, England**

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15 **Abstract**

16 A comprehensive organic geochemical investigation of the Hawsker Bottoms
17 outcrop section in Yorkshire, England has provided new insights about environmental
18 conditions leading into and during the Toarcian oceanic anoxic event (T-OAE; ~183 Ma).
19 Rock-Eval and molecular analyses demonstrate that the section is uniformly within the
20 early oil window. Hydrogen index (HI), organic petrography, polycyclic aromatic
21 hydrocarbon (PAH) distributions, and tricyclic terpane ratios mark a shift to a lower
22 relative abundance of terrigenous organic matter supplied to the sampling locality during
23 the onset of the T-OAE and across a lithological transition. Unlike other ancient intervals
24 of anoxia and extinction, biomarker indices of planktonic community structure do not
25 display major changes or anomalous values. Depositional environment and redox
26 indicators support a shift towards more reducing conditions in the sediment porewaters
27 and the development of a seasonally stratified water column during the T-OAE. In
28 addition to carotenoid biomarkers for green sulfur bacteria (GSB), we report the first
29 occurrence of okenane, a marker of purple sulfur bacteria (PSB), in marine samples
30 younger than ~1.64 Ga. Based on modern observations, a planktonic source of okenane's
31 precursor, okenone, would require extremely shallow photic zone euxinia (PZE) and a
32 highly restricted depositional environment. However, due to coastal vertical mixing, the
33 lack of planktonic okenone production in modern marine sulfidic environments, and
34 building evidence of okenone production in mat-dwelling Chromatiaceae, we propose a
35 sedimentary source of okenone as an alternative. Lastly, we report the first parallel
36 compound-specific $\delta^{13}\text{C}$ record in marine- and terrestrial-derived biomarkers across the
37 T-OAE. The $\delta^{13}\text{C}$ records of short-chain *n*-alkanes, acyclic isoprenoids, and long-chain *n*-

38 alkanes all encode negative carbon isotope excursions (CIEs), and together, they support
39 an injection of isotopically light carbon that impacted both the atmospheric and marine
40 carbon reservoirs. To date, molecular $\delta^{13}\text{C}$ records of the T-OAE display a negative CIE
41 that is smaller in magnitude compared to the bulk organic $\delta^{13}\text{C}$ excursion. Although
42 multiple mechanisms could explain this observation, our molecular, petrographic, and
43 Rock-Eval data suggest that variable mixing of terrigenous and marine organic matter is
44 an important factor affecting the bulk organic $\delta^{13}\text{C}$ records of the T-OAE.

45

46 Keywords: Toarcian oceanic anoxic event; lipid biomarkers; okenane; photic zone
47 euxinia, stable carbon isotopes; Hawsker Bottoms

48 **1. Introduction**

49

50 Several transient episodes of enhanced deposition and preservation of organic-
51 rich sediments punctuated the Mesozoic Era. A combination of factors may have caused
52 these intervals, known as oceanic anoxic events (OAEs), including greenhouse conditions
53 and enhanced marine productivity (e.g. Schlanger and Jenkyns, 1976; Jenkyns, 1980;
54 1988; 2010; Trabucho-Alexandre et al., 2010). The first Mesozoic OAE was the Early
55 Jurassic Toarcian OAE (T-OAE; ~183 Ma), which was associated with elevated
56 extinction rates, enhanced weathering rates, warm temperatures, ocean acidification, and
57 a negative carbon isotope excursion (CIE) (Hesselbo et al., 2000; Cohen et al., 2004;
58 Bambach, 2006; Hesselbo et al., 2007; Jenkyns, 2010; Kiessling and Simpson, 2011).
59 The duration of the T-OAE is not precisely known but may have lasted on the order of
60 several hundred thousand years (Kemp et al., 2005; Suan et al., 2008; Kemp et al., 2011).
61 The Karoo and Ferrar igneous provinces, which erupted at 183 ± 1 Ma, may have
62 coincided with the Pliensbachian-Toarcian extinction (Pálffy and Smith, 2000; Courtillot
63 and Renne, 2003). However, better radiometric dating of the volcanism and OAE are
64 required to confidently link these two events. Although multiple mechanisms have been
65 proposed to account for the Toarcian negative CIE, including methane hydrate
66 dissociation, upwelling of isotopically light waters, thermogenic release of methane, and
67 biomass burning (Hesselbo et al., 2000; Schouten et al., 2000; McElwain et al., 2005; van
68 de Schootbrugge et al., 2005; Finkelstein et al., 2006), the source of the isotopically light
69 carbon remains unclear.

70 The analysis of sedimentary organic matter provides the opportunity to evaluate
71 environmental and ecological responses to carbon cycle perturbations, as well as
72 potentially constraining the perturbation itself. Previous organic geochemical work across
73 the T-OAE has indicated changes in planktonic community structure and redox
74 chemistry, particularly the development of photic zone euxinia (PZE) (e.g. Farrimond et
75 al., 1989; 1994; Schouten et al., 2000; Pancost et al., 2004; Schwark and Frimmel, 2004;
76 Bowden et al., 2006; van Breugel et al., 2006). However, since biomarker records can
77 reflect local responses, additional comprehensive organic geochemical studies from
78 multiple locations are required to build a global perspective of ecological and
79 environmental change associated with the T-OAE. Here, we investigate the temporal
80 variation of lipid biomarkers, Rock-Eval data, organic petrography, and compound-
81 specific carbon isotopes from the Lower Jurassic section at Hawsker Bottoms, Yorkshire,
82 England.

83

84 **2. Geologic Setting and Site Description of Hawsker Bottoms, Yorkshire, England**

85

86 A well-studied section of the Toarcian OAE is located on the Yorkshire coast in
87 northern England. We analyzed sample splits spanning 14 m of an organic-rich, lower
88 Jurassic outcrop section in the Cleveland Basin at Hawsker Bottoms previously studied
89 by Hesselbo et al. (2000). The lithology is dominated by black shales containing discrete
90 levels of calcite concretions and constitutes the Jet Rock *sensu stricto* (Hesselbo and
91 Jenkyns, 1995). The sections around Hawsker Bottoms have been used for defining the
92 ammonite biostratigraphy of the Toarcian (Howarth, 1992).

93 The Early Jurassic paleogeography of the area, although somewhat uncertain, is
94 depicted in published paleogeographic maps (e.g. Bradshaw et al., 1992). The Cleveland
95 Basin of North Yorkshire was part of a system of shallow epicontinental seas and small
96 extensional tectonic basins linked to the Central Graben via the Sole Pit Basin. The
97 region formed part of the broad epicontinental sea that covered much of northwest
98 Europe. Marine sedimentation was initiated during the Late Triassic, and a succession of
99 marine siliciclastic mudstones accumulated during the Early Jurassic.

100 The Grey Shale Member of the Whitby Mudstone Formation consists of
101 bioturbated, silty mudstones with beds of calcareous siderite concretions. The mudstones
102 have thin sharp-based beds, wave ripple, and starved ripple laminations (Wignall et al.,
103 2005; Ghadeer and Macquaker, 2011). Grain size and bioturbation intensity decrease
104 toward the top of the unit, and sediment color darkens. The Jet Rock Member consists of
105 dark, organic matter-rich, fissile mudstones with abundant ammonites and horizons of
106 calcareous nodules. The boundary between these two members of the Whitby Mudstone
107 Formation likely represents an increase in water depth in the basin.

108 The early Toarcian (*D. tenuicostatum* Zone) was a period of major basin
109 subsidence throughout England. Organic matter content fluctuates through the Grey
110 Shales, but increasing levels of organic matter are present from the *D. semicelatum*
111 Subzone to the *C. exaratum* Subzone (*H. falciferum* Zone). Minor shoaling cycles with
112 striped siltstone laminae suggest that water depths were on the order of tens of meters
113 (Powell, 2010). Similarly, sedimentary structures suggest deposition during storms by the
114 effects of waves (Wignall et al., 2005; Ghadeer and Macquaker, 2011). Thus, bottom
115 water conditions were more energetic than is commonly thought, where the water column

116 was likely shallower than 50 m. Consequently, Hawsker Bottoms likely represents an
117 inner continental shelf environment, which physical oceanographers define as the region
118 where turbulence from the surface and bottom boundary layers effectively homogenizes
119 the whole water column (Lentz and Fewings, 2012). Accordingly, inner shelf
120 environments are typically a few meters to tens of meters deep

121 The abundance of ammonites in the shales indicates that the water column was at
122 times oxygenated and favorable to nektonic faunas (Powell, 2010). The abundance of thin
123 beds with tops homogenized by bioturbation suggests that long-term, persistent bottom
124 water anoxia did not occur in the basin (Ghadeer and Macquaker, 2011). Besides
125 deposition as bedload by geostrophic flows and density currents, additional sediment was
126 supplied by suspension settling. Textural analyses have shown that much of the
127 sedimentary organic matter was delivered to the seafloor as fecal pellets, flocs, or other
128 organo-mineralic aggregates (Ghadeer and Macquaker, 2011). The contribution of a
129 biogenic component to rock composition varies, and the differences have been attributed
130 to a changing balance of primary production relative to dilution and length of transport
131 path during deposition (Macquaker and Taylor, 1996; Wignall et al., 2005; Ghadeer and
132 Macquaker, 2011).

133

134 **3. Methods**

135 Powdered rock samples were analyzed by Rock-Eval pyrolysis. The total organic
136 carbon (TOC; %), T_{\max} (°C), S_1 , and S_2 were determined and used to calculate the
137 hydrogen index (HI) and production index (PI). Kerogen isolates from four samples
138 across the section were mounted onto slides in duplicate and assessed optically under

139 white light and fluorescent light using a Zeiss research microscope and a Zeiss x 40
140 Plank-Neofluar objective. A Zeiss Axioskop, Axio Image D1, and a Zeiss 18 filter set
141 were used to take photomicrographs and fluorescence images.

142 Powdered samples (~ 5 g) were extracted using a Dionex ASE 200 Accelerated
143 Solvent Extractor with a solvent mixture of dichloromethane:methanol 9:1 (v/v).
144 Elemental sulfur was removed from the total lipid extract (TLE). Asphaltenes were
145 separated from the maltene fraction, which was then separated into saturated, aromatic,
146 and polar fractions by silica gel chromatography. The saturated and aromatic fractions
147 were analyzed by gas chromatography-mass spectrometry (GC-MS) and gas
148 chromatography-metastable reaction monitoring-mass spectrometry (GC-MRM-MS).
149 Carbon isotopic measurements of saturated hydrocarbons were made by gas
150 chromatography/combustion/ isotope ratio mass spectrometry (GC-C-IRMS) using a
151 ThermoFinnigan Delta Plus XP coupled to a ThermoFinnigan Trace GC. The mean value
152 of triplicate analyses are reported here in per mil (‰) relative to Vienna Pee Dee
153 belemnite (VPDB), and the standard deviation from the mean value was 0.4‰ or less. A
154 detailed description of methods is included in the supplementary online material (SOM).
155

156 **4. Results and Discussion**

157 *4.1 Rock-Eval analysis*

158 Rock-Eval results provided insight into the thermal maturity and type of organic
159 matter preserved in the Hawsker Bottoms sediments. The TOC percentage (Hesselbo et
160 al., 2000) was plotted for comparison with the HI, PI, and T_{\max} (fig. 1). The narrow range
161 of the PI (0.11-0.18) and T_{\max} (429-440°C) parameters indicate that thermal maturity is

162 uniform through the section and at the early stage of oil generation (Peters et al., 2005).
163 Molecular indices of thermal maturity further substantiate this conclusion (see section
164 4.2).

165 The HI data reveal that the type of organic matter undergoes a transition that
166 appears to coincide with a lithological transition from medium grey shale to dark grey
167 thin-bedded shale within the limitations of our sampling resolution. The low HI values
168 below -2.5 m are characteristic of type III kerogen, whereas the higher HI values above -
169 3 m are characteristic of type II kerogen (Peters et al., 2005). Type III kerogen is
170 dominated by either terrigenous or highly degraded organic matter, and type II kerogen is
171 typically derived from marine organic matter (Peters et al., 2005).

172 Published HI values from other T-OAE localities also increase across the
173 initiation of the CIE (Prauss et al., 1991; Schouten et al., 2000; Röhl et al., 2001;
174 Sabatino et al., 2009; Suan et al., 2011). Previous workers have attributed the HI
175 variability to different degrees of organic matter degradation under varying redox
176 conditions (Schouten et al., 2000; Röhl et al., 2001; Sabatino et al., 2009). Alternatively,
177 others have argued that the HI variability represents a shift in the composition of the
178 organic matter (Suan et al., 2011). Petrographic and molecular evidence for the presence
179 of plant-derived material in these sediments (see section 4.3) supports the conclusion that
180 lower HI values at the bottom of the section are due to a larger abundance of terrigenous
181 organic matter relative to marine organic matter.

182

183 *4.2 Molecular indicators of thermal maturity*

184 The thermal history of the section was further assessed according to molecular
185 thermal maturity parameters (fig. 2). The C₃₁ hopane 22S/(22S+22R) ratio was constant
186 through the section and exhibited a narrow range between 0.58 and 0.59. A value of
187 ~0.55 represents the endpoint which is reached around the main phase of oil generation
188 (Peters et al., 2005). The C₃₀ hopane $\beta\alpha/(\beta\alpha+\alpha\beta)$ ratio ranged from 0.08 to 0.11, which is
189 close to values indicative of a mature source rock (Peters et al., 2005). The C₂₉ sterane
190 $\alpha\alpha\alpha$ 20S/(20S+20R) ratio varied from 0.52 to 0.57, which is comparable to the endpoint
191 value of 0.52-0.55 (Peters et al., 2005). In summary, molecular thermal maturity
192 indicators corroborate the Rock-Eval results, further supporting a uniform thermal
193 maturity within the early window of oil generation.

194 Some biomarker-based thermal maturity parameters can be influenced by
195 additional factors such as source and diagenetic effects (Moldowan et al., 1986; Dahl et
196 al., 1993; Peters et al., 2005; Bennett and Olsen, 2007; French et al., 2012). Indeed, two
197 of the thermal maturity parameters presented in figure 2 exhibit some variation tracking
198 changes in lithology and source input, despite the multiple lines of evidence supporting
199 constant thermal maturity through the section. The Ts/(Ts+Tm) ratio, where Tm is C₂₇ 17
200 α -trisorhopane and Ts is C₂₇ 18 α -trisorhopane, varied from 0.41 to 0.59, whereas the
201 diasterane/sterane ratio of C₂₇₋₂₉ compounds ranged from 1.08 to 1.44. Both ratios deviate
202 from a relatively constant pattern in the lower ~2 meters of the section. This pattern is
203 explained by changes in lithology and/or organic matter source input, which is consistent
204 with lower HI values and additional evidence supporting variable terrigenous organic
205 input (see section 4.3).

206

207 4.3 Biomarker and petrographic evidence of terrigenous organic matter input

208 A combination of molecular and petrographic analyses was performed to evaluate
209 the relative contribution of terrigenous organic matter through the sampling interval (fig.
210 3). Previous work has suggested a terrigenous source for C₁₉ and C₂₀ tricyclic terpanes,
211 which has led to the use of C₁₉/C₂₃ and C₂₀/C₂₃ tricyclic terpane ratios to identify input of
212 terrigenous organic matter (Noble et al., 1986; Peters et al., 2005). These two ratios
213 display higher values in the lowest part of the section and decrease after the lithological
214 transition at -2.5 meters, indicating relatively greater terrigenous organic matter input in
215 the lowermost part of the sampling interval.

216 A wide range of PAHs was detected in the Hawsker Bottoms samples, including
217 phenanthrene, fluoranthene, pyrene, benzo[a]anthracene, triphenylene, chrysene,
218 benzo[b]fluoranthene, benzo[k]fluoranthene, benzo[e]pyrene, indeno[c,d]pyrene,
219 dibenzo[a,h]anthracene, benzo[g,h,i]perylene, coronene, and retene. PAHs are a diverse
220 set of compounds with multiple documented sources including products of pyrolysis,
221 combustion, hydrothermal activity, and igneous intrusion as well as direct inputs from
222 algae, fungi, vascular plants, and extraterrestrial organics (Kawka and Simoneit, 1990;
223 George, 1992; Jiang et al., 2000; Sephton et al., 2005; Grice et al., 2007; Marynowski and
224 Simoneit, 2009). PAHs have been used to reconstruct the history of wildfires, higher
225 plant input, and anthropogenic activity, where peri-condensed, unsubstituted PAHs are
226 markers for combustion of organic matter (Hites et al., 1977; Venkatesan and Dahl, 1989;
227 Killops and Massoud, 1992; Krüge et al., 1994; Jiang et al., 1998; Arinobu et al., 1999;
228 Finkelstein et al., 2005; Peters et al., 2005; Marynowski and Simoneit, 2009). Some
229 sedimentary PAHs, such as phenanthrene, chrysene, and triphenylene, are more affected

230 by diagenesis or additional sources (Jiang et al., 1998; Grice et al., 2007), so they were
231 not included in the total PAH sum plotted in figure 3b. Although the patterns are not
232 identical, enhanced concentrations of total PAH co-occur with elevated C_{19}/C_{23} and
233 C_{20}/C_{23} tricyclic terpane ratios in the bottom 2 meters of the sampling interval. Some
234 PAHS, such as retene, are thought to derive from higher plants, in particular coniferous
235 resin (Wakeham et al., 1980; Ellis et al., 1996; Jiang et al., 1998; Grice et al., 2005;
236 Peters et al., 2005), although algal and bacterial sources have been reported as well (Wen
237 et al., 2000). Retene was detected in all samples, and its concentration was plotted
238 separately as a marker of higher plant input in figure 3c. Retene was more abundant in
239 the lowest interval of the section, which is consistent with the total PAHs, tricyclic
240 terpane ratios, and HI data.

241 Microscopic analysis of four kerogen samples adds an additional line of evidence
242 supporting stratigraphic variations in kerogen type and organic matter sources. According
243 to the petrographic results, the kerogen is comprised of as much as 80% of terrigenous
244 organic matter in the bottom of the section and about 25-40% in the remainder of the
245 sampling interval (fig. 3d). Since the elemental, molecular, and isotopic composition of
246 organic matter from higher plants is distinct from marine organic matter, HI can reflect
247 the compositional difference of distinct types of organic matter (e.g. Talbot and
248 Livingstone, 1989). Assuming constant HI values for terrigenous and marine organic
249 matter end-members, the percent terrigenous organic matter was estimated through the
250 section using the linear relationship between the percent terrigenous macerals measured
251 by optical microscopy and the corresponding HI (see SOM for more details). Calculated

252 values of $f_{Terr\ OM}$ ranged from 12 to 85%, where the highest values were found in the
253 lowermost part of the section (fig. 3e).

254 The stratigraphic change in the relative supply of terrigenous organic matter ,
255 which is supported by the HI, molecular, and petrographic results, may have been driven
256 by effects related to rising sea level (Hesselbo, 2008; Suan et al., 2011). Considering that
257 increased rates of continental weathering across the T-OAE would have enhanced
258 delivery of terrigenous material (Cohen et al., 2004), the opposite trend recorded in our
259 data might be best explained by progressive remoteness from the coastline on a gently
260 sloping shelf during sea level transgression (e.g. Macquaker et al., 2010). However, sea
261 level related effects might not be the only factor responsible for the changing signal in
262 relative abundance of terrigenous and marine organic matter. For example, enhanced
263 marine export productivity and/or enhanced marine organic matter preservation could
264 have diluted the input of terrigenous organic matter, thereby changing the relative
265 apparent contribution.

266

267 *4.4 Biomarker indicators of source and community structure*

268 Since some compounds or compound classes are associated with a particular
269 biological source, metabolism, or physiology, molecular distributions can be informative
270 about changes in microbial community structure. Unlike other intervals of ocean anoxia
271 associated with mass extinction events (e.g. Xie et al., 2005; Cao et al., 2009), algal- and
272 bacterial-derived biomarkers indicative of community structure did not vary significantly
273 through the CIE (fig. 4). Instead, some biomarker indices that typically reflect
274 community structure were more affected by source input at Hawsker Bottoms.

275 The regular sterane/17 α -hopane ratio is used as an indicator of the relative
276 contributions of eukaryote and bacterial biomass. The regular sterane/hopane ratio was
277 calculated using regular C₂₇₋₂₉ steranes and 17 α C₂₉₋₃₃ hopanes. The regular sterane/17 α -
278 hopane ratio exhibited low values below -2.5 m, whereas it was more elevated (>0.5) and
279 relatively constant in the top 10 m of the section. While this offset could be interpreted as
280 a shift from a bacterially dominated environment in the lower part of the section to a
281 eukaryotic environment above -3 m, it is more likely that a change in the organic matter
282 source input is driving the regular sterane/17 α -hopane ratio variability. As well as
283 containing low total steroid abundances, terrigenous organic matter can deliver hopanes
284 derived from soil bacteria, thereby lowering the regular sterane/17 α -hopane ratio (Peters
285 et al., 2005; Handley et al., 2010; Sãenz et al., 2011; French et al., 2012).

286 Similarly, small deviations are found at the bottom of the section for the C₂₇/C₂₇₋
287 ₃₀, C₂₈/C₂₇₋₃₀, and C₂₉/C₂₇₋₃₀ sterane ratios. These ratios included regular steranes as well
288 as diasteranes and are commonly used as indicators of the relative contribution from red
289 algae biomass, chlorophyll-c algae, and green algae, respectively. However, they can also
290 be affected by the delivery of C₂₉ steranes derived from land plants (Moldowan et al.,
291 1985; Peters et al., 2005). Indeed, C₂₉ sterane was the dominant sterane in samples from
292 the bottom of the section, where it represented nearly half of the total C₂₇₋₃₀ steranes. The
293 C₃₀/C₂₇₋₃₀ sterane ratio, on the other hand, which is an indicator of marine pelagophyte
294 algae, was constant throughout the section and represented only a minor proportion of the
295 total steranes abundance.

296 The 2 α -methylhopane index (2-MHI) has been used as an indicator of
297 cyanobacterial input (Summons et al., 1999), although additional sources were later

298 reported (Rashby et al., 2007). The 3 β -methylhopane index (3-MHI) is considered a
299 marker for aerobic proteobacteria, including methanotrophs and acetic acid bacteria
300 (Zundel and Rohmer, 1985; Talbot et al., 2003; Farrimond et al., 2004; Talbot and
301 Farrimond, 2007). The 2-MHI and 3-MHI were invariant, and the 3-MHI was in the
302 range of average Phanerozoic marine values (~1-3%) (Farrimond et al., 2004; Cao et al.,
303 2009). Likewise, the 2-MHI also lacked elevated values. Based on elevated 2-MHI, 3-
304 MHI, and nitrogen isotope anomalies, previous workers have reported an increased
305 contribution of diazotrophic cyanobacteria and methanotrophic bacteria during other
306 OAEs (e.g. Kuypers et al., 2004; Cao et al., 2009; Sepúlveda et al., 2009; Luo et al.,
307 2011). The low and invariant contribution of these microbial groups indicates that
308 environmental conditions suitable for their predominance did not prevail at this locality.
309 Thus, enhanced cyanobacterial diazotrophy may not fully explain the previously reported
310 depleted bulk organic $\delta^{15}\text{N}$ at this location during the T-OAE (Jenkyns et al., 2001).

311

312 *4.5 Indicators of redox change and depositional environment*

313 A suite of biomarkers was used to assess changes in water column stratification
314 and redox potential, including the gammacerane index, C₃₅ homohopane index (C₃₅ HHI),
315 pristane/phytane (Pr/Ph) ratio, and the concentration of aromatic carotenoid derivatives
316 (fig. 5). Although some of these parameters can also be influenced by diagenesis, source
317 input, and thermal maturity, they display patterns consistent with a shift towards more
318 intense reducing conditions at least in the sediment porewaters and potentially in the
319 overlying water column.

320 Although the biological sources of gammacerane are not fully known (Peters et
321 al., 2005), it is a diagenetic product of tetrahymanol, a compound found in bacterivorous
322 ciliates thriving at the chemocline of stratified water bodies (ten Haven et al., 1989;
323 Sinninghe Damsté et al., 1995). Thus, the occurrence of gammacerane, expressed as the
324 gammacerane index = $[gammacerane / (gammacerane + 17\alpha, 21\beta C_{30} \text{ hopane})] * 100$, has
325 been used to infer changes in water column stratification in ancient environments.
326 Gammacerane was detected in all of the analyzed samples but became more prominent in
327 those deposited during the OAE, starting at the onset of thin-bedded shales.

328 Elevated abundances of gammacerane during the T-OAE may reflect the
329 development of seasonal water column stratification, possibly due to stronger seasonality
330 and/or deepening of the water column. In contrast, prior to the T-OAE, seasonality may
331 have been weaker, or the water column may have been too shallow to stratify, even
332 during warm months, due to turbulent mixing. The development of seasonal water
333 column stratification at Hawsker Bottoms during the T-OAE would have aided the
334 development of water column oxygen-depletion, particularly during warm and productive
335 months. However, gammacerane enrichments alone do not necessitate water column
336 anoxia, particularly given the association of its precursor, tetrahymanol, with suboxic
337 waters in the modern (Wakeham et al., 2007; 2012).

338 The C_{35} HHI and Pr/Ph ratio are recorders of depositional redox conditions in
339 sediments. The C_{35} HHI records the degree of preservation of the extended side chain of
340 C_{35} hopanes derived from intact bacteriohopanepolyols (BHPs) (Köster et al., 1997;
341 Peters et al., 2005). Higher C_{35} HHI values are characteristic of oxygen-depleted marine
342 depositional environments. Pristane and phytane in ancient marine rock extracts and oils

343 are largely, but not exclusively, derived from the chlorophyll phytyl side chain from
344 photoautotrophs. Redox conditions influence the diagenetic pathway of the phytyl side
345 chain. Reducing conditions promote the conversion of phytol to phytane, and oxic
346 conditions promote the conversion of phytol to pristane (Didyk et al., 1978; Peters et al.,
347 2005). The C₃₅ HHI nearly doubled in samples deposited during the OAE compared to
348 those deposited prior to the event. Values of the Pr/Ph ratio >3 recorded at the bottom of
349 the section are suggestive of deposition of terrigenous organic matter under oxic
350 conditions. The Pr/Ph ratio values near or below 1 during the T-OAE, together with the
351 elevated C₃₅ HHI, suggest intensification of reducing conditions in the sediment
352 porewaters during deposition.

353 Biomarkers for anaerobic phototrophic green sulfur bacteria (GSB) have been
354 used to argue for the development of PZE during the T-OAE (Schouten et al., 2000;
355 Pancost et al., 2004; Bowden et al., 2006) and other OAEs (e.g. Cao et al., 2009) based
356 on the physiological requirement of co-occurring reduced sulfur species and light. We
357 detected 2,3,6-aryl isoprenoids, isorenieratane, and chlorobactane in all samples. Unlike
358 previous studies of the T-OAE or any Phanerozoic organic geochemical study of marine
359 samples, trace concentrations of okenane were also identified by GC-MRM-MS in
360 samples above -3 m, whereas it was below detection limit in samples from the lowest 2
361 meters of the section. Okenone, a photosynthetic pigment belonging to the PSB family
362 Chromatiaceae, is the only known precursor of okenane (Brocks and Schaeffer, 2008).
363 All compounds were compared with an authentic carotenoid standard and an extract from
364 the Barney Creek Formation (BCF; fig. 6; Brocks et al., 2005). Normalizations of the C₄₀

365 aromatic carotenoid derivatives against the mass of TOC and TLE reveal a similar pattern
366 of elevated concentrations during the anoxic event compared to the pre-event baseline.

367 In total, all of the organic geochemical redox indicators point towards a shift
368 towards more reducing conditions broadly corresponding with the initiation of the
369 negative CIE. However, sedimentological features, such as starved wave and combined
370 flow ripples, indicate that this area was an energetic, shallow inner shelf environment on
371 the order of tens of meters and probably no deeper than 50 m, where enough oxygen was
372 present in the water column on some timescale to sustain nektonic fauna, including
373 ammonites, and allow for bioturbation at the sediment-water interface through the OAE
374 (Wignall et al., 2005; Powell, 2010; Ghadeer and Macquaker, 2011). We explore
375 different scenarios to reconcile these apparent opposing lines of evidence.

376 First, the geochemical and sedimentological signals recorded in the rock record
377 are a composite of many processes occurring on different timescales. In the modern
378 ocean, highly productive coastal and continental margin sediments and the overlying
379 water column oscillate between oxic and anoxic conditions over different timescales (e.g.
380 Burdige, 2007). Enhanced productivity and export of organic matter, which are important
381 features of Mesozoic OAEs (e.g. Erba, 2004; Jenkyns, 2010), would have increased the
382 oxygen demand in the water column and sediment porewaters during productive months.
383 During the T-OAE, anoxic conditions may have been restricted to the sediment
384 porewaters during seasons of low productivity, allowing bioturbation to occur when
385 bottom waters were better oxygenated. Conversely, oxygen-deficient waters may have
386 expanded seasonally to the water column during intervals of high productivity and
387 enhanced stratification, as implied by the gammacerane index.

388 Second, the molecular, paleontological, and sedimentary indicators of redox
389 chemistry apply to different parts of the depositional environment, diagenetic pathways,
390 and have different sensitivities along the redox spectrum. With the exception of the GSB
391 and PSB carotenoid markers, the geochemical parameters reported here do not require
392 strict anoxia or euxinia in the water column. The gammacerane index pattern supports the
393 development of seasonal stratification during the OAE, which would have promoted
394 oxygen depletion of the water column. However, the possibility remains that even during
395 intervals of high productivity and stratification, water column oxygen concentrations at
396 this location were depleted but high enough to sustain organisms with physiological
397 oxygen requirements, thereby explaining the fossil and sedimentary evidence.
398 Additionally, the Pr/Ph ratio and C₃₅ HHI pertain primarily to sedimentary redox
399 conditions opposed to water column redox structure. Therefore, the occurrence of intact
400 aromatic carotenoid derivatives merits further discussion to assess water column redox
401 chemistry.

402 To date, okenane has only been reported in Paleoproterozoic rock extracts and
403 lacustrine Cenozoic extracts (Brocks et al., 2005; Zhang et al., 2011). Given the
404 atmospheric pO_2 during the Mesozoic was near present atmospheric levels (Berner,
405 2006), its detection in marine samples of this age requires careful interpretation. The PSB
406 family Chromatiaceae blooms in a range of anoxic environments with light and reduced
407 sulfur species, including stratified lakes, fjords, coastal lagoons, estuaries, and coastal
408 microbial mats, but not all Chromatiaceae produce okenone (Brocks and Schaeffer, 2008
409 and references therein). Okenone-producing planktonic Chromatiaceae dwell in water
410 columns where the chemocline is above 25 m and in 75% of the reported cases less than

411 12 m (Brocks and Schaeffer, 2008). Notably, all of the modern chemocline depth
412 observations for okenone production are based on stratified lake systems. Thus, the lack
413 of okenone in modern marine sulfidic environments presents a “no analogue problem” for
414 ancient marine samples containing okenane that were deposited under atmospheric pO_2
415 close to modern levels.

416 Transient free sulfide has been reported in the water column of intense upwelling
417 zones, including the Arabian Sea, Namibian coast, and the Peruvian coast (e.g Dugdale et
418 al., 1977; Brüchert et al., 2003; Naqvi et al., 2006; Schunck et al., 2013). However, these
419 episodes are typically short lived. In contrast, sulfidic waters persist in some restricted
420 marine basins and fjords, including the Black Sea, Cariaco Basin, Saanich Inlet, and the
421 Framvaren and Effingham Fjords. However, these two types of marine environments
422 (non-restricted, transiently sulfidic and restricted, permanently sulfidic) fail to represent
423 suitable modern analogues for Hawsker Bottoms on several counts. Isorenieratene has
424 been measured in the water column and sediments of some restricted marine basins,
425 particularly fjords and the Black Sea (e.g. Sinninghe Damsté and Schouten, 2006), but
426 GSB carotenoids have not been detected in transiently sulfidic upwelling systems.
427 Furthermore, okenone has not been reported in the water column or sediments of any
428 modern marine transiently or permanently sulfidic environment, with the exception of the
429 upper sediments of Kyllaren fjord, a small, highly restricted basin (Smittenberg et al.,
430 2004; Sinninghe Damsté and Schouten, 2006). Like okenone, modern planktonic marine
431 occurrences of chlorobactene are also limited to semi-enclosed water masses that are not
432 representative of fully marine conditions (e.g. Naeher et al. 2012). Interestingly, multiple
433 emerging lines of evidence suggest the occurrence of a “cryptic sulfur cycle” in some

434 OMZs, with a potential role for photosynthetic sulfide oxidation (Canfield et al., 2010;
435 Stewart et al., 2012). However, the presence of GSB and/or PSB and their respective
436 carotenoids have yet to be reported in modern OMZs.

437 Furthermore, the physical oceanographic processes determining the degree of
438 vertical mixing, hence stratification and redox gradient stability, are markedly different
439 between inner shelf environments and sulfidic, silled basins, which are highly restricted
440 and in many classic modern examples, are an order of magnitude or more deeper than
441 estimated paleodepths of Hawsker Bottoms. Although the water column at Hawsker
442 Bottoms became deeper with the sea level transgression across the T-OAE, the
443 depositional environment remained relatively shallow because of its location on a gently
444 sloping shelf (e.g. Macquaker et al., 2010). Consequently, turbulent mixing at the surface
445 and bottom boundary layers would have prevented a stable sulfidic chemocline from
446 developing. On the other hand, considering the limited occurrence of planktonically
447 produced okenone in modern lakes, a planktonic source of okenone at Hawsker Bottoms
448 would imply that, rather than an inner shelf environment, Hawsker Bottoms was a highly
449 restricted coastal basin not reflective of fully marine conditions.

450 Alternatively, we argue that okenane at Hawsker Bottoms was likely derived from
451 benthic microbial mats based on the lack of modern analogues of okenone-production in
452 marine sulfidic environments, the dynamics of inner shelf physical mixing, and building
453 evidence of okenone-producing, mat-dwelling Chromatiaceae (e.g. Caumette et al., 1991;
454 Airs et al., 2001; Caumette et al., 2004; Meyer et al., 2011). Furthermore, planktonic
455 Chlorobiaceae are not the exclusive source of isorenieratene and chlorobactene. Previous
456 work has documented additional non-planktonic GSB sources of isorenieratene and

457 chlorobactene, including microbial mats (e.g. Wahlund et al., 1991; Brocks and
458 Summons, 2003; Beatty et al., 2005; Bühring et al., 2011). Although a mixed planktonic
459 and mat origin of the carotenoids cannot be ruled out, it is more likely that the GSB and
460 PSB carotenoids detected in Hawsker Bottoms samples share a source. Previous studies
461 of sedimentary structures in the Toarcian shales of Yorkshire have attributed wavy
462 laminations to microbial mats (O'Brien, 1990), thereby further supporting a sedimentary
463 origin of GSB and PSB carotenoid derivatives detected in Hawsker Bottoms samples.
464 Similar wavy laminated fabrics have been reported in coeval shales in northern European
465 T-OAE sections (Trabucho-Alexandre et al., 2012). Unfortunately, a sedimentary source
466 of the GSB and PSB carotenoid derivatives offers little information about water column
467 redox chemistry. However, this interpretation does not preclude the development of a
468 suboxic, anoxic, or euxinic water column in this region on some timescale during the T-
469 OAE. Instead, additional inorganic geochemical data is required to better assess the water
470 column redox conditions and degree of basin restriction (Algeo and Tribovillard, 2009).

471

472 *4.6 Compound specific stable carbon isotopic data*

473 A limited number of compound specific $\delta^{13}\text{C}$ records of the T-OAE are currently
474 available in the literature. Here, we report the first long-chain *n*-alkane $\delta^{13}\text{C}$ records of
475 the T-OAE. Compound specific $\delta^{13}\text{C}$ analyses of marine- and terrestrial-derived lipids
476 reveal a shift towards lighter $\delta^{13}\text{C}$ values (fig. 7). Short-chain *n*-alkanes, as well as
477 pristane and phytane are typically used as marine indicators, whereas long-chain *n*-
478 alkanes primarily reflect terrigenous sources. The *n*-C₁₇, *n*-C₁₈, and *n*-C₁₉ alkanes
479 displayed a negative excursion of ~2–3‰, which is consistent with the ~2–4‰ negative

480 excursions documented in the partial n -C₁₆₋₂₀ alkane records from the Toarcian
481 Posidonienschiefer in southwest Germany (Schouten et al., 2000). Pristane and phytane
482 encode a muted excursion (~ 1.5 – 2%) compared to short-chain alkane records, and they
483 also have smaller CIE magnitudes compared to the pristane and phytane isotopic records
484 from the Toarcian Paris Basin and the Posidonienschiefer (~ 3 – 4%) (Schouten et al.,
485 2000; van Breugel et al., 2006). On the other hand, long-chain n -alkanes (n -C₂₇, n -C₂₈,
486 and n -C₂₉), which are primarily but not exclusively derived from epicuticular waxes of
487 vascular plants (Eglinton and Hamilton, 1967), display the largest compound specific
488 negative CIE (~ 4 – 5%). The molecular isotopic records appear to register the initiation of
489 the negative CIE earlier than in the bulk organic record, and within the CIE, the
490 compound-specific $\delta^{13}\text{C}$ values remain fairly stable while the bulk curve becomes
491 gradually depleted. However, these features could be due to sampling resolution
492 differences. Higher resolution molecular isotopic records are required to better address
493 the timing and structure of the isotopic excursion recorded in different carbon reservoirs.

494 The absolute magnitude of the bulk organic CIE (~ 5 – 7%) is larger than the CIEs
495 recorded in the molecular records from Yorkshire (this study; ~ 1.5 – 5%), the Paris Basin
496 ($\sim 3\%$; van Breugel et al., 2006), and the Posidonienschiefer (~ 2 – 4% ; Schouten et al.,
497 2000). Identifying the reason behind the CIE magnitude offsets is critical for placing
498 bounds on the magnitude of isotopically light carbon added into the system. Bulk organic
499 matter is comprised of an array of molecularly and isotopically heterogeneous
500 constituents. In addition to environmental perturbations, organic matter source mixing
501 can contribute to bulk organic $\delta^{13}\text{C}$ excursions (e.g. Pancost et al., 1999). The comparison
502 of short- and long-chain n -alkane isotopic compositions demonstrates that, unlike the

503 modern, terrigenous organic matter is isotopically heavier than marine organic matter
504 during the Toarcian (fig. 7), which is consistent with previous Toarcian studies (Vetö et
505 al., 1997; Schouten et al., 2000). Multiple lines of evidence presented in section 4.3
506 highlight a significant transition in the terrigenous organic matter input at Hawsker
507 Bottoms. Indeed, the bottom 2 meters of the study interval are dominated by terrigenous
508 organic matter and are isotopically heavier than the overlying interval that is dominated
509 by marine organic matter. Therefore, an undetermined component of the bulk organic
510 CIE magnitude may be attributed to source mixing effects.

511 Additional factors could also contribute to the difference in magnitudes between
512 bulk and molecular CIEs. For instance, it is possible that the full CIE was not captured in
513 the molecular isotopic records due to a lower sampling resolution compared to the bulk
514 organic $\delta^{13}\text{C}$ records. Alternatively, water availability can modulate the magnitude of the
515 CIE recorded in vascular leaf waxes, as has been discussed for the Paleocene Eocene
516 Thermal Maximum (PETM; e.g. Schouten et al., 2007; Smith et al., 2007), but unlike the
517 PETM, the ratio of angiosperms and conifers would not account for the *n*-alkane and bulk
518 organic CIE magnitude offset because the rise of angiosperms postdates the Early
519 Jurassic (e.g. Heimhofer et al., 2005). Additionally, thermal maturation could influence
520 the $\delta^{13}\text{C}$ of individual compounds, which become isotopically heavier with increasing
521 thermal maturity (Clayton, 1991; Clayton and Bjorøy, 1994; Tang et al., 2005).

522 Nevertheless, while multiple mechanisms may account for the difference between
523 molecular and bulk organic CIE magnitudes, it is significant that a negative CIE is
524 recorded in both marine- and terrestrial-derived lipids, albeit to different degrees. The
525 parallel isotopic change in marine and terrestrial carbon pools recorded at Hawsker

526 Bottoms further supports previous studies suggesting that the T-OAE was a global carbon
527 cycle perturbation where isotopically light carbon entered the atmospheric, terrigenous,
528 and marine carbon reservoirs (Hesselbo et al., 2000; 2007; Al-Suwaidi et al., 2010;
529 Caruthers et al., 2011; Gröcke et al., 2011). Multiple sources of isotopically light carbon
530 have been proposed, including methane hydrate dissociation, regional upwelling of
531 isotopically light waters in stratified epicontinental seas, thermogenic release of methane
532 from organic-rich strata in contact with dykes, biomass burning, or a combination of
533 these mechanisms (Hesselbo et al., 2000; Schouten et al., 2000; McElwain et al., 2005;
534 van de Schootbrugge et al., 2005; Finkelstein et al., 2006).

535 Although our study does not provide evidence in support of a specific forcing
536 mechanism, it allows us to narrow down potential mechanisms. A deep-water source of
537 isotopically light carbon is unlikely because of the CIE observed in land plant
538 biomarkers. The lack of evidence for bacterial methanotrophy in our section suggests that
539 methane hydrate dissociation did not supply appreciable methane to the sampling
540 locality. The organic matter source transition complicates the interpretation of PAH
541 abundances as tracers of biomass burning, so a different sampling locality without an
542 organic matter source transition should be studied to test the biomass burning mechanism
543 using PAHs. A cascade of mechanisms rather than a single mechanism likely initiated the
544 T-OAE. However, our results indicate that the influence of source mixing on the bulk
545 organic $\delta^{13}\text{C}$ has been previously underestimated and could have potentially affected
546 other bulk isotopic systematics such as nitrogen. Therefore, previous estimates regarding
547 the magnitude of this global perturbation of the carbon cycle should be revisited.

548

549 **5. Conclusions**

550 We use a multiproxy approach based on bulk geochemistry, lipid biomarkers, and
551 compound-specific stable isotopes to elucidate environmental and ecological changes
552 associated with the T-OAE at Hawsker Bottoms in Yorkshire, England. Molecular
553 indicators and Rock-Eval results suggest that thermal maturity is uniformly in the early
554 oil generation window throughout the sampling interval. The HI data, organic
555 petrography, PAH distribution, and tricyclic terpane ratios indicate a transition in the
556 relative input of terrigenous vs. marine organic matter across a lithological transition. The
557 shift to lower relative abundance of terrigenous organic matter was likely a result of sea
558 level related effects such as coastal proximity, changes in marine organic matter
559 preservation, and/or dilution effects from increased marine productivity.

560 Organic geochemical redox and depositional environment indicators point
561 towards an overall shift towards more reducing conditions in sediment porewaters and the
562 development of seasonal stratification during the OAE. Previous sedimentological
563 observations require that the water column was not completely anoxic throughout the
564 entire T-OAE, which may seem contradictory at first to the detection of GSB and PSB
565 carotenoids. However, here we present the first occurrence of okenane, a carotenoid
566 marker of PSB, in marine samples younger than the Paleoproterozoic (1.64 Ga). This
567 unexpected finding challenges the interpretation of GSB and PSB carotenoids as markers
568 of PZE in the context of Hawsker Bottoms due to inner shelf vertical mixing, the lack of
569 modern analogues of okenone-production in marine sulfidic environments, and the
570 emerging evidence of okenone-producing mat-dwelling Chromatiaceae. Therefore, in
571 combination with previous reports of microbial wavy lamination in Toarcian shales of

572 Yorkshire and coeval shales in northern Europe, we argue that okenane, and potentially
573 chlorobactane and isorenieratane, was most likely mat-derived at Hawsker Bottoms.

574 The compound-specific $\delta^{13}\text{C}$ records of short-chain *n*-alkanes, acyclic
575 isoprenoids, and long-chain *n*-alkanes support a carbon cycle perturbation that affected
576 both the atmospheric and marine systems, which precludes the recycling of isotopically
577 light CO_2 from anoxic waters as the sole mechanism responsible for the T-OAE negative
578 CIE. Notably, compound specific $\delta^{13}\text{C}$ records of the T-OAE, including the new data
579 presented here from Yorkshire and previous molecular data from the Paris Basin and the
580 Posidonienschiefer, encode negative CIEs that are smaller in magnitude compared to bulk
581 organic $\delta^{13}\text{C}$ records. Many mechanisms could contribute to this observation, particularly
582 variable mixing of terrigenous and marine organic matter, which is supported by the
583 multiple lines of evidence for a transition in organic matter source. Identifying the
584 mechanisms behind the CIE magnitude offsets is important for estimating the magnitude
585 of isotopically light carbon injected into the surface carbon reservoirs.

586

587 Acknowledgements

588 We thank Alison Cohen and Emma Grosjean for their contributions to the early
589 stages of this study, Lorraine Eglinton for petrographic analysis, Carolyn Colonero for
590 laboratory assistance, and three anonymous reviewers for valuable comments that
591 improved this work. Funding support for work at MIT was provided by grants from the
592 NASA Astrobiology Institute and the NASA Exobiology Program to RES and an NSF
593 graduate fellowship to KLF. This project was partly funded by a NSERC Discovery

594 Grant (# 288321) and NERC Standard Grant (NE/H021868/1) to DRG. JTA is funded by
595 the NERC Standard Grant (NE/H021868/1) to DRG.

596

597 Figure Captions:

598 **Fig. 1. Rock-Eval Analysis.** A) Total organic carbon (TOC; %) (Hesselbo et al., 2000);
599 B) Hydrogen index (HI; mg HC/g TOC); C) Production index (PI); D) T_{max} (°C).

600

601 **Fig. 2. Molecular Indicators of Thermal Maturity.** A) The C_{31} hopane 22S/(22S+22R)
602 ratio; B) C_{30} hopane $\beta\alpha/(\beta\alpha+\alpha\beta)$ ratio; C) C_{29} sterane $\alpha\alpha\alpha$ 20S/(20S+20R) ratio; D)
603 Ts/(Ts+Tm) ratio; E) diasteranes/steranes ratio. Note the lack of variation in A, B and C
604 across the section compared to the minor variations exhibited by D and E due to changes
605 in source input and/or lithology.

606

607 **Fig 3. Terrigenous Organic Matter Indicators.** A) Ratios of C_{19}/C_{23} and C_{20}/C_{23}
608 tricyclic terpanes; B) total concentration of PAHs normalized by TOC and TLE; C)
609 concentration of retene normalized by TOC and TLE; D) measured contribution of
610 terrigenous organic matter (OM) by petrographic analysis; E) calculated percentage of
611 terrigenous organic matter ($f_{Terr\ OM}$) based on linear regression ($f_{Terr\ OM} (\%) = -0.279 * HI$
612 $+ 136$; $R^2=0.90$) of petrographic measurements of terrigenous organic matter (Fig. 3D)
613 and HI (Fig. 1B).

614

615 **Fig. 4. Source and Community Indicators.** A) The ratio of regular steranes/17 α -
616 hopanes; B) C_{27}/C_{27-30} steranes; C) C_{28}/C_{27-30} steranes; D) C_{29}/C_{27-30} steranes; E) 2 α -
617 methylhopane and 3 β -methylhopane indices.

618

619 **Fig. 5. Redox and Depositional Environmental Indicators.** A) Gammacerane index; B)
620 C_{35} homohopane index ($C_{35}HHI$); C) pristane/phytane (Pr/Ph) ratio; D) concentration of
621 isorenieratane; E) concentration of chlorobactane; F) concentration of okenane. The
622 concentrations of C_{40} carotenoids are semi-quantitative and were normalized against TOC
623 and TLE.

624

625 **Fig. 6 Gas chromatography - metastable reaction monitoring - mass spectrometry**
626 **(GC-MRM-MS) chromatogram for the identification of aromatic carotenoid**
627 **derivatives.** MRM chromatograms displaying the 554→134 transition characteristic for
628 chlorobactane and okenane in an authentic standard of combined C_{40} carotenoids (A) and
629 in sample ENR004 at 5.98 m (C). Plots B and D are MRM chromatograms displaying the
630 546→134 MRM transition characteristic of isorenieratane, renieratane, and
631 renierapurpurane in an authentic standard of combined C_{40} carotenoid (B) and in sample
632 ENR004 (D). Compound abbreviations were used for labels: chlorobactane (Ch);
633 okenane (Ok); isorenieratane (Iso); renieratane (Ren); renierapurpurane (Rpurp).

634

635 **Fig. 7 Carbon Isotopic records.** A) Previously reported $\delta^{13}\text{C}_{\text{org}}$ from Hawsker Bottoms
636 (Hesselbo et al., 2000; Kemp et al., 2005); B) $\delta^{13}\text{C}$ of n-C₁₇₋₁₉ short-chain n-alkanes; C)
637 $\delta^{13}\text{C}$ of pristane and phytane; D) $\delta^{13}\text{C}$ of n-C₂₇₋₂₉ long-chain n-alkanes. Note the different
638 x-axis scales.
639

640 References

641

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Figure 1

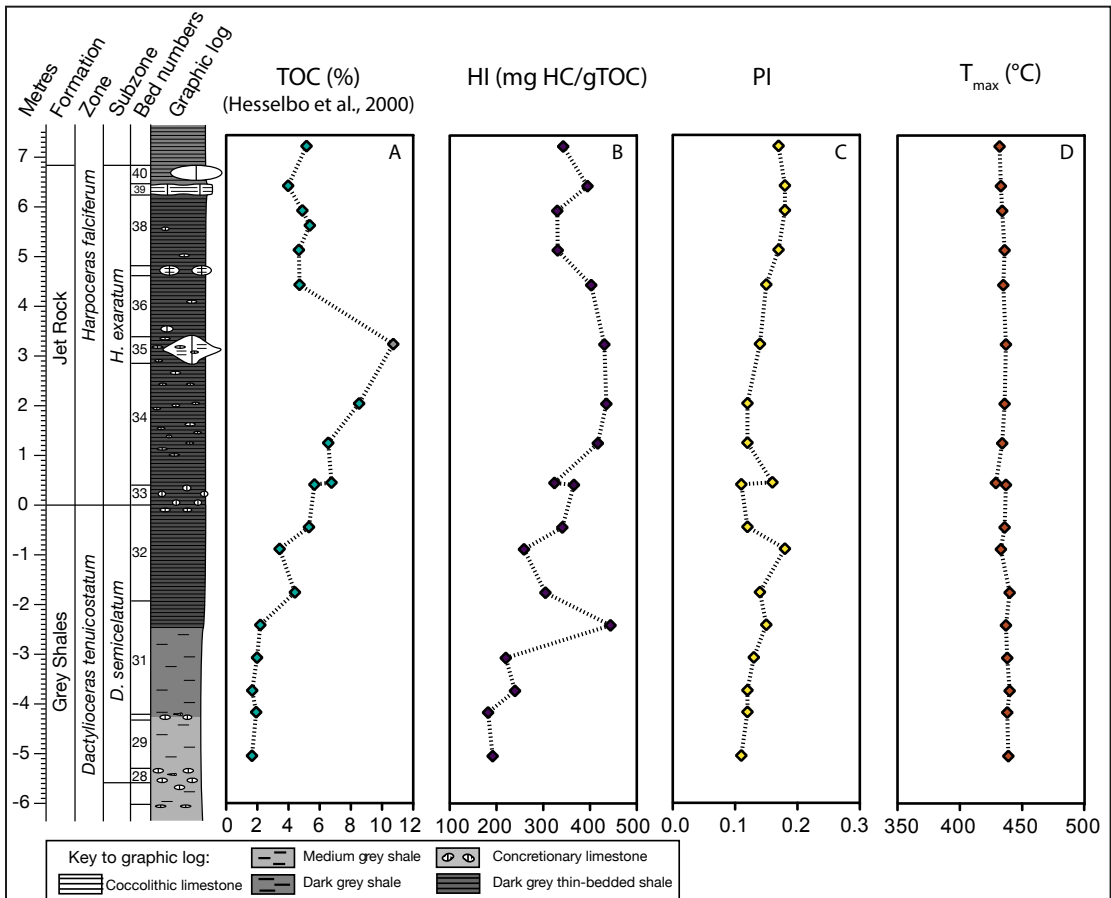


Figure 2

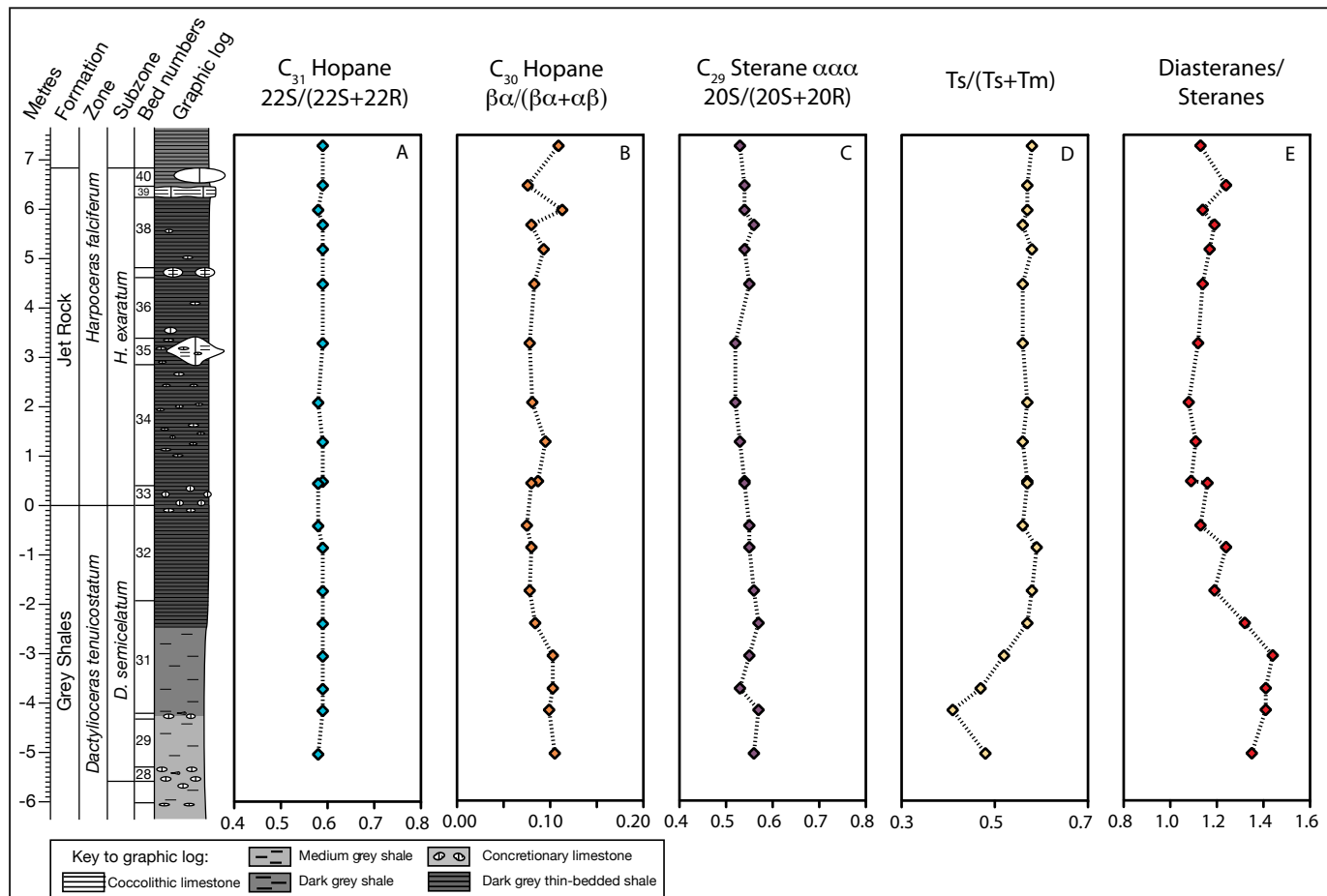


Figure 3

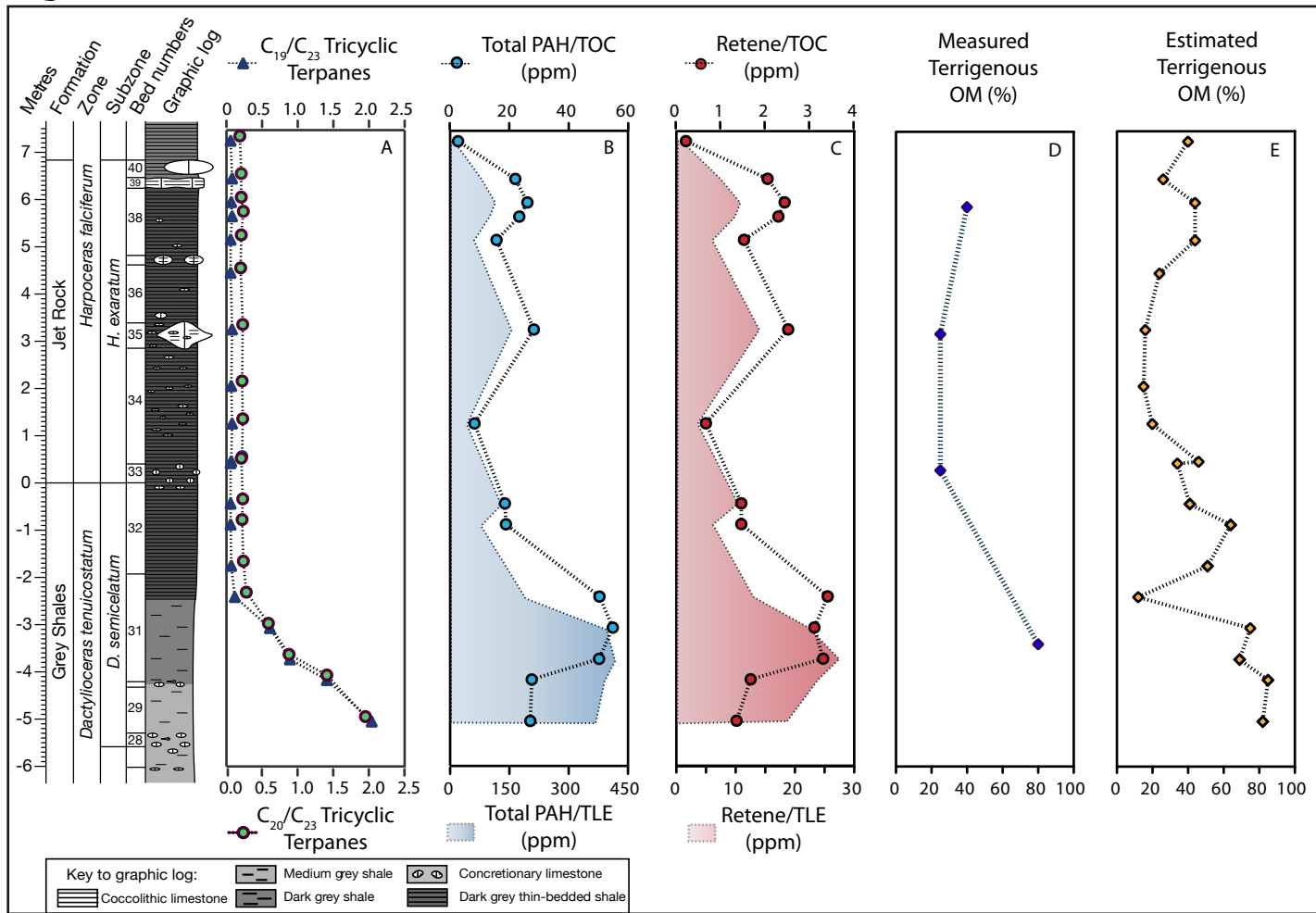


Figure 4

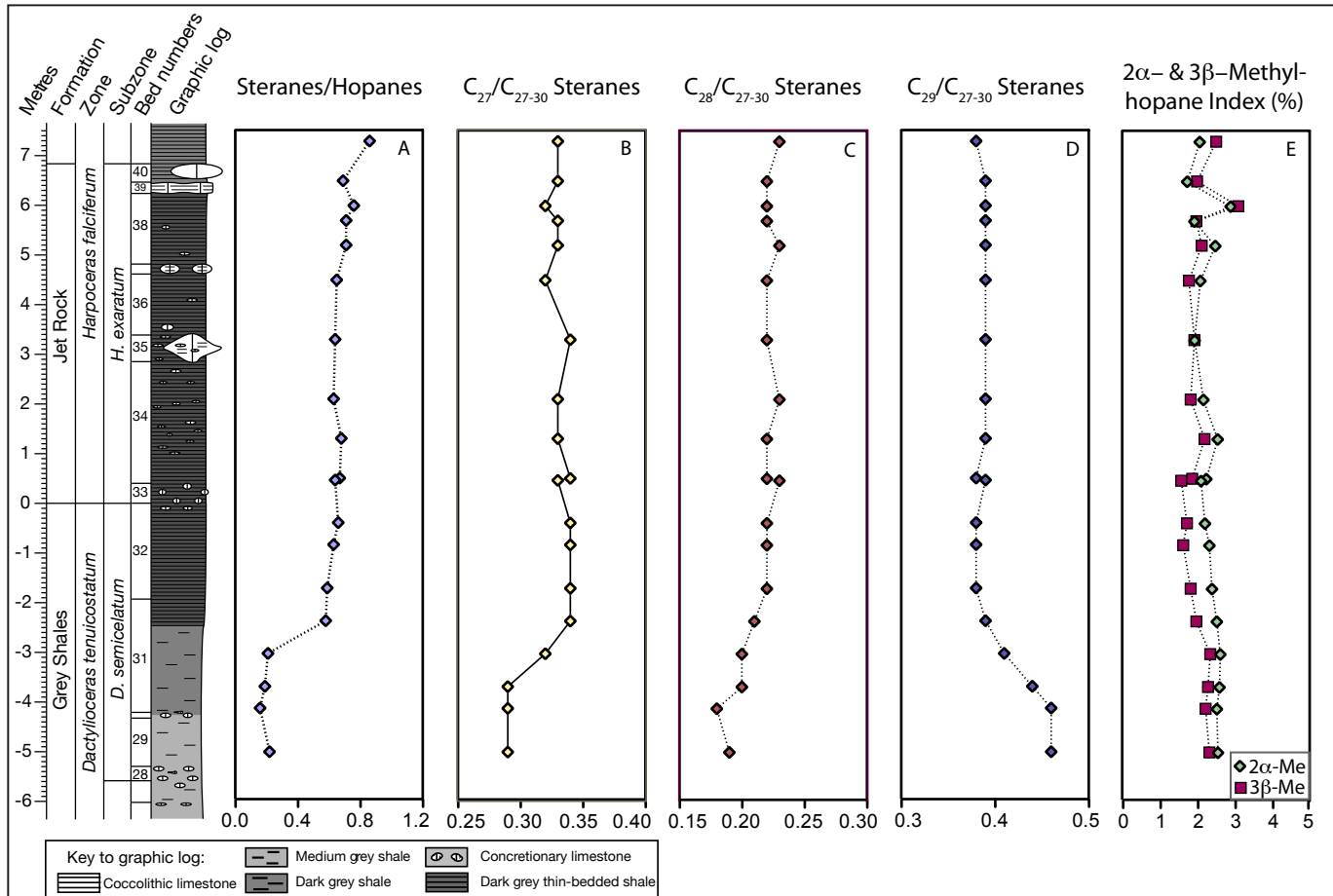


Figure 5

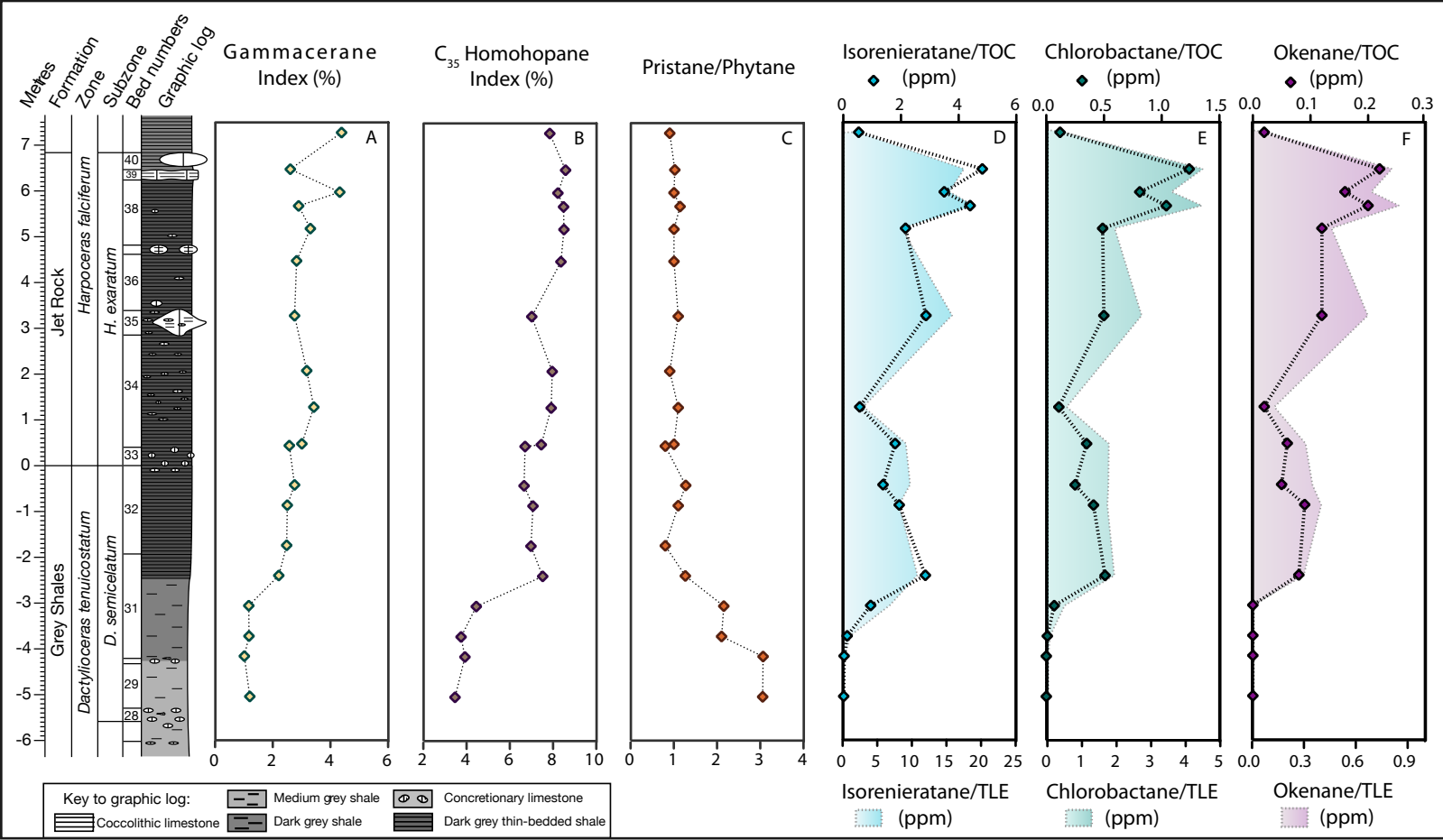


Figure 6

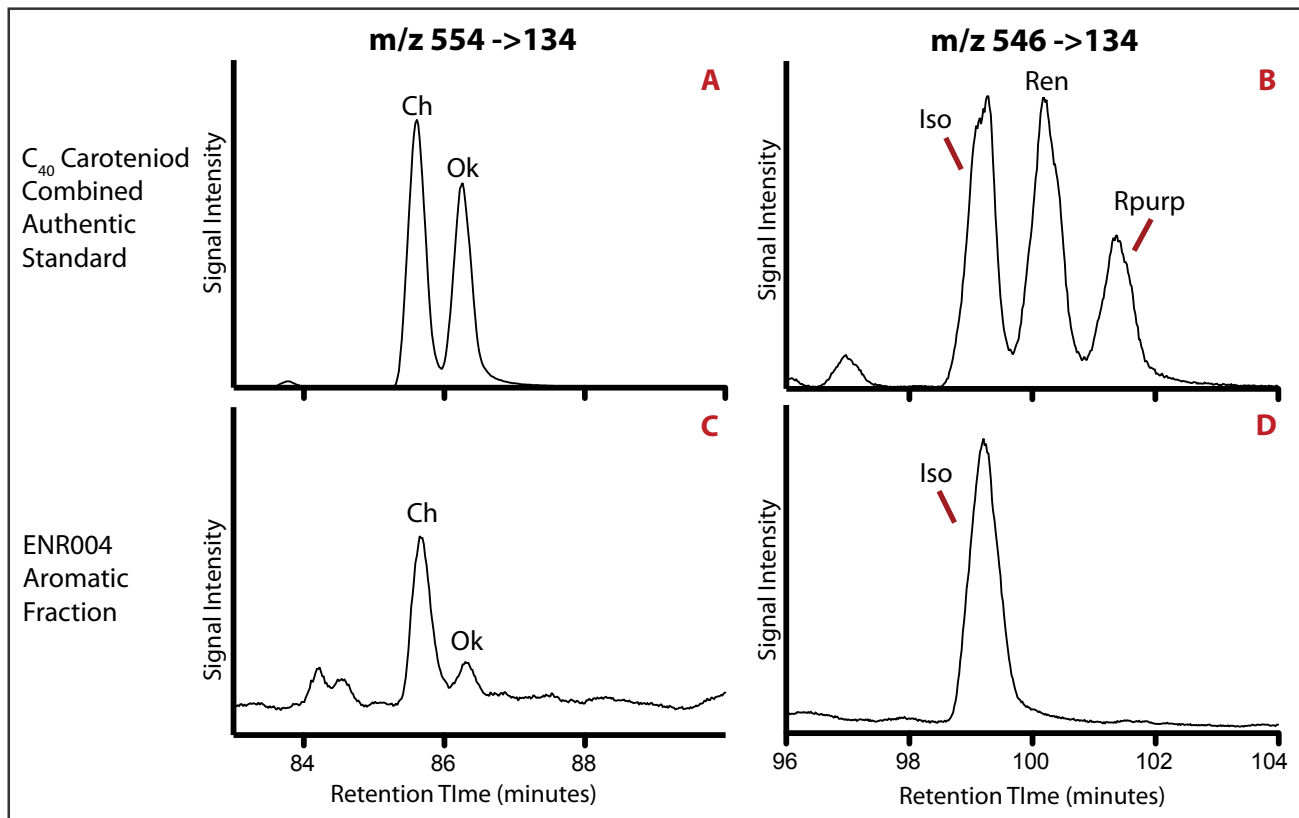
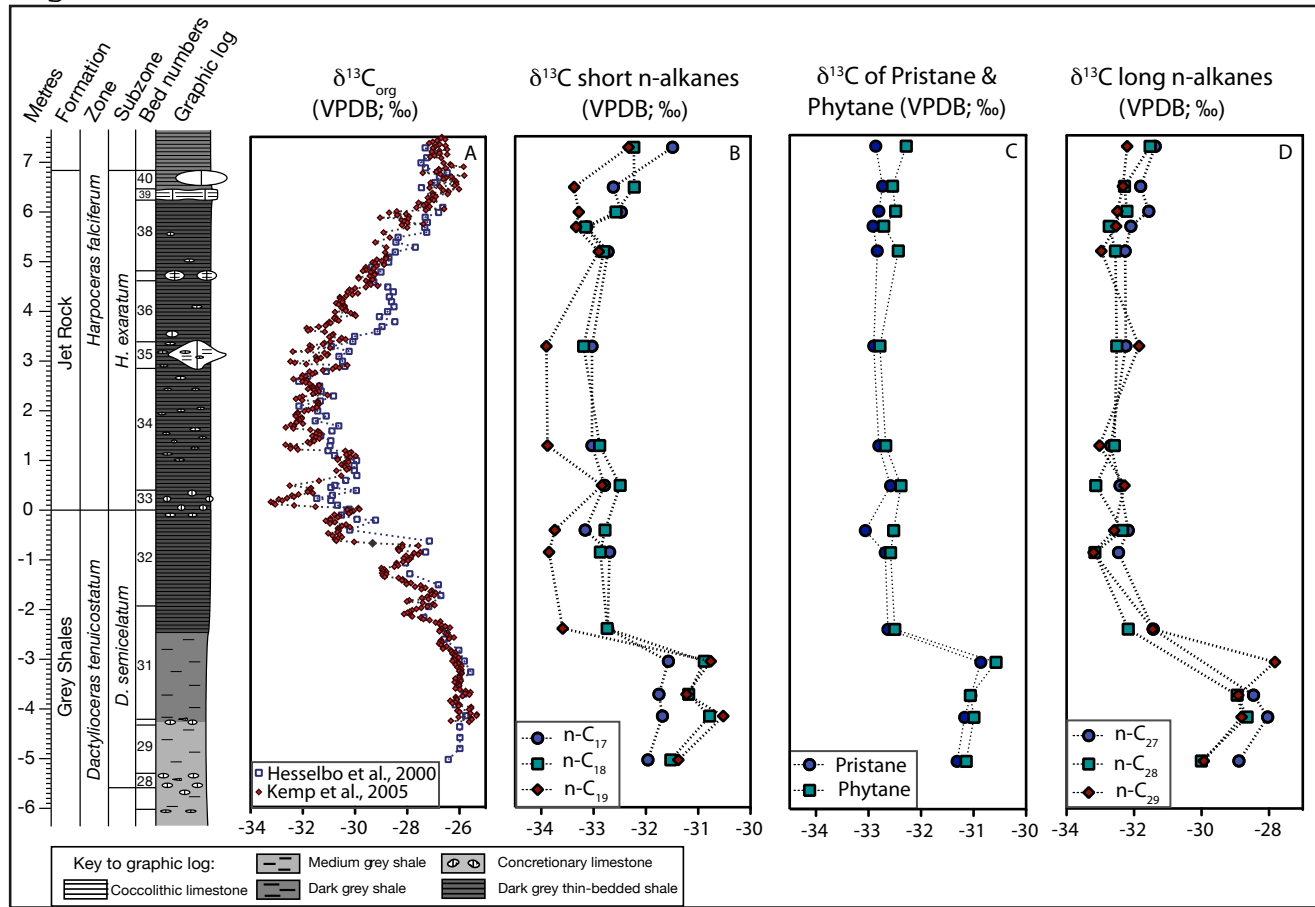


Figure 7



Supplementary Online Material (SOM)

3.1 Rock-Eval pyrolysis

Powdered sediment samples (~1 g) were analyzed at the University of Newcastle on a Rock-Eval pyrolysis instrument. The total organic carbon (TOC %), T_{\max} (°C), S_1 , and S_2 were determined. The S_1 and S_2 are expressed in mg hydrocarbons (HC) per gram dry rock. These parameters were used to calculate the hydrogen index (HI) ($HI = [100 * S_2] / TOC$; expressed in mg hydrocarbon (HC)/g TOC) and the production index (PI) ($PI = S_1 / [S_1 + S_2]$). The Rock-Eval TOC values and the bulk $\delta^{13}C_{org}$ have been previously reported (Hesselbo et al., 2000).

3.2 Organic petrography

In order to examine the nature of the organic matter, particularly the fraction of terrigenous organic matter, four samples across the section were prepared for optical analysis of the kerogen. Kerogen was isolated from the sample material remaining after lipid extraction as described in section 3.3. The mineral matrix was removed by the sequential addition of HCl and HF. Samples were centrifuged and rinsed between acid treatments, and the residual matter was rinsed with water and methanol. A subsample of each kerogen sample was mounted onto a slide in duplicate and assessed optically under white light and fluorescent light using a Zeiss research microscope and a Zeiss x 40 Plan-Neofluar objective. A Zeiss Axioskop, Axio Image D1, and a Zeiss 18 filter set were used to take photomicrographs and fluorescence images.

The fraction of terrigenous organic matter was estimated for the samples that were not analyzed by organic petrography using the linear relationship between the percent terrigenous organic matter measured by optical microscopy and the corresponding HI:

$$f_{Terr\ OM} (\%) = -0.279 * HI + 136 \quad (1)$$

where the linear regression had an R^2 value of 0.90 and $f_{Terr\ OM}$ represents the terrigenous organic matter as a percentage.

3.3 Biomarker extraction and analysis

Powdered samples (~ 5 g) were extracted using a Dionex ASE 200 Accelerated Solvent Extractor at 1000 psi and 100°C, with a solvent mixture of dichloromethane:methanol 9:1 (v/v). The total lipid extract (TLE) was reacted with acid-activated copper shots to remove elemental sulfur. Asphaltenes were separated (3x) from the maltene fraction by precipitation in n-pentane at 4°C, and after centrifugation at 3000 rpm for 10 minutes. The maltene fraction was then separated into saturated, aromatic, and polar fractions by silica gel chromatography using hexane, 1:1 (v/v) hexane/dichloromethane, and 7:3 (v/v) dichloromethane/methanol.

The saturated fractions were screened by gas chromatography-mass spectrometry (GC-MS) in full scan using an Agilent 6890 GC equipped with a HP6890 autosampler and interfaced to an Agilent 5973 mass spectrometer. Saturated hydrocarbons were also analyzed by gas chromatography-metastable reaction monitoring-mass spectrometry (GC-MRM-MS) on a Micromass Autospec Ultima mass spectrometer coupled with an Agilent 6890N GC. The analysis was carried out with a 60 m J&W Scientific DB-1 fused silica capillary column (internal diameter: 0.25 mm; 0.25 m film thickness) in pulsed splitless mode. The initial GC oven temperature was programmed to 60°C (held for 2

minutes), ramped to 150°C at 10°C/minute, and then to 315°C at 3°C/minute (held for 24 minutes). The ion source was in EI mode at a temperature of 250°C, an ionization energy of 70 eV, and acceleration voltage of 8000 kV. Tricyclic terpanes and hopanes were identified by MRM using the molecular ion to the m/z 191 transitions. Likewise, steranes were identified by MRM using the molecular ion to m/z 217 transitions.

The aromatic fraction was analyzed by GC-MS in SIM modes. Prior to analysis, 400 ng of an aromatic internal standard, deuterated phenanthrene, was added to each sample. The GC was fitted with a DB-5 stationary phase column, and the GC oven temperature was ramped from 60°C to 150°C at 20°C/minute, and then to 330°C at 4°C/minute (held for 27 minutes). The aryl isoprenoids and isorenieratane was identified in the m/z 134 ion chromatograms and quantified using the internal standard. Absolute quantification is not possible without taking into account relative response factors but our approach does allow an internally consistent estimation across the sample set.

The aromatic fraction was also analyzed by GC-MS in full scan and MRM modes on a Micromass Autospec Ultima mass spectrometer coupled with an Agilent 6890N GC autospec. The GC was fitted with a DB-5 stationary phase column. Polycyclic aromatic hydrocarbons (PAHs) were identified and quantified in full scan mode by their mass spectra and by comparison with a mix of authentic standards, with the exception of retene, coronene, and triphenylene, which were identified by their mass spectra and relative retention time. Aromatic carotenoid derivatives were also analyzed by GC-MRM-MS using parent-daughter reactions. Isorenieratane, okenane, and chlorobactane were identified in characteristic MRM transitions by comparison of retention times to an extract from the Barney Creek Formation (BCF) and a standard mix of hydrogenated

carotenoids containing chlorobactane, okenane, isorenieratane, renieratane, and renierapurpane. Using the MRM data, okenane and chlorobactane were quantified against the GC-MSD quantified isorenieratane. The aromatic carotenoid derivative and PAH concentrations were normalized against mass of TLE and TOC.

Compound specific carbon isotopic measurements of saturated hydrocarbons were made by gas chromatography/combustion/ isotope ratio mass spectrometry (GC-C-IRMS) using a Thermo Finnigan Delta plus XP coupled to a Thermo Finnigan Trace GC. The initial oven temperature was programmed to 60°C (held for 3 minutes), ramped to 180°C at 10°C/min, and then to 320°C at 4°C/min (held for 20 minutes). All samples were bracketed by pulses of in house calibrated reference CO₂ gas and Oztech calibrated reference CO₂ gas. A standard mix of *n*-alkanes (mix A; Arndt Schimmelmann, Indiana University) was analyzed twice a day to monitor the instrument condition. The mean value of triplicate analyses are reported here in permil (‰) relative to Vienna Pee Dee belemnite (VPDB), and the standard deviation from the mean value was better than 0.4‰.