

1 **Re-Pt-Os Isotopic and Highly Siderophile Element Behavior in Oceanic**
2 **and Continental Mantle Tectonites**
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18 **INTRODUCTION**
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20 Tectonically-emplaced mantle rocks, such as ophiolites, abyssal peridotites and orogenic
21 peridotite massifs, provide a principle constraint on the composition of and processes in the
22 Earth's upper mantle (Bodinier and Godard 2003). In the past, these 'mantle tectonites' have
23 sometimes received different names because their history and origin has been unclear. Mantle
24 tectonites are now understood to reflect a range of geologic environments regarding their
25 emplacement and their origin (e.g., Dilek and Furnes 2013). The advantage of these rocks
26 compared to mantle xenoliths is the large-scale exposure of textural and compositional relations
27 between different rock types that can be used to identify processes such as melting, magma or
28 fluid transport, chemical reactions, mixing or deformation at a range of spatial scales. A
29 disadvantage of most mantle tectonites is that they commonly display substantial chemical
30 modification of some elements, resulting from widespread serpentinization at low temperatures.
31 In some cases, this may also affect abundances of several of the highly siderophile elements
32 (HSE: Re, Au, PGE: Os, Ir, Ru, Rh, Pt, Pd), however, this can be tested by comparison with
33 unaltered rocks of similar composition. As is discussed in Reisberg and Luguet (2015, this
34 volume), Harvey et al. (2015, this volume) and Aulbach et al. (2015, this volume), peridotite
35 xenoliths have their own alteration issues regarding sulfides and chalcophile elements.
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37 Numerous studies have obtained Os isotope and/or highly siderophile element abundance
38 data on many different types of mantle tectonites. Some of these studies have focused on large-
39 scale chemical and isotopic variations, others on grain size-scale compositional variations to
40 understand small-scale distribution processes. These studies have, together, significantly
41 advanced the understanding of the processes that fractionate the HSE in the mantle at different
42 spatial scales and have provided insights into the behavior of sulfide in the mantle – the phase
43 that typically hosts the vast majority of the strongly chalcophile elements, including the HSE.
44 Osmium isotopes and Re-Os model ages have provided tools to directly date melting of mantle
45 tectonites and have changed views on the efficacy of mixing and melting processes in the mantle.
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47 Here, we review these advances, which have mostly taken place in the past 15 years,
48 aided by new developments in isotope dilution and ICP-MS based techniques and the application
49 of in situ laser ablation ICP-MS. First, we provide a brief summary of the current views about
50 geodynamic environments of different mantle tectonites. Work on Os isotopes and HSE
51 abundances in mantle tectonites of different geodynamic settings will be reviewed subsequently.
52 In the Discussion, we summarize the views on processes and chemical behavior of Os isotopes
53 and the HSE in mantle tectonites.

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57 **BREVIA OF CONCEPTS, TERMINOLOGY AND ANALYTICAL CAVEATS**
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59 **Re-Pt-Os parameters**
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61 In this chapter, we normalize isotopic abundance ratios to ^{188}Os following currently
62 accepted conventions: $^{187}\text{Os}/^{188}\text{Os} = \text{measured } ^{187}\text{Os}/^{188}\text{Os}$, $^{186}\text{Os}/^{188}\text{Os} = \text{measured } ^{186}\text{Os}/^{188}\text{Os}$,
63 $^{187}\text{Os}/^{188}\text{Os}_i = \text{initial } ^{187}\text{Os}/^{188}\text{Os}$ at age t . Up to the late 1990s some workers used the $^{187}\text{Os}/^{186}\text{Os}$
64 ratio to compare variations in ^{187}Os in natural materials. After it became clear that some minerals
65 and rocks also display variations of radiogenic ^{186}Os from the decay of ^{190}Pt (Walker et al. 1997;
66 Brandon et al. 1998, 1999), it was suggested to use ^{188}Os as the stable reference isotope (e.g.,
67 Shirey and Walker 1998). In the present work, conversion of early $^{187}\text{Os}/^{186}\text{Os}$ and $^{187}\text{Re}/^{186}\text{Os}$
68 data to $^{187}\text{Os}/^{188}\text{Os}$ and $^{187}\text{Re}/^{188}\text{Os}$, respectively, was performed by multiplication with a value of
69 $^{186}\text{Os}/^{188}\text{Os} = 0.1203$. This value for $^{186}\text{Os}/^{188}\text{Os}$ was commonly obtained by early less precise and
70 accurate isotopic ratio determinations. Later high-precision measurements of Os isotopic
71 compositions of mantle-derived rocks using N-TIMS and faraday cup detection have yielded
72 lower $^{186}\text{Os}/^{188}\text{Os}$ for the Earth's mantle and the bulk silicate Earth (0.119838 ± 0.000003 , 2 s.d.,
73 Brandon et al. 2006). The measurement of Os isotopic ratios via OsO_3^- ions requires that the raw
74 data is corrected for interferences produced by the minor isotopes of O. Fractionation of Os
75 isotopes during mass spectrometric measurements is commonly corrected assuming $^{192}\text{Os}/^{188}\text{Os} =$
76 3.0827 (Luck and Allègre 1983, Shirey and Walker 1998). High-precision Os isotopic data
77 require more elaborate measurement and correction protocols (e.g., Brandon et al. 2005a, 2006,
78 Luguet et al. 2008a; Chatterjee and Lassiter, 2015).
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80 The deviation of $^{187}\text{Os}/^{188}\text{Os}$ of a sample from an arbitrary 'average' chondritic
81 composition (present $^{187}\text{Os}/^{188}\text{Os} = 0.12700$, $^{187}\text{Re}/^{188}\text{Os} = 0.40186$) at age t is given as γOs_t and
82 was calculated using the equation and parameters given in Shirey and Walker (1998). The decay
83 constant of ^{187}Re used for calculations is $1.666 * 10^{-11} \text{ year}^{-1}$ (Smoliar et al. 1996; Selby et al.
84 2007). The average chondritic $^{187}\text{Os}/^{188}\text{Os}$ and $^{187}\text{Re}/^{188}\text{Os}$ have no specific meaning other than as
85 a reference for comparing different materials. Initial $^{186}\text{Os}/^{188}\text{Os}$ of samples were calculated using
86 $\lambda_{^{190}\text{Pt}} = 1.48 * 10^{-12} \text{ year}^{-1}$ (Brandon et al. 2006). Rhenium depletion model ages T_{RD} (Ch) and Re-
87 Os model ages T_{MA} (Ch) have been defined previously relative to a chondritic evolution model
88 (Ch) using the parameters mentioned above (Walker et al. 1989; Shirey and Walker 1998).
89 Alternatively, these model ages may be calculated relative to the Re-Os evolution of the primitive
90 mantle model composition, e.g., T_{MA} (PM). The primitive mantle has a slightly higher $^{187}\text{Os}/^{188}\text{Os}$
91 (0.1296) and $^{187}\text{Re}/^{188}\text{Os}$ (0.4346) than the 'average' chondrite reference values (Meisel et al.
92 2001).
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94 **Normalization of concentration data**
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96 In the literature, normalizations of HSE abundances in mantle rocks are sometimes
97 performed relative to mean abundances in CI chondrites using data from compilations (e.g.,
98 Anders and Grevesse 1989; Lodders 2003; Horan et al. 2003). One disadvantage of this approach
99 is that the HSE composition of the earth's mantle (and of the bulk Earth) likely does not match CI
100 chondrites (Walker et al. 2002a, 2002b; Horan et al. 2003; Becker et al. 2006; Fischer-Gödde et
101 al. 2010; Fischer-Gödde et al. 2011). However, it does have the advantage of using a measureable
102 reference frame for normalization. An alternative approach to assess igneous fractionation of the
103 HSE in mantle and crustal rocks is to normalize to a primitive mantle model composition (PM,
104 sometimes also referred to as primitive upper mantle, PUM and bulk silicate earth, BSE), and to
105 arrange elements according to their incompatibility as it is commonly performed for lithophile
106 elements (e.g., Hofmann 1988). The HSE concentrations in PM used for normalization and the

107 sequence of HSE in normalized concentration diagrams are those given in Becker et al. (2006)
108 and Fischer-Gödde et al. (2011). This theoretically has the advantage of providing comparison
109 with the primitive mantle composition. Alternatively, a composition of ‘average depleted spinel
110 lherzolites’ (based on mantle xenoliths and tectonites) has been defined in the literature for
111 comparative purposes (Pearson et al. 2004). Here we use, in different situations to reflect
112 different aims, both normalization to primitive mantle and to ‘average’ chondrite values
113 calculated with equal weighting from ordinary, enstatite and carbonaceous chondrites, from data
114 compiled in Walker (2009) and from Fischer-Gödde et al. (2010). We also use both logarithmic
115 and linear scales to best display the variations present in each particular figure. The sequence of
116 HSE in normalized concentration diagrams of terrestrial rocks commonly follows the sequence of
117 increasing enrichment in basalts and komatiites (i.e. $Os < Ir < Ru < Rh < Pt < Pd < Au < Re \approx S$),
118 which is similar, but not always identical, to the depletion in many peridotites. Elemental patterns
119 in some peridotites, that differ from this general depletion sequence, reflect re-enrichment in Re,
120 Au, Pd and multi-stage histories.

122 **Precision and accuracy of concentration data and analytical issues**

124 Previous studies have indicated that some of the early analytical techniques used to
125 determine HSE abundances or Re-Os systematics did not always produce complete recovery of
126 Ir, Os and Ru, even at test portion masses of > 10 g (e.g., Shirey and Walker 1998; Meisel and
127 Moser 2004; Becker et al. 2006, Lorand et al. 2008, Meisel and Horan 2015, this volume). In the
128 discussion of processes that fractionate the HSE, we will primarily focus on more recent data that
129 have been obtained either by Carius tube digestion at enhanced temperatures ($T > 230^{\circ}C$), by
130 high-pressure asher (typically $> 300^{\circ}C$), or by improved NiS fire assay techniques (Gros et al.
131 2002). If isotopic ratios were analyzed by ICP-MS or, in the case of Os, N-TIMS, these methods
132 yield combined analytical uncertainties (1 s.d.) of concentrations that may range between better
133 than a few % for Re and 15 % for Au for well-homogenized whole rock powders of lherzolites
134 and test portion masses of about 2 grams or more (Meisel and Moser 2004; Pearson et al. 2004;
135 Becker et al. 2006; Lorand et al. 2008; Fischer-Gödde et al. 2011). Heterogeneity of abundances
136 of carrier phases of the HSE in powders of some peridotites is a well-known problem (‘nugget
137 effect’). In addition to the nugget effect, complete and reproducible digestion of refractory
138 platinum group element minerals (PGM) in some harzburgites or dunites, may represent a
139 challenge. Incomplete digestion of refractory alloy phases may bias ratios of Os, Ir and Ru. For
140 further details, see Meisel and Horan (2015, this volume).

143 **HIGHLY SIDEROPHILE ELEMENTS IN MANTLE TECTONITES FROM** 144 **DIFFERENT GEODYNAMIC SETTINGS**

146 **Summary of mantle tectonites and their geodynamic settings**

148 Mantle tectonites include peridotite sections of ophiolites, abyssal peridotites and
149 orogenic peridotites that often, but not exclusively, occur in orogenic belts (also known as
150 peridotite massifs, alpine or alpinotype peridotites). These different mantle tectonites can be
151 distinguished by their geodynamic setting, and associated emplacement history and pressure-
152 temperature (P, T) evolution, but also by their chemical composition. Most of the rocks
153 concerned record a relatively simple cooling history from lithospheric mantle conditions (T of
154 $1000-1300^{\circ}C$ and P of the garnet, spinel or plagioclase lherzolite stability field) to some lower T
155 and P equivalent to crustal conditions. Owing to their origin from in situ lithospheric or
156 asthenospheric mantle conditions, these rocks are sometimes also referred to as ‘high-temperature
157 peridotites’. In contrast, ‘low-temperature’ orogenic peridotites are former high-temperature
158 peridotites that have been subducted as part of a package of crustal rocks in collision zones (e.g.,
159 the Alpe Arami peridotite, Nimis and Trommsdorff 2001; peridotites of the Western Gneiss

160 region, Norway, Brueckner et al. 2010; Zermatt-Saas ophiolite, Barnicoat and Fry 1986). Low-
161 temperature peridotites were partially re-equilibrated at high P/T conditions, but in some cases,
162 this partial re-equilibration is hardly noticeable and chemical and textural features inherited from
163 the high-temperature history of the peridotites predominate (e.g., at the Lanzo peridotite massif;
164 Pelletier and Müntener 2006).

165
166 Improved understanding of the geodynamic evolution of passive continental margins
167 (Dilek et al. 2000), the transition to ocean spreading and the role of ocean spreading rate in the
168 lithological composition of the oceanic crust (Dick et al. 2006) have led to improved
169 interpretations of the origin and geodynamic environments of mantle tectonites and ophiolites
170 (Dilek et al. 2000; Dilek and Furnes 2013). It is now understood that high-temperature peridotite
171 tectonites derived from continental lithospheric mantle may be exhumed during slow extension of
172 continental lithosphere and the formation of sedimentary basins or small ocean basins. Well-
173 known examples are the island of Zabargad in the Red Sea (Brueckner et al. 1988), the Pyrenean
174 peridotite bodies in southern France (Vielzeuf and Kornprobst 1984; Bodinier et al. 1988), the
175 peridotite bodies of NW Italy (Ivrea-Verbano Zone, Lanzo; Ernst 1978; Sinigoi et al. 1983;
176 Shervais and Mukasa 1991; Mazzucchelli et al. 2009) and some of the mantle tectonites in the
177 Alps and in Italy that sometimes have been referred to as ‘ophiolites’ (for instance the External
178 Ligurian peridotites; Rampone et al. 1995). Some mantle tectonites were exhumed in oceanic
179 environments as indicated by their alteration and association with ophiolitic breccias, basalts,
180 gabbros and cherts. Such rocks, for instance the Internal Ligurian peridotites of the Tethys ocean,
181 do not show the classical Penrose-type ophiolite sequence and are most similar to exhumed
182 mantle in modern ultraslow spreading environments, e.g., like parts of the SW Indian ridge or the
183 Gakkal ridge (Dick et al. 2000, 2006; Michael et al. 2003). The classical Penrose-type ophiolite
184 stratigraphy, which is believed to be representative of moderate to fast spreading ocean ridges, is
185 represented by the Samail ophiolite in Oman and the Troodos ophiolite (Cyprus). However, it
186 should be noted that these ophiolite complexes were at least partly affected by convergent plate
187 margin processes (Dilek et al. 2000; Dilek and Furnes 2013).

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189 Indeed, many ophiolites probably formed close to subduction zones and were later
190 incorporated into the crust by collision of terranes or continental fragments. Evidence for the
191 proximity of subduction zones is mostly derived from the composition of associated igneous
192 rocks such as calcalkaline basalts or boninites. To what degree subduction processes affected the
193 mantle tectonites is not always clear. For instance, mantle rocks in the northwestern segments of
194 the ophiolites in Oman may have been influenced by supra-subduction zone melting processes or
195 by migration of magmas that formed in subarc mantle, as is indicated by the abundance of
196 podiform chromitite deposits in these rocks and the calcalkaline and boninitic affinities of the
197 crustal rocks (Boudier et al. 2000; Ishikawa et al. 2002). In contrast, the southern massifs of the
198 Samail ophiolite in Oman show little evidence for such rocks and the crust is predominantly
199 MORB-like in composition (Koga et al. 2001; Pallister and Knight 1981). Some peridotite
200 massifs contain abundant pyroxenite layers which sometimes carry chemical and isotopic
201 evidence for the significant presence of recycled crust components (e.g., Beni Bousera, Ronda,
202 Bohemian massif; Pearson et al., 1991a, 1991b, 1993; Becker, 1996a, 1996b). Such compositions
203 only occur in mantle tectonites from areas that may have undergone lithospheric delamination
204 and previous episodes of subduction. Some ‘ophiolites’, such as the Ligurian ophiolites (N Italy)
205 and similar complexes in the Alps, were not affected by convergent processes and are more
206 properly assigned to purely extensional environments (e.g., Piccardo and Guarnieri 2010;
207 Rampone et al. 1995, 1996).

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209 In the following sections, we will describe the HSE and Os isotopic characteristics of
210 different types of mantle tectonites in the context of their formation environments, as far as these
211 have been constrained. These sections contain basic information on the formation environment
212 and evolution of the ultramafic bodies together with the Re-Pt-Os isotopic and HSE composition

213 of their various mantle lithologies. We will proceed from abyssal peridotites and other mantle
 214 tectonites exhumed in extensional geodynamic environments to peridotite massifs and ophiolites
 215 affected by magmatic processes at convergent plate margins. Interpretations of these
 216 compositions will then follow in the Discussion. The geological settings covered, locations,
 217 available HSE data and key references are summarized in Table 1.
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Table 1. Locations, geological settings, available data and references for samples discussed in this chapter

Setting and location		HSE data	Key HSE references
<i>Abyssal peridotites</i>			
Atlantic - North	Kane Fracture Zone	¹⁸⁷ Os, HSE	Snow & Schmidt, 1998; Rehkämper et al., 1999; Brandon et al., 2000; Luguët et al., 2001; Becker et al., 2006
	15° 20' N Fracture Zone	¹⁸⁷ Os, HSE, Se-Te	Harvey et al., 2006; Marchesi et al., 2013
	Azores, North Atl Ridge	¹⁸⁷ Os	Roy-Barman & Allègre, 1994
Atlantic - South	Shaka, 59° S, Bouvet, Dinguian, Islas Orcadas FZs	¹⁸⁷ Os	Snow and Reisberg, 1995
Indian	SWIR	¹⁸⁷ Os (WR, sulf), HSE	Roy-Barman & Allègre, 1994; Snow & Schmidt, 1998; Luguët et al., 2001; Alard et al. 2005; Warren & Shirey, 2012
Pacific	East Pacific Rise, Hess Deep	¹⁸⁷ Os, HSE	Roy-Barman & Allègre, 1994; Snow & Schmidt, 1998; Rehkämper et al., 1999
Arctic	Gakkel Ridge	¹⁸⁷ Os (WR, sulf), HSE	Liu et al., 2008; 2009; Warren & Shirey, 2012;
	Lena Trough	¹⁸⁷ Os	Lassiter et al., 2014
<i>Oceanic mantle tectonites</i>			
Italy	Internal Ligurides	¹⁸⁷ Os, HSE	Rampone et al., 1996; Snow et al., 2000; Luguët et al., 2004; Fischer-Gödde et al. 2011
<i>Ophiolites – Little or no subduction influence</i>			
Oman	Samail	¹⁸⁷ Os, HSE	Hanghøj et al., 2010
Chile	Taitao	¹⁸⁷ Os	Schulte et al., 2009
<i>Ophiolites – Uncertain origin</i>			
Tibet	Luobusa, Dongqiao	¹⁸⁷ Os (WR, PGM), HSE	Zhou et al., 1996; Becker et al., 2006; Shi et al., 2007; Pearson et al., 2007
Finland	Jormua, Outokumpu	¹⁸⁷ Os	Tsuru et al., 2000; Walker et al., 1996
Austria	Eastern Alps	¹⁸⁷ Os	Meisel et al., 1997
Turkey, Tethyan	Marmaris, Tekirova, Koycegiz	¹⁸⁷ Os, HSE	Aldanmaz et al., 2012
Cuba	Mayari-Cristal	¹⁸⁷ Os (PGM)	González-Jiménez et al., 2009; Marchesi et al., 2011
<i>Ophiolites – Convergent margin origin</i>			
Cyprus	Troodos	¹⁸⁷ Os, HSE	Büchl et al., 2002; 2004;
Scotland	Unst, Shetland	¹⁸⁷ Os, HSE	Prichard and Lord, 1996; O'Driscoll et al., 2012
Philippines	Zambales	HSE	Zhou et al., 2000
Iraq	Qalander	HSE	Ismail et al., 2014
Egypt	Eastern Desert ophiolite	¹⁸⁷ Os (PGM)	Ahmed et al., 2006
California	Feather River, Josephine, others	HSE, ¹⁸⁷ Os (PGM)	Agranier et al., 2007; Meibom et al., 2002; Pearson et al. 2007

Convergent margin tectonites (High temperature)

Spain	Ronda	¹⁸⁷ Os, HSE	Reisberg et al., 1991; Reisberg and Lorand, 1995; Gueddari et al., 1996; Lorand et al., 2000; Fischer-Gödde et al., 2011; Marchesi et al., 2014
Morocco	Beni Bousera	¹⁸⁷ Os, HSE	Gueddari et al., 1996; Roy-Barman et al., 1996; Pearson et al., 2004
Lower Austria, Czech Republic	Bohemian Massif	¹⁸⁷ Os, HSE	Becker et al., 2001; 2004; 2006; Ackerman et al., 2013

Continental/Continent-ocean transitional tectonites

France	Lherz	¹⁸⁷ Os, HSE, Se-Te, PGM	Reisberg & Lorand, 1995; Becker et al., 2006; Luguët et al., 2007; Lorand et al., 2010; Riches & Rogers, 2011; König et al., 2014
France: other Pyrenees	e.g. Turon, Freychinede, Fontete Rouge	¹⁸⁷ Os, HSE	Reisberg & Lorand, 1995; Lorand et al., 1999; Becker et al., 2006; Fischer-Gödde et al., 2011
Italy	Ivrea Zone: Baldissero, Balmuccia	¹⁸⁷ Os, HSE, Se-Te	Wang et al., 2013; Wang and Becker, 2015
Italy	External Ligurides	¹⁸⁷ Os, HSE	Rampone et al., 1995; Snow et al., 2000;
Swiss Alps	Totalp	¹⁸⁷ Os, HSE	van Acken et al., 2008; 2010a; 2010b
Switzerland	Helvetic domain	¹⁸⁷ Os	Meisel et al., 1996
Japan	Horoman	¹⁸⁷ Os, HSE	Rehkamper et al., 1999; Saal et al., 2001
Italy	Lanzo	¹⁸⁷ Os, HSE	Lorand et al., 2000; Becker et al., 2006; Fischer-Gödde et al., 2011

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HSE in abyssal peridotites from spreading oceanic lithosphere

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Rocks from slow spreading ridges share many characteristics with mantle tectonites exhumed in passive continental margin or transitional oceanic environments (see later sections). That is, a spectrum of peridotite compositions is often present, including lherzolites, harzburgites and replacive dunites. However, in some cases (e.g., 15° 20' N fracture zone, Atlantic Ocean; Harvey et al. 2006), a greater degree of serpentinisation is present, sometimes with little primary mineralogy remaining, possibly due to the nature of emplacement and exposure of abyssal peridotites, either with little overlying crust (slow-ultra slow spreading) or bounded by transform faults. This can be important for the budgets of the HSE, as is discussed in the first section of the discussion.

Abyssal peridotites contain major and trace element evidence for significant melt depletion, and isotopic evidence for that melt extraction being ancient, with long-term depletion of incompatible elements. Early studies found Os isotope evidence for this depletion, with ¹⁸⁷Os/¹⁸⁸Os ratios between 0.1208 and 0.1304 in abyssal peridotite whole-rocks from several global localities (Martin, 1991; Roy-Barman and Allègre, 1994; Snow and Reisberg, 1995). These ratios range from close to the estimate for the primitive upper mantle (0.1296; Meisel et al., 2001) to values which equate to Re depletion at ~1.2 Ga (T_{RD}, see Fig. 4), assuming all Re was stripped from the residue during melting (Shirey and Walker, 1998). In reality, Re remains present in abyssal peridotites, although typically at much lower abundances than in the PM. This means that the actual age of depletion is older than calculated for a T_{RD} age, because evolution of ¹⁸⁷Os/¹⁸⁸Os didn't cease entirely after depletion. However, all abyssal peridotite sample suites display evidence for recent open system behavior, most probably in the form of Re addition (e.g., Harvey et al., 2006), but also sometimes Os loss and enrichment in ¹⁸⁷Os (Snow and Reisberg, 1995). This is apparent in the sub-horizontal trends within suites which show similar ¹⁸⁷Os/¹⁸⁸Os over a range of ¹⁸⁷Re/¹⁸⁸Os ratios (Fig. 1), and is consistent with the evidence for extensive

248 serpentinisation during fluid-rock interaction (e.g., Harvey et al., 2006). This seawater
249 interaction can also be coupled with elevated and lowered $^{87}\text{Sr}/^{86}\text{Sr}$ and $^{143}\text{Nd}/^{144}\text{Nd}$ ratios,
250 respectively (Snow and Reisberg, 1995). However, despite the extremely radiogenic isotopic
251 composition of seawater ($^{187}\text{Os}/^{188}\text{Os} \sim 1.05$; Levasseur et al., 1998), the modelled effects of
252 seawater interaction on Os isotopes are small except at very high fluid/rock ratios (Fig. 17), due
253 to the very low Os concentration in seawater (~ 11 fg/g; Levasseur et al., 1998) compared to
254 mantle samples (~ 1 -5 ng/g). Moreover, a comparison of the rims and cores of abyssal peridotites
255 from Gakkel Ridge in the Arctic Ocean found no systematic difference in Os contents and only a
256 very small increase in $^{187}\text{Os}/^{188}\text{Os}$ from core to rim (Liu et al., 2008). A possible alternative
257 source of radiogenic Os is by reaction with percolating melts from enriched lithologies. There is,
258 however, a much larger effect of seawater interaction on Re/Os ratios (cf. ~ 7.3 pg/g Re in
259 seawater; Anbar et al., 1992), with examples of sample rims reset while sample cores display a
260 co-variation between Re/Os and Al_2O_3 contents, which must be a primary melt depletion feature
261 (Liu et al., 2008).

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263 In general, the processes of alteration mean that the measured Re-Os elemental and
264 isotopic values may not accurately represent the long-term history of abyssal peridotites, casting
265 doubt on the accuracy of T_{RD} ages. Nonetheless, all abyssal peridotite suites consist primarily of
266 samples with Os isotope ratios ranging from close to the PM estimate to sub-chondritic values
267 (Fig. 1), reflecting long-term evolution in a low Re/Os environment following ancient melt
268 depletion. Seawater interaction can only increase $^{187}\text{Os}/^{188}\text{Os}$, so both alteration and minor
269 ingrowth of ^{187}Os since depletion would only serve to reduce the apparent age.

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271 Snow and Reisberg (1995) proposed an ‘uncontaminated’ range for abyssal peridotites of
272 0.1221 to 0.1270, with a mean of 0.1246. Both Snow and Reisberg (1995) and Roy-Barman and
273 Allègre (1994) identified that this range was less radiogenic than the range of early MORB
274 analyses. Further analyses of samples from a forearc region and from slow or ultra-slow
275 spreading ridges have significantly extended the known range of Os isotope compositions;
276 whole-rock $^{187}\text{Os}/^{188}\text{Os}$ of 0.119, 0.117 and 0.114 were found, respectively, from the Izu-Bonin
277 forearc (Parkinson et al., 1998), ODP Hole 1274a ($15^\circ 20'$ N transform, mid-Atlantic; Harvey et
278 al., 2006) and Gakkel Ridge (Arctic; Liu et al., 2008). The unradiogenic samples of the forearc
279 setting were first thought to indicate that subduction zones might be ‘graveyards’ for ancient
280 depleted mantle (Parkinson et al., 1998). While mantle in subduction zones may be extremely
281 depleted, the findings from the $15^\circ 20'$ N transform and Gakkel Ridge indicate that such portions
282 of ancient depleted mantle are likely present throughout the upper mantle.

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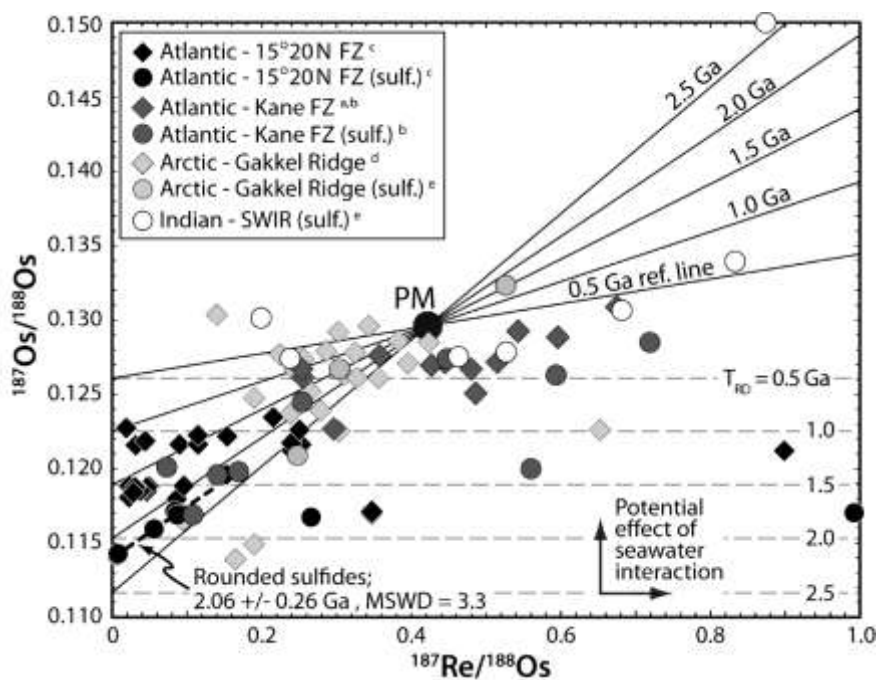
284 Sulfide compositions display greater Os isotopic variation than whole-rocks (some
285 plotting at $^{187}\text{Re}/^{188}\text{Os}$ ratios up to ~ 12) and can be divided into two broad groups: rounded
286 intragranular grains and more skeletal interstitial sulfides (Alard et al., 2005; Harvey et al., 2006).
287 The latter typically have higher Re/Os and more radiogenic Os isotope signatures (see
288 Discussion), and the rounded, included sulfides possess the least radiogenic $^{187}\text{Os}/^{188}\text{Os}$, lower
289 than the host whole-rocks, reflecting depletion and isolation since an ancient melting episode.

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291 Assuming that isochron information is typically compromised due to recent open-system
292 behavior (see Fig. 1), then minimum Re depletion ages must be utilized; these are shown by the
293 horizontal dashed lines in Fig. 1. The least radiogenic whole-rocks from Gakkel and sulfides
294 from the $15^\circ 20'$ N transform equate to T_{RD} ages in excess of 2 Ga. The six rounded sulfides from
295 one sample from Hole 1274a actually display a near-isochronous relationship. The age of this
296 errorchron is ~ 2.05 Ga, consistent with the T_{RD} ages for these sulfides. Sulfides from South-West
297 Indian ridge peridotites (Warren and Shirey, 2012) typically have more radiogenic compositions,
298 closer to the PM value, and their sub-horizontal array suggests relatively recent resetting of their
299 Re/Os ratios (Fig. 1). However, when combined with data from Alard et al. (2005) and with Pb

300 isotope data (Warren and Shirey, 2012), the broader array appears to give an age approaching 2
 301 Ga.

302
 303 As well as constraints on the ^{187}Os evolution of the convecting mantle, the combined
 304 ^{186}Os - ^{187}Os systematics of abyssal peridotites from the Kane transform area of the Atlantic Ocean
 305 have been studied (Brandon et al., 2000). The average $^{186}\text{Os}/^{188}\text{Os}$ of these samples is 0.1198353
 306 \pm 0.0000007, identical to the mean from alloys and chromitites (Walker et al., 1997; Brandon et
 307 al., 2006), indicating the general absence of significant fractionation of Pt and Os in the abyssal
 308 and ophiolite environments. The Kane samples display a co-variation of $^{187}\text{Os}/^{188}\text{Os}$ with Pt/Os
 309 ratio which would likely not have been preserved if recent melting had taken place (Brandon et
 310 al., 2000). No covariation of $^{187}\text{Os}/^{188}\text{Os}$ and Re/Os, due to seawater interaction, exists. The
 311 variability of $^{187}\text{Os}/^{188}\text{Os}$ could either be ascribed to differing ages of depletion or to variable
 312 degrees of depletion, perhaps with garnet present in which Re is thought to be compatible
 313 (Righter and Hauri, 1998). Brandon et al. (2000) proposed that Re is only depleted by about 40%
 314 in these rocks, therefore requiring very ancient melt depletion to produce the most unradiogenic
 315 samples. This ancient melting is not evident in $^{143}\text{Nd}/^{144}\text{Nd}$, indicating decoupling of the two
 316 isotope systems, perhaps due to the Nd budget being predominantly hosted by clinopyroxene
 317 which is continually involved in partial melting, whereas the Os budget is likely dominated by
 318 included sulfides which are isolated from moderate degrees of partial melting and thus retain an
 319 ancient signature (Brandon et al., 2000). The later work of Harvey et al. (2006), outlined above,
 320 supports the influence of shielded sulfides, which control much of the whole-rock Os signature.
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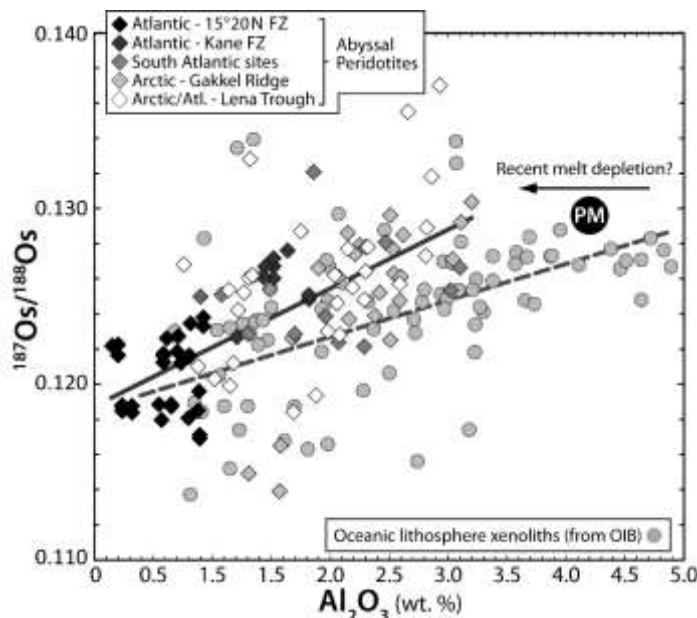


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324 **Figure 1.** Re-Os isochron diagram for separated sulfides and whole-rock abyssal peridotites from the
 325 Atlantic, Arctic and Indian Oceans. Sub-horizontal trends within all sample suites indicate recent open
 326 system behavior (most probably Re addition) but sub-PM Os ratios strongly predominate in all suites,
 327 reflecting long-term evolution in a low Re/Os environment following ancient melt depletion. Sulfides
 328 display the greatest Os isotope variations, with the least radiogenic values and some radiogenic values
 329 ($^{187}\text{Os}/^{188}\text{Os}$: \sim 0.167) plotting with $^{187}\text{Re}/^{188}\text{Os}$ ratios up to \sim 12. Age reference lines are shown as solid
 330 lines; Re depletion (T_{RD}) ages are shown as horizontal dashed lines. The dashed line sub-parallel to the age
 331 reference lines is the best fit line for six rounded sulfides from a single Hole 1274a abyssal peridotite.
 332 Whole-rock data for Indian and Pacific peridotites are not shown due to the paucity of available data. Refs:
 333 ^a Brandon et al. (2000), Becker et al. (2006); ^b Alard et al. (2005); ^c Harvey et al. (2006); ^d Liu et al. (2008);
 334 ^e Warren and Shirey (2012).

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The apparent disconnect between abyssal peridotites and their overlying crust found in the early Os isotope abyssal studies (also see Discussion) is clearly seen in refractory Macquarie Island peridotites (Southern Ocean) and their surprisingly enriched overlying crust (Dijkstra et al., 2010). Here, a slow spreading and low productivity ridge would not be expected to account for the 20-25% near fractional melting suggested by the very high Cr numbers for spinel (0.40-0.49) in the peridotites. Although many authors have suggested a minor or absent role for abyssal peridotites in the generation of oceanic crust (e.g., Liu et al., 2008; Dijkstra et al., 2010), a compilation of abyssal peridotite data by Lassiter et al. (2014), including new analyses of Lena Trough peridotites ($^{187}\text{Os}/^{188}\text{Os}$: 0.118-0.130, average 0.1244), is remarkably similar to the distribution of $^{187}\text{Os}/^{188}\text{Os}$ in xenoliths entrained in ocean island basalts (Fig. 2 and references in caption). The authors argue that this range of $^{187}\text{Os}/^{188}\text{Os}$ for both suites represents the composition of the convecting mantle, and is inconsistent with a refractory ‘slag’ hypothesis for abyssal peridotites (cf. Rampone and Hofmann, 2012). One issue with this interpretation, however, is that OIB xenoliths likely do not represent a deep source mantle for those melts, and instead might sample the lithospheric mantle which is plausibly genetically related to abyssal peridotites. Nonetheless, on an $^{187}\text{Os}/^{188}\text{Os}$ - Al_2O_3 diagram (sometimes called an ‘aluminachron’, Fig. 2, where Al_2O_3 is used as a proxy for melt- and Re-depletion), abyssal peridotites and OIB xenoliths produce best-fit lines with similar ‘initial’ values, but differing slopes (the intersection of the correlation with the $^{187}\text{Os}/^{188}\text{Os}$ axis at $\text{Al}_2\text{O}_3 = 0$ yields the initial $^{187}\text{Os}/^{188}\text{Os}_i$ at the time of the partial melting event). The similarity of the most depleted ‘initial’ values suggests that the age of Re depletion is similar for the two suites. So rather than the different slopes reflecting different depletion ages, the steeper trend of the abyssal suite instead suggests additional recent depletion of Al during partial melting to form new oceanic crust (Lassiter et al., 2014). This argues for a role for abyssal peridotites in the formation of mid-ocean ridge basalts. It remains possible, however, that the trends instead represent mixing between melts and residues and that the differing slopes reflect different conditions (e.g. depth, fS_2 etc.) of such mixing.



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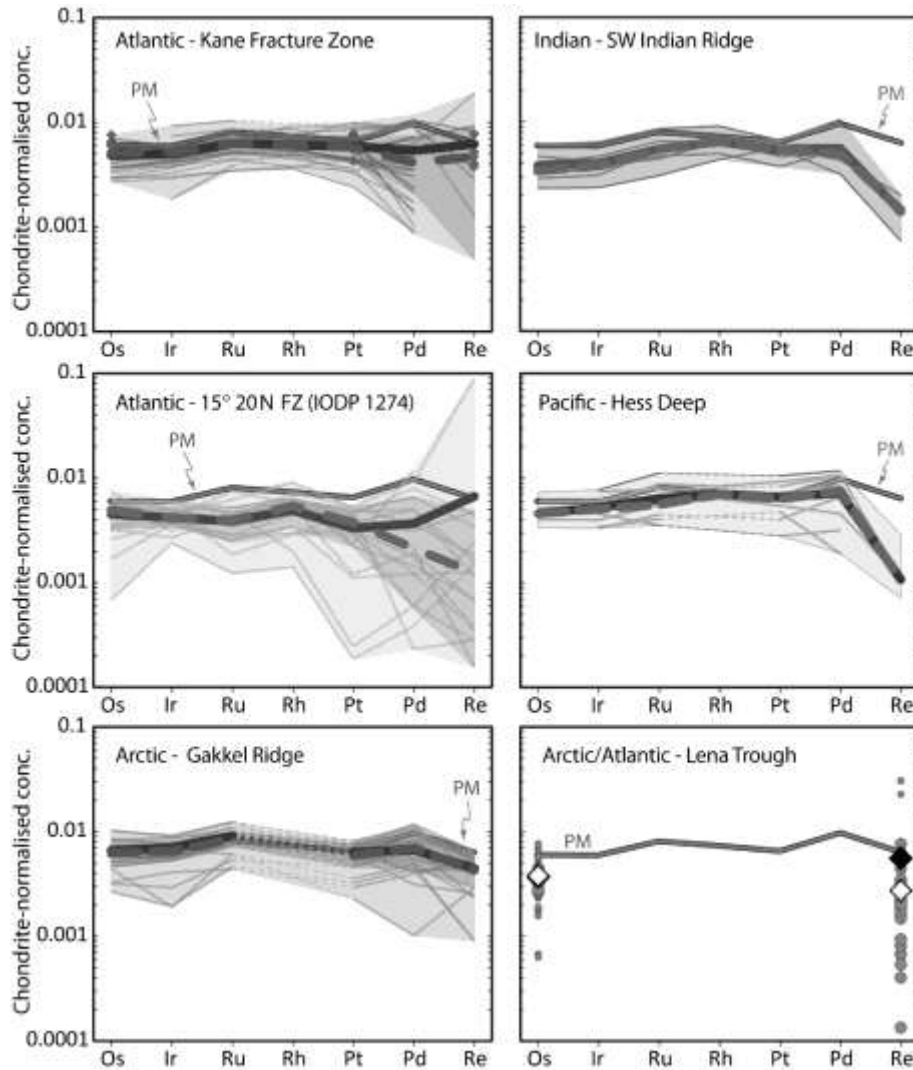
Figure 2. $^{187}\text{Os}/^{188}\text{Os}$ - Al_2O_3 diagram (after Reisberg and Lorand, 1995, see also Fig. 4b) for whole-rock abyssal peridotites from the Atlantic and Arctic Oceans compared to xenoliths entrained by ocean island basalts (after Lassiter et al., 2014). There is considerable scatter in both the abyssal and OIB xenolith datasets, possibly reflecting variable ages of melt depletion, or recent resetting of $^{187}\text{Os}/^{188}\text{Os}$ by seawater or melt interaction. Overall, the best-fit lines for the two suites have similar ‘initial’ values, suggesting similar mean ages of depletion, but the abyssal peridotite trend is significantly steeper. Rather than representing an older age, this likely reflects depletion of Al during recent melting to form oceanic crust,

371 which would not therefore affect $^{187}\text{Os}/^{188}\text{Os}$. Data sources as in Fig. 1, except Lena Trough and South
372 Atlantic abyssal peridotites from Lassiter et al. (2014) and Snow and Reisberg (1995), respectively. Circles
373 represent peridotite xenoliths entrained in ocean island basalts (Hassler and Shimizu, 1998; Widom et al.,
374 1999; Becker et al., 2006; Bizimis et al., 2007; Simon et al., 2008; Ishikawa et al., 2011).

375

376 Analyses of the range of HSE in abyssal peridotites showed that they are not present in
377 strictly chondritic proportions (Snow and Schmidt, 1998), and thus may not be consistent with
378 the theory that HSE in the silicate Earth were derived from a late veneer of primitive chondritic
379 material, after core formation had ceased (Chou, 1978). Snow and Schmidt (1998) proposed that
380 mantle HSE patterns instead reflected remixing of the outer core into the mantle. However,
381 subsequent analyses using improved digestion techniques (Becker et al., 2006) cast doubt on the
382 robustness of the Os, Ir and Ru data in that study (obtained by NiS fire assay), reducing the
383 magnitude of the observed non-chondritic signature. Moreover, later work highlighted the
384 importance of metasomatism and melt-rock reaction processes in producing non-chondritic HSE
385 patterns in mantle rocks. Rehkämper et al. (1999) found that abyssal peridotites broadly
386 contained HSE in chondritic proportions and that HSE ratios were inconsistent with an outer core
387 input. Where non-chondritic ratios were identified in the Horoman peridotite, a petrogenetic
388 model showed that these ratios were consistent with sulfide addition associated with melt
389 percolation. Alard et al. (2000) then identified PPGE-rich (Pt-group) and IPGE poorer (Ir-group;
390 Barnes et al. 1985) interstitial sulfides that were introduced during melt infiltration. These
391 sulfides have the potential to produce non-chondritic HSE patterns in whole-rocks and also have
392 suprachondritic Re/Os and $^{187}\text{Os}/^{188}\text{Os}$.

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Figure 3. Chondrite-normalized concentration diagrams of the HSE in abyssal peridotites from the Atlantic, Indian, Pacific and Arctic Oceans. Average chondrite values were calculated with equal weighting from ordinary, enstatite and carbonaceous chondrites, from data compiled in Walker (2009) and from Fischer-Gödde et al. (2010). Thick solid lines denote mean compositions, thick broken lines are median values. The lightly shaded fields cover the whole range of compositions, while the darker shade represents only compositions within one standard deviation of the mean for that data set (not used for Indian and Pacific, where $n \leq 8$). For the Lena Trough (F), the black diamond denotes the mean and the white diamond is the median, smaller circles fall outside one standard deviation. References as for Figures 1 and 2, plus Kane FZ and Pacific – Rehkämper et al. (1999); Kane and Indian - Luguët et al. (2001; 2003); Kane, Indian, Pacific - Snow & Schmidt (1998); 15° 20 N FZ - Marchesi et al. (2013).

407 A study of Kane fracture zone peridotites (Atlantic Ocean) identified a range of HSE
 408 systematics in different lithologies (Luguët et al., 2003). Harzburgites have low Pd/Ir ratios and
 409 are sulfide-poor. Refertilised harzburgites often have higher concentrations of Pd, while
 410 lherzolites have approximately chondritic proportions of HSE and between 100 and 300 $\mu\text{g/g}$ S,
 411 which encompasses the estimate for the PM (250 \pm 50 $\mu\text{g/g}$; Lorand, 1990; McDonough and
 412 Sun, 1995; Palme and O'Neill, 2014). Peridotites from the 15°20' N fracture zone are typically
 413 more depleted (Marchesi et al., 2013) than those from Kane (Figure 3), and show complete
 414 consumption of sulfide in some cases, presumably with HSE (particularly the IPGE) then hosted
 415 by PGM. Both this study, and an earlier one looking at two sites with differing alteration from
 416 Gakkel Ridge (Liu et al., 2009), found there to be no significant mobilisation of the HSE during

417 serpentinisation, but S contents were reduced. The same is also true of weathering, except for Re
418 and Pd in some cases. For the 15°20' N fracture zone, there was also no observed mobilisation of
419 HSE by sulfur- and Si-undersaturated melt, which is somewhat surprising given that S-
420 undersaturated melt would be expected to dissolve sulfide. Presuming that sulfide was dissolved
421 into the melt, the implication from the 15°20' N samples is that all HSE are retained until sulfide
422 is almost exhausted. However, this finding may be dependent on the phase relations in any given
423 sulfide system, as fractionation of sulfide melt from solid sulfide would be expected to
424 fractionate PPGE from IPGE (e.g., Mungall et al., 2005). In contrast to studies advocating melt
425 percolation as a means to fractionate HSE (e.g., Alard et al., 2000), Liu et al. (2009) contend that
426 supra-chondritic Ru/Ir and Pd/Ir in Gakkel peridotites cannot be reconciled with melt enrichment
427 and therefore instead support an inherent primitive origin for such ratios.

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430 **HSE in mantle tectonites from continental extensional domains and continent-ocean** 431 **transitions**

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433 Mantle tectonites exhumed in passive continental margin or transitional oceanic
434 environments share many characteristics with similar rocks from ultraslow spreading ridges (see
435 abyssal peridotite section). These environments often display the complete spectrum of peridotite
436 compositions, including lherzolites, harzburgites and replacive dunites and because of their
437 compositional variety, mantle rocks from these environments have been the focus of detailed
438 petrological and geochemical studies of lithophile element behavior, HSE abundances and Re-Os
439 isotopic studies. Many of these tectonites have been exhumed in the course of the development of
440 small oceanic and sedimentary basins in the Alpine-Mediterranean realm (Piccardo and Guarnieri
441 2010).

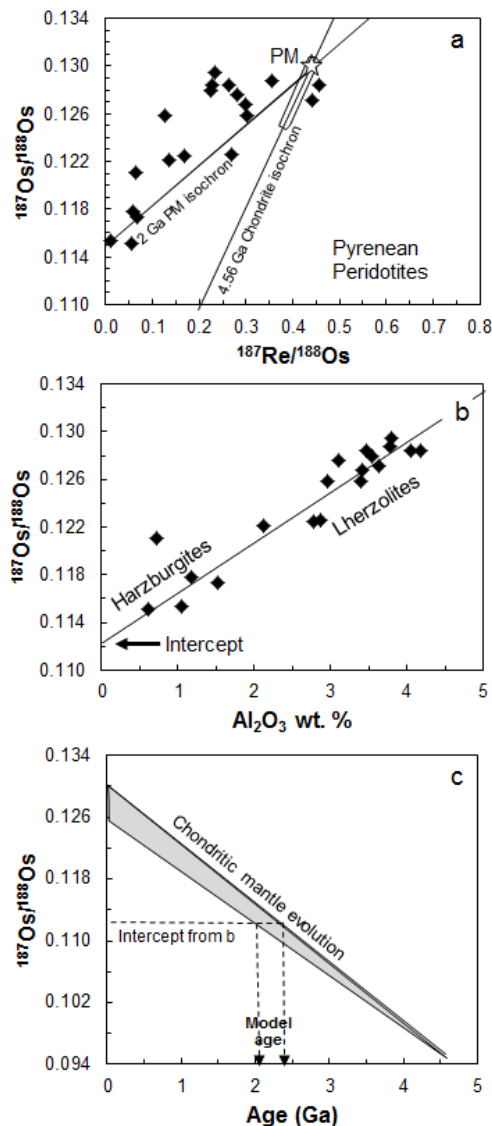
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443 **Pyrenees.** In the Pyrenees, numerous small, serpentinized peridotite bodies (typically km²
444 size or less) occur as lenses in high-grade gneiss-granulite-sediment rock associations (e.g., at
445 Lherz, Bestiac, Turon de Tecouere). The mantle and lower crustal rocks were presumably
446 exhumed during extension and subsequent compressional movements between Iberia and the
447 European plate in the Mesozoic to Cenozoic (Vielzeuf and Kornprobst 1984). The mantle rocks
448 are predominantly variably serpentinized spinel lherzolites and harzburgites, with occasional
449 spinel and garnet facies pyroxenitic banding (Bodinier et al. 1987, 1988). Melt infiltration
450 affected incompatible trace elements, such as the light REE, in the mantle rocks to a variable
451 extent (Vasseur et al. 1991). The small ultramafic body near the village of Lherz (Lers), the type
452 locality of lherzolite, has been studied in detail and has yielded textural and geochemical
453 evidence that the lherzolites in that body formed by reactive infiltration of incompatible element-
454 depleted melt into older harzburgites (Le Roux et al. 2007). The peridotites at Lherz are a key
455 example that shows how reactive transport of basic silicate melt may re-enrich depleted mantle
456 rocks in incompatible major elements via precipitation of pyroxenes, a process called
457 refertilization. The pyroxenites may represent leftover cumulates and reaction products from
458 these processes. However, mechanical mixing of pyroxenite and harzburgite has also been
459 proposed as a mechanism capable of producing the refertilisation at Lherz which is commonly
460 attributed to melt reaction (Riches and Rogers, 2011).

461

462 Early Re-Os work on peridotites from Pyrenean ultramafic bodies by Reisberg and
463 Lorand (1995) yielded positive correlations between measured ¹⁸⁷Os/¹⁸⁸Os and ¹⁸⁷Re/¹⁸⁸Os (the
464 Re-Os isochron diagram, Fig. 4a), as well as Al₂O₃ contents (Fig. 4b), respectively. Al₂O₃
465 contents have been used as a preferred melt extraction index (see also Fig. 2) and proxy for the
466 Re/Os ratio, because Re abundances are typically believed to have been partially affected by
467 serpentinization, whereas Al is largely considered immobile during alteration processes (Reisberg
468 and Lorand 1995, Shirey and Walker 1998). The positive correlation of ¹⁸⁷Os/¹⁸⁸Os with Al₂O₃
469 was interpreted to reflect past melt extraction, assuming the mantle rocks were cogenetic and

470 their different Al_2O_3 , Re contents and Re/Os ratios reflected different degrees of partial melting.
 471 The intersection of the 'initial' $^{187}\text{Os}/^{188}\text{Os}$ and a chondritic evolution curve then gives a model
 472 age of 2.3 Ga (Reisberg and Lorand 1995). Figures 4b and 4c show this model age concept using
 473 the range of measured $^{187}\text{Os}/^{188}\text{Os}$ in bulk rocks of chondrites (Walker et al. 2002; Fischer-Gödde
 474 et al. 2010). The same approach was applied by Reisberg and Lorand (1995) to peridotites from
 475 the Ronda peridotite massif (see below). The ancient Re-Os model ages of these peridotite
 476 massifs, their coincidence with Sm-Nd model ages of overlying crustal rocks and their
 477 geodynamic position have been used to argue that these bodies represent fragments of exhumed
 478 subcontinental lithospheric mantle that have undergone Proterozoic melt extraction (Reisberg and
 479 Lorand 1995; Burnham et al. 1998). It is plausible to infer that the melt extraction processes may
 480 have occurred in an ocean ridge environment and consequently, the model ages would record the
 481 ancient formation of lithospheric mantle from asthenosphere.
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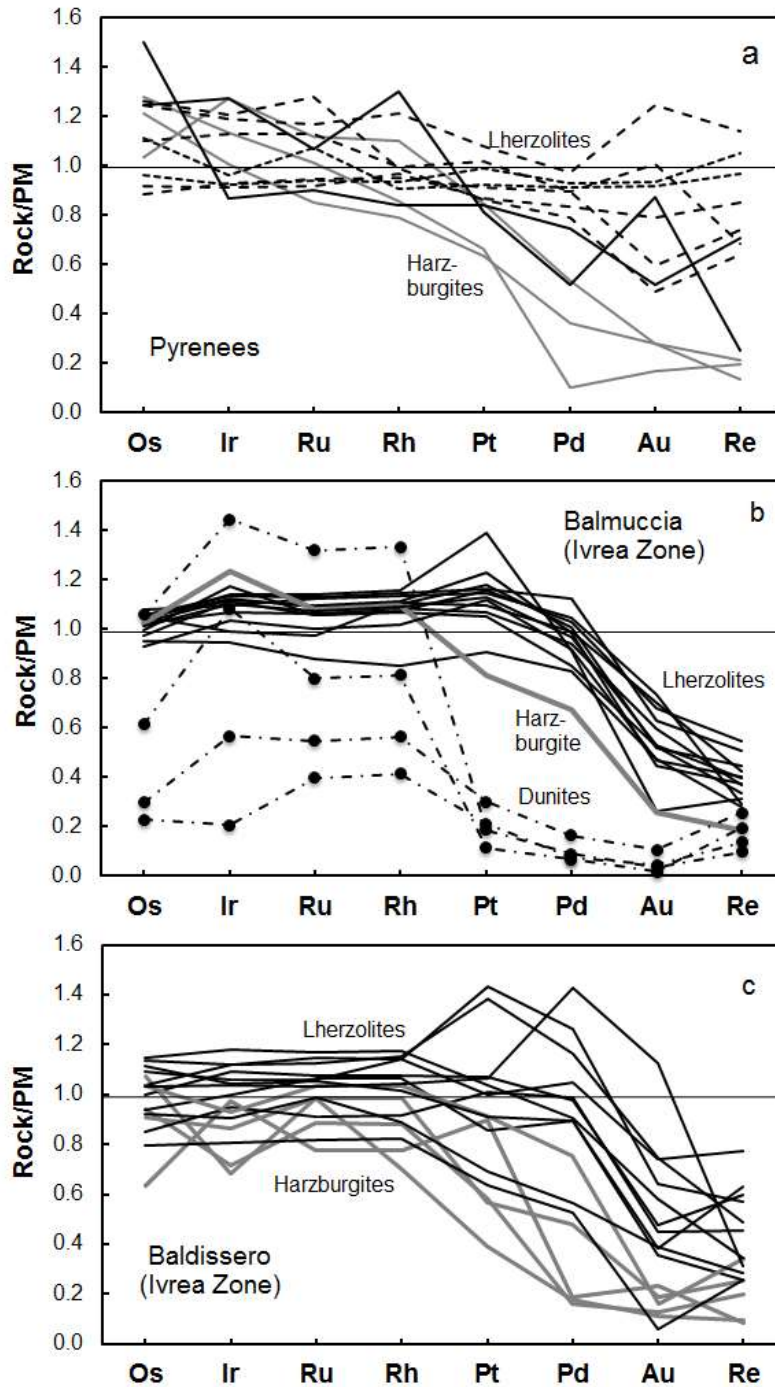


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 484 **Figure 4.** a) Re-Os isochron diagram showing measured $^{187}\text{Os}/^{188}\text{Os}$ and $^{187}\text{Re}/^{188}\text{Os}$ of peridotites from
 485 Pyrenean ultramafic bodies. Data sources: Reisberg and Lorand (1995); Burnham et al. (1998); Becker et
 486 al. (2006). Also shown is a 4.56 Ga chondritic reference isochron (Shirey and Walker 1998) and the
 487 primitive mantle model of Meisel et al. (2001) with an associated 2 Ga isochron. The field outlined on the
 488 isochron represents the spectrum (2 s.d.) of $^{187}\text{Os}/^{188}\text{Os}$ of bulk rocks of chondrites (Walker et al. 2002a;
 489 Brandon et al. 2005a, 2005b; Fischer-Gödde et al. 2010). b) $^{187}\text{Os}/^{188}\text{Os}$ - Al_2O_3 diagram showing data from
 490 Pyrenean peridotites (from Reisberg and Lorand 1995). c) $^{187}\text{Os}/^{188}\text{Os}$ evolution diagram of the range of

491 chondritic compositions from a). Intersection of the $^{187}\text{Os}/^{188}\text{Os}$ of the correlation in b) at $\text{Al}_2\text{O}_3 = 0$ yields
492 an initial $^{187}\text{Os}/^{188}\text{Os}_i$ that intersects the evolution curves in c) to yield Re-Os model ages of the inferred
493 melt extraction that may have resulted in the development of the correlation in a) and b). The assumption
494 behind this model age is that partial melting of the peridotites occurred approximately at the same time. If
495 the assumption is incorrect, peridotites with the lowest $^{187}\text{Os}/^{188}\text{Os}$ may still yield a model age of melt
496 extraction.

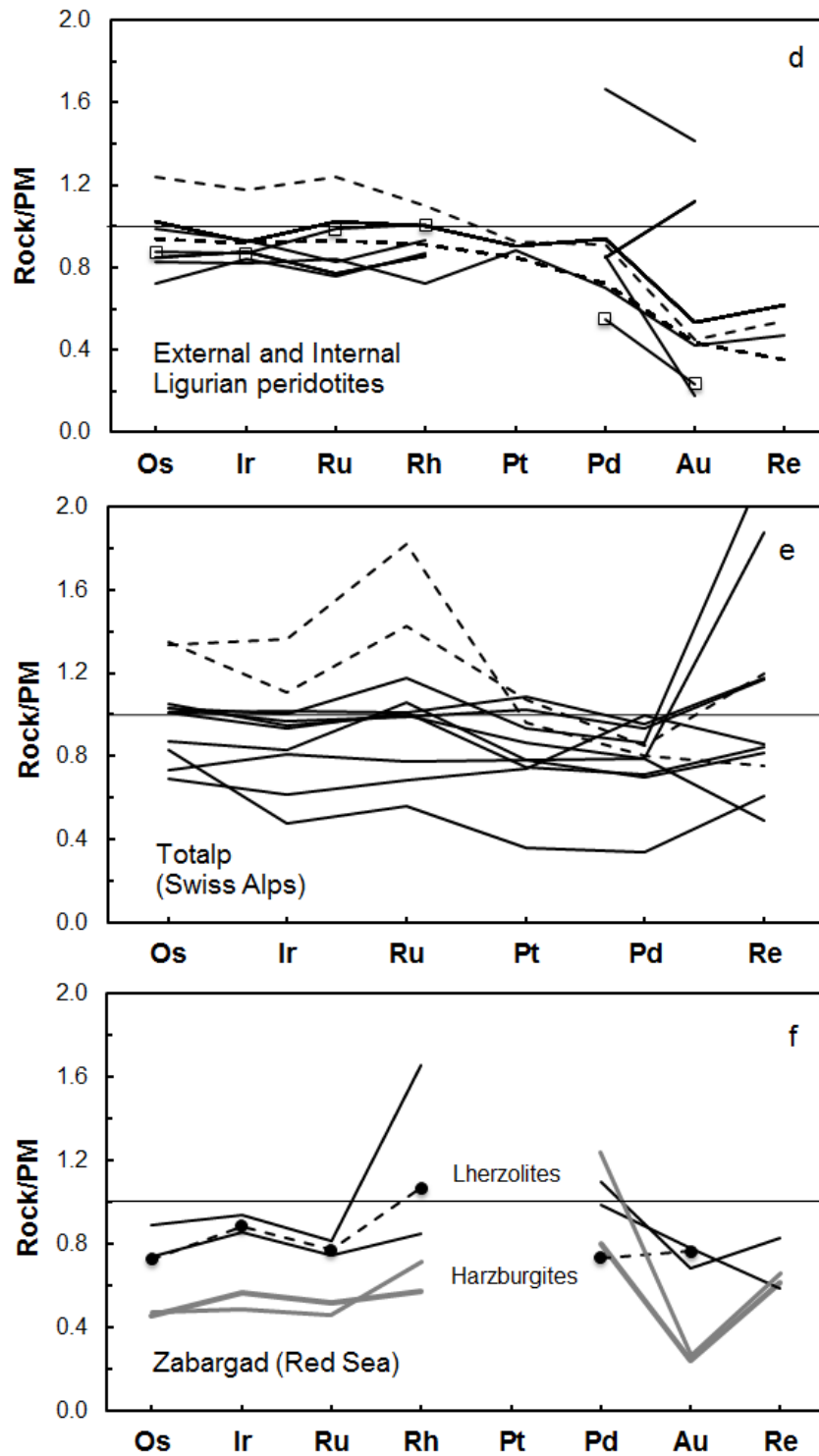
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498 Subsequently published HSE concentration data for the same and additional samples
499 show some features that are not only characteristic of peridotites from the Pyrenees, but also of
500 mantle tectonites from many other locales. Here we will outline the differences between
501 lherzolites and harzburgites, because these different lithologies have been well studied for their
502 bulk rock compositions, as well as their sulfide and other accessory phase mineralogy and
503 mineral compositions. The lherzolites (Fig. 5a) display limited abundance variations for Os, Ir,
504 Ru and Rh, and variable abundances of Pt, Pd, Au, Re and the chalcogen elements S, Se and Te
505 (Pattou et al. 1996; Lorand et al. 1999, 2008, 2010, 2013; Becker et al. 2006; Luguét et al. 2007;
506 Fischer-Gödde et al. 2011; König et al. 2012, 2014; Wang and Becker 2013).



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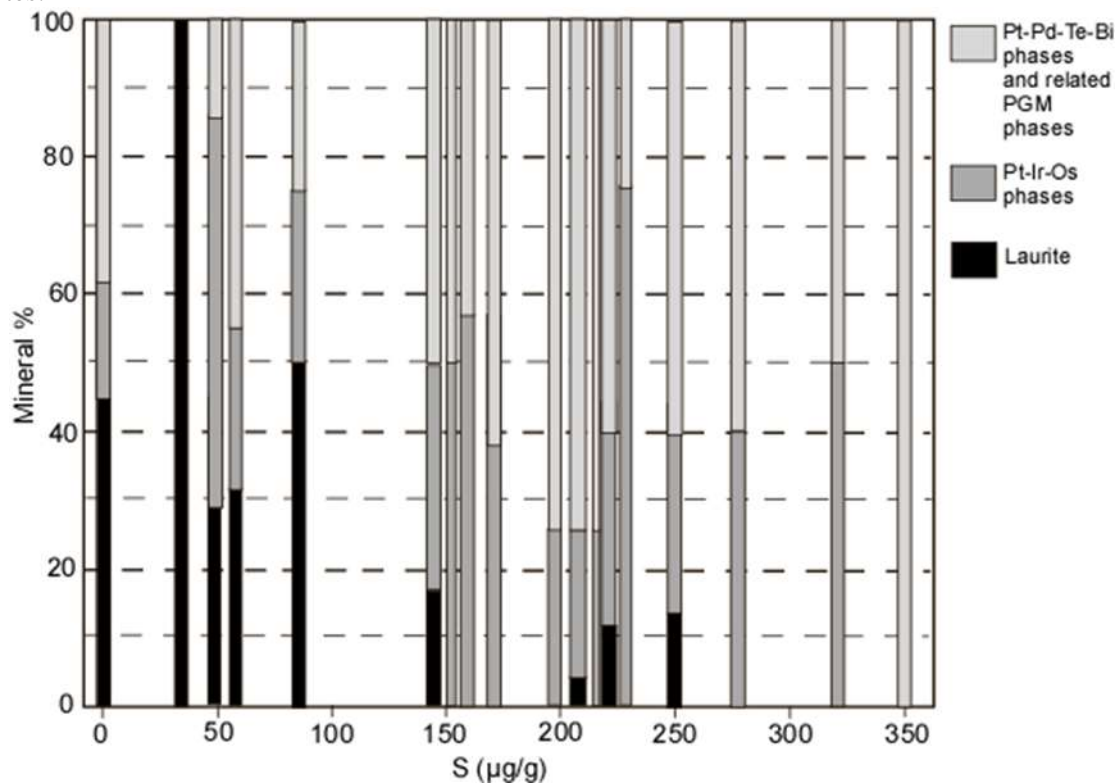
508 **Figure 5.** Primitive mantle normalized concentration diagrams of the HSE in representative peridotites
 509 from continental extensional and transitional oceanic environments. Note the linear concentration scale. a)
 510 Pyrenees (S France) - data from Fischer-Gödde et al. (2011). b) Balmuccia (N Italy) - Dash-dotted lines are
 511 tabular dunites, gray line harzburgite. c) Baldissero (N Italy) - Gray lines are harzburgites. Balmuccia and
 512 Baldissero samples from Wang et al. (2013). The primitive mantle values in this and subsequent diagrams
 513 are those given in Meisel et al. (2001); Becker et al. (2006) and Fischer-Gödde et al. (2011).



514

515 **Figure 5.** Continued. Note the different concentration scale compared to a-c. d) External and Internal
 516 Ligurian peridotites (Luguet et al. 2004; Fischer-Gödde et al. 2011). e) Totalp (Swiss Alps) - No Au or Rh
 517 data are available for these samples (van Acken et al. 2010a) and thus these elements were omitted from
 518 the diagram. f) Zabargad Island (Red Sea) - Schmidt et al. (2000). Lherzolites are either solid or dashed
 519 black lines in order to distinguish different patterns. Gray lines: harzburgites.

521 The highest abundances of Pt, Pd and Re in the Pyrenean lherzolites occur in samples
 522 that yield $^{187}\text{Os}/^{188}\text{Os}$ and major element compositions similar to estimates of the composition of
 523 the primitive mantle (Meisel et al. 2001; Becker et al. 2006). Ratios of the HSE in these samples
 524 suggest broadly chondritic proportions of the HSE, with the exception of Ru and Pd, which are
 525 suprachondritic compared to other HSE. In contrast, harzburgites (Fig. 5a) from the Lherz body
 526 are commonly strongly depleted in Rh, Pt, Pd, Re and chalcogens, whereas abundances of the Ir
 527 group PGE (IPGE; Os, Ir and Ru) were retained at similar to slightly higher levels than in
 528 lherzolites.



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 530

531 **Figure 6.** Proportions of different types of platinum group minerals in harzburgites and lherzolites
 532 from Lherz. The S content may be used as an indicator of the fertility of the rocks (modified from Lorand
 533 et al. 2010),

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535 Study of the accessory phase mineralogy of peridotites from the Pyrenees has indicated
 536 the presence of variable proportions of different sulfide types (pentlandite, pyrrhotite,
 537 chalcopyrite, pyrite), alloy phases (Os-rich, Pt-rich, Au-rich) and other types of platinum-group
 538 metal phases such as Pt-bearing tellurides (Fig. 6) (Luguet et al. 2007; Lorand et al. 2008, 2010;
 539 Lorand and Luguet, 2015). The majority of these phases are likely low temperature exsolution
 540 products that formed during cooling of once homogeneous high-temperature phases such as
 541 sulfide liquids and monosulfide solid solution. The exsolution origin of such phases is reflected
 542 in strong chemical fractionations of some HSE (notably Pt, but sometimes also Pd and Au) and
 543 related elements (e.g., Bi, Te, Se and S) that are only observed on the grain scale, but not in
 544 corresponding bulk rocks. However, some alloy phases, for instance Pt-Ir- and IPGE-rich alloys
 545 may have been inherited from previous episodes of high degrees of melting (Lorand et al. 2010).
 546 The significance of these observations are discussed further below and by Lorand and Luguet
 547 (2015, this volume), and Luguet and Reisberg (2015, this volume).

548

549 ***Balmuccia, Baldissero and Lanzo peridotite bodies.*** In northern Italy, several peridotite
550 bodies occur that also represent fragments of continental lithospheric mantle in an extensional
551 continental margin setting. The peridotite bodies at Balmuccia, Baldissero and Lanzo were all
552 derived from the southern European passive continental margin that had developed following the
553 Variscan orogeny. Towards the end of the Variscan orogeny during the lower Carboniferous and
554 upper Permian, the lower crust and presumably also existing continental lithospheric mantle were
555 flooded with MORB like magma from the asthenosphere (Quick et al., 2009; Snoko et al., 1999;
556 Voshage et al., 1990). The peridotite bodies of Balmuccia and Baldissero, mostly spinel
557 lherzolites with subordinate harzburgites, discordant dunites and pyroxenites, show different
558 distributions of their T_{RD} (Fig. 10 in Wang et al. 2013). In Balmuccia the model ages of the
559 lherzolites show a single distribution peak of Paleozoic model ages, with a harzburgite yielding
560 the only Proterozoic model age (Note: samples with $^{187}\text{Os}/^{188}\text{Os} < 0.1254$ yield Precambrian
561 $T_{RD}[\text{PM}]$ model ages, see Fig. 7). At Baldissero, a bimodal distribution of T_{RD} occurs with a
562 Paleozoic and a Proterozoic peak (Fig. 10 in Wang et al. 2013). Lithophile element, Re-Os, Sm-
563 Nd isotopic and HSE abundance data and textural relations can be interpreted such that depleted
564 Proterozoic mantle (the harzburgites) were variably refertilized by MORB-like magma during the
565 Paleozoic (Mazzucchelli et al. 2009, Mukasa and Shervais 1999, Obermiller 1994, Rivalenti et al.
566 1995, Wang et al. 2013). The greater compositional homogeneity of peridotites from Balmuccia
567 compared to those from Baldissero (Fig. 5b, c) suggests that the former body was fluxed and re-
568 equilibrated with melt more efficiently than the latter. IPGE concentrations in harzburgites in
569 both bodies are lower than in lherzolites, which is opposite to observations from some other
570 suites of peridotites (Pearson et al. 2004; Becker et al. 2006). Re-Os data suggests that some of
571 the Cr-diopside-rich websterites at Balmuccia may have formed during these or earlier events of
572 reactive melt infiltration. However, most Al-augite-rich clinopyroxenites yielded Jurassic model
573 ages (Wang and Becker 2015c). Spinel and plagioclase bearing lherzolites from the Lanzo
574 peridotite massif are similar to lherzolites from Baldissero in their HSE patterns (not shown in
575 Fig. 5) and in their distribution of $^{187}\text{Os}/^{188}\text{Os}$ data (Fig. 7a, Becker et al. 2006, Fischer-Gödde et
576 al. 2011).

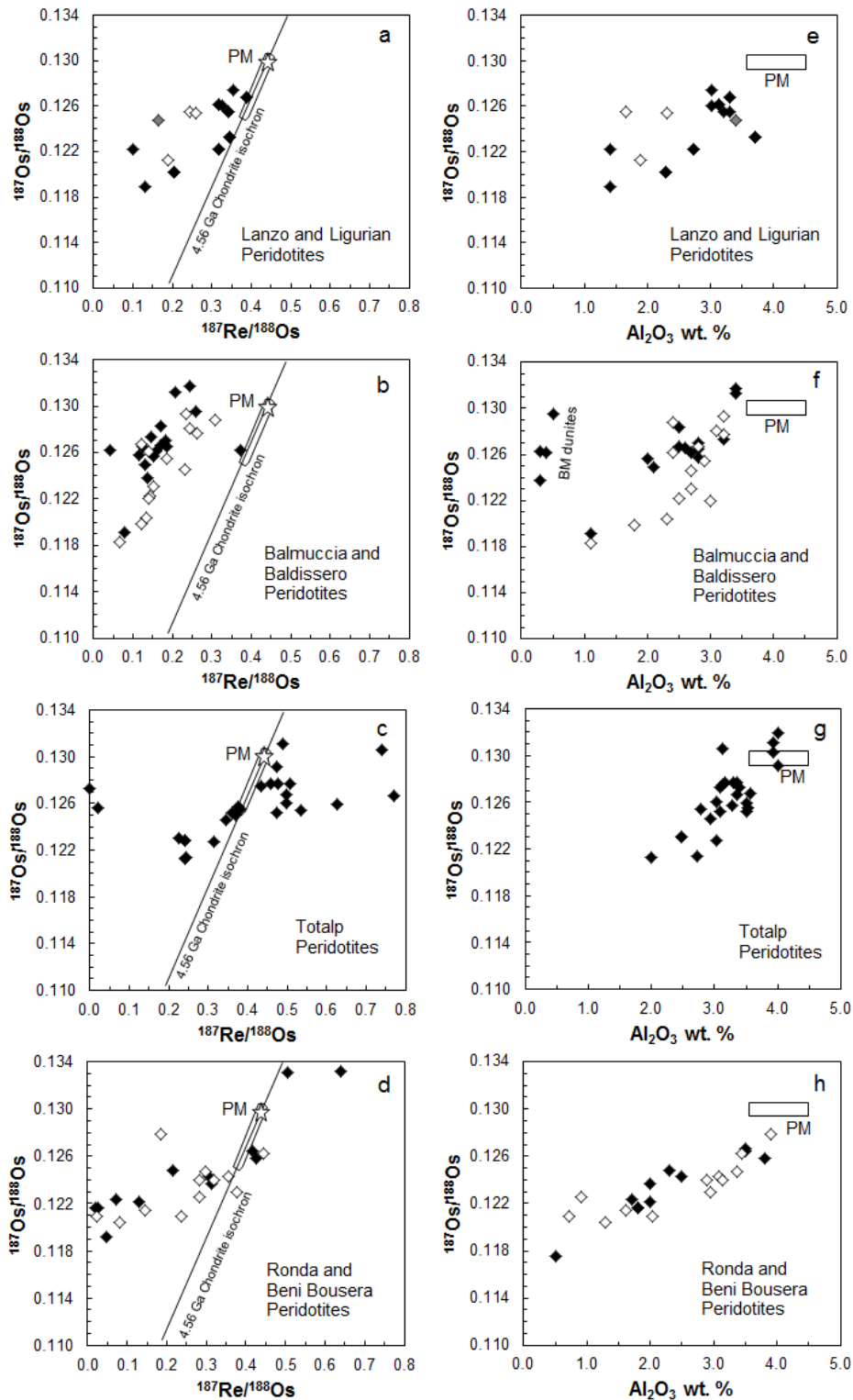
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578 ***External and Internal Ligurian peridotites.*** The External Ligurian peridotites are now
579 recognized to represent mantle rocks of the subcontinental lithospheric mantle of the south
580 European realm (but more distal than Lanzo, Balmuccia and Baldissero), presumably exhumed
581 during the early- to mid-Mesozoic (Rampone et al. 1995; Piccardo and Guarnieri 2010). In
582 contrast, the Internal Ligurian peridotites have been interpreted to derive from depleted mantle of
583 ultraslow spreading ocean floor of the Jurassic Tethys Ocean (Rampone et al. 1996; 1998;
584 Piccardo and Guarnieri 2010). In both cases, plagioclase-spinel lherzolites are the predominant
585 rock type (with subordinate pyroxenites).

586
587 Detailed petrological and geochemical work in these and other studies has shown that the
588 Ligurian peridotites have been variably affected by melt infiltration and refertilization (Rampone
589 et al. 2004). In spite of the somewhat different tectonic setting, the Re-Os and HSE composition
590 of External and Internal Ligurian peridotites is similar to other lherzolites (Figs. 5d and 7a; Snow
591 et al. 2000; Luguët et al. 2004; Fischer-Gödde et al. 2011). Mantle lherzolites and pyroxenites
592 with evidence for melt infiltration and chemical characteristics similar to lherzolite massifs from
593 N Italy have been described from the suture zone in the Alps (e.g., Totalp, Swiss Alps; van
594 Acken et al., 2008; 2010a; 2010b). The Totalp lherzolite body is notable for its Re-rich
595 composition and slightly suprachondritic Re/Os of the lherzolites (Figs. 5e, 7c), which is different
596 from most other peridotite tectonites. The Re-rich composition of the lherzolites and associated
597 pyroxenites can be related to infiltration of melt with MORB-like isotopic compositions,
598 presumably during the Mesozoic or late Paleozoic.

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Zabargad peridotite. The peridotite body of Zabargad Island in the Red Sea represents a young example of subcontinental lithospheric mantle, exhumed during post-Miocene extension of the Red Sea (Bonatti et al. 1986, Piccardo et al. 1993).



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Figure 7. a-d) Re-Os isochron diagrams showing measured compositions of peridotites from different mantle tectonites. Peridotite data from Reisberg and Lorand (1995), Pearson et al. (2004), Becker et al. (2006), van Acken et al. (2010a), Fischer-Gödde et al. (2011), Wang et al. (2013). Further details see Fig. 4. e-h) $^{187}\text{Os}/^{188}\text{Os}$ - Al_2O_3 diagrams. Estimates for the Al_2O_3 content of PM are given as a range from 3.5 to 4.5 wt. %, rather than a single value (McDonough and Sun 1995; Palme and O'Neill 2014 and references

609 therein). Solid diamonds: Lanzo (a, e), Balmuccia (b, f), Totalp (c, g), Ronda (d, h). Open diamonds:
610 Internal Ligurides (a, e), Baldissero (b, f), Beni Bousera (d, h). Note that depleted peridotites tend to have
611 low $^{187}\text{Os}/^{188}\text{Os}$, however, this is not the case for dunites from Balmuccia (BM).
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614 Spinel-bearing lherzolites, amphibole harzburgites and dunites display evidence for
615 metasomatism by fluids or hydrous melts which led to the formation of amphibole harzburgites
616 (Piccardo et al. 1993). The HSE patterns and S abundances of the lherzolites are similar to
617 comparable rocks elsewhere. However, Cu is notably depleted in these lherzolites (around 10
618 $\mu\text{g/g}$). Amphibole-bearing dunite and harzburgite have higher than expected abundances of Pd,
619 Au, Re and S (Fig. 5f; Schmidt et al. 2000). An orthopyroxenite and a plagioclase wehrlite
620 display high PGE and Au abundances, but low Re, S and Cu abundances (Schmidt et al. 2000).
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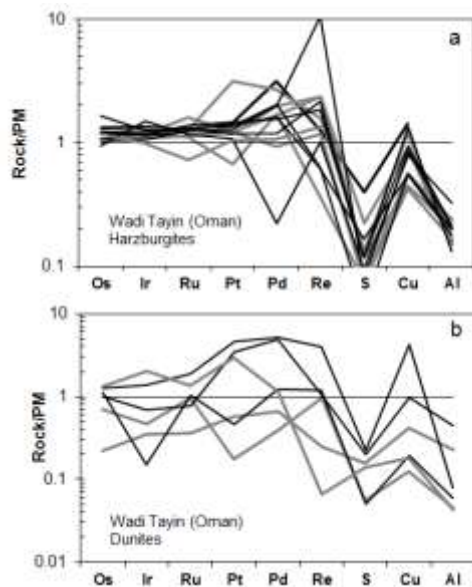
622 **Horoman peridotite.** The Horoman peridotite body in Japan comprises outcrops of
623 layered dunite, harzburgite and lherzolite that have been interpreted to be the result of variable
624 degrees of melt-peridotite reaction that occurred during percolative melt transport in the mantle.
625 Dunites, harzburgites and spinel- and plagioclase-bearing lherzolites at Horoman are believed to
626 have undergone variable degrees of pyroxene dissolution into percolating olivine-saturated
627 magma (Takahashi 1992; Takazawa et al. 1992, 1996, 1999). Despite the occurrence of highly
628 unradiogenic Pb in the Horoman peridotite (Malaviarachchi et al. 2008), abundances of the HSE
629 and $^{187}\text{Os}/^{188}\text{Os}$ data in lherzolites and harzburgites (Rehkämper et al. 1997; Saal et al. 2001) are
630 similar to data from peridotites elsewhere. The correlation of Re abundances with MgO in the
631 peridotites may be the result of refertilization processes (Saal et al. 2001).
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633 **HSE in ophiolites that formed at fast spreading ridges with little or no influence from** 634 **subduction processes** 635

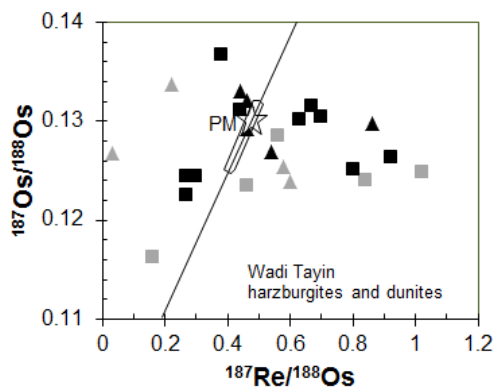
636 **Oman ophiolite, Wadi Tayin Section.** The crustal and mantle section of Wadi Tayin in
637 the SE part of the Samail ophiolite (Oman) represents one of the best exposed examples of fast-
638 spreading oceanic crust and upper mantle on Earth (Pallister and Hopson, 1981; Hanghøj et al.,
639 2010). Geochemical studies of the crustal rocks in the section indicate that the crust mostly
640 comprises normal mid-ocean ridge-type basalts and gabbros (Koga et al. 2001, Pallister and
641 Knight 1981). Part of the ophiolite likely formed at an ocean spreading center about 90-95 Ma
642 ago, but must have been incorporated into an active subduction-collision zone that led to changes
643 in magma compositions in the NW part of the ophiolite (Searle and Cox 1999, Tilton et al. 1981).
644

645 A study of PGE and Re abundances and $^{187}\text{Os}/^{188}\text{Os}$ in the lower crustal gabbros
646 indicated low Re concentrations and systematically higher PGE concentrations compared to
647 MORB (Peucker-Ehrenbrink et al. 2012). The Os isotopic compositions of some gabbros may
648 have been affected by circulation of seawater. The HSE abundances and $^{187}\text{Os}/^{188}\text{Os}$ of parts of
649 the exposed mantle section were studied across an 11 km transect from the exposed Moho into
650 high- and then low-temperature peridotites underneath (Hanghøj et al. 2010). Platinum group
651 element concentration data on harzburgites of similar composition have also been published by
652 Lorand et al. (2009). The high-temperature peridotites likely represent textures and compositions
653 of the mantle inherited from the ocean ridge stage, whereas the low temperature peridotites
654 underneath may represent mantle modified by deformation, re-equilibration and fluid transport
655 during obduction of the ophiolite. The mantle rocks at Wadi Tayin comprise serpentinized
656 harzburgites and replacive dunites that are strongly enriched in fluid-mobile incompatible
657 lithophile elements (e. g., Rb, Pb), which may reflect late alteration or, alternatively, retention of
658 small quantities of melt during peridotite-melt interaction (Hanghøj et al. 2010). The strong
659 fractionation of the REE in most of these samples is significantly greater than in abyssal
660 peridotites and suggests that these rocks can be regarded as highly depleted melting residues in
661 which the LREE were strongly depleted by fractional melting (Hanghøj et al. 2010). The dunites

662 are usually interpreted as forming by magmatic dissolution-precipitation processes that dissolve
 663 pyroxenes and increase the modal amount of olivine (Kelemen et al. 1995). Harzburgites
 664 typically show consistent HSE abundances with IPGE greater than most abyssal peridotites,
 665 slight depletion in Pt and enrichment in Pd. Dunites, however, show far greater variability,
 666 including their Os/Ir ratio, and range from moderately depleted abundances of Re, Pd and Pt to
 667 variable enrichments of Re, Pd and Pt, sometimes a factor of 2-3 times higher than values
 668 commonly observed in lherzolites (Fig. 8). The enrichments of Re, Pd and Pt in the harzburgites
 669 and dunites may have resulted from shallow precipitation of magmatic sulfide from S-saturated
 670 magmas, although S concentrations in the mantle rocks are low (typically a few tens of ug/g,
 671 Hanghøj et al. 2010) compared to Pd, Re and Cu abundances (Fig. 8). The initial $^{187}\text{Os}/^{188}\text{Os}_i$ (at
 672 90 Ma) of the harzburgites and dunites are remarkable in that they display a large range from as
 673 low as 0.113 to suprachondritic values of 0.15 in dunites (Fig. 9). As in other mantle tectonites,
 674 most samples are in the chondritic to subchondritic range, however, some samples with
 675 suprachondritic $^{187}\text{Os}/^{188}\text{Os}$ either require interaction with magma with radiogenic $^{187}\text{Os}/^{188}\text{Os}$, or
 676 have lost a substantial amount of their original inventory of Re.
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 680 **Figure 8.** Primitive mantle-normalized logarithmic concentration diagrams of the HSE in a) harzburgites
 681 and b) dunites from Wadi Tayin, Samail ophiolite, Oman (Hanghøj et al. 2010). Black patterns are rocks
 682 with high-temperature fabrics, gray patterns are ‘low-temperature’ rocks from the lower part of the mantle
 683 section. Sulfur, Cu and Al are included for comparison with Re and Pd (see text).

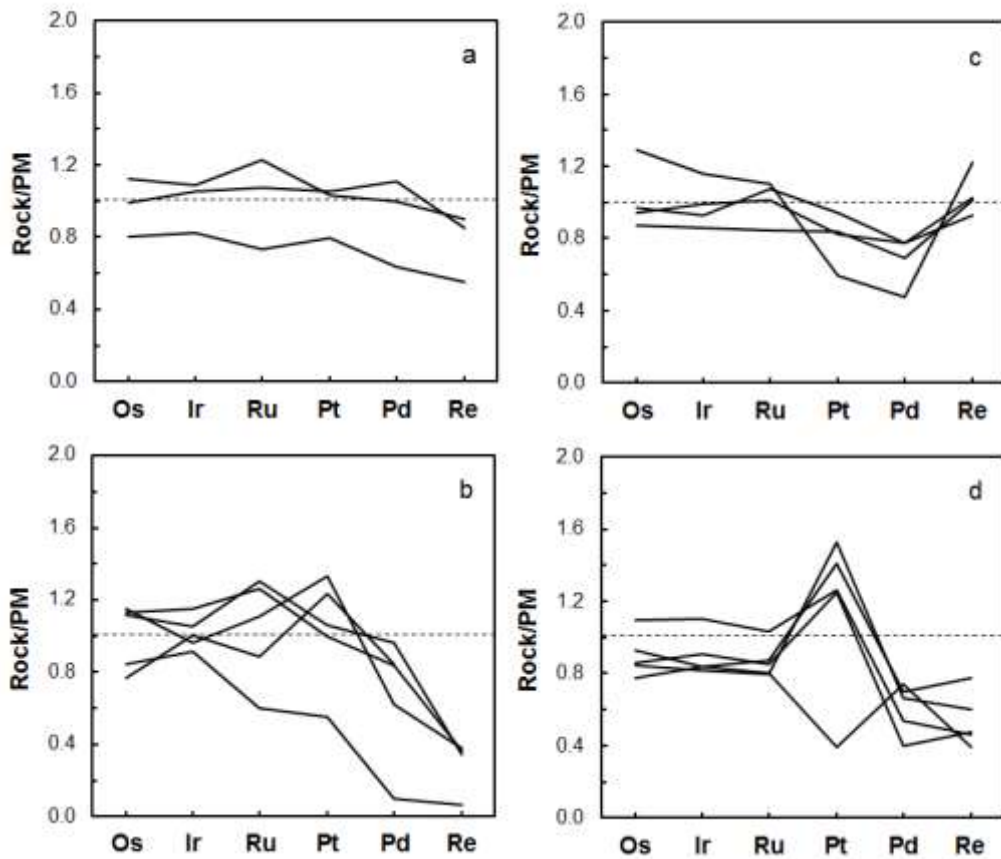


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 685 **Figure 9.** Re-Os isochron diagram for harzburgites (squares) and replacive dunites (triangles) from the
 686 Wadi Tayin section in the Samail ophiolite (Oman). The ophiolite formed part of a spreading center 95 Ma
 687 ago. Black symbols are rocks with high-temperature fabrics, gray symbols are ‘low-temperature’ rocks

688 from the lower part of the mantle section. Dunites with $^{187}\text{Os}/^{188}\text{Os}$ of up to 0.15 are not shown. Data from
689 Hanghøj et al. (2010). Chondrite field, 4.56 Ga reference isochron and PM composition as in Fig. 4.
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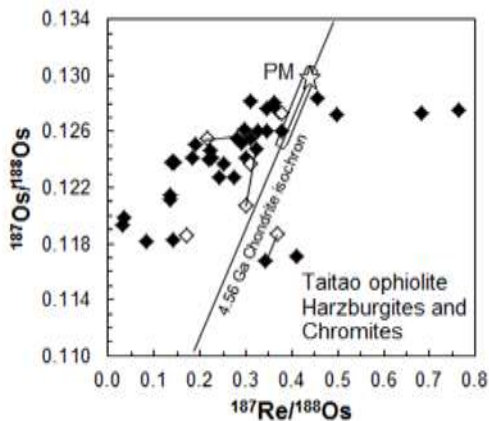
691 ***Taitao ophiolite (Chile).*** The Taitao ophiolite on the Taitao Peninsula in S. Chile is
692 believed to represent part of the oceanic lithosphere formed about 6 Ma ago on the Chile Ridge,
693 which is presently subducting under South America (Guivel et al. 1999). The ophiolite was
694 obducted during or soon after its magmatic formation and was affected by hydrothermal
695 alteration and a metamorphic overprint related to subduction, obduction and contact
696 metamorphism imposed by young granitoid intrusions. The Taitao ophiolite displays a somewhat
697 dismembered Penrose style sequence of serpentinized harzburgites, gabbros, basic dikes, pillow
698 basalts and sediments (Schulte et al. 2009 and references therein). The chemistry of the basic
699 rocks hints that at least some of these magmas may have been affected by subduction zone
700 processes, similar to basalts from the active Chile Ridge (Klein and Karsten 1995). The
701 serpentinized harzburgites display some variability in their HSE patterns ranging from samples
702 that display variable depletions of Re and Pd, depletion of Pd but not Re, and samples showing
703 positive or negative anomalies of Pt relative to Ru and Pd (Fig. 10; Schulte et al. 2009). Basic
704 rocks tend to have very low abundances of IPGE, with variable positive Pt anomalies and strong
705 enrichment of Re (Schulte et al. 2009). Measured $^{187}\text{Os}/^{188}\text{Os}$ range from 0.117 to 0.128, with
706 many samples scattering around a 1.6 Ga reference line in an isochron diagram (Figs. 11).
707 Because of the relatively large range in $^{187}\text{Os}/^{188}\text{Os}$ and the strongly depleted major element
708 composition of the harzburgites, the slope in the $^{187}\text{Os}/^{188}\text{Os}$ - Al_2O_3 diagram (Fig. 12) is different
709 from other suites of peridotites (Figs. 2 and 7). Schulte et al. (2009), however, interpreted the
710 HSE data of the mantle rocks to reflect a two-stage partial melting history at 1.6 Ga and 6 Ma
711 ago. Textural evidence indicates that some harzburgites may have been affected by melt
712 impregnation processes, which may have led to some of their chemical and isotopic variability.
713 The initial $^{187}\text{Os}/^{188}\text{Os}_i$ (6 Ma) of the basic rocks ranges from chondritic to suprachondritic (γOs_i
714 = -1 to +342). The suprachondritic composition may either reflect the presence of a rhenium-
715 enriched component in the mantle source or the influence of seawater/altered crust during the
716 emplacement of the magmas.

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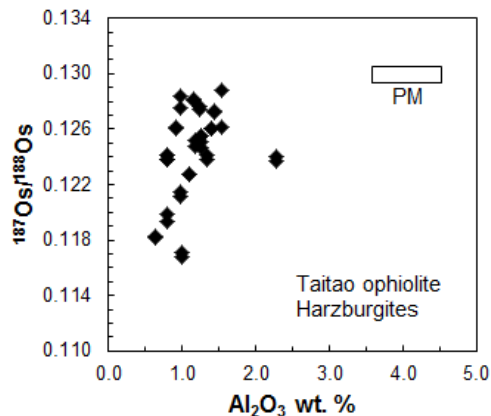
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Figure 10. Primitive mantle-normalized concentration diagrams of the HSE in mantle rocks from the Taitao ophiolite. Four different types of patterns can be distinguished: a) relatively little fractionated peridotites; b) rocks with stronger depletions of Pd and Re (the sample with low abundances represents a pyroxenite); c) samples that display depleted Pd and re-enrichment of Re; d) rocks that display strong positive or negative anomalies of Pt. Data from Schulte et al. (2009).



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Figure 11. Re-Os isochron diagram for peridotites (solid symbols) and chromites (open symbols) in harzburgites from the Taitao ophiolite (6 Ma old). Tie lines connect chromites and corresponding bulk rocks, indicating small-scale Os isotopic disequilibrium in these mantle rocks. Data from Schulte et al. (2009),



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735 **Figure 12.** $^{187}\text{Os}/^{188}\text{Os}$ - Al_2O_3 diagram for whole rock harzburgites from the Taitao ophiolite. Note the large
736 range in $^{187}\text{Os}/^{188}\text{Os}$ in these depleted mantle rocks. Data from Schulte et al. (2009).
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738 **High-temperature orogenic peridotites from convergent plate margin settings**

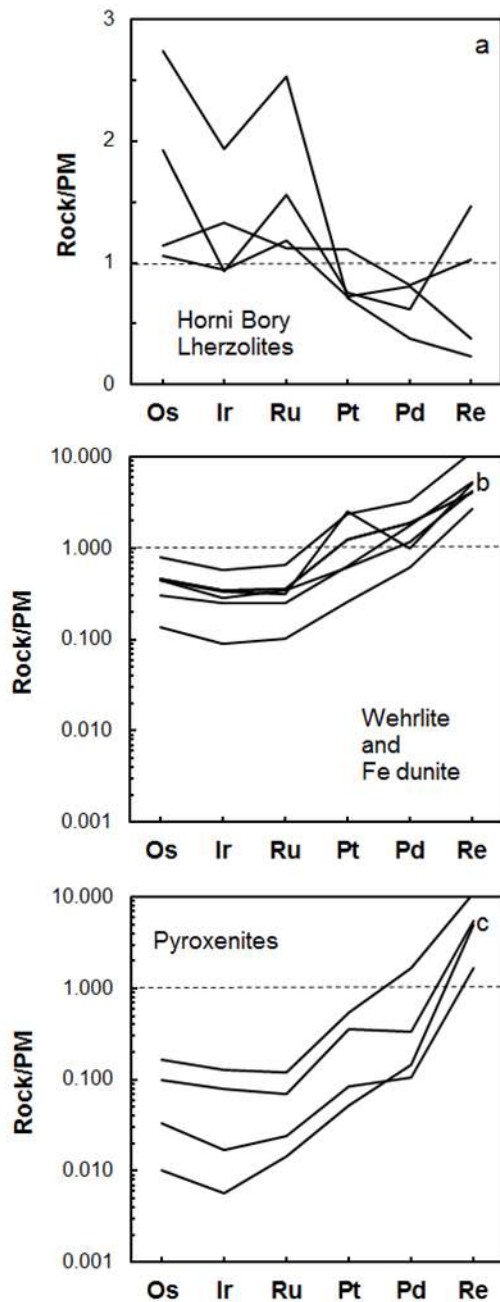
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740 Many ophiolites have originally been emplaced near subduction zones and commonly
741 even their mantle sections were affected by magmas that formed in supra-subduction zone
742 environments (see below). Among high-temperature orogenic peridotites, evidence for the
743 influence on mantle rocks by magmas that may have formed in convergent plate margin settings
744 is not very common and, in fact, is somewhat ambiguous. Here, we discuss examples of mantle
745 tectonites that were emplaced during or in the aftermath of subduction and collision processes. In
746 the case of the Ronda and Beni Bousera ultramafic massifs these bodies represent mantle
747 exhumed during the collapse of the Betic orogenic belt in the western Mediterranean during the
748 Cenozoic (van der Wal and Vissers 1993; Blichert-Toft et al. 1999). In the southern Bohemian
749 massif, similar processes occurred during collapse of the core zone of the Variscan belt during
750 the Carboniferous (Medaris Jr et al. 2005). The principal evidence is mostly derived from
751 geodynamic reconstructions in combination with the lithophile element and isotope geochemistry
752 of peridotites and pyroxenites. Notably garnet bearing pyroxenite layers in these peridotite
753 massifs show strong evidence that they formed from magmas with crustal geochemical and
754 isotopic signatures (e.g., Eu anomalies, enrichments of LREE, graphite with $\delta^{13}\text{C}$ suggestive of
755 organic protoliths, sediment-like Sr-Nd-Pb isotopic compositions; (Pearson et al. 1991a; 1991b;
756 1993; Becker 1996a).
757

758 **Ronda (Southern Spain).** The Ronda peridotite has been a classic study area of mantle
759 processes (Frey et al. 1985, Reisberg and Zindler 1986, Reisberg et al. 1989). It shows a
760 transition from garnet lherzolite to spinel lherzolite and plagioclase-bearing peridotites (Obata
761 1980). Initially the peridotites were regarded as residues of partial melting (Frey et al. 1985);
762 however, later the significance of melt infiltration into continental lithospheric mantle was
763 recognized and the latter process also may have caused partial re-equilibration of the peridotites
764 at shallow pressure-temperature conditions (Bodinier et al. 2008). Re-Os model ages of depleted
765 peridotites yield an average age of melt extraction in these rocks of 1.3 ± 0.4 Ga (Reisberg et al.
766 1991, Reisberg and Lorand, 1995; Becker et al. 2006). The HSE patterns (not shown) and
767 $^{187}\text{Os}/^{188}\text{Os}$ (Fig. 7d, h) of the peridotites are similar to data on fertile to depleted peridotite
768 tectonites exhumed in extensional tectonic settings (Lorand et al. 2000; Becker et al. 2006;
769 Fischer-Gödde et al. 2011). Pyroxenite layers from Ronda have suprachondritic Re/Os and
770 $^{187}\text{Os}/^{188}\text{Os}$, and Pd and Pt are enriched relative to IPGE (Marchesi et al. 2014; Reisberg et al.
771 1991). The depletion of Re in some pyroxenites relative to Pd (Fig. 21c) may reflect multi-stage
772 melting processes (Marchesi et al. 2014).
773

774 ***Beni Bousera (Morocco).*** The Beni Bousera peridotite massif crops out on the
775 Moroccan side of the Alboran Sea and shares a similar history with the Ronda body. Re-Os and
776 HSE concentration data on peridotites are comparable with data from Ronda (Fig. 7d, h, Kumar
777 et al. 1996; Pearson et al. 2004; Pearson and Nowell 2004; Luguët et al. 2008b; Fischer-Gödde et
778 al. 2011). Studies of the Re-Os systematics in pyroxenite layers from Beni Bousera yielded
779 highly variable Re/Os and $^{187}\text{Os}/^{188}\text{Os}$, the latter reflecting radiogenic ingrowth, but also partly
780 incorporation of unradiogenic Os from reaction with the host peridotites (Kumar et al. 1996,
781 Pearson and Nowell 2004). The Re-Os model ages cluster near 1.3 Ga, similar to results from
782 some peridotites, and similar to Lu-Hf ages of some, but not all pyroxenites. The spectrum of Re-
783 Os model ages and Lu-Hf isochron ages is consistent with other evidence that suggests a complex
784 multi-phase history of both the Ronda and the Beni Bousera bodies (Loubet and Allègre 1982,
785 Marchesi et al. 2014). Luguët et al. (2008b) and Marchesi et al. (2014) found variations of Pt/Os
786 and Re/Os in some bulk rocks and sulfides from pyroxenites at Beni Bousera and Ronda,
787 respectively. These rocks were interpreted to represent likely equivalents of the sources of mantle
788 plume-derived picrite and komatiite lavas with elevated ^{186}Os signatures (Brandon and Walker
789 2005 and Discussion section).

790
791 ***Southern Bohemian Massif (Lower Austria, Czech Republic).*** In the Bohemian
792 Massif, kilometer-sized bodies comprised of serpentized high-temperature garnet lherzolites,
793 spinel harzburgites and dunites occur enclosed in high-pressure granulites and amphibolite facies
794 gneisses (e.g., Carswell and Jamtveit 1990; Becker 1996b; 1997; Medaris Jr et al. 2005). As in
795 the peridotite massifs of the Betic cordillera, the garnet pyroxenite layers in the peridotites show
796 chemical and isotopic compositions that suggest that they precipitated from basic magmas that
797 formed in mantle contaminated by recycled sedimentary material (Becker 1996a). Detailed Re-
798 Os work on layered peridotite-pyroxenite rocks indicates that peridotite-derived Os and Cr are
799 mobilized during melt-rock reaction that led to the formation of layered pyroxenite-dunite rocks
800 (Becker et al. 2001; 2004). The pyroxenites in these rocks show suprachondritic initial
801 $^{187}\text{Os}/^{188}\text{Os}$ which may be inherited from subducted materials as indicated by initial Sr-Nd
802 isotopic compositions. The variation of $^{187}\text{Os}/^{188}\text{Os}_i$ in modally layered lithologies indicates Os
803 isotopic disequilibrium on the cm-scale resulting from magmatic infiltration processes. Another,
804 yet different, example of metasomatic overprint that affected HSE abundances in peridotites in
805 the Bohemian Massif are Mg-rich peridotites with relatively high IPGE contents (e.g., up to 10
806 ng/g Os), but not quite as high Pt, Pd and Re abundances (Ackerman et al. 2013). These rocks
807 occur with pyroxenites and Fe-rich cumulate rocks with high Pt, Pd and Re abundances (Fig. 13).

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809 In peridotites from Ronda, Beni Bousera and Lower Austria, measured $^{187}\text{Os}/^{188}\text{Os}$ are
810 subchondritic or chondritic, similar to peridotites from extensional tectonic settings. Pyroxenites
811 show high, but variable Re/Os and suprachondritic γOs_i . However, unlike some data on lithophile
812 elements, these features are not necessarily indicative of the influence of subducted crust or
813 subduction zone fluids. High Re/Os (and γOs_i) seem to be a hallmark of mantle pyroxenites and
814 may be acquired by magmatic fractionation in the crust or during melting and transport of
815 magmas in the mantle (e.g., Pearson and Nowell 2004; van Acken et al. 2010b; Marchesi et al.
816 2014; Wang and Becker 2015c). This topic will be discussed in later sections.



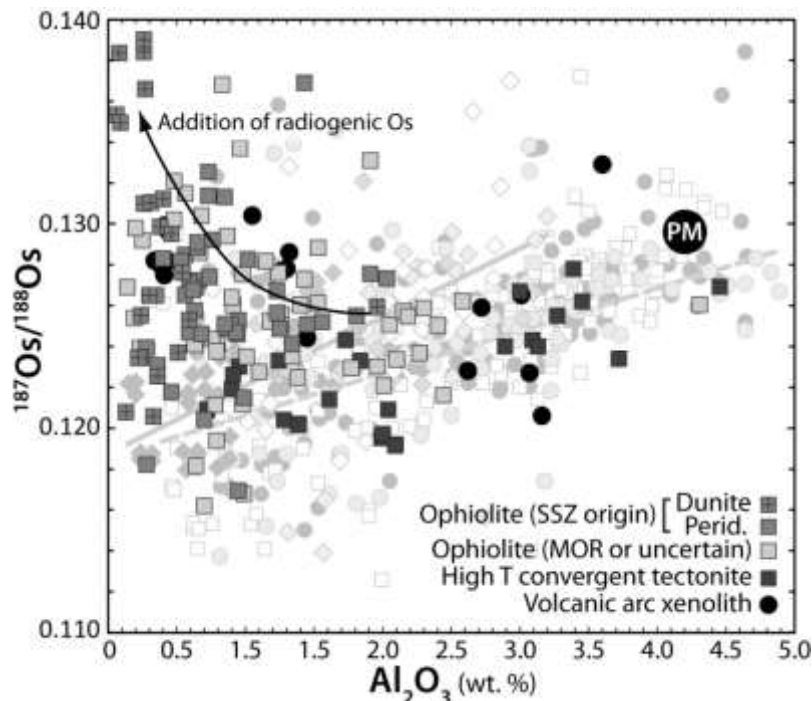
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Fig. 13. Primitive mantle-normalized HSE concentration diagrams of a) mantle peridotites (lherzolite); b) metasomatic Fe-dunite-wehrlite rocks; c) pyroxenites from the Horni Bory peridotite massif (Bohemian Massif, Czech Republic). Note the linear scale in a). Data from Ackerman et al. (2013).

Highly siderophile elements in peridotites and melt-reacted lithologies of ophiolites influenced by convergent plate margin magmatism

In comparison to ophiolites with little subduction influence, convergent plate margin ophiolites typically comprise more depleted harzburgitic mantle sections and thicker ultramafic sequences in the lower crust. This is due to the greater degree of partial melting that usually occurs in the fluid-fluxed supra-subduction zone setting. However, the presence of hydrous

831 melts and fluids also promotes the formation of melt-reacted lithologies such as dunites,
 832 pyroxenites and, in particular, chromitites, in the mantle sections of ophiolites from convergent
 833 plate margins. Such melt-rock reaction, and the lithologies it produces, is diverse and depends
 834 principally on the melt/rock ratio and the degree of saturation of silica and sulfide in the melt.
 835 The variable impact on sulfide is, of course, critical to the behavior of the HSE, and melt-rock
 836 reaction is thus a major process by which HSE are fractionated and heterogeneity is generated.
 837 This fractionation of the HSE, including that which occurs during chromitite formation, likely
 838 plays an important role in defining the HSE characteristics of magmas at Earth's surface,
 839 particularly those of convergent margin ophiolites and in volcanic arc systems (e.g., Dale et al.,
 840 2012b).
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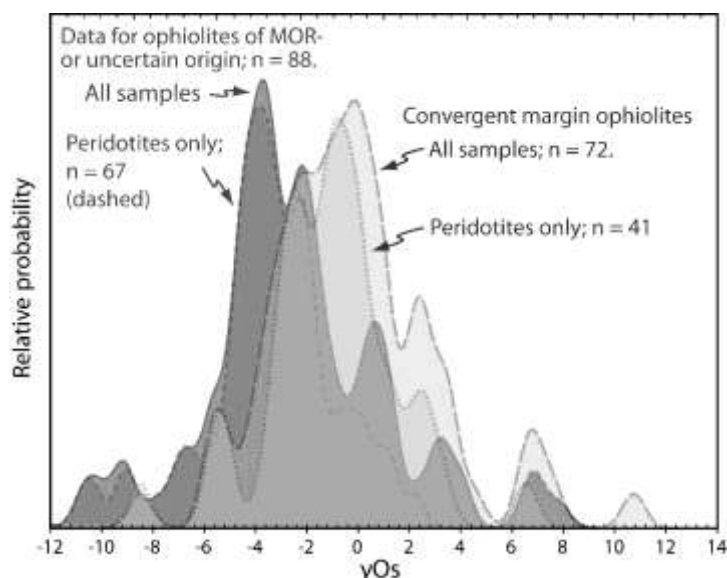


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 844 **Figure 14.** $^{187}\text{Os}/^{188}\text{Os}$ - Al_2O_3 diagram for ophiolite ultramafic rocks (predominantly harzburgites, but also
 845 dunites), high-temperature convergent tectonites and sub-arc mantle xenoliths (see legend for symbols).
 846 Also shown for comparison are abyssal peridotites (diamonds), ocean island basalt mantle xenoliths (light
 847 grey circles), continent-ocean transitional tectonites (white squares) and sub-continental lithospheric
 848 xenoliths (mid-grey circles). There is considerable scatter in all datasets, partly reflecting variable ages of
 849 melt depletion, but also probably recent resetting of $^{187}\text{Os}/^{188}\text{Os}$ by seawater or melt interaction. The most
 850 Al-depleted ophiolitic samples (particularly those from convergent margin settings) and subduction zone-
 851 related ultramafics have more radiogenic ^{187}Os isotope compositions than peridotites from other settings.
 852 This presumably reflects a flux of radiogenic Os, or possibly a time-integrated addition of Re, related to the
 853 flux from the subducting slab, although greater melt-rock ratios in this environment may also play a part.
 854 Crustal contamination during emplacement is also possible. In part, the decoupling of $^{187}\text{Os}/^{188}\text{Os}$ from
 855 Al_2O_3 is due to the formation of dunitic rocks by melt-rock reaction, but many peridotites in convergent
 856 settings also possess more radiogenic Os than expected for a given Al content. Data sources for ophiolites:
 857 Snow et al. (2000); Kepezhinskas and Defant (2001); Büchl et al. (2004); Becker et al. (2006); Schulte et
 858 al. (2009); Hanghøj et al. (2010); Aldanmaz et al. (2012); O'Driscoll et al. (2012). High-T convergent
 859 margin tectonites: Reisberg et al. (1991); Roy-Barman et al. (1996); Becker et al. (2001, 2006); Pearson et
 860 al. (2004); Marchesi et al. (2014). Sub-arc xenoliths: Brandon et al. (1996); McInnes et al. (1999); Widom
 861 et al. (2003). Abyssal peridotites – see Figs. 1 & 2. Ocean island basalt mantle xenoliths – see Fig. 2.
 862 Continent/continent-ocean transition tectonite: Reisberg and Lorand (1995); Meisel et al. (1996); Roy-
 863 Barman et al. (1996); Rehkämper et al. (1999); Snow et al. (2000); Saal et al. (2001); Becker et al. (2006);
 864 Luguët et al. (2007); van Acken et al. (2008); Riches and Rogers (2011); Wang et al. (2013).
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At the same time, there is the potential for sulfide to be exhausted during moderate to high degrees of mantle melting, particularly if sulfur solubility is increased (Jugo, 2009) due to an elevated oxygen fugacity of the sub- or back-arc mantle, relative to typical depleted MORB mantle (e.g., Carmichael, 1991; Kelley and Cottrell, 2009). Given the extremely chalcophile nature of the HSE (e.g., Mungall and Brenan, 2014; with the possible exception of Re; Brenan, 2008), sulfide exhaustion would cause HSE behavior to depart significantly from the typical mid-ocean ridge setting where sulfide is thought to remain in the residue.

Commonly, convergent margin ophiolites contain substantial units of podiform chromitite, enveloped in dunite, which require high degrees of melt depletion and are probably formed through a process of melt-rock reaction, particularly when a hydrous melt is present and the melt/rock ratio is high, or as cumulates from melts formed through high degrees of melting (Ballhaus, 1998; Zhou et al., 1998). Chromitites are known to contain variable but high concentrations of HSE (Prichard and Lord, 1996), particularly the IPGE, indicating their presence in high concentrations in the chromitite-forming melts. Further concentration of HSE occurs primarily because chromitites contain associated platinum-group mineral grains (PGM) which form due to a local oxygen fugacity-induced reduction in solubility of the HSE (Finnigan et al., 2008). This reduction in oxygen fugacity occurs locally around chromite crystals because of their preference for trivalent transition metal cations, particularly Cr^{3+} and Fe^{3+} ions. The IPGE have lower solubilities in silicate melts than PPGE, on the order of tens versus hundreds of ng/g (e.g., O'Neill et al., 1995; Borisov and Walker, 2000; Brenan et al., 2005; Ertel et al., 2006), and hence Os, Ir and Ru are particularly enriched in PGM from chromitites. Although chromitites and platinum-group minerals (PGM) are covered more comprehensively in O'Driscoll & González-Jiménez (2015, this volume), we include a brief Os isotope summary in the Discussion because ophiolitic chromitites are a major source of PGM, and they have a direct bearing on determining both the Os isotopic composition of the convecting mantle and the degree of mantle heterogeneity. Here we focus mainly on HSE behavior in the range of mantle lithologies present in ophiolites, rather than the specifics of PGM mineralogy and its role in HSE behavior (cf. O'Driscoll & González-Jiménez, 2015, this volume).



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Figure 15. Probability density plot of $\gamma\text{Os}_{\text{initial}}$ in ophiolitic ultramafic rocks (peridotites, dunites and chromitites), grouped according to geological setting of formation. $\gamma\text{Os}_i = \left(\frac{^{187}\text{Os}}{^{188}\text{Os}}_{\text{sample}} / \frac{^{187}\text{Os}}{^{188}\text{Os}}_{\text{chondrite initial}} - 1 \right) \times 100$. The absolute values in this plot should be treated with caution as these include a correction for ingrowth based on the measured $^{187}\text{Re}/^{188}\text{Os}$, which, in some cases, may have been perturbed during/since emplacement. In addition, this plot is based on a limited number of different

904 ophiolites, with several ophiolites contributing a disproportionate number of data: Troodos (Cyprus),
905 Samail (Oman), Shetland (UK), Taitao (Chile) and Jormua (Finland) ophiolites account for 129 of the 160
906 analyses. Given the different Os isotope records preserved by PGM grains from different ophiolites
907 (Pearson et al., 2007), much of this difference could merely reflect large-scale mantle heterogeneity.
908 Nonetheless, the overall offset between the two categories is two to four gamma units, which may
909 represent a real difference generated by addition of radiogenic Os in the supra-subduction zone
910 environment. Data sources given in Fig. 14, except Becker et al. (2006) and Dijkstra et al. (2010).

911

912 ***Troodos Ophiolite (Cyprus).*** Two complementary studies of melt percolation in the
913 Troodos ophiolite found fractionated HSE abundances and variable $^{187}\text{Os}/^{188}\text{Os}$ in a range of
914 residual and melt reaction products (Büchl et al., 2002, 2004). A sequence of spinel lherzolites,
915 minor dunites and clinopyroxene-bearing harzburgites was found to have a large range of initial
916 $^{187}\text{Os}/^{188}\text{Os}$ (at 90 Ma) from subchondritic (0.1168) to mildly suprachondritic (0.1361); a second
917 unit, consisting of harzburgites, dunites and chromitites, has an even larger and more radiogenic
918 range of 0.1234 to 0.1546. The subchondritic values can readily be explained by ancient melt
919 depletion of Re (>800 Ma), as for abyssal peridotites and most other mantle rocks. The
920 suprachondritic Os compositions, as with those from the Oman ophiolite described earlier
921 (Hanghøj et al., 2010) and many other ophiolites (see Figs. 14 & 15), require the addition of a
922 radiogenic melt component (unless samples have experienced significant recent Re loss), likely
923 during the formation of the Troodos around 90 Ma ago. The ultimate source of this radiogenic
924 Os is not known, and could relate to seawater contamination prior to concentration in chromitites
925 (because a radiogenic signature is also evident in the most Os-rich chromitite samples) or to
926 crustal contamination during emplacement, but the former at least is difficult to reconcile with
927 the very low Os concentrations in seawater (Levasseur et al., 1998). Another possible
928 mechanism, that would be applicable to both mid-ocean ridge and supra-subduction ophiolites, is
929 the production of radiogenic melts due to preferential sampling of radiogenic interstitial sulfides
930 (Alard et al., 2005; Harvey et al., 2011) or due to the presence of enriched domains in the mantle
931 (cf. pyroxenites; Reisberg et al., 1991; Pearson and Nowell, 2003). However, melting of
932 enriched domains is not consistent with the refractory boninitic melt that typically produces HSE-
933 and Cr-rich chromitites. Given the apparent global distinction in Os isotopes between ophiolites
934 of convergent and mid-ocean ridge origin (Fig. 15), the most plausible explanation for a
935 significant part of the radiogenic signature is a flux from the subducting slab, with Os mobilized
936 in oxidized chlorine-rich fluids (Brandon et al., 1996; Becker et al., 2004). In this scenario,
937 despite the extreme fractionation of Re from Os in oceanic crust, the low Os contents and
938 relatively young age of subducted mafic crust would suggest that a sedimentary input may be
939 required to provide sufficient radiogenic Os to impart that signature on the Os-rich mantle.

940

941 The process(es) of dunite formation also induces significant HSE fractionation.
942 Harzburgites, which could be simple residues of melting or, as Büchl et al. (2002) conclude, the
943 product of melting during melt-percolation at low melt/rock ratios, have largely uniform IPGE
944 patterns and concentrations that only range by roughly a factor of two (Fig. 16). Palladium and
945 Re abundances do, however, vary over approximately an order of magnitude in harzburgites
946 (Büchl et al., 2002). In contrast, a dunite rim and core, the product of high melt/rock ratios,
947 together with a websterite and a boninite all display high and remarkably uniform concentrations
948 of Pt (6.5 – 12.2 ng/g), moderately variable Pd and Re, and two or more orders of magnitude
949 variation in Os content. Qualitatively, it seems that dunites and websterite could be produced by
950 some sort of mixing process between harzburgite and boninitic melt, retaining high Pt but
951 removing/diluting Os; requiring Os to be mobilised. This is supported by modelling of HSE
952 ratios (dominated by mixing of harzburgitic and magmatic sulfides) and REE in clinopyroxene
953 during open system melting (Büchl et al., 2002).

954

955 ***Shetland Ophiolite Complex (UK).*** Harzburgites from Unst, Shetland, have Os isotope
956 compositions ranging from γOs of 2 to -6 (using an O-chondrite reference frame; $^{187}\text{Re}/^{188}\text{Os} =$
957 0.422, $^{187}\text{Os}/^{188}\text{Os} = 0.1283$). Most Os isotope ratios are consistent with an ambient convecting

958 mantle signature (see section on Os isotope heterogeneity in Discussion) but there is evidence of
959 both melt depletion at ~1.2 Ga and also radiogenic Os addition for some samples (O'Driscoll et
960 al., 2012).

961

962 Dunites have a wider range of $^{187}\text{Os}/^{188}\text{Os}$ than harzburgites (-22 to 12), reflecting the
963 effects of melt-rock reaction involved in their formation (O'Driscoll et al., 2012). Chromitites
964 have the narrowest range of $^{187}\text{Os}/^{188}\text{Os}$, from $\gamma\text{Os} +0$ to +3.5. This relative homogeneity is
965 perhaps surprising given the higher melt/rock ratios involved in producing chromitite, but this is
966 set against the extremely high Os concentrations, and low Re abundances, that allow for accurate
967 estimation of the initial Os isotope composition. In part, the range for dunites (and harzburgites)
968 may reflect difficulties in age correcting over 492 Ma (as this is dependent on measured Re and
969 Os concentrations – with the potential for recent disturbance). Overall, however, a radiogenic Os
970 flux is required to explain the supra-chondritic γOs values. As discussed for the Troodos
971 Ophiolite, there are various possible sources of the radiogenic Os, but a flux from the downgoing
972 slab may be the most plausible mechanism.

973

974 Shetland Ophiolite samples display huge variations in HSE concentrations, with some
975 chromitites containing up to ~100 $\mu\text{g/g}$ Pt (Prichard and Lord, 1996; O'Driscoll et al., 2012)
976 while some dunites contain less than 100 pg/g Pt. The most HSE-rich chromitites (from Cliff)
977 have Ir and Ru contents that are roughly two orders of magnitude higher than the range of
978 chromitites analysed from the Qalander, Luobusa and Zambales ophiolites (Fig. 16). Moreover,
979 these chromitites have unusual HSE patterns with PPGE/IPGE ratios greater than unity and Pd
980 concentrations up to 156 $\mu\text{g/g}$ (O'Driscoll et al., 2012), compared with typical IPGE-rich
981 chromitites which have Pd and Pt contents approximately four orders of magnitude lower (Zhou
982 et al., 1996; Zhou et al., 2000; Ismail et al., 2014; Zhou et al., 2014). The range of HSE
983 abundances between chromitites from different localities is, in itself, huge. The two other
984 localities analysed have more typical HSE patterns, albeit in one case also enriched by one to two
985 orders of magnitude. The degree of P-PGE enrichment has been linked to the thickness and
986 sulfide content of the ultramafic dunite sequence and ultimately to the degree of melting, and, in
987 the case of the extremely PPGE-enriched Cliff chromitites, also linked to hydrothermal
988 redistribution from surrounding ultramafics (Prichard and Lord, 1996).

989

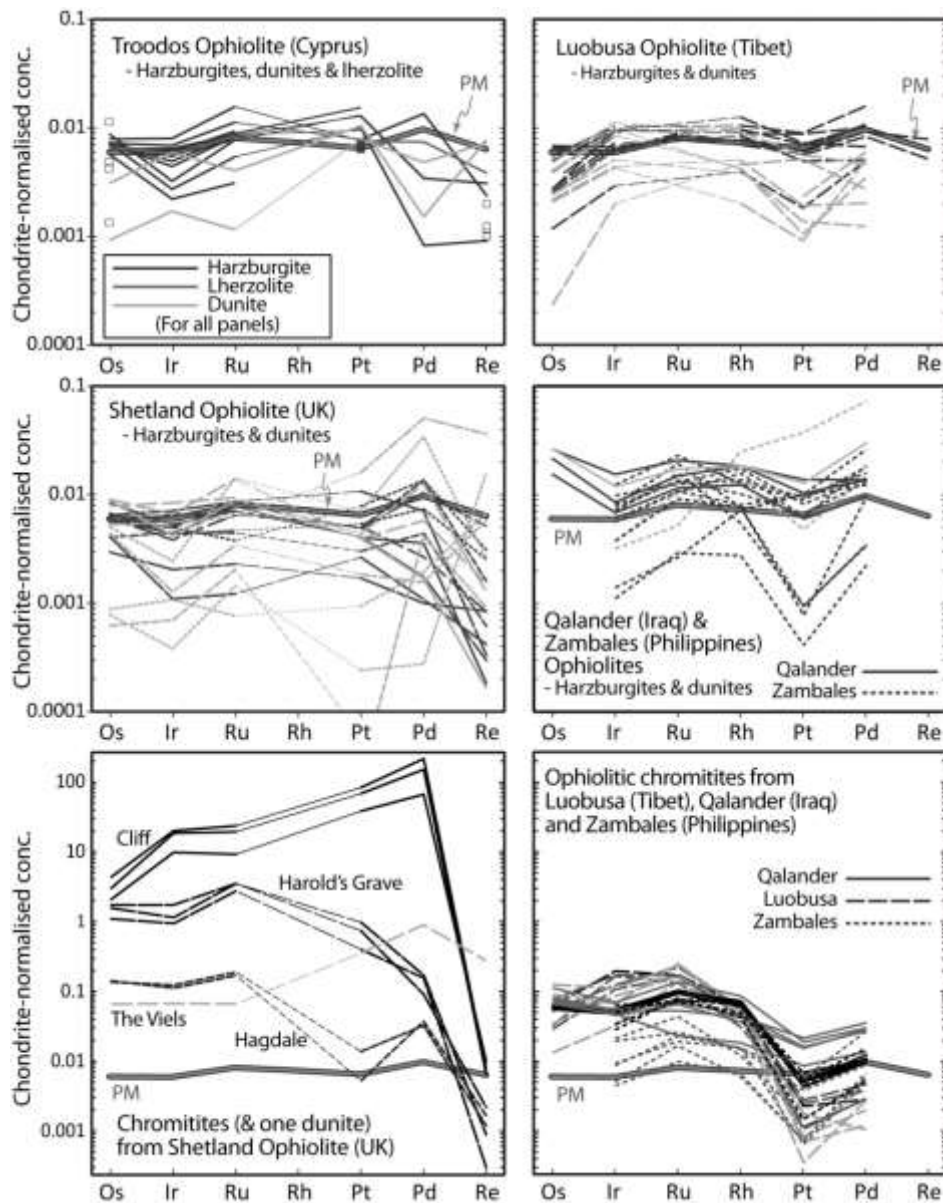
990 There are also large variations in the HSE concentrations and patterns of dunites, which
991 show an overall depletion in Pt, relative to IPGE, and are enriched in Pd in many cases. Rhenium
992 concentrations are low in almost all harzburgite, dunites and chromitites, although enrichment in
993 Re does also occur in some dunites.

994

995 **Zambales Ophiolite (Philippines).** The Zambales Ophiolite contains two distinct blocks,
996 which differ in the composition of their chromitites. The Acoje Block contains chromitites with
997 high-Cr spinel, while the Coto Block is characterized by more Al-rich spinel (Zhou et al., 2000).
998 A comparative study of these two blocks found variations and similarities in the HSE budget of
999 the two chromitite types. As in other studies (e.g., Ahmed et al., 2006; Ismail et al., 2014) high-
1000 Cr chromitites are found to be richer in HSE than those with high-Al spinel. In this case,
1001 however, the IPGE contents vary significantly (e.g., Ru = 8-38 ng/g for Coto, and 62-70 ng/g for
1002 Acoje), while Pt and Pd contents and ratios are similar in the two types (Fig. 16) (Zhou et al.,
1003 2000). Dunites are also found to vary, particularly in Pt content, with the Acoje Block having
1004 more Pt-rich dunites. These spinel compositions and HSE contents are linked to the parental
1005 magmas of the chromitites. The Cr-rich Acoje chromitites were likely generated by interaction
1006 with a refractory boninitic melt, while the Coto chromitites probably had a more tholeiitic source.
1007 Boninitic melts are typically sulfide undersaturated, and thus may form with, and retain, high
1008 HSE abundances, compared to tholeiitic melts which are commonly saturated in sulfide thus
1009 inducing its precipitation and a reduction in HSE content of the remaining melt (Zhou et al.,
1010 2014).

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Qalander Ophiolite (Iraq). The Qalander Ophiolite is a poorly preserved mélangé-type complex, containing serpentinised dunites and harzburgites which surround two types of podiform chromitite; high-Al and high-Cr. The harzburgites and dunites analysed have comparable HSE patterns overall (Ismail et al., 2014), broadly similar to PM estimates (Becker et al., 2006), except offset to higher concentrations (Fig. 16) particularly for Os (4-9 ng/g Ir, 10-17 ng/g Os). As with other chromitite occurrences, Cr-rich and Al-rich types have differing relative proportions of HSE, although they almost all possess high IPGE/PPGE ratios (see Zhou et al., 2014; cf. Shetland, above). Cr-rich chromitites are the most strongly enriched in IPGE, and have the highest IPGE/PPGE ratios. Al-rich chromitites have significantly higher PPGE concentrations, above those of peridotite, while the Cr-rich type has PPGE at the low end of the peridotite range.



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Figure 16. Chondrite-normalized concentration diagrams of the HSE in ophiolites of convergent margin (Troodos, Shetland, Zambales) or uncertain origin. PM estimate shown for comparison. See Fig. 1 for normalization values. White squares for Troodos denote Re-Os analyses of dunites. Qalander & Zambales

1030 chromitites: black lines – Cr-rich, grey lines – Al-rich; Luobusa chromitites: black – massive, grey -
1031 disseminated. It is not clear why there is a discrepancy in the Os data for Luobusa, across two studies.
1032 Given that Becker et al. (2006) used high-temperature acid digestion and isotope dilution, these Os data
1033 should be used in the first instance; the other HSE data is broadly comparable between the two studies.
1034 (References: Zhou et al., 1996; Zhou et al., 2000; Büchl et al., 2002; Büchl et al., 2004; O'Driscoll et al.,
1035 2012; Ismail et al., 2014).
1036

1037 ***Egyptian ophiolites and podiform chromitites, Oman N massifs.*** The Os isotope
1038 composition of PGM from chromitites of the Proterozoic Eastern Desert ophiolite, Egypt and in
1039 the Phanerozoic Oman ophiolite were analysed by Ahmed et al. (2006). It was found that PGM
1040 from different regions of each ophiolite have distinct $^{187}\text{Os}/^{188}\text{Os}$ ratios, from sub- to broadly
1041 chondritic in some regions, to significantly suprachondritic in others (0.1293 for the Proterozoic
1042 Eastern Desert ophiolite and up to 0.1459 for the Oman ophiolite). At the same time, there are
1043 also distinct compositions of the chromitites themselves, with (i) concordant lensoid forms with
1044 intermediate-Cr spinel, which are relatively PGE-poor, and (ii) discordant, dyke-like chromitites,
1045 with high Cr spinel, which are PGE-rich. The authors conclude that the variety of chromitites,
1046 and the Os-HSE signatures that they contain, reflects the variety of formation processes. The
1047 radiogenic chromitites of the Eastern Desert are thought to be affected by crustal contamination,
1048 whereas the radiogenic, Cr- and HSE-rich chromitites from Oman reflect high degree melting and
1049 an input from a subducting slab, most likely in a supra-subduction zone setting (Ahmed et al.,
1050 2006), although here we note that some workers prefer a MOR origin and obducted emplacement
1051 method for the Oman ophiolite (see earlier section).
1052

1053 ***Feather River ophiolite (California).*** A suite of serpentinitised peridotites from the Feather
1054 River ophiolite has been compared with serpentinitised abyssal peridotites and used as a means of
1055 establishing the chemical impacts of serpentinitisation at a range of water/rock ratios and depths in
1056 the mantle (Agranier et al., 2007). The serpentinites have elevated concentrations of seawater-
1057 derived fluid mobile elements, such as boron, although typically lower than abyssal peridotites.
1058 In contrast to many abyssal peridotites, however, Feather River serpentinites do not have
1059 corresponding seawater-affected supra-chondritic $^{187}\text{Os}/^{188}\text{Os}$ ratios (measured range: 0.1175 –
1060 0.1279). Nonetheless, there is a probable covariation between Os abundance and Os isotope
1061 composition in Feather River rocks, albeit over this limited range of $^{187}\text{Os}/^{188}\text{Os}$ compared to
1062 abyssal rocks. Agranier et al. (2007) contend that the serpentinites formed at lower water/rock
1063 ratios (greater depth) than is typical for abyssal rocks, and are therefore more representative of
1064 bulk serpentinitised lithosphere.
1065

1066 In summary, melt percolation in the supra-subduction zone environment generates
1067 substantial lithological heterogeneity, which is accompanied by significant Os isotope and HSE
1068 variability, both between lithological groups (harzburgites, dunites, chromitites, pyroxenites) and
1069 within groups. There is compelling evidence for addition of melt-derived radiogenic ^{187}Os to
1070 parts of the mantle sections of ophiolites (see above and Figs. 14 and 15), most probably due to a
1071 degree of Os fluxing from the downgoing slab, although other possibilities exist. However, the
1072 precise mechanism for such a transfer is not yet clear. The process of melt-rock reaction during
1073 melt percolation results in a decoupling of Al_2O_3 and $^{187}\text{Os}/^{188}\text{Os}$ (Fig. 14), which for other suites
1074 is considered a fairly robust method for determining the approximate ages of depletion for suites
1075 of peridotites, where measured Re contents are often unreliable (Meisel et al., 2001; Lassiter et
1076 al. 2014).
1077

1078 **Highly siderophile elements in the mantle sections of ophiolites of uncertain origin**

1079
1080 ***Luobusa ophiolite (Tibet).*** Chondrite-normalised HSE concentrations for harzburgites,
1081 dunites and chromitites from the Luobusa ophiolite are presented in Fig.16. The concentrations
1082 of Ir, Pt and Pd are broadly comparable between two different studies (Zhou et al., 1996; Becker
1083 et al., 2006), but the low Os/Ir ratios of the Ni-S fire assay data of Zhou et al. (1996) are not

1084 supported by the high temperature (345°C) isotope dilution data of Becker et al. (2006),
1085 suggesting either different petrogenetic histories for the two sample sets or an unidentified
1086 analytical issue for Os in the Zhou et al. data. To err on the side of caution, we will assume the
1087 latter here and disregard the very low Os/Ir ratios in the harzburgites and chromitites.
1088

1089 The harzburgites appear to represent residua after MORB extraction (Zhou et al., 1996).
1090 The HSE abundances are similar to the PM mantle estimate (Becker et al., 2006), and do not
1091 indicate significant melt depletion, except perhaps for Pt (although data for Re – the most
1092 incompatible HSE – is only available for two samples). The Cr-numbers of Cr-spinel in melt-
1093 reacted dunitic rocks are higher than those in the harzburgites, suggesting interaction of a
1094 boninitic melt with the residual peridotite, which also removed pyroxene (Zhou et al., 1996). As
1095 a result, melts became more boninitic and saturated in Cr-spinel, which precipitated to form
1096 chromitite pods within the dunite zones. The inferred boninitic melts suggest a subduction-
1097 related origin for this ophiolite. Chromitites have distinct, strongly fractionated HSE patterns
1098 with high IPGE/PPGE ratios (e.g., normalized Ir/Pt ratios ~100). The concentrations of IPGE in
1099 the chromitites are an order of magnitude or more greater than those of the harzburgites, while Pt
1100 abundances are approximately five times lower in the chromitites than the harzburgites, and are
1101 comparable to the dunites (Fig. 16). These concentrations and patterns are similar to other Cr-
1102 rich chromitites from the Qalander and Zambales ophiolites (Zhou et al., 2000; Ismail et al.,
1103 2014). Dunites have similar PPGE contents to the chromitites, but without the enrichment in
1104 IPGE, due, presumably, to a lack of PGE saturation, and consequent PGM formation, during
1105 dunite formation.
1106

1107 ***Jormua ophiolite (Finland).*** Serpentinites, the oxides they contain, and podiform
1108 chromitites have all been analysed for Re-Os abundances and Os isotopes (Tsuru et al., 2000).
1109 As with most abyssal peridotites that have undergone serpentinisation, Os concentrations,
1110 although somewhat variable (1.5 to 11.7 ng/g) are broadly similar to those of the convecting
1111 upper mantle. Rhenium abundances are more variable; most samples are depleted in comparison
1112 with PM (Becker et al., 2006) but some experienced (probably recent) Re enrichment. Whole-
1113 rock samples have experienced open-system behavior, with respect to Re-Os isotopes, but
1114 chromite to Cr-rich magnetite separates have extremely low Re/Os and largely homogenous
1115 initial $^{187}\text{Os}/^{188}\text{Os}$ values, with a mean γOs of approximately -5, suggesting closed-system
1116 behavior. Other parts of the ophiolite contain chromitites with γOs between +1 and +3. The
1117 authors conclude that the positive values may indicate the presence of MORB-type and
1118 subcontinental lithospheric mantle sources. Addition of radiogenic Os by melt percolation may
1119 be another mechanism to explain the Os isotope data.
1120

1121 ***Outokumpu ophiolite (Finland).*** The Cr-rich nature of residual chromites and boninite-like
1122 volcanic rocks suggest a supra-subduction origin for this ophiolite, but an origin in a continental
1123 rift zone has also been proposed (Walker et al., 1996). The key conclusion of an Os isotope
1124 study (Walker et al., 1996), mainly of chromite, was that this mantle section displayed broadly
1125 chondritic $^{187}\text{Os}/^{188}\text{Os}$ ratios, and hence Re and Os abundances, which were used to support the
1126 ‘late veneer’ model (Chou, 1978). In detail, however, there were variations from a ‘residual’
1127 sub-chondritic laurite (Ru (Os,Ir) S₂) to fluid addition with a composition of around 0.4 γOs . In
1128 this case, however, the radiogenic signature is thought to be derived either from seawater
1129 contamination or from a crustal input during emplacement, akin to that proposed for the Eastern
1130 Desert Ophiolite, Egypt (see previous section).
1131

1132 ***Tethyan ophiolites (Turkey).*** Harzburgites and dunites from Tethyan ophiolites at Koycegiz,
1133 Marmaris, Tekirova, Adrasan and Lake Salda in Turkey have been analysed by Aldanmaz et al.
1134 (2012). Both mid-ocean ridge and supra-subduction zone geochemical signatures have been
1135 identified in different parts of the ophiolites, and these have differing HSE systematics. The mid-
1136 ocean ridge harzburgites have broadly chondritic Os/Ir and supra-chondritic Pd/Ir and Rh/Ir,

1137 similar to PM estimates (Becker et al., 2006), although some PPGE/IPGE enrichment is ascribed
1138 to sulfide addition. They also have a sub-chondritic range of $^{187}\text{Os}/^{188}\text{Os}$ of 0.1223 to 0.1254, and
1139 have correspondingly depleted Re/Os ratios (Aldanmaz et al., 2012). In contrast, the peridotites
1140 of supra-subduction zone affinity have more variable HSE patterns and a wider range of
1141 $^{187}\text{Os}/^{188}\text{Os}$ from 0.1209 to 0.1318, which is -5.26 to 3.27 in γOs units, relative to O-chondrite
1142 evolution. The greater heterogeneity of supra-subduction zone peridotites, compared to those of
1143 mid-ocean ridge affinity, reflects a more complex evolution.

1144
1145 ***Eastern Alps ophiolites (Austria).*** Peridotitic units of Eastern Alps ophiolites (the Reckner,
1146 Hochgrossen, Kraubath, Steinbach and Bernstein peridotites; including two chromitites) have
1147 been found to have remarkably uniform $^{187}\text{Os}/^{188}\text{Os}$ ratios (~0.1266-0.1281), clustering around
1148 the chondritic evolution curve (Meisel et al., 1997), with the exception of one locality (Dorfertal)
1149 which has an Os isotope composition consistent with a minimum age of Re depletion of ~1.6 Ga.
1150 The authors considered the uniformity of Os composition to be somewhat surprising given the
1151 uncertain age and affinity of the samples. One important finding of that study was the robustness
1152 of Os isotopes, given a high degree of serpentinisation, compared with other geochemical data,
1153 and even petrographic and field methods.

1154
1155 ***Mayari-Cristal ophiolite (Cuba).*** The key finding of a study of PGM in the Mayari-Cristal
1156 ophiolite was the scale of Os isotope heterogeneity present within single hand specimens, thin
1157 sections and down to a scale of several millimeters that separated two PGM with contrasting Os
1158 isotope ratios ($^{187}\text{Os}/^{188}\text{Os}$: 0.1185 and 0.1232; Marchesi et al., 2011), which equate to Re
1159 depletion ages of 1370 and 720 Ma, respectively (O-chondrite reference). Given that the budget
1160 of Os for these PGM is thought to be sourced from several m^3 of mantle, this has intriguing
1161 implications for mixing (or the lack thereof) of distinct percolating melts in the mantle (Marchesi
1162 et al., 2011).

1164 DISCUSSION

1166 **Influence of low-temperature alteration processes on the HSE in bulk rocks and minerals**

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1168
1169 Here we briefly discuss the influence of low-temperature (non-magmatic) processes on
1170 the bulk rock, sulfide and PGM composition of mantle tectonites. Ultrabasic rocks affected by
1171 oxidative weathering are usually not used for bulk rock chemical analyses to study high-
1172 temperature processes. Sulfides are at least partially oxidized by these processes, thus, it is
1173 expected that the abundances of chalcophile elements will be disturbed in non-systematic ways.
1174 Because areas of ultramafic rocks affected by oxidative weathering are easily identified by their
1175 brown color, stemming from ferric iron bearing secondary weathering products, such altered
1176 areas can be normally identified and removed.

1177
1178 ***The influence of serpentinization on HSE abundances and $^{187}\text{Os}/^{188}\text{Os}$.***
1179 Serpentinization represents another common low temperature alteration process of ultrabasic
1180 rocks. Serpentinization reactions occur during the reaction of igneous and metamorphic ultrabasic
1181 rocks with seawater or freshwater under a range of geologic conditions and temperatures (e.g.,
1182 Evans et al. 2013 and references therein). For instance, these processes occur today in oceanic
1183 mantle exposed on the seafloor and at greater depth where heated seawater moves within deep-
1184 reaching fractures. Similar processes occurred in ultramafic parts of ophiolites during their
1185 exhumation on or beneath past seafloors, during tectonic obduction or by reaction with fluids and
1186 meteoric water of variable origin during continental collision (Hirth and Guillot 2013). During
1187 serpentinization of peridotites, water reacts with olivine, pyroxenes, spinel (to a lesser extent) and
1188 sulfides that formed at high temperatures. Depending on temperature and progress of reaction,
1189 the new minerals formed include serpentine minerals (chrysotile, lizardite, at higher temperatures

1190 antigorite), magnetite and other secondary minerals such as brucite (see for example Bach et al.,
1191 2004).

1192

1193 The influence of serpentinization on the abundances of HSE in mantle tectonites has not
1194 been studied in much detail. Early Re-Os studies of serpentinized peridotites (e.g., Snow and
1195 Reisberg 1995) have emphasized that serpentinization of peridotites in the oceanic lithosphere
1196 occurs under reducing conditions. Because of the low fO_2 environment caused by the local
1197 production of hydrogen and methane (Evans et al. 2013), secondary sulfides (heazlewoodite,
1198 millerite, godlevskite), Fe-Ni alloy phases (awaruite) and native metals (Au, Cu) may form
1199 (Klein and Bach, 2009) and thus, the HSE are able to retain a low valence. The extent to which
1200 the HSE are retained in these secondary phases compared to the original abundances in the
1201 unaltered bulk rocks and how much of the HSE may be lost into the fluids is poorly constrained.
1202 The similarities of abundances of Os, Ir, Ru, Rh, Pt and Pd in fresh and variably serpentinized
1203 peridotites with similar lithophile element composition have been used to argue that
1204 serpentinization at reducing conditions results in only minor changes in the abundances of these
1205 elements in serpentinized ultramafic bulk rocks that are difficult to resolve from analytical or
1206 intrinsic variations in such rocks (e.g., Becker et al. 2006; Fischer-Gödde et al. 2011; Foustoukos
1207 et al., 2015; Marchesi et al., 2013; van Acken et al. 2008). This contention is supported by
1208 abundances of these elements in serpentinized komatiites, which often preserve correlations
1209 between PGE and Mg or Ni, which were unequivocally produced by igneous fractionation
1210 processes (e.g., Brüggmann et al. 1987; Puchtel et al. 2004, 2005).

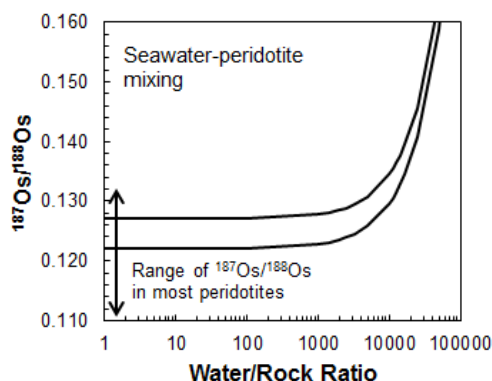
1211

1212 The influence of serpentinization on Re and Au abundances is more difficult to predict,
1213 as no systematic studies exist and the applicability of experimental studies of Re behavior in
1214 specific hydrothermal fluids is difficult to evaluate (Pokrovski et al. 2014, Xiong and Wood
1215 1999). Compared to Pd, Re is often depleted in serpentinized harzburgites, as expected for
1216 strongly depleted residues of partial melting; however, it may also be more enriched than Pd in
1217 normalized concentration diagrams (e.g., Figs. 3, 5, 9, 10, 16). It is difficult to judge if these
1218 abundances reflect secondary addition of Re from seawater (which has very low Re abundances)
1219 that has dissolved sulfides elsewhere, or, if re-enrichment of Re occurred before alteration (e.g.,
1220 by precipitation of liquid sulfide from silicate melts, as may be plausible from observations of
1221 unaltered peridotites). Similar uncertainties arise in serpentinized lherzolites. Correlations of Re
1222 with indicators of melt extraction or refertilization such as Al, Ca or $Mg/(Mg + Fe^{2+})$ in
1223 peridotites have been interpreted as evidence for limited mobilization of Re by low-temperature
1224 alteration processes (e.g., Becker et al. 2006). In mantle pyroxenites that were affected by
1225 variable degrees of serpentinization, Re seems to be unaffected by alteration because it is
1226 typically systematically more enriched than Pd and Pt. Such a behavior is expected from crystal
1227 fractionation products of basic melts (van Acken et al. 2010b). The behavior of gold during
1228 serpentinization of mantle peridotites has not been studied systematically either. Although Au, in
1229 some cases, follows Pd and Re in its geochemical behavior in unaltered peridotites (Fischer-
1230 Gödde et al. 2011), it shows scattered distributions in element variation diagrams that are not
1231 well understood. Because of the known mobility of Au in hydrothermal systems in basic and
1232 ultrabasic rocks (Pokrovski et al. 2014) and the enrichment of Au in some serpentinite-hosted
1233 sulfide deposits (e.g., the Lost City hydrothermal field, Mid Atlantic Ridge), it is to be expected
1234 that Au may be rather mobile during serpentinization.

1235

1236 The question of whether or not the Os budget of serpentinized peridotites can be
1237 measurably affected by radiogenic ^{187}Os from seawater has been discussed in several publications
1238 (e.g., Alard et al. 2005, Brandon et al. 2000, Harvey et al. 2006, Martin 1991, Roy-Barman and
1239 Allègre 1994, Snow and Reisberg 1995, Standish et al. 2002). Cenozoic seawater has highly
1240 variable and mostly very radiogenic $^{187}Os/^{188}Os$ ranging between 0.5 and 1 (Peucker-Ehrenbrink
1241 and Ravizza 2000), however, the concentration of Os in seawater is extremely low (about 3.8 fg/g
1242 Os, (Sharma et al. 1997). These low abundances are in stark contrast to the ng/g levels of Os in

1243 peridotites. Figure 17 illustrates the effects of simple peridotite-seawater mixing, assuming
 1244 $^{187}\text{Os}/^{188}\text{Os}$ of 0.122 and 0.127 and 3.9 ng/g Os in unaltered peridotite and modern seawater with
 1245 $^{187}\text{Os}/^{188}\text{Os}$ of 1 and 3.8 fg/g Os. Very high water-rock ratios of 10^3 to 10^4 are required in order to
 1246 disturb the $^{187}\text{Os}/^{188}\text{Os}$ of peridotite bulk rocks at the % level or higher. Lower values of
 1247 $^{187}\text{Os}/^{188}\text{Os}$ in seawater, such as 0.5, would not alter this conclusion. For comparison, water-rock
 1248 ratios of significantly less than 100 have been calculated for rock units of the Oman ophiolite
 1249 (McCulloch et al. 1981). Some workers have suspected that Mn hydroxide films in cracks and on
 1250 surfaces may pose a problem because these phases tend to scavenge Os from seawater (Martin
 1251 1991, Roy-Barman and Allègre 1994). Although leaching studies of serpentinized peridotites
 1252 have not yielded clear indications of contamination, it is preferable to remove such surfaces or
 1253 avoid such rocks altogether. Most abyssal peridotites are strongly serpentinized, yet they are
 1254 characterized by chondritic to subchondritic $^{187}\text{Os}/^{188}\text{Os}$, similar to unaltered or weakly
 1255 serpentinized post-Archean peridotite xenoliths or other tectonites. Thus there appears to be no
 1256 need to invoke late addition of radiogenic Os by serpentinization. Positive linear correlations of
 1257 $^{187}\text{Os}/^{188}\text{Os}$ with Al_2O_3 contents in serpentinized peridotites provide the strongest argument
 1258 against a significant influence of serpentinization on $^{187}\text{Os}/^{188}\text{Os}$ in such rocks (Reisberg and
 1259 Lorand 1995). These correlations are a primary magmatic feature of mantle rocks (e.g., Handler
 1260 et al. 1997; Peslier et al. 2000; Meisel et al. 2001; Gao et al. 2002).
 1261



1262
 1263 **Fig. 17.** The influence of contamination with seawater on $^{187}\text{Os}/^{188}\text{Os}$ values of peridotites. Typical water-
 1264 rock ratios during alteration of ophiolites are < 100 . Because of the large difference in the concentrations
 1265 of ^{188}Os , even a small increase in $^{187}\text{Os}/^{188}\text{Os}$ of altered peridotites caused by addition of radiogenic Os
 1266 from seawater ($^{187}\text{Os}/^{188}\text{Os} = 1$) would require unrealistically high water/rock ratios. For details on end
 1267 member compositions, see text.

1268
 1269 Suprachondritic $^{187}\text{Os}/^{188}\text{Os}$ occasionally occur in bulk rocks of strongly serpentinized
 1270 abyssal peridotites (Standish et al. 2002) and from serpentinized harzburgites and dunites of
 1271 ophiolite sections and peridotite massifs (e.g., Becker et al. 2001, Büchl et al. 2002, Hanghøj et
 1272 al. 2010). Standish et al. (2002) reported small-scale variations of $^{187}\text{Os}/^{188}\text{Os}$ in serpentinized
 1273 harzburgites and dunites. In the latter study, isotopic differences in chromite ($^{187}\text{Os}/^{188}\text{Os} = 0.124$ -
 1274 0.148) compared to bulk rocks ($^{187}\text{Os}/^{188}\text{Os} = 0.118$ - 0.158) were interpreted to result from
 1275 serpentinization and the addition of seawater-derived radiogenic Os in the altered portion of the
 1276 rocks. Considering the Os concentration differences between seawater and peridotites, it is not
 1277 clear how sufficient ^{187}Os can be added from seawater to raise the $^{187}\text{Os}/^{188}\text{Os}$ to values higher
 1278 than 0.15. The Os isotopic data in Standish et al. (2002) cannot be reconciled with low-
 1279 temperature alteration in a simple way, because Cr rich spinels sometimes have more radiogenic
 1280 Os than their bulk rocks, and samples with the highest $^{187}\text{Os}/^{188}\text{Os}$ are characterized by unusually
 1281 low Os concentrations (below 1 ng/g). Other workers have interpreted chondritic to slightly
 1282 suprachondritic initial $^{187}\text{Os}/^{188}\text{Os}$ in serpentinized dunites and harzburgites to result from the
 1283 interaction between magmas with suprachondritic $^{187}\text{Os}/^{188}\text{Os}$ and mantle rocks, which, because
 1284 of magmatic dissolution of sulfide liquid or chromite, may also cause a decrease of Os

1285 abundances in peridotites (Becker et al. 2001, Büchl et al. 2002, Hanghøj et al. 2010). Alard et al.
1286 (2005) and Harvey et al. (2006) have interpreted different generations of sulfides in serpentinized
1287 peridotites from the Atlantic Ocean to reflect magmatic impregnation from percolating magma
1288 with suprachondritic $^{187}\text{Os}/^{188}\text{Os}$, similar to observations from continental peridotites (Burton et
1289 al. 1999; Alard et al. 2002; Harvey et al. 2011; Reisberg and Luguet 2015, this volume). To
1290 conclude, the effects of serpentinization on the $^{187}\text{Os}/^{188}\text{Os}$ of serpentinized peridotite are likely
1291 minor and difficult to resolve from Os isotopic heterogeneities in mantle rocks inherited from
1292 high-temperature igneous processes.

1293

1294 *Low-temperature decomposition of primary sulfides in peridotites.* Work on sulfide
1295 compositions in peridotites and results of experimental data at typical mantle P-T conditions also
1296 noted that sulfides in peridotites, in particular sulfides on grain boundaries, display exsolution
1297 assemblages from a homogeneous sulfide phase, typically monosulfide solid solution (mss, (e.g.,
1298 Lorand and Luguet, 2015, this volume). The result of these decomposition processes, which
1299 depends on the cooling history, is a heterogeneous assemblage of intergrown sulfides (commonly
1300 pentlandite, pyrrhotite and chalcopyrite), and other minerals, notably platinum metal bearing
1301 alloys and Te-, Bi-, Se-rich phases (Alard et al. 2000; Lorand et al. 2010, 2013; Luguet et al.
1302 2003, 2004, 2007). Because of these subsolidus processes, it is not uncommon that some
1303 elements (e.g., Pt, Te, Au) become strongly redistributed from sulfides into other trace phases in
1304 which they are a major element (e.g., Pt alloys, tellurides, selenides). As a consequence these
1305 elements may display negative anomalies in normalized concentration diagrams of exsolved
1306 sulfide phases (Alard et al. 2000, Lorand et al. 2010) that are not present on the bulk rock scale.
1307 A detailed discussion of phase assemblages and their composition will be given elsewhere in this
1308 volume (Harvey et al., 2015; Lorand and Luguet, 2015).

1309

1310 **The influence of melt infiltration and partial melting on HSE abundances in mantle** 1311 **tectonites**

1312

1313 Since the early 1980s, studies of lithophile element geochemistry and Sr-Nd-Pb isotope
1314 compositions have shown that mantle tectonites have undergone variable degrees of partial
1315 melting during past melting events. Typically this is indicated by their depletion in moderately
1316 and highly incompatible elements (e.g., Frey et al. 1985; Johnson et al. 1990) and unradiogenic
1317 Sr and radiogenic Nd isotopic compositions (e.g., Jacobsen and Wasserburg 1979; Polvé and
1318 Allègre 1980; Reisberg and Zindler 1986). The compositional pattern of major elements in
1319 mantle tectonites is such that most abyssal peridotites and ophiolites genetically related to
1320 convergent plate margins are harzburgites (and subordinate dunites), whereas lherzolites tend to
1321 occur more often in ultra-slow spreading environments, subcontinental settings or continent-
1322 ocean transitions. These compositional differences mirror different degrees of partial melting in
1323 these settings and are broadly consistent with the polybaric melting column model of upwelling
1324 upper mantle (Langmuir et al. 1992). The model predicts that below mid-ocean ridges or other
1325 regions of shallow mantle upwelling such as back arc basins, the highest degrees of melting and
1326 harzburgitic residues are expected at the top of the mantle, whereas lherzolites should occur at
1327 greater lithospheric depth where less melting occurs.

1328

1329 Subsequent work has established that many peridotites show petrologic and geochemical
1330 evidence for a multi-stage history of high-temperature processes (summarized by Bodinier and
1331 Godard 2003). These multi-stage processes include melt extraction and later melt infiltration and
1332 reaction with existing peridotite, which induces chemical changes in mantle rocks that range from
1333 kinetically controlled fractionation of incompatible trace elements (e.g., Vasseur et al. 1991) to
1334 significant modal mineralogical changes (Le Roux et al. 2007). The latter processes are capable
1335 of converting harzburgites into lherzolites (“refertilization”) by stagnation of magma or repeated
1336 influx of magma saturated in a multiphase assemblage of pyroxene(s) ± Al phase (plagioclase,
1337 spinel or garnet) + sulfide in deeper parts of the lithospheric mantle. Melt-rock reaction in

1338 shallow mantle tends to produce tabular dunite, rather than lherzolites (Kelemen et al. 1995,
1339 1997) or plagioclase-pyroxene bearing impregnations, dikes and pockets in otherwise depleted
1340 harzburgite (Edwards and Malpas 1996, Seyler et al. 2004). As a consequence of these
1341 processes, the inventory of incompatible elements and their isotopic composition in these
1342 metasomatically modified rocks is mostly derived from the magma that produced these changes
1343 (for instance, the LREE-depleted compositions of lherzolites from Lherz in the Pyrenees and
1344 their isotopic compositions must have been inherited from the infiltrating magma, Le Roux et al.
1345 2007). Melt infiltration and chemical reaction with peridotite has been recognized as an important
1346 process in many mantle tectonites from different tectonic settings (e.g., Pyrenees, Ronda,
1347 Ligurides, Ivrea Zone, Lanzo, Horoman, abyssal peridotites, ophiolites). It may be ubiquitous in
1348 melting columns, mantle diapirs and in the deep lithosphere and should be considered normal for
1349 open-system melting environments. In the following, we first discuss some general compositional
1350 constraints from peridotites that may be linked to melting processes. We then address the
1351 influence of reactive melt infiltration on sulfide-silicate equilibration and discuss partitioning of
1352 the HSE.

1353

1354 *Behavior of the HSE during partial melting of harzburgites and lherzolites.* A general
1355 observation is that harzburgites have similar abundances of Os, Ir and Ru (IPGE, Barnes et al.
1356 1985) to lherzolites, whereas concentrations of other PGE, Re and Au are typically much lower
1357 in harzburgites than in lherzolites (Figs. 3, 5, 9, 10, 16). On the other hand, basalts and komatiites
1358 often have higher chondrite-normalized concentrations of Pt-group PGE (PPGE: Rh, Pt, Pd,
1359 Barnes et al. 1985), Au and Re than IPGE (Bezou et al. 2005; Brüggemann et al. 1987; Hertogen et
1360 al. 1980; Puchtel et al. 2004; Rehkämper et al. 1999). These studies have pointed out that the
1361 main host phase of the HSE in lherzolites at high temperatures should be sulfide. Thus, the
1362 stronger depletion of Rh, Pt, Pd, Au, Re and sulfur in harzburgites compared to lherzolites likely
1363 reflects the consumption of sulfide in peridotite during high degrees of melting (e.g., Barnes et al.
1364 1985; Keays 1995; Lorand 1988; Morgan 1986). The details of sulfide dissolution and HSE
1365 partitioning into basic magma have remained unclear, particularly for melting processes at P-T
1366 conditions that should yield lherzolite residues. Many workers have advocated sulfide-silicate
1367 partitioning (e.g., Brenan et al. 2015, this volume, and references therein). For chalcophile
1368 element partitioning, the assumption has been that during local partial melting in the mantle, a
1369 homogeneous sulfide liquid or solid should coexist in equilibrium with silicate melt, olivine,
1370 pyroxenes and an Al-rich phase (Keays 1995; Morgan 1986; Rehkämper et al. 1999). The amount
1371 of sulfide liquid dissolved into the silicate melt is controlled by ambient pressure, temperature
1372 and FeO content of the melt (Jugo et al. 2005; Mavrogenes and O'Neill 1999; O'Neill and
1373 Mavrogenes 2002). Another partitioning process, mss-liquid sulfide partitioning, that was also
1374 proposed to control HSE abundances (Bockrath et al. 2004) will be discussed below. During
1375 melting of upwelling asthenosphere or deep lithosphere, at temperatures $>1250^{\circ}\text{C}$, it is expected
1376 that mantle rocks and coexisting magmas were chemically and isotopically equilibrated, as is
1377 commonly assumed for lithophile elements (Hofmann and Hart 1978).

1378

1379 At high degrees of melting, partitioning of Os, Ir, Ru, Rh and Pt may be controlled by the
1380 solubility of alloys of these elements in silicate melt (Pearson et al. 2004; Fonseca et al. 2011;
1381 2012; Mungall and Brenan 2014; Brenan et al. 2015, this volume). The significance of this for
1382 the composition of harzburgites will be discussed later. Here, we specifically focus on processes
1383 during low and moderate degrees of melting in the deeper regions of the melting column where
1384 sulfide should be stable in the residue and sulfide-silicate partitioning has been proposed as the
1385 main control on the distribution of the HSE and other chalcophile elements (Barnes et al. 1985;
1386 Morgan 1986). However, it has been unclear if sulfide exists as a solid phase (mss), liquid
1387 sulfide, or both. Recent improvements in the accuracy and precision of liquid sulfide-silicate
1388 partition coefficients ($D^{\text{sulf/sil}}$) indicate values in the range of 10^5 to 10^6 and 10^4 for the PGE and
1389 Au, respectively (Li and Audétat 2013; Mungall and Brenan 2014; Brenan et al. 2015), whereas
1390 Re is much less chalcophile ($D^{\text{sulf/sil}} \approx 300\text{-}800$, Brenan 2008; Fonseca et al. 2007). Assuming a

1391 simple fractional melting process (batch melting yields similar results as long as the elements are
1392 not highly incompatible), element concentrations in the residues can be calculated according to
1393 the mass balance equation $C_r = C_o (1-F)^{(1/D^b)-1}$, with C_r = concentration of an element in the
1394 residue, C_o = total concentration of an element in the bulk system (residue + melt), D^b = bulk
1395 partition coefficient of an element between residue and melt, F = melt fraction. As long as sulfide
1396 is present in the mantle residue and it is equilibrated with silicates and silicate melt, the high
1397 $D^{\text{sulf/sil}}$ require nearly constant concentrations of all PGE in peridotites (Fig. 18a), because bulk
1398 partition coefficients of the PGE in lherzolites are $\gg 1$: At 0.02 wt. % S in fertile lherzolite and
1399 35 wt. % S in monosulfide solid solution, $D_{\text{PGE}}^b > 0.00057 \times 10^5 + 0.9994 \times 0.1 = 57$, assuming
1400 $D_{\text{Pd}}^{\text{sil.min./sil.melt}} < 0.1$ with other PGE likely having higher $D^{\text{sil.min./sil.melt}}$ (Mungall and Brenan 2014).
1401 Gold should also be retained in lherzolites that have lost a significant fraction of melt ($D_{\text{Au}}^b \geq$
1402 $0.00057 \times 5 \times 10^3 = 3$, assuming $D_{\text{Au}}^{\text{sil.min./sil.melt}} < 0.01$ (Mungall and Brenan 2014), whereas Re
1403 should be moderately depleted at relevant $f\text{O}_2$ in normal upper mantle (FMQ-1), as its D^b is
1404 always below 1 in cases where no garnet occurs in the residue ($D_{\text{Re}}^b \leq 0.00057 \times 800 + 0.9997 \times$
1405 $0.1 = 0.6$, assuming $D_{\text{Re}}^{\text{sil.min./sil.melt}} < 0.1$ (no garnet), Brenan 2008; Mallmann and O'Neill 2007).
1406

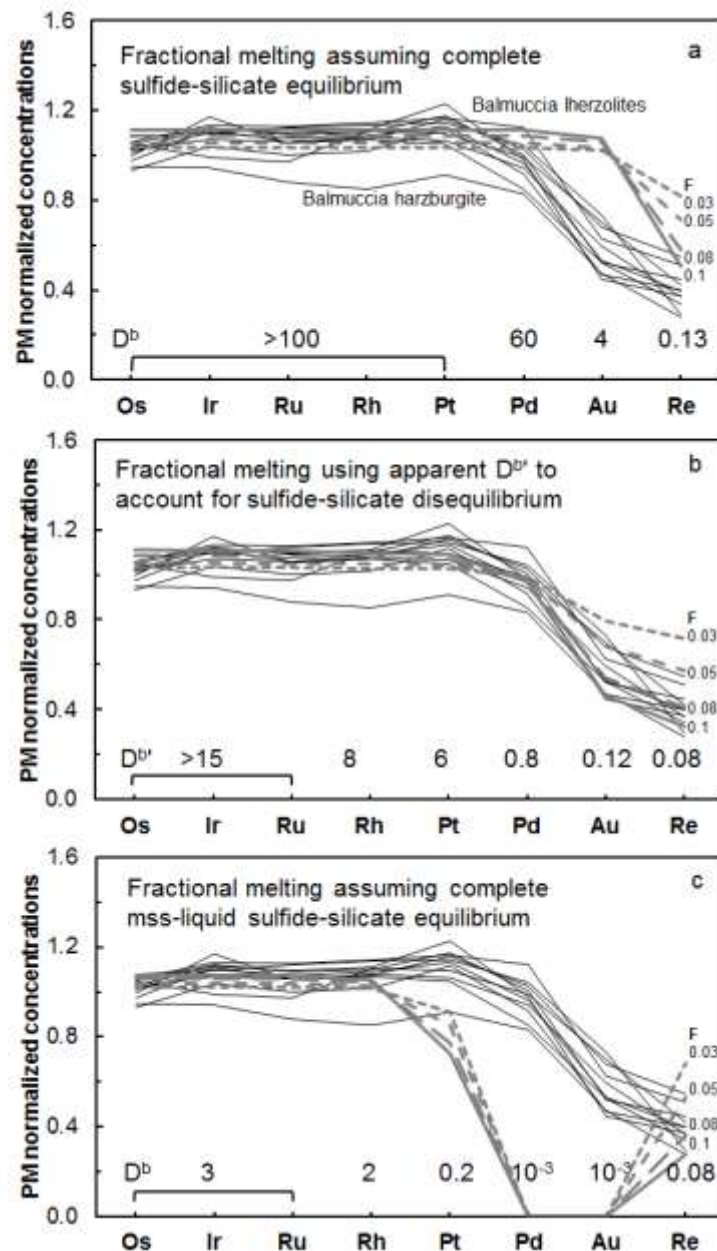
1407 The situation in mantle rocks, however, has been found to be more complicated; one
1408 indication being the difficulty in reproducing peridotite HSE patterns by sulfide-silicate
1409 equilibrium partitioning (Fig. 18a). In the following, we discuss evidence suggesting that many
1410 mantle peridotites are in chemical disequilibrium regarding chalcophile element partitioning at
1411 the scale of hand specimen to grain boundaries. An alternative partitioning scenario, such as
1412 mss-sulfide liquid-silicate liquid equilibrium, is also discussed below.
1413

1414 ***Melt infiltration at high temperatures induce chemical disequilibrium of chalcophile***
1415 ***elements in mantle peridotites.*** Studies of chalcophile element abundances in sulfides of
1416 different textural position, in mantle xenoliths and in peridotite tectonites, have shown that
1417 significant compositional differences may exist between sulfides that occur as inclusions in
1418 olivine (and sometimes pyroxenes and spinel) and sulfides present at grain boundaries. The
1419 former are rich in Ir-group PGE and depleted in Pd, Au and Re, while the latter may or may not
1420 be depleted in IPGE and have higher Pd, Re and Cu (Alard et al. 2000; 2002; Luguet et al. 2001;
1421 2003; 2004). Although these different assemblages are sometimes complicated by internal
1422 separation into multi-phase assemblages (pentlandite, pyrrhotite and other phases) that occurred
1423 late during slow cooling, it is clear from their different compositions that included and grain
1424 boundary sulfides were not chemically equilibrated during their formation. The sulfide
1425 assemblages on grain boundaries are sometimes associated with pyroxene-spinel assemblages
1426 that have been interpreted to have formed during melt infiltration and refertilization. From this
1427 observation, it follows that reactive melt infiltration likely led to sulfur saturation in these
1428 magmas and precipitation of the sulfides located on grain boundaries (e.g., Alard et al. 2000).
1429 The reaction of silicate melts and sulfide segregation processes are not only indicated by the
1430 different sulfide assemblages in the peridotites, but also by the HSE abundances in
1431 mineralogically zoned boundaries between pyroxenites and host peridotites and disequilibrium
1432 sulfide assemblages in mantle pyroxenites (see section on mantle pyroxenites below).
1433

1434 Some authors have proposed that sulfide melts may be mobile in mantle rocks, and thus
1435 may change the Re-Os and PGE systematics of mantle rocks (Gaetani and Grove 1999). The
1436 existing data on peridotites, however, do not support pervasive or wide-spread sulfide melt
1437 mobility, as linear correlations between $^{187}\text{Os}/^{188}\text{Os}$, Re and S abundances and lithophile elements
1438 such as Al, Ca or Mg in peridotites would not be maintained over long periods of time in the
1439 mantle (Fig. 4, 7; e.g., Becker et al. 2006; Meisel et al. 2001; Reisberg and Lorand, 1995; Wang
1440 and Becke,r 2013), although minor mobility is not precluded due to scatter in the datasets. The
1441 role of fluids as metasomatic agents in the redistribution of HSE and other chalcophile elements
1442 has been invoked in some cases (e.g., Lorand and Alard, 2010). One possibility is that such
1443 fluids are the end products left after crystallization of mantle-derived melts or, if they are of

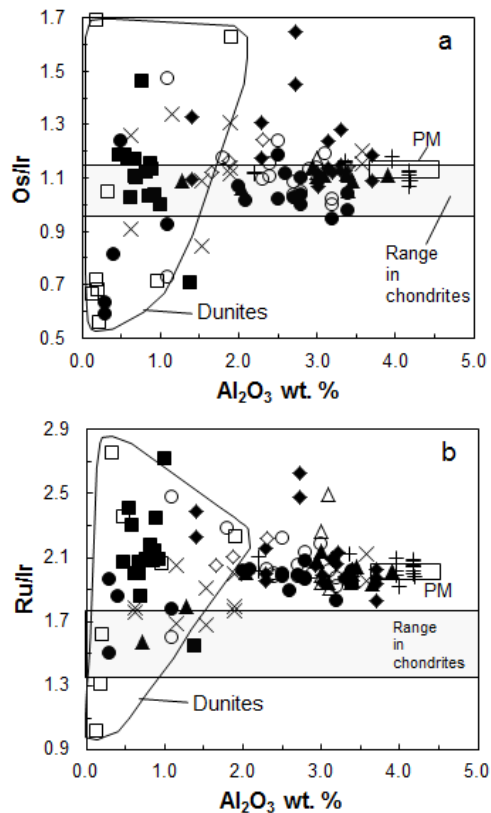
1444 external origin, may have been derived from crustal sources at lower temperatures during the
 1445 exhumation history of mantle tectonites. Regardless of the origin of the fluids, what is not yet
 1446 clear is the effect of these small-scale observations on the mass balance of bulk rocks. In
 1447 summary, silicate melts are the main metasomatic agents that, by way of coupled precipitation of
 1448 sulfide melt, pyroxenes and an Al phase, clearly produce significant modifications of HSE
 1449 abundances and $^{187}\text{Os}/^{188}\text{Os}$ at magmatic temperatures in the mantle.
 1450

1451 Detailed surveys of the accessory mineral inventory of peridotites (e.g., Fig. 6) have
 1452 revealed the occurrence of Pt-Ir alloys, Ru-Os-bearing sulfides and Os-Ir-Ru alloy phases
 1453 (Luguet et al. 2007; Lorand et al. 2010; O'Driscoll and González-Jiménez, 2015, this volume).
 1454 These phases are expected to become stabilized by decreasing fS_2 shortly before or during the
 1455 exhaustion of liquid sulfide in harzburgite residues at moderate to high degrees of melting (e.g.,
 1456 Fonseca et al. 2012; Mungall and Brenan 2014; Brenan et al., 2015, this volume). Thus, their
 1457 occurrence in harzburgites (e.g., at Lherz; Luguet et al. 2007) is not unexpected.



1458
 1459
 1460

1461 **Figure 18.** Primitive mantle-normalized concentration diagrams of the HSE in residues of fractional
 1462 melting of fertile peridotite in comparison to lherzolites and a harzburgite from the Balmuccia peridotite
 1463 massif (data from Wang et al. 2013). The latter are shown here as an example, because concentration data
 1464 of HSE and lithophile elements in lherzolites are relatively homogeneous and lithophile incompatible
 1465 element data suggest that these rocks are residues of fractional melting (see text). The linear concentration
 1466 scale was used to show details of the fractionation between Pt, Pd, Au and Re. Shown are the effects of
 1467 equilibrium and disequilibrium distribution of the HSE between rock and coexisting melt and different
 1468 melt fractions F . a) Ideal sulfide-silicate equilibrium partitioning. Bulk partition coefficients D^b
 1469 were calculated based on sulfide-silicate and mineral-silicate melt partition coefficients at fO_2 near FMQ-1
 1470 (Brenan 2008; Fonseca et al. 2007; Mallmann and O'Neill 2007; Mungall and Brenan 2014). b) Apparent
 1471 bulk partition coefficients $D^{b'}$ were estimated to account for mixing and the disequilibrium distribution
 1472 between sulfides and silicates during open system melting (see text). c) The effects of monosulfide solid
 1473 solution (mss)-liquid sulfide-silicate partitioning, assuming equilibrium among all phases. Mss-sulfide melt
 1474 partition coefficients from Ballhaus et al. (2006), Brenan (2002), Li et al. (1996), Mungall et al. (2005).
 1475 Note that for the PGE, some silicate mineral-silicate melt partition coefficients (e.g., pyroxenes) are not
 1476 well-constrained. In such cases partition coefficients for olivine were used. Thus D^b for Pd and Au in c)
 1477 may be higher if these elements are more compatible in pyroxenes and in the Al phase.
 1478



1479 **Figure 19.** a) Os/Ir- Al_2O_3 and b) Ru/Ir- Al_2O_3 in peridotite tectonites. Representative lherzolites and
 1480 harzburgites from continental extensional and transitional oceanic environments (Balmuccia: solid circles,
 1481 Baldissero: open circles, Lherz: x, Turon de la Tecuere: +, Lanzo: solid diamond, Internal Ligurides: open
 1482 diamond). Also shown are harzburgites (solid squares) and dunites (open squares) from the Wadi Tayin
 1483 section of the Oman ophiolite, dunites from Balmuccia (solid circles within the Dunite fields, see also
 1484 Figs. 5, 8b) and lherzolites from Ronda (open triangle) and Beni Bousera (solid triangle). For data sources of
 1485 peridotites see Fig. 5 and text. Chondritic range from Horan et al. (2003) and Fischer-Gödde et al. (2010).
 1486 Primitive mantle model from Becker et al. (2006). The data show relatively homogeneous ratios in
 1487 lherzolites and larger variations in harzburgites and in replacive dunites (see text for details).
 1488

1489

1490

1491 However, such phases have also been detected in lherzolites from Lherz that formed by
 1492 refertilization, albeit they occur in smaller proportions than in harzburgites (Lorand et al. 2010).

1493 If the alloy phases were indeed inherited from more depleted parent rocks, their presence in some
1494 lherzolites may also reflect chemical disequilibrium between these phases and the more abundant
1495 sulfide minerals that were precipitated as sulfide liquid from silicate melt. The impact of such
1496 inherited and presumably 'residual' alloy phases on bulk rock budgets of lherzolites that formed
1497 by refertilization appears rather limited. For instance, the bulk rock Os/Ir ratios of lherzolite
1498 tectonites is rather homogeneous and overlaps chondritic values (Fig. 19a, Becker et al. 2006;
1499 Fischer-Gödde et al. 2011; Liu et al. 2009; Pearson et al. 2004; Wang et al. 2013). Because of the
1500 different solubilities of Os and Ir metal in silicate melt (e.g., Mungall and Brenan 2014),
1501 chondritic Os/Ir are not a priori maintained in residual peridotites at higher degrees of melting (as
1502 witnessed by the larger scatter of this ratio in harzburgites). Pt/Ir and Pt/Os in lherzolites range
1503 from chondritic to mildly subchondritic. Only rarely do lherzolites display enrichments of Pt that
1504 are decoupled from Pd, Au and Re (e.g., Fig. 5b, c) and might be ascribed to the excess presence
1505 of Pt minerals. In this context, it is noteworthy that ratios of Ir, Os and Ru in mantle tectonites
1506 tend to be more scattered in harzburgites than in lherzolites (Fig. 19). The difference in
1507 homogeneity of the different rock types may either reflect digestion problems in the laboratory, i.
1508 e. the difficulty of complete dissolution of refractory platinum group metal alloys in harzburgites
1509 (Meisel and Horan, 2015, this volume, and references therein), or it may be due to dissolution of
1510 refractory alloy phases in coexisting sulfur-undersaturated melt at high temperatures.
1511

1512 *Osmium isotopic disequilibrium within mantle peridotites.* Evidence for small-scale
1513 chemical disequilibrium regarding chalcophile elements is provided by Re-Os data that suggest
1514 that grain- to hand specimen-scale Os isotopic disequilibrium is common in the mantle. Burton et
1515 al. (1999) found that different mineral separate fractions from mantle xenoliths showed differing
1516 $^{187}\text{Os}/^{188}\text{Os}$ that were not related by isochronous behavior. Leaching experiments of powders of
1517 refertilized mantle xenoliths and tectonites show that $^{187}\text{Os}/^{188}\text{Os}$ frozen in during the Archean or
1518 Proterozoic survived Phanerozoic refertilization, most likely because of the preservation of
1519 ancient chromite or olivine that contained inclusions of HSE carrier phases (Chesley et al. 1999;
1520 Becker et al. 2006; Wang et al. 2013). Alard et al. (2002; 2005) showed that the sulfide
1521 populations with different PGE compositions also display systematic differences in Re/Os and
1522 $^{187}\text{Os}/^{188}\text{Os}$. In peridotite xenoliths and abyssal peridotites, sulfides on grain boundaries tend to
1523 have chondritic to suprachondritic Re/Os and $^{187}\text{Os}/^{188}\text{Os}$, whereas sulfides in inclusions also
1524 display subchondritic values (Harvey et al., 2006; Harvey et al., 2011; Warren and Shirey, 2012).
1525 The heterogeneous $^{187}\text{Os}/^{188}\text{Os}$ in different bulk rocks of essentially all suites of peridotites,
1526 xenoliths or tectonites from different geodynamic environments (e.g., Figs. 1, 2, 4, 7, 9, 11; and
1527 Reisberg and Luguet 2015, this volume) also represents a manifestation of disequilibrium on the
1528 scale of hand specimen and outcrops. In principle, such variation may have been caused by
1529 differences in the age of partial melting and melt infiltration. However, evidence for grain-scale
1530 initial Os isotopic heterogeneity at times of melt infiltration (in cases where the timing can be
1531 constrained) suggest that mixing of residues and melts with different $^{187}\text{Os}/^{188}\text{Os}$ during reactive
1532 melt infiltration did not result in full Os isotopic equilibrium. A good example are the ultramafic
1533 tectonites in the Pyrenees and in the Italian and Swiss Alps (Baldissero, Balmuccia, Lanzo,
1534 Totalp), where episodic melt infiltration into Proterozoic continental lithospheric mantle during
1535 Paleozoic and Mesozoic extension only partially re-equilibrated $^{187}\text{Os}/^{188}\text{Os}$ values. All these data
1536 and observations suggest that disequilibrium must have been maintained even at high
1537 temperatures in the upper mantle and in the presence of silicate melt. The widespread
1538 heterogeneity of initial $^{187}\text{Os}/^{188}\text{Os}$ at the grain boundary- to centimeter-scale in mantle rocks also
1539 suggests that sulfide liquids are efficiently trapped even during recrystallization processes.
1540

1541 Alongside evidence from textures and lithophile elements (e.g., Le Roux et al. 2007;
1542 Mazzucchelli et al. 2009; Müntener et al. 2005; Rivalenti et al. 1995), the extent of re-
1543 equilibration is manifested in the scatter of HSE abundances displayed by different suites of
1544 peridotites, in the abundance of harzburgite rocks in outcrops and in the distribution of Re-Os
1545 model ages in these bodies. At Lherz, Lanzo and Baldissero Re depletion ages of peridotites

1546 display bimodal distributions of Proterozoic and Phanerozoic ages, with harzburgites or depleted
1547 lherzolites typically showing older model ages (i. e., lower measured $^{187}\text{Os}/^{188}\text{Os}$) than lherzolites
1548 (Reisberg and Lorand 1995; Burnham et al. 1998; Becker et al. 2006; Fischer-Gödde et al. 2011;
1549 Wang et al. 2013). In contrast, at Balmuccia and Totalp, depleted lherzolites and harzburgites are
1550 rare and display Proterozoic Re depletion ages. Model ages of fertile lherzolites at these locales
1551 range from Phanerozoic to future ages (van Acken et al. 2008, 2010; Wang et al., 2013). Of note
1552 is that the scatter of the concentrations of Os, Ir and Ru in fertile peridotites at these localities is
1553 more limited than in other lherzolite bearing tectonites (compare Fig. 5b with 5a and 5c).

1554

1555 Osmium isotopic heterogeneity is also prevalent in abyssal peridotites, which are
1556 commonly presumed to represent melting residues of MORB-type magmas. Harvey et al. (2006)
1557 have shown that sulfides in harzburgites from the 15°20' N fracture zone (Atlantic Ocean)
1558 preserve small-scale isochronous relationships that date back to the Paleo-Proterozoic. Such
1559 preservation of early- to mid-Proterozoic $^{187}\text{Os}/^{188}\text{Os}$ values in bulk rocks and sulfides has also
1560 been reported in other abyssal peridotites (Parkinson et al. 1998; Alard et al. 2005; Liu et al.
1561 2008; Warren and Shirey 2012). Further evidence of small-scale disequilibrium is apparent in
1562 studies of platinum-group minerals from ophiolites. Platinum group minerals from the Mayari-
1563 Cristal Ophiolite, Cuba, have been found to have diverse $^{187}\text{Os}/^{188}\text{Os}$ ratios even on the scale of a
1564 single thin section (Marchesi et al., 2011). The most extreme example found was the presence of
1565 two PGM only a few millimeters apart, with $^{187}\text{Os}/^{188}\text{Os}$ ratios of 0.1185 and 0.1232 (Marchesi et
1566 al., 2011), which give T_{RD} ages of 1370 and 720 Ma, respectively (ordinary chondrite reference
1567 evolution line; Walker et al. 2002). The mechanism of formation for such PGM is not well
1568 known, but given that the budget of Os for these PGM is thought to be sourced from at least
1569 several m^3 of mantle (total Os equivalent to $\sim 1\text{m}^3$ mantle), this would imply little if any mixing of
1570 percolating melts, or a lack of equilibration between mineral grains and subsequent percolating
1571 melts.

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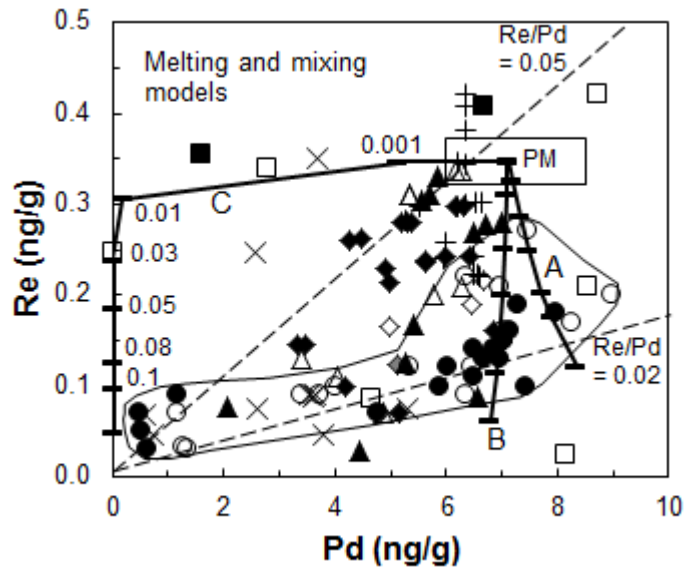
1573 ***The influence of disequilibrium between mantle and magmas on HSE distributions.***
1574 The predicted behavior of the HSE can be compared with HSE patterns of peridotites. Relatively
1575 'constant' concentrations have been noted for the Ir group PGE in many studies of lherzolite
1576 tectonites. However, Rh and Pt display a tendency towards higher concentrations in lherzolites
1577 (e.g., Fischer-Gödde et al. 2011). In some (but not all) suites of peridotites, Pd correlates with
1578 fertility indicators such as Al_2O_3 abundances (e.g., Becker et al. 2006). Some workers (e.g.,
1579 Lorand et al. 1999, Pearson et al., 2004) have noted that the variable depletion of Pd in lherzolites
1580 is difficult to reconcile with partial melting and very high sulfide-silicate partition coefficients ($>$
1581 10^4 to 10^5). The smooth depletion of Pd, Au and Re relative to other HSE in lherzolites from
1582 Balmuccia and elsewhere (e.g., Fig. 18) is inconsistent with equilibrium partitioning and the
1583 liquid sulfide-liquid silicate partitioning data. It is also difficult to explain by other equilibrium
1584 partitioning processes involving sulfides, e.g., mss-liquid sulfide (see below). Furthermore,
1585 concentrations of Os, Ir and Ru in peridotite tectonites of similar lithophile element composition
1586 display considerable scatter (e.g., Fig. 5), as do Os isotopic compositions. For lherzolites, at
1587 least, the different concentrations cannot entirely be an artifact of heterogeneous distribution of
1588 sulfide grains within sample powders or the rock (Meisel and Moser 2004; Meisel and Horan
1589 2015, this volume). Instead, these concentration variations may reflect the compositional
1590 variability of sulfide grains in the rock; as indicated by variable Ir and Ru concentrations in
1591 peridotitic sulfides (e.g., Alard et al. 2000). As there is indisputable evidence for widespread, or
1592 even ubiquitous, chemical and isotopic disequilibrium of the HSE in peridotites, it is plausible
1593 that the distribution of chalcophile elements between peridotite and magma is partly controlled
1594 by the composition of sulfide liquids from infiltrating primitive magmas and partly by mixing
1595 processes between such liquids and sulfide liquids already present in the rocks (e.g., Lorand et al.
1596 1999; Alard et al. 2000; Pearson et al. 2004; Lorand et al. 2010).

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1598 In the melting model shown in Fig. 18b apparent sulfide-silicate partition coefficients
1599 were used to match the patterns of peridotites from the Balmuccia peridotite massif. Apparent
1600 partition coefficients take into account the extent to which the HSE composition of peridotites
1601 displays the effects of mixing, and thus the influence of the original infiltrating melt
1602 compositions, rather than just sulfide melt-silicate melt equilibrium. It is clear that the
1603 fractionations inherited from the melt contribute to the lowering of D^b , compared to the
1604 equilibrium case. The differences will be particularly notable for Pd and Au. As Pd in depleted
1605 lherzolites is commonly slightly depleted, the apparent bulk distribution coefficient for this
1606 element should be < 1 and apparent sulfide-silicate distribution coefficients in the model in Fig.
1607 18b would be about 1300; far lower than the 10^5 to 10^6 range for sulfide-silicate equilibrium
1608 (Mungall and Brenan 2014). For Pt and Rh apparent partition coefficients may also be lower.
1609 Gold abundances in depleted lherzolites are lower than in fertile lherzolites and this, coupled with
1610 the slight enrichment of Au in primitive basaltic magmas, suggests that Au also has an apparent
1611 bulk distribution coefficient < 1 . Consequently, apparent sulfide-silicate distribution coefficients
1612 for Au are significantly lower (about 200 in the case of Fig. 18b) than equilibrium values (4000-
1613 10000; Mungall and Brenan 2014). Rhenium and other moderately chalcophile elements with
1614 equilibrium sulfide-silicate partition coefficients < 1500 are not sensitive enough to identify
1615 chemical disequilibrium, as the influence of the silicate mineral-silicate melt partition coefficients
1616 is substantial. Combined sulfide-silicate and silicate mineral-silicate melt partition coefficients of
1617 these elements yield bulk partition coefficients < 1 , whether or not equilibrium is assumed.
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1619 Figure 20 displays the variation of Re concentrations versus Pd concentrations in various
1620 suites of mantle tectonites (note that in more strongly serpentinized peridotites, such as from the
1621 Oman ophiolite, Re may also be affected by late-stage alteration). Both elements tend to correlate
1622 in harzburgites and in depleted lherzolites, however, in more fertile rocks, Re displays larger
1623 variations (0.07 to 0.4 ng/g) at relatively constant Pd (5 to 9 ng/g). The most likely explanation
1624 for this observation is that sulfide and other HSE carrier populations in harzburgites and depleted
1625 lherzolites reflect mixing and full disequilibrium, whereas pre-existing phases in fertile
1626 lherzolites may have partially reacted and equilibrated with a larger fraction of silicate melt and
1627 sulfide liquid. The data also suggest that HSE carriers in fertile peridotites of some suites (e.g.,
1628 Balmuccia and Baldissero) must be more depleted in Re than other suites, which may be a
1629 property of the melts that precipitated sulfides during reactive infiltration. The curved trend
1630 defined by some data in Fig. 20 may be related to the quantity of melt that reacted and
1631 precipitated sulfide liquid in the rock. The systematic behavior of Pd, Au, Re and of other
1632 chalcophile elements such as S, Se, Te, Cu and Ag in most peridotites and in MORB (Wang and
1633 Becker 2015b) indicates that the relative depletion and enrichments of these elements in
1634 peridotites and in MORB may be described by apparent bulk partition coefficients. Melt
1635 compositions calculated by this approach may yield similar concentrations of Pd, Au and Re as in
1636 primitive MORB, although the latter almost certainly require a more complicated fractionation
1637 history (e.g., Langmuir et al. 1992; Rehkämper et al. 1999; Bezos et al. 2005; Mungall and
1638 Brenan 2014; Wang and Becker 2015c).

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Figure 20. Concentrations of Re and Pd in peridotite tectonites and evolution of the composition of residues in different melting models. Symbols as in Fig. 19, melting curves A to C calculated using parameters from Fig. 18 and the text. A: equilibrium liquid sulfide-silicate partitioning (Fig. 18a), B: disequilibrium distribution, taking into account the effect of mixing of different types of sulfide with different partitioning histories (Fig. 18b), C: mss-liquid sulfide-silicate partitioning (Fig. 18c). Different Re/Pd ratios in lherzolites are indicated by dashed lines. None of the melting models yields a satisfactory match for the data distribution of different peridotite suites. In this diagram, ideal binary mixing processes without chemical reaction should result in linear correlations; e.g., mixing of ‘residual’ Re- and Pd-depleted sulfide liquid with Re-Pd-rich sulfide liquid precipitated from percolating magma. Most peridotites from Lanzo display such a trend along a Re/Pd of 0.05. Depleted lherzolites and harzburgites from Baldissero and Balmuccia also display a linear trend albeit at a lower Re/Pd, presumably because the infiltrating magma was more depleted in Re and other incompatible elements. In fertile lherzolites the data is scattered, likely because of the predominance of sulfides derived from infiltrating magma and partial chemical equilibration. Chemical equilibration tends to decouple variations of Re and Pd because of their very different partitioning behavior at low to moderate degrees of melting (Brenan et al. 2015, this volume).

Symbol key: Balmuccia: solid circle (dunites at low Re and Pd concentrations), Baldissero: open circle, Lherz: x, Turon de la Tecuere: +, Lanzo: solid diamond, Internal Ligurides: open diamond, External Ligurides: gray diamond, Ronda: open triangle, Beni Bousera: solid triangle. Also shown are harzburgites (solid square) and dunites (open square) from the Wadi Tayin section of the Oman ophiolite, For data sources of peridotites see Fig. 5 and text.

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An alternative model of HSE partitioning during mantle melting was presented by Bockrath et al. (2004) and Ballhaus et al. (2006). These authors proposed that residual mss may coexist with liquid sulfide over a significant pressure-temperature range in the mantle. Partitioning between these phases may control the HSE abundances in residues and silicate melts. However, because of uncertainties in the position of the sulfide liquidus in different experimental studies, the stability of mss in the asthenosphere or deeper lithosphere is debated (see Fonseca et al., 2012; Mungall and Brenan 2014). The relevance of mss-liquid sulfide partitioning in the upper mantle can be evaluated on the basis of existing partitioning data for chalcophile elements and the composition of mantle rocks, basalts and their sulfides. Melting models of bulk rock compositions of lherzolites that employ mss-liquid sulfide partition coefficients (Fig. 18c) display a poor match for Pt, Pd and Au. However, it must be acknowledged that bulk partition coefficients are strongly influenced by the silicate mineral-silicate melt partition coefficients. Only for olivine-silicate melt partitioning does sufficient data exist for Pt, Pd and Au (see equations 11-13 in Mungall and Brenan, 2014, which yield low $D^{\text{olivine/silicate}}$ melt for these elements at fO_2 of 10^{-9} to 10^{-10} bar). Pyroxene-silicate melt partition coefficients for these

1679 elements are poorly constrained, and thus D^b may be higher. As for sulfide liquid-silicate
1680 partition models, Re fits well because its D^b is strongly controlled by the large mass fraction of
1681 silicates and the well-determined mineral-silicate melt partition coefficients.

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1684 In principle, mss-liquid sulfide partitioning may account for the different patterns of Ir
1685 group and Pt group PGE in sulfide inclusions and sulfides on grain boundaries in peridotites (e.g.,
1686 Ballhaus et al. 2006). However, the behavior of Re concentrations in sulfide inclusions versus
1687 grain boundary sulfides argues against this process. Equilibrium mss-liquid sulfide partitioning
1688 would predict higher Re and Os concentrations in residual sulfides compared to coexisting
1689 sulfide liquids, because both elements are compatible in mss ($D_{Os}^{mss/sul\ liq} = 3-7$, $D_{Re}^{mss/sul\ liq} = 3$,
1690 Brenan 2002; Ballhaus et al. 2006). Although sulfide inclusions in silicates of peridotites may
1691 have higher Ir and Os than sulfides on grain boundaries (e.g., Alard et al. 2000, Alard et al.
1692 2002), Re is depleted in the former and enriched in the latter, commonly accompanied by
1693 correlated Re/Os (Alard et al. 2005). Recently, it has been proposed that some harzburgites
1694 contain sulfides with high Se/Te ratios similar to what is expected from mss-liquid sulfide
1695 partitioning (König et al. 2014; 2015). However, because of the low concentrations of these
1696 elements, the mass balance of such phases in strongly depleted peridotites is difficult to constrain,
1697 and they may also reflect precipitation of sulfide from somewhat more fractionated magma with
1698 high Se/Te and Re/Os (Wang and Becker, 2015a). Work on Cu and Ag abundances in peridotites
1699 has shown that the relative behavior of these elements in bulk rock lherzolites is consistent with
1700 the systematics predicted by sulfide liquid-silicate partitioning but not with mss-liquid sulfide
1701 partitioning (Wang and Becker 2015b).

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1703 The differing $^{187}Os/^{188}Os$ of the two sulfide populations suggests that sulfides precipitated
1704 on grain boundaries during melt infiltration did not equilibrate with included sulfides, which is a
1705 basic requirement for equilibrium mss-sulfide liquid-silicate melt partitioning models. Thus, as
1706 shown before in the discussion of sulfide liquid-silicate melt partitioning, none of the proposed
1707 partitioning processes that are potentially relevant during partial melting yields a satisfactory
1708 quantitative description of the HSE composition of many mantle peridotites. Sulfide melt-silicate
1709 melt partitioning seems to be the best match for the observed HSE pattern in lherzolite bulk
1710 rocks. However, at least for Pd, Au, Re and S, their ratios in lherzolites may be mostly inherited
1711 from the melts that infiltrated depleted precursor rocks (e.g., harzburgites; Fig. 20). The origin of
1712 the HSE fractionation in the infiltrating melts and their sulfide liquids will be discussed below.

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1714 ***HSE fractionation during the formation of mantle pyroxenites.*** Mantle pyroxenites are
1715 important because they represent products of magmatic fractionation in the mantle and thus yield
1716 information on the composition of relatively ‘primitive’ magmas (Bodinier and Godard 2003).
1717 Pyroxenites are cumulates that formed by reactive infiltration and fractional crystallization of
1718 primitive to more evolved basic magmas. Websterites (‘Cr diopside suite’) and orthopyroxenites
1719 sometimes display mineralogically zoned reaction domains with peridotites, which have formed
1720 due to melt infiltration into the surrounding peridotite (e.g., Becker et al., 2004; Bodinier et al.,
1721 1987, 2008). Quite often, clinopyroxenites (‘Al augite suite’) appear to have formed from more
1722 evolved compositions and the absence of reaction zones may indicate their formation at
1723 shallower levels (e.g., Sinigoi et al., 1983; Suen and Frey, 1987).

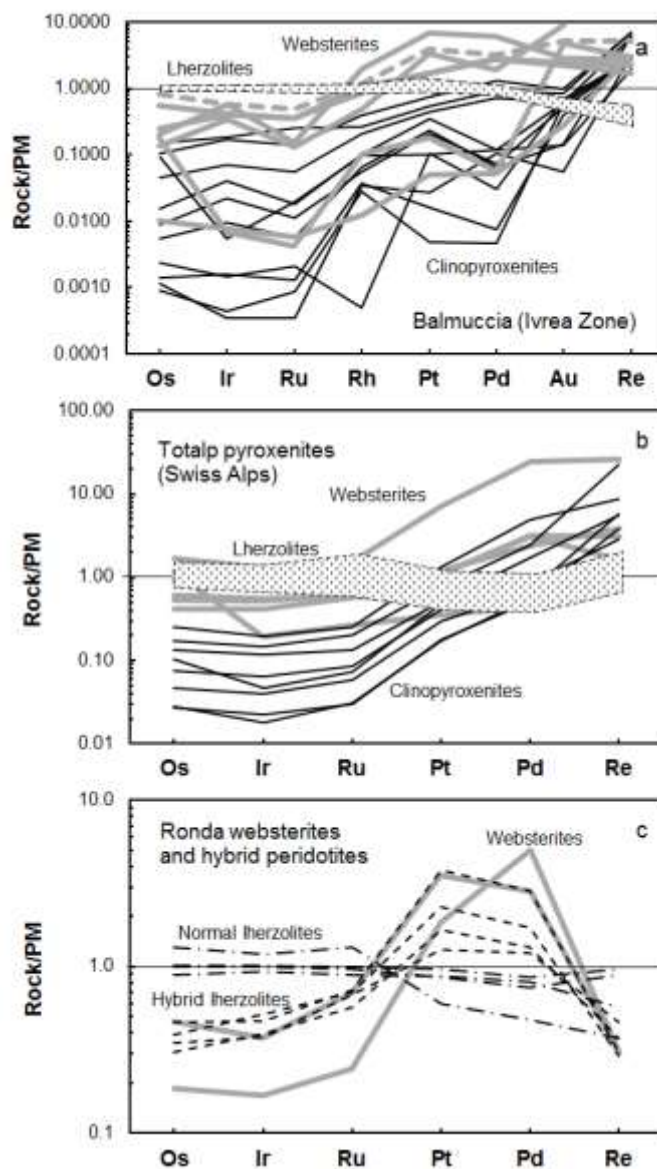
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1725 Only limited data are available for HSE abundances and Os isotopic compositions in
1726 mantle pyroxenites from tectonites, including pyroxenites from Ronda (Reisberg et al. 1991,
1727 Reisberg and Lorand, 1995; Marchesi et al. 2014), Beni Bousera (Kumar et al. 1996, Pearson and
1728 Nowell 2004; Luguët et al. 2008b), Lower Austria (Becker et al. 2001, 2004), Troodos (Büchl et
1729 al. 2002), Totalp (van Acken et al. 2008, van Acken et al. 2010b), Hori Bory (Ackerman et al.
1730 2013) and Balmuccia (Wang and Becker 2015c). The HSE patterns of pyroxenites in mantle
1731 tectonites are broadly similar to data from sulfides in pyroxenite xenoliths. In general, the relative
1732 fractionation of the HSE is similar to that in basalts, but with higher concentrations of Os, Ir, Ru,

1732 Rh, Pt and Pd than in MORB. Websterites and orthopyroxenites often display HSE patterns that
 1733 are less strongly fractionated than clinopyroxenites (Fig. 21).
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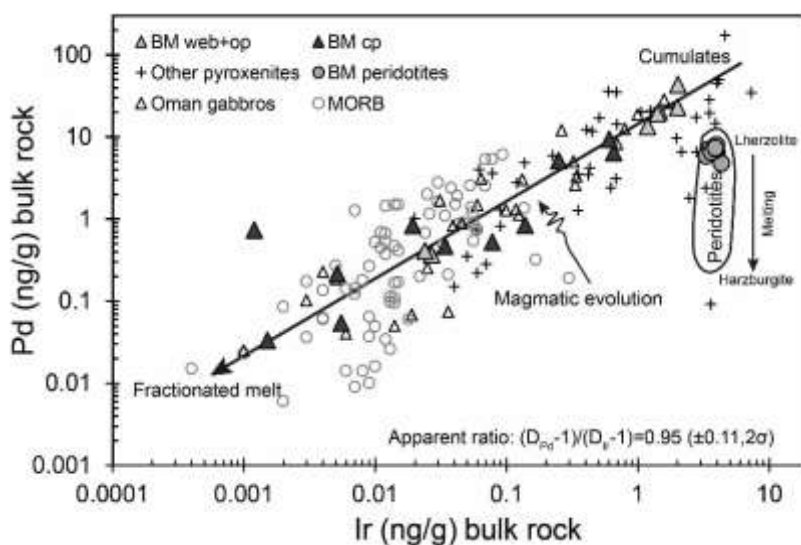
1735 Concentrations of S and Re in pyroxenites are similar or lower than in MORB, but often
 1736 higher than in lherzolites. Abundances of other HSE in pyroxenites are similar or lower than in
 1737 lherzolites (Fig. 21). Some pyroxenites display a depletion of Re relative to Pd, which may have
 1738 been caused by multi-stage melting (Marchesi et al. 2014). The occurrence of



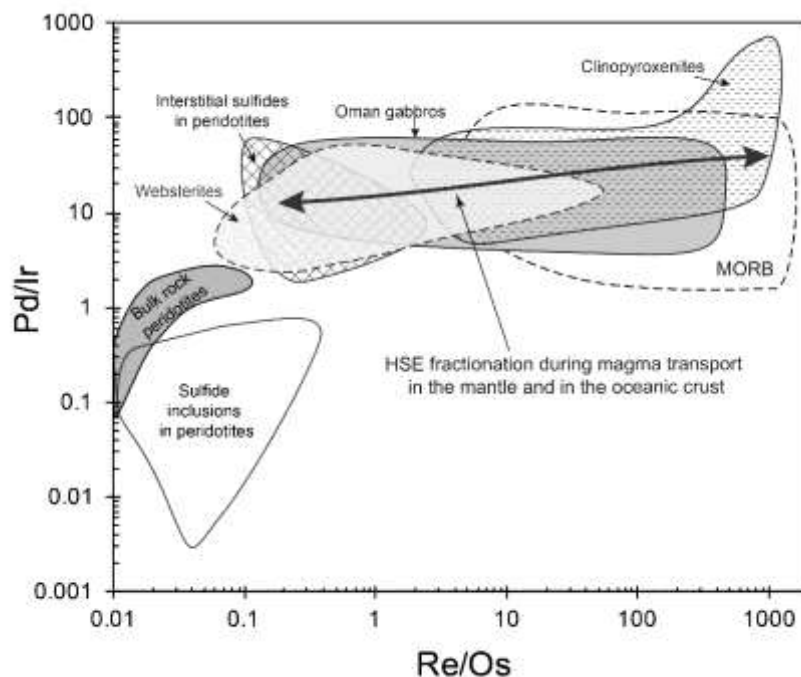
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 1741 **Figure 21.** Primitive mantle-normalized concentration diagrams of mantle pyroxenites from peridotite
 1742 massifs. Websterites: gray lines, clinopyroxenites: black lines. a) Balmuccia (Wang and Becker 2015c)
 1743 Balmuccia lherzolites from Wang et al. (2013). b) Totalp (van Acken et al. 2010b) Totalp lherzolites from
 1744 van Acken et al. (2010a). c) Ronda (Marchesi et al. 2014): Hybrid lherzolites (dashed lines) were also
 1745 affected by reactive infiltration of magma, but differ in composition from the pyroxenites and normal
 1746 lherzolites. Typical Ronda lherzolites (dash-dotted lines) from Fischer-Gödde et al. (2011).
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1749 centimeter-scale Os isotopic heterogeneity between alternating pyroxenite-peridotite layers
 1750 (Becker et al. 2001, 2004; Büchl et al. 2002; van Acken et al. 2008) is another indication of the

1751 difficulty of small-scale Os isotopic equilibration between silicate melt and existing sulfide
 1752 populations. A study of a zoned clinopyroxenite-websterite-orthopyroxenite rock from Lower
 1753 Austria that represents a former reaction zone between high-temperature silicate melt and
 1754 peridotite has shown that Sr and Nd isotopic compositions were equilibrated across a 10 cm
 1755 distance of the rock at the time of its formation (Becker et al. 2004). In contrast, both γ_{Os_i} and Os
 1756 concentrations display strong gradients over the same distance, indicating disequilibrium. HSE
 1757 compositions of sulfides in single thin sections of Totalp pyroxenites vary from those with Ru/Ir,
 1758 Pd/Ir and Re/Ir similar to peridotitic sulfides, to those with high ratios of these elements, typical
 1759 of melt compositions (van Acken et al. 2010b). The detailed processes that resulted in the close
 1760 association of these different sulfide populations are not yet clear, but they suggest that
 1761 disequilibrium among sulfides may be common in mantle pyroxenites as well as peridotites.
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1768 **Figure 22.** a) Pd-Ir diagram of bulk rock concentrations in mantle peridotites, pyroxenites (BM =
1769 Balmuccia, Ivrea Zone), MORB and gabbros of the oceanic crust. Web: websterite, op: orthopyroxenite,
1770 cp: clinopyroxenite. The correlation suggests that, with the exception of a few gabbros and MORB, mantle-
1771 derived magmatic rocks define a continuum between melt compositions and pyroxenites ('cumulates').
1772 Most magmatic products are offset from the peridotite data, indicative of disequilibrium between magmas
1773 and bulk peridotite. b) Pd/Ir-Re/Os diagram showing the limited range of fractionation of Pd/Ir in
1774 magmatic products compared to Re/Os. The Pd/Ir data are consistent with similar bulk distribution
1775 coefficients of these elements during magmatic processing in mantle and crust (see a). Most magmatic
1776 rocks shown in (b) define fields that overlap with or lie along the extension of grain boundary sulfides from
1777 peridotites, indicating a common origin of grain boundary sulfides and mantle-derived igneous rocks. Both
1778 diagrams are modified from Wang and Becker (2015c). Data sources: Oman gabbros, Peucker-Ehrenbrink
1779 et al. (2012); MORB: Hertogen et al. (1980), Rehkämper et al. (1999b), Bezos et al. (2005), Lissner et al.
1780 (2014); BM pyroxenites and peridotites, Wang et al. (2013, 2015c); other pyroxenites are from Totalp, van
1781 Acken et al. (2008, 2010b); Beni Bousera, Luguët et al. (2008); Ronda, Marchesi et al (2014); Horní Bory
1782 (Ackerman et al. 2013); Dramala massif, Pindos ophiolite (Sergeev et al., 2014); interstitial sulfides and
1783 sulfide inclusions in peridotites, Alard et al. (2005), Harvey et al. (2006).

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1786 A comparison of Re/Os and Pd/Ir in pyroxenites with data on ocean ridge basalts and
1787 gabbros from the lower oceanic crust indicates considerable overlap (Fig. 22). This observation
1788 suggests that significant fractionation of HSE ratios in magmas already occurs by precipitation of
1789 sulfide liquid during magmatic transport and reaction in the mantle (Wang and Becker 2015c). In
1790 contrast to Re/Os, which shows large variations in magmatic products over several orders of
1791 magnitude, the variation of Pd/Ir in the latter is much more limited and Pd and Ir show similar
1792 bulk partitioning behavior. Because of the segregation of sulfide liquid from magmas during
1793 magmatic transport in the mantle, the HSE compositions of basaltic magmas may preserve little
1794 direct information on HSE concentrations of deeper parts of the melting region. Figure 22a also
1795 shows that the data fields defined by most magmatic products, particularly the basalts, are offset
1796 from the bulk compositions of peridotites, but overlap with ratios in grain boundary sulfides from
1797 peridotites. A similar observation was made for variations of Se/Te (Wang and Becker 2015c).
1798 This observation may provide the best indication so far that most magmas that contribute to the
1799 oceanic crust did not fully equilibrate with the bulk rock of mantle peridotite residues.

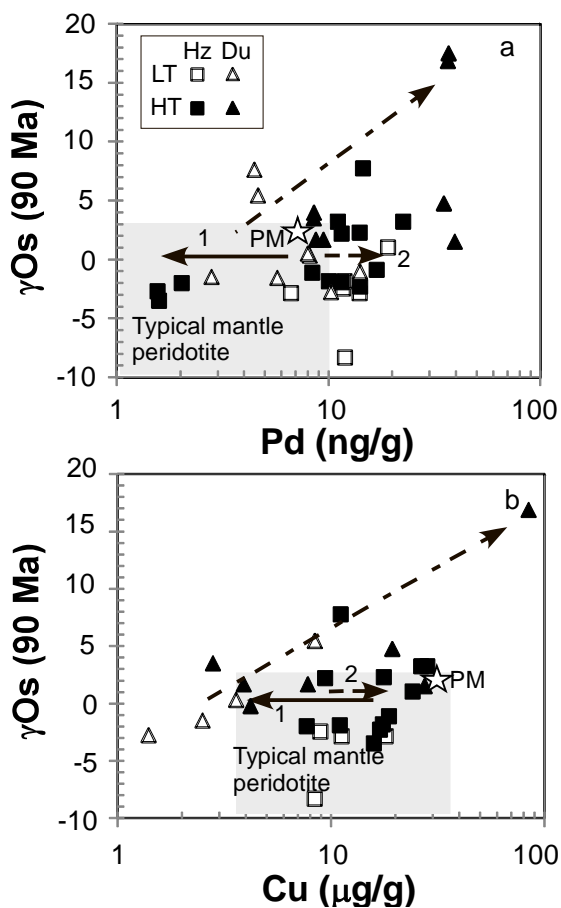
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1802 ***HSE fractionation during the formation of harzburgites and replacive dunites.*** Data
1803 on HSE and other chalcophile elements in harzburgites show that many of these rocks have high
1804 abundances of IPGE and lower abundances of Rh, Pt and Pd (e.g., Pearson et al. 2004; Becker et
1805 al. 2006; Luguët et al. 2007). These IPGE-PPGE fractionations are generally consistent with
1806 fractionation of melting residues at moderate to high (15 to 30 %) degrees of partial melting
1807 (Mungall and Brenan 2014; Brenan et al., 2015, this volume, and references therein). The
1808 incongruent breakdown of liquid or solid sulfide occurs at advanced degrees of melting at low fS_2
1809 and may play an important role in the stabilization of Os-Ir-Ru and Pt-Ir alloy phases that have
1810 been found in such rocks (Lorand et al. 1999; Luguët et al. 2007; Lorand et al. 2010; Fonseca et
1811 al. 2012; Mungall and Brenan 2014; Brenan et al. 2015). With progressive melting in the absence
1812 of a Fe-Ni-rich sulfide phase, all Re, Au and Pd should be dissolved in coexisting melts, provided
1813 that residues and melts were equilibrated. The abundances of Os, Ir, Ru, Rh and Pt, and their
1814 fractionation in harzburgite residues (e.g., Fig. 20) should be controlled by the solubility of these
1815 elements in sulfur-bearing silicate melts and the stability of Os-Ir, Ru-Os and Pt-Ir phases
(Mungall and Brenan 2014).

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1818 However, harzburgites may show variations in HSE abundances that are not entirely
1819 consistent with a simple melting history as envisioned before. Normalized abundances of Re and
1820 S in harzburgites are sometimes higher than normalized abundances of Pd (Figs. 5, 10). These
1821 patterns have been interpreted either in terms of precipitation of secondary sulfides from
1822 infiltrating melts with high Re/Os and fractionated HSE patterns (Chesley et al. 1999, Pearson et
al. 2004, Becker et al. 2006; Wang and Becker 2015a). Alternatively, enrichments of Re and S

1823 compared to Pd and Pt (and of Se relative to Te) in some harzburgites have been interpreted to
 1824 reflect the presence of mss of residual origin (König et al. 2014). The former explanation is
 1825 consistent with magmatic re-enrichment processes of incompatible elements (e.g., light rare earth
 1826 elements) in some of these rocks. Some harzburgites display lower abundances of IPGE than
 1827 expected for depleted mantle peridotite, e.g., < 3 ng/g Ir, instead of 4 to 5 ng/g expected for
 1828 residues of moderate to high degrees of melting (Figs. 5, 10). In order to understand this
 1829 behavior, it is useful to recall that even at high temperatures most peridotites likely contain
 1830 unequilibrated sulfide melt (maybe also mss), with a range of HSE concentrations. Complete
 1831 dissolution of some of these sulfide droplets (but not others) into sulfur-undersaturated melt,
 1832 without concurrent precipitation of IPGE alloy phases, will result in a net decrease of the
 1833 abundances of all HSE. This process almost certainly plays an important role in the formation of
 1834 some replacive dunites and associated harzburgite-lherzolite-pyroxenite rock assemblages
 1835 (Becker et al. 2001, 2004, Büchl et al. 2002, 2004, Hanghøj et al. 2010, Wang et al. 2013). For
 1836 instance, the variable IPGE abundances and strong depletions of Pt, Pd, Re and other chalcophile
 1837 elements in discordant dunite bodies in lherzolites at Balmuccia indicate that the magmas were
 1838 undersaturated in sulfur, which caused the dissolution of sulfides from the lherzolitic protoliths of
 1839 the dunites (Fig. 5, Wang et al. 2013).
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 1843 **Figure 23.** The enrichment of chalcophile elements in harzburgites and dunites from Wadi Tayin (Oman
 1844 ophiolite, Hanghøj et al. (2010). a) γ_{Os_i} -Pd diagram shows that in most harzburgites and dunites Pd is
 1845 enriched in comparison to typical mantle peridotites. b) γ_{Os_i} -Cu diagram indicates that Cu in dunites
 1846 loosely correlates with γ_{Os_i} . In general, Cu is less enriched than Pd. Open symbols are low-temperature
 1847 rocks, filled symbols high-temperature rocks (see Fig. 9). Arrow 1 indicates the expected depletion
 1848 behavior due to melting, 2, redistribution of Pd due to dissolution and precipitation of sulfides and the
 1849 dash-dotted arrow indicates correlated changes in γ_{Os_i} , and Cu concentrations resulting from melts with

1850 suprachondritic Os isotopic composition. For Pd this correlation breaks down, presumably because of local
1851 sulfide segregation from coexisting magma.

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1854 The harzburgites from Wadi Tayin (Oman ophiolite) display normal abundances of IPGE
1855 and tend to show primitive mantle-like or even slightly suprachondritic abundances of Pt, Pd and
1856 Re (Lorand et al. 2009; Hanghøj et al. 2010). Some of the harzburgites show selective
1857 enrichments of Pt that also have been noted from abyssal peridotites and other ophiolites (Fig.
1858 10) and peridotite massifs (Fig. 5). The Pt enrichments may indicate the precipitation of Pt-
1859 enriched sulfide liquid from silicate melt that may have dissolved Pt from destabilized Pt-Ir
1860 alloys at high degrees of melting. Dunites from Wadi Tayin are similarly enriched in HSE, but
1861 show more fractionated Re/Os and PPGE/IPGE ratios. Because the dunites are thought to reflect
1862 pathways of olivine-saturated magmas, the enrichments of Pt, Pd and Re in dunites and
1863 harzburgites likely reflect sulfide segregation from magmas enriched in these elements (Fig. 23).
1864 Although this process appears to have occurred pervasively, the initial $^{187}\text{Os}/^{188}\text{Os}$ (at around 90-
1865 95 Ma) in the mantle section at Wadi Tayin were not equilibrated (Fig. 23). The high abundances
1866 of Pt, Pd and Re in otherwise incompatible element depleted mantle rocks suggest that sulfide
1867 saturation may play an important role in the uppermost mantle underneath fast-spreading ocean
1868 ridges. Dunites from the Troodos ophiolite also display 'melt-like' HSE compositions (Büchl et
1869 al. 2002). A common property of dunites is that their initial $^{187}\text{Os}/^{188}\text{Os}$ extends to
1870 suprachondritic values (γOs_i ranging from -3 to +17, e.g., Fig. 23 and Becker et al. 2001),
1871 suggesting that some of the parent magmas had suprachondritic Os isotopic compositions.
1872 However, as the case of the dunites from Balmuccia shows, not all dunites are characterized by
1873 an enrichment of Pt, Pd and Re and melt like HSE patterns.

1874

1875 PGE enrichments also occur in podiform chromitites, which are magmatic precipitates
1876 associated with dunites and harzburgites in ophiolites that formed in the proximity of convergent
1877 plate margins. Because chromitites may represent economically relevant sources of PGE, these
1878 high-temperature magmatic ore deposits will be discussed in Barnes and Ripley (2015, this
1879 volume).

1880

1881

1882 **Summary – Mantle melting and mantle-magma interaction – different sides of the same**
1883 **coin**

1884

1885 Models of partial melting of mantle tectonites must consider the natural open-system
1886 behavior relevant for melting column models, diapiric upwelling of partially molten mantle or
1887 conversion of lithospheric mantle to asthenosphere by melt infiltration (as was suggested to have
1888 occurred in the magmatic history of some mantle tectonites, e.g., Müntener et al. 2005). Thus,
1889 melt infiltration and melting should occur more or less simultaneously, provided that porous flow
1890 permits melt infiltration. The composition of the residues will change with time until external
1891 processes cause upwelling and melting to stop and the mantle to cool. The HSE concentration
1892 and $^{187}\text{Os}/^{188}\text{Os}$ data on mantle tectonites with well-constrained ages (e.g., Oman ophiolite) show
1893 that the extent of sulfide-silicate equilibrium in these melting processes must be limited. Several
1894 different types of sulfide (presumably mostly liquids, but also mss and other solid phases at lower
1895 temperatures) may exist at high temperatures in peridotite (see also Lorand and Lugué 2015, this
1896 volume). Residual sulfides with subchondritic $^{187}\text{Os}/^{188}\text{Os}$ occur as inclusions in silicates and are
1897 inherited from ancient melting processes. These sulfides may represent residual sulfide liquids or
1898 mss, or both. Sulfide liquids with chondritic to suprachondritic $^{187}\text{Os}/^{188}\text{Os}$ and higher Re/Os and
1899 Pd/Ir are precipitated from infiltrating silicate melt and mostly reflect the composition of these
1900 melts with variable reaction with peridotite. Hybrid sulfide liquids may form locally where
1901 magmas and peridotite react and magmas became oversaturated in sulfur. In addition, relic PGM
1902 phases such as Pt-Ir alloys inherited from depleted protoliths may survive these magmatic

1903 processes. An important aspect of melt infiltration in the lherzolite stability field is the co-
1904 precipitation of sulfides with pyroxene \pm Al phase assemblages. Only such a process can explain
1905 correlations of Re, Re/Os and sulfur concentrations with fertility indicators such as Al₂O₃. As it is
1906 likely that the same processes were also responsible for the correlations between ¹⁸⁷Os/¹⁸⁸Os and
1907 Al₂O₃ in many suites of mantle peridotites, the mass balance with inherited Re-depleted sulfides
1908 suggests that the infiltrating melts had suprachondritic ¹⁸⁷Os/¹⁸⁸Os (the origin of such melts will
1909 be discussed later). This notion is supported by Os isotopic measurements on grain boundary
1910 sulfides in peridotites and by initial Os isotopic compositions of most mantle pyroxenites (Alard
1911 et al., 2002; Alard et al., 2005; Harvey et al., 2010, 2011; Harvey et al., 2015, this volume; Wang
1912 and Becker, 2015).

1913
1914 Different modeling approaches, both complicated and simple may produce appropriate
1915 HSE compositions of basalts from model mantle compositions (e.g., Rehkämper et al. 1999;
1916 Bezos et al. 2005, Harvey et al. 2011, Mungall and Brenan 2014). As discussed here and
1917 elsewhere (e.g., Lorand et al. 1999, Pearson et al. 2004, Lorand and Alard 2010, Fischer-Gödde
1918 et al. 2011, König et al. 2014, Wang and Becker 2015a), models that employ equilibrium
1919 distribution of the HSE between mantle phases have difficulties in accounting for some of the
1920 detailed compositional variations of the compatible HSE in bulk peridotites. Studies of HSE in
1921 bulk rocks of mantle peridotites and pyroxenites and their trace phases indicate that in high
1922 temperature magmatic processes in the mantle, disequilibrium between different HSE host phases
1923 and silicates may be the rule (e.g., Burton et al. 1999, Alard et al. 2000, 2002, 2005). In spite of
1924 these complexities, a useful assessment of the bulk distribution behavior of the HSE is possible
1925 and their relative behavior is consistent with abundance data in komatiites and basalts. The data
1926 on bulk rocks and sulfides of mantle pyroxenites and sulfides from grain boundaries in peridotite
1927 tectonites and in xenoliths indicate that infiltrating melts show relative fractionation of the HSE
1928 and S similar to the fractionation pattern of basalts, with mantle normalized abundances of S \approx
1929 Re > Au > Pd > Pt \geq Rh > Ru > Ir \geq Os. The HSE data on peridotites and pyroxenites suggest
1930 that the composition of infiltrating melts also affects the composition of peridotites (e.g., Fig. 5,
1931 7, 20). Notably, enrichments and depletions of Re in peridotites may be caused by precipitation
1932 of sulfides with suprachondritic Re/Os. If the abundances of Re, Au, Pd, Pt and other chalcophile
1933 elements in mantle peridotites are predominantly controlled by sulfide segregation from primitive
1934 basic magma, the question arises, which partition process produced the relative fractionation
1935 among these elements in these magmas to begin with? The answer may lie in the increasing
1936 importance of alloy solubility in silicate melt during moderate to high degrees of melting in the
1937 shallow mantle, near or beyond the exhaustion of sulfide in the residues. At these conditions, the
1938 concentrations of the HSE in silicate melts may be controlled by residual PGE alloys, the
1939 different solubility of Pt, Rh, Ru, Ir and Os and possibly silicate mineral-oxide-melt partitioning
1940 (Mungall and Brenan 2014; Brenan et al. 2015, this volume). Thus, basic melt infiltrating the
1941 asthenosphere and lithosphere at greater depth likely carries the HSE and ¹⁸⁷Os/¹⁸⁸Os signature of
1942 oceanic crust produced in previous Wilson cycles. This conclusion is consistent with
1943 suprachondritic initial ¹⁸⁷Os/¹⁸⁸Os of mantle pyroxenites and some peridotites that were affected
1944 by melt infiltration and coexisting harzburgites with subchondritic ¹⁸⁷Os/¹⁸⁸Os, which may
1945 represent ancient remnants of shallow oceanic mantle.

1946
1947

1948 **Os isotopic heterogeneity in the mantle**

1949

1950 The compatibility of Os during partial mantle melting, and the existence of two
1951 radioactive decay systems producing isotopes of Os, makes it an ideal element with which to
1952 investigate mantle heterogeneity (Hart and Ravizza, 1996; Burton et al., 1999). The relative
1953 compatibility of Os and Re is primarily controlled by their differing preference for sulfide over
1954 melt (See section above: Behaviour of HSE during partial melting). This produces strong
1955 fractionation of moderately incompatible Re from compatible Os during partial melting of the

1956 mantle, giving rise to very high Re/Os ratios in crust-forming melts (see Gannoun et al., 2015,
1957 this volume) and correspondingly low, sub-chondritic $^{187}\text{Os}/^{188}\text{Os}$ ratios in depleted mantle. In
1958 turn, crust recycled back into the mantle is potentially traceable due to its distinct Os isotope
1959 signature. Likewise, small degree melts within the mantle may also produce variations in Re/Os
1960 and thus, over time, in $^{187}\text{Os}/^{188}\text{Os}$. Due to the chalcophile affinity of Os, Re-Os isotope
1961 variations can provide different, yet complementary, information to lithophile isotope systems,
1962 and can sometimes display behavior that is decoupled from lithophiles (e.g., Class et al., 2009).

1963 The ^{190}Pt - ^{186}Os decay system, in contrast to the Re-Os system, does not typically produce
1964 resolvable differences in $^{186}\text{Os}/^{188}\text{Os}$ ratios in mantle rocks due to the much smaller decay
1965 constant compared to ^{187}Re , and due to the lower degree of fractionation between parent and
1966 daughter. Only in specific cases of high-degree melting do Pt concentrations significantly exceed
1967 those of the mantle, such as in some volcanic arc settings (Dale et al., 2012b) and in komatiites
1968 (e.g., Puchtel and Humayun, 2001; Fiorentini et al., 2011); but in the latter case Os in the melt
1969 approaches mantle concentrations and thus fractionation of Pt and Os remains limited. Recycled
1970 crust has only moderately high Pt/Os (Dale et al., 2009; Peucker-Ehrenbrink et al., 2012) which
1971 is not sufficient to produce anomalous compositions given the subsidiary Os concentrations of
1972 crust, relative to mantle. Nevertheless, ^{186}Os enrichments have been identified in some intraplate
1973 magmas (Brandon et al., 1998; 2003; Puchtel et al., 2005) and in a later section we briefly discuss
1974 whether mantle processes are a plausible mechanism by which to produce these enrichments.

1975

1976 In this section, we focus on broad-scale mantle heterogeneity, whereas disequilibrium on a
1977 hand specimen scale, or smaller, is covered in the previous section on ‘Os isotopic
1978 disequilibrium’.

1979

1980 **$^{187}\text{Os}/^{188}\text{Os}$ mantle composition and heterogeneity.** The bulk Os isotope composition of the
1981 silicate Earth was likely set by late accretion of material with a bulk primitive composition, after
1982 core formation had ceased (Kimura et al., 1974; Chou, 1978). However, neither the $^{187}\text{Os}/^{188}\text{Os}$
1983 composition (Meisel et al., 2001) nor the relative HSE abundances of PM estimates (Becker et
1984 al., 2006) match those of any known chondrite group. This difference has been reconciled by (i)
1985 late accretion of differentiated planetesimal core material and primitive chondritic material
1986 (Fischer-Gödde and Becker, 2012), (ii) by a hybrid model for the enrichment of Earth’s HSE
1987 involving late accretion to a fractionated mantle signature (which may be a residue from metal-
1988 silicate segregation, cf. Righter et al., 2008; Walker, 2009), or (iii) by mantle processes
1989 accounting for the combination of non-chondritic ratios involving Ru and Pd and chondritic
1990 ratios of other HSE in fertile Iherzolites (e.g., Lorand et al., 2010). See Day et al. (2015 this
1991 volume) for further discussion.

1992

1993 The processes of continental crust production and incomplete rehomogenisation of recycled
1994 oceanic crust have likely both served to reduce the $^{187}\text{Os}/^{188}\text{Os}$ of the peridotitic mantle below
1995 that of the primitive mantle. Thus, heterogeneous distribution of ^{187}Os in the mantle is due to the
1996 timing and degree of melt depletion and the presence of enriched domains, which may either be
1997 recycled surface materials or domains within the mantle fertilized by low-degree melts.

1998

1999 A compilation of $^{187}\text{Os}/^{188}\text{Os}$ data for global peridotites (excluding pyroxenites), grouped
2000 according to the tectonic settings used in this chapter and in this volume, is shown in Fig 24, and
2001 a summary of the averages and ranges for each setting/sample type is shown in Table 2. Cratonic
2002 and circum-cratonic xenoliths, which won’t be discussed further here, are both typically strongly
2003 unradiogenic, reflecting their severe and early melt depletion and subsequent isolation from the
2004 convecting mantle (see Aulbach et al. 2015, this volume, and references therein). All major
2005 tectonite and xenolith groups (continental/continent-ocean transitional tectonites, high-T
2006 convergent tectonites, ophiolites, abyssal peridotites, oceanic mantle xenoliths, sub-continental
2007 lithosphere xenoliths and sub-arc xenoliths) have a considerable ‘peak’ in probability of
2008 $^{187}\text{Os}/^{188}\text{Os}$ between 0.125 and 0.128, indicating a degree of effective large-scale homogenization

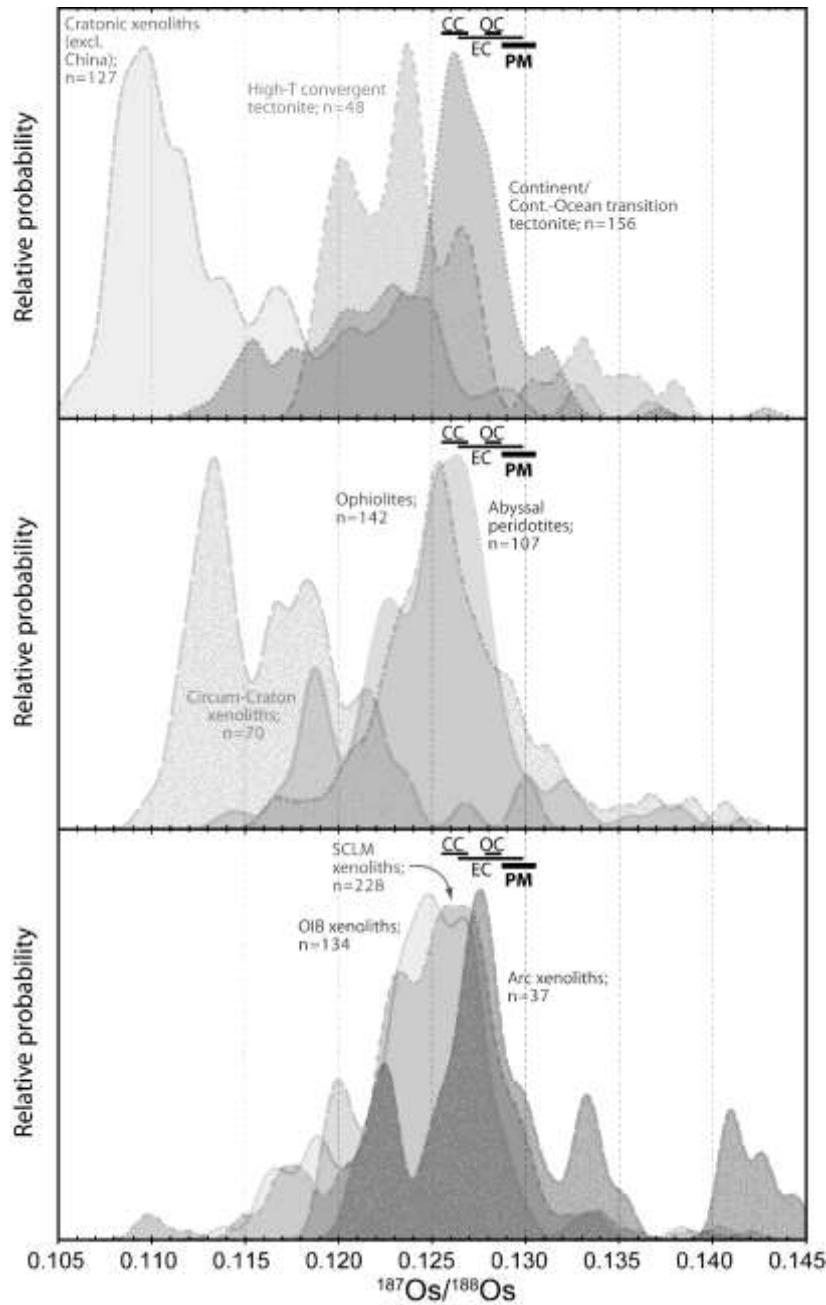
2009 in the convecting mantle and younger lithosphere, albeit incomplete. Moreover, most groups
 2010 have remarkably similar total ranges of $^{187}\text{Os}/^{188}\text{Os}$ (when excluding up to 3% of the most
 2011 extreme data), between 0.026 and 0.029 units, with the exception of high-T convergent margin
 2012 tectonites ($n = 48$) which have a range of 0.023, and sub-continental lithospheric mantle
 2013 xenoliths, with a larger range of 0.037 (although in this latter case the primary data may be
 2014 compromised by secondary processes such as weathering and reaction with host melts. Greater
 2015 than 85% of samples from each tectonic setting fall within a narrower range of $^{187}\text{Os}/^{188}\text{Os}$ of
 2016 around 0.015 units (the range of each group varies from 0.013 for all ophiolites, to 0.019 for
 2017 continental/continent-ocean transition tectonites).

2018 Table 2. Summary of compiled $^{187}\text{Os}/^{188}\text{Os}$ data for mantle tectonites, by setting and sample type
 2019
 2020
 2021

Sample type	Mean $^{187}\text{Os}/^{188}\text{Os}$	Mode $^{187}\text{Os}/^{188}\text{Os}$	Low	High	Main range (% included)	<i>n</i>
Abyssal peridotites	0.1243	0.1261	0.1139	0.1382	0.024 (100%)	107
Continent/cont-ocean transition	0.1255	0.1262	0.1126	0.1372	0.025 (97%)	156
High-T convergent margin	0.1259	0.1237	0.1184	0.1411	0.023 (100%)	48
Ophiolites (all*)	0.1271	0.1252	0.1162	0.1418	0.026 (97%)	142
Arc xenoliths	0.1315	0.1277	0.1206	0.1498	0.029 (97%)	37
OIB xenoliths	0.1244	0.1248	0.1138	0.1339	0.026 (99%)	134
Sub-continental xenoliths	0.1260	0.1257-67	0.1094	0.1464	0.037 (98%)	228

2022 * 2 Ga Finland ophiolite localities omitted due to long-term isolation from convecting mantle
 2023

2024 In detail, however, each grouping displays a variable distribution of Os isotope composition,
 2025 and the positions of the modal and mean $^{187}\text{Os}/^{188}\text{Os}$ compositions differ between many of the
 2026 groupings. One caveat here is that the data plotted on Fig. 24 are present-day measured
 2027 $^{187}\text{Os}/^{188}\text{Os}$ ratios, to reflect the current degree of overall mantle heterogeneity, and thus do not
 2028 account for any isolation of portions of lithosphere sampled in this dataset. If these portions were
 2029 exposed to gradual convective stirring then some of the ‘older’ depletion ages may have been
 2030 remixed with more radiogenic ambient mantle. Not all components of the compilation, therefore,
 2031 necessarily reflect the composition of the ‘convecting’ mantle.
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Figure 24. Probability density plots of present-day $^{187}\text{Os}/^{188}\text{Os}$ ratios in whole-rock samples grouped according to the tectonic settings discussed in this chapter: Ophiolites, abyssal peridotites, continent/continental-ocean transitional tectonite, high temperature convergent tectonite. Xenoliths from the subcontinental lithospheric mantle, oceanic lithosphere, cratonic lithosphere and circum-cratonic lithosphere are also shown (see Aulbach et al. 2015, this volume, and Luguet & Reisberg, 2015, this volume, for a discussion of HSE in these xenolith groups). Ranges for primitive mantle (Meisel et al., 2001) and major chondrite groups also shown; CC – carbonaceous, OC – ordinary, EC – enstatite (Walker et al., 2002a). A universal uncertainty of 0.00125 was applied to each datum to avoid bias towards more precise analyses and to provide sufficient smoothing for the smaller datasets, where used. For data sources see Fig. 14, except cratonic xenoliths: Walker et al. (1989); Pearson et al. (1995a); Pearson et al. (1995b); Shirey and Walker (1995); Chesley et al. (1999); Meisel et al. (2001); Pearson et al. (2004); Becker et al. (2006); Maier et al. (2012), and circum-craton xenoliths: Pearson et al. (2004); Luguet et al. (2009); Aulbach et al. (2014).

2050 All tectonite groups have ranges that extend to sub-chondritic and supra-chondritic
2051 $^{187}\text{Os}/^{188}\text{Os}$ ratios, although some extend broadly equally in each sense, while others have a
2052 pronounced skew towards less or more radiogenic values. For instance, the ophiolite record has a
2053 modal $^{187}\text{Os}/^{188}\text{Os}$ of ~ 0.1255 , with a broadly equal number of data extending in each sense down
2054 and up to values of 0.115 and 0.143, respectively (Fig. 24). At least half of the data fall between
2055 0.1225 and 0.128. In contrast, the dataset for continental/continent-ocean transitional tectonites
2056 shows a modal $^{187}\text{Os}/^{188}\text{Os}$ of ~ 0.126 , close to that of ophiolite ultramafics, but with a range
2057 extending down to 0.112 and up to 0.133, with a lower mean value than for ophiolites (Fig. 24).
2058 The abyssal peridotite samples of the convecting mantle show a remarkably similar probability
2059 profile to the continental/transitional tectonites, with a modal $^{187}\text{Os}/^{188}\text{Os}$ of ~ 0.126 , and a range
2060 from 0.1125 to 0.140; possibly with similar subsidiary peaks at 0.1225 and perhaps even at 0.115
2061 (although this most unradiogenic peak appears important for continental/transitional tectonites,
2062 but likely is not significant for abyssal peridotites, given the sample size).
2063

2064 The ‘tails’ to low and high $^{187}\text{Os}/^{188}\text{Os}$ reflect, respectively, ancient melt-depleted domains
2065 and enriched domains which have not fully re-homogenised with the rest of the convecting
2066 mantle through convecting stirring and potentially melt percolation and infiltration. The
2067 distribution of the data is further mentioned below in the context of platinum-group mineral
2068 studies. Qualitatively, at least, re-enrichment of ophiolitic mantle is supported by the observation
2069 that convergent margin ophiolites appear to have more radiogenic ^{187}Os than mid-ocean ridge
2070 ophiolites (Fig. 15), and by the absence of a skew to old depleted values in the overall ophiolite
2071 $^{187}\text{Os}/^{188}\text{Os}$ distribution (Fig. 24; cf. abyssal peridotite and ophiolite curves). The relatively
2072 radiogenic distribution of sub-arc xenoliths is also consistent with the process of re-enrichment in
2073 the subduction zone environment.
2074

2075 *The chromitite and PGM record of Os isotope mantle composition and heterogeneity.*

2076 Here, we focus only on the Os isotope evidence from PGM, rather than the systematics of PGM
2077 formation and composition (see O’Driscoll & Gonzáles-Jiménez 2015, this volume, for a
2078 comprehensive review). The utility of chromitites, and the PGM that they typically contain, is
2079 that they are Os-rich, Re-poor and tend to be largely robust to subsequent alteration processes
2080 caused by metamorphism and/or fluid-rock interaction. The very low Re/Os ratios mean that
2081 their $^{187}\text{Os}/^{188}\text{Os}$ isotope composition is almost ‘frozen in’ at the point of formation, or at worst
2082 require very small corrections for radiogenic ingrowth, even over periods of 3 Ga or greater
2083 (Malitch and Merkle, 2004). For these reasons, they have been used to estimate the Os
2084 composition of the convecting mantle, to assess mantle heterogeneity and to identify potential
2085 major mantle melting events through Earth’s history. One caveat to this use, however, is that
2086 chromitite formation occurs in zones of high melt flow, and these melts may have imparted a
2087 radiogenic $^{187}\text{Os}/^{188}\text{Os}$ signature on the chromitite, thus rendering it no longer entirely
2088 representative of the ‘average’ upper mantle (e.g., O’Driscoll et al., 2012; see also Convergent
2089 Ophiolite section above).
2090

2091 A global suite of ophiolitic chromites was used to provide an estimate of the average
2092 $^{187}\text{Os}/^{188}\text{Os}$ composition of the convecting mantle (Walker et al., 2002b). Linear regression of the
2093 isotope data relative to the age of the chromite provided an evolution curve with a present-day
2094 $^{187}\text{Os}/^{188}\text{Os}$ composition of 0.1281. Although the uncertainties overlap, this best estimate equates
2095 to approximately 5% less ingrowth of ^{187}Os over the life of the Earth when compared to the PM
2096 (0.1296; Meisel et al., 2001). This is presumably due to continental crust extraction and the
2097 presence of recycled oceanic crust in the mantle, which has not (yet) been efficiently
2098 rehomogenised. A study of over 700 detrital PGM from the Josephine Ophiolite, California,
2099 found a Gaussian distribution of $^{187}\text{Os}/^{188}\text{Os}$ ratios from 0.119 to 0.130 (Meibom et al., 2002).
2100 This was interpreted to represent long-term heterogeneity (melt-enriched and -depleted
2101 endmembers) which has been partially erased and homogenised by metasomatism and melt-rock
2102 reaction processes. Further work on a range of global ophiolites, however, indicated a more

2103 complex distribution of Os isotope ratios in Earth's mantle. Over 1000 detrital PGM from
2104 ophiolites in California, Urals, Tibet and Tasmania revealed a variety of $^{187}\text{Os}/^{188}\text{Os}$ distributions,
2105 from close to Gaussian to skewed towards old, unradiogenic values in the case of Urals, and a
2106 bimodal distribution for both Tibet and Tasmania (Pearson et al., 2007). It was proposed that the
2107 apparent 'peaks' in probability for certain $^{187}\text{Os}/^{188}\text{Os}$ ratios are consistent across different
2108 ophiolites and across other geological settings such as cratonic xenoliths, and that these peaks
2109 reflect global signatures produced by major global mantle melting episodes throughout Earth's
2110 history which match the implied crustal growth record from zircon ages. The composition of the
2111 major peak in $^{187}\text{Os}/^{188}\text{Os}$ for PGM is 0.1276 (Pearson et al., 2007; adjusted to present-day in
2112 Dale et al., 2009b), although the mean composition is likely significantly lower because of the
2113 skewed distribution to less radiogenic values. Perhaps notably, when considering representative
2114 analyses of convecting mantle composition, this upper limit of $^{187}\text{Os}/^{188}\text{Os}$ composition from
2115 PGM analysis is less radiogenic than the average of analysed chromites (0.1281; Walker et al.,
2116 2002b), even though many of the PGM are also sourced from supra-subduction zone ophiolites
2117 and therefore may be subject to the same process of radiogenic Os addition. Also of note is the
2118 fact that ultramafics from most of the tectonic settings have 'peak' values that are slightly less
2119 radiogenic than the 'peak' value from PGM (see Fig. D7; $^{187}\text{Os}/^{188}\text{Os}$ ~0.1265, compared to
2120 0.1276).

2121

2122 In summary, although global compilations have inherent bias towards exposed and well-
2123 studied areas, all the larger datasets ($n > 100$) for mantle settings that have not been isolated for
2124 long periods (cf. cratons), have very similar modal $^{187}\text{Os}/^{188}\text{Os}$ compositions of between 0.125
2125 and 0.127, and mean compositions between 0.1243 and 0.1271. Such values equate to around 8
2126 to 18% less ingrowth of ^{187}Os over the life of the Earth than for PM evolution (cf. Meisel et al.,
2127 2001), presumably largely due to crustal extraction and long-term isolation – although the exact
2128 degree of mantle Re depletion is dependent on the timing of this extraction. These values are
2129 somewhat higher than the 5% estimated from chromitites (see above, cf. Walker et al. 2002b),
2130 but some of this discrepancy is due to the omission of pyroxenites and other enriched lithologies
2131 from this data compilation. The small variance in the isotopic ranges for each setting appears
2132 noteworthy in terms of gauging mantle mixing efficiency, but is beyond the scope of this review.

2133

2134 ***$^{186}\text{Os}/^{188}\text{Os}$ mantle composition and heterogeneity.*** Platinum-group minerals and
2135 chromitites have been used as recorders of the $^{186}\text{Os}/^{188}\text{Os}$ evolution of the mantle. Many PGM
2136 are IPGE-rich and have low Pt/Os and hence faithfully record the $^{186}\text{Os}/^{188}\text{Os}$ of the mantle at the
2137 time when those PGM formed. Brandon et al. (2006) used Os-rich PGM data, together with
2138 chondrite analyses, to constrain the terrestrial evolution of $^{186}\text{Os}/^{188}\text{Os}$ from an initial of
2139 ~0.1198269 +/- 0.0000014 (2 sigma) at 4.567 Ga to a present-day value of 0.1198382 +/-
2140 0.0000028.

2141

2142 The potential for large-scale heterogeneity generated by the ^{190}Pt - ^{186}Os system is far smaller
2143 than that of the ^{187}Re - ^{187}Os system, and in most cases is beyond what is distinguishable given
2144 current analytical capabilities. Nevertheless, anomalously radiogenic $^{186}\text{Os}/^{188}\text{Os}$ ratios have
2145 been found in some high-degree melts in intraplate settings in Hawaii, Gorgona Island and
2146 Kostomuksha, Russia (Brandon et al., 1998; Brandon et al., 2003; Puchtel et al., 2005), coupled
2147 with only limited ^{187}Os enrichment. Possible mechanisms to generate such signatures are
2148 discussed below.

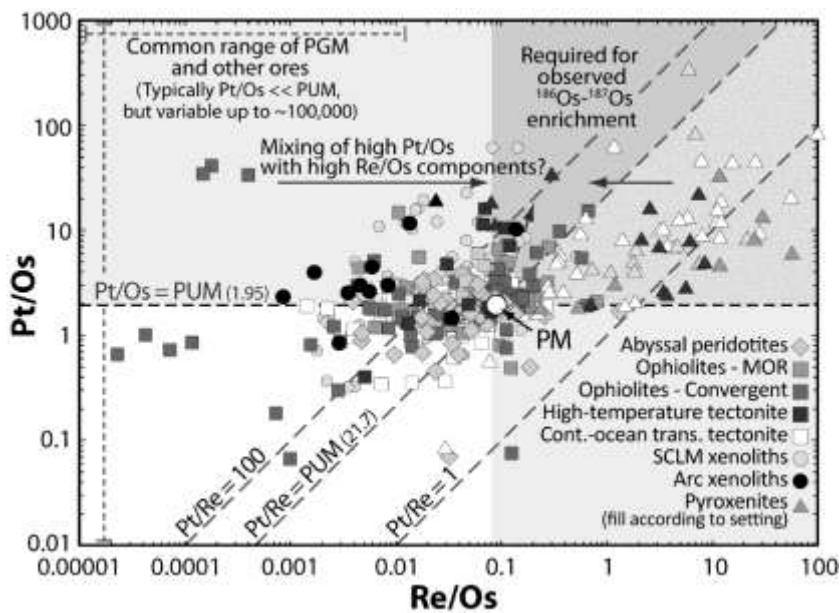
2149

2150 The range of Pt/Os ratios found in the supra-subduction zone environment indicates that
2151 there must be huge ^{186}Os variations on a lithological and mineral scale, if those materials were
2152 isolated. Alaskan-Uralian complexes (see Johan, 2002 for details) also display a large range of
2153 Pt/Os ratios, but these are beyond the scope of this chapter. Chromitites from ophiolites typically
2154 possess very low Pt/Os ratios (~0.1, compared with 1.95 for the PM), but can sometimes have
2155 Pt/Os of >10 (see Ophiolite sections). Platinum group minerals from within chromitites and

2156 other PGE-saturated ores can have even more extreme Pt/Os; laurites (Ru (Os, Ir)₂), may have
 2157 ratios of <0.01 (González-Jiménez et al., 2009) while PtFe alloys can have Pt/Os of >100,000
 2158 (Walker et al., 1997). Extremely high Pt/Os ratios, such as those of the Meratus Ophiolite,
 2159 Borneo (up to 2000), evolve to much higher ¹⁸⁶Os/¹⁸⁸Os compositions than those of the bulk
 2160 mantle, and because PGM are largely robust to subsequent processes, they may show
 2161 isochronous behavior and can be used to date ophiolitic complexes (Coggon et al., 2011). These
 2162 PGM, after ingrowth over as little as 200 Ma, have ¹⁸⁶Os/¹⁸⁸Os ratios that range from a slightly
 2163 sub-PM value of 0.119801 to 0.120315. As a guide to the magnitude of this difference, it is at
 2164 least 30 times greater than the difference between the bulk mantle and the highest ¹⁸⁶Os/¹⁸⁸Os
 2165 mantle melt yet discovered (0.000015; Brandon et al., 1999). These data will be discussed
 2166 further in the subsequent section on the production of HSE-Os signatures in mantle melts.

2167
 2168 A recent study of Eoarchaeon chromitites from south-west Greenland found ¹⁸⁶Os/¹⁸⁸Os data
 2169 proposed to reflect mantle melt depletion events in Earth's earliest history, during the Hadean at
 2170 approximately 4.1 Ga and possibly as old as 4.36 Ga (Coggon et al., 2013). In so doing, Coggon
 2171 et al. (2013) also inferred that the late veneer must have occurred prior to this time, consistent
 2172 with the message of an 'early' late veneer from studies of basaltic meteorites from different
 2173 parent bodies (Dale et al., 2012a).

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2178 **Figure 25.** Pt/Os vs. Re/Os for mantle rocks and the implications for the generation of ¹⁸⁶Os-¹⁸⁷Os
 2179 enrichments. The dark shaded field denotes the sense of relative fractionation of the Pt-Os and Re-Os
 2180 systems. The actual required Pt/Os ratio to explain the ¹⁸⁶Os enrichment in some intraplate magmas is ~10
 2181 or greater, over an ingrowth time of 1.5 Ga. This is obviously dependent on the age and on whether the
 2182 high Pt/Os component is only part of a composite source (which would require higher Pt/Os ratios or an
 2183 older age). Rocks possessing the required high Pt/Os, but only mildly elevated Re/Os are only a minor
 2184 component of the current mantle database. Data sources as in Fig. 14.

2185
 2186

2187 **The role of recycled oceanic lithosphere in producing HSE and Os isotope signatures in**
 2188 **magmas**

2189

2190 At least part of the compositional variability observed in mantle melts at Earth's surface is
 2191 derived from heterogeneity in the mantle. The biggest single process by which such
 2192 heterogeneous chemistry is generated must be that of recycling of oceanic lithosphere through

2193 subduction (e.g., Hofmann and White, 1982). In addition, there are other processes, such as melt
2194 percolation within the mantle and lithosphere (e.g., Halliday et al., 1995) that potentially play an
2195 important role in producing the variety of magma compositions that we observe at Earth's
2196 surface. Many instances of melt percolation may ultimately be sourced from enriched recycled
2197 material, but this is not a requirement in producing variations in fertility in the mantle. Here, we
2198 focus on the composition of recycled ultramafic and mafic lithosphere within the mantle, and its
2199 impact within the source regions of oceanic magmas.

2200

2201 ***Oceanic alteration.*** Prior to subduction, the oceanic lithosphere gains variable amounts of
2202 water and trace elements during seawater interaction or hydrothermal alteration, resulting in the
2203 formation of serpentine minerals, at the expense of olivine. This alteration can, in more extreme
2204 cases, be accompanied by elevated $^{187}\text{Os}/^{188}\text{Os}$ and the loss of Os relative to the other IPGE (see
2205 abyssal peridotite section), but typically, abyssal peridotites retain mantle-like HSE proportions
2206 and $^{187}\text{Os}/^{188}\text{Os}$ ratios. Regardless of the precise HSE signature, serpentinisation permits water
2207 transport deep into subduction zones and beyond into the deep mantle. Together with the
2208 hydrous mafic crust, this provides fluxes of fluids from the downgoing slab into the mantle
2209 wedge at a range of depths, as well as retention of water beyond the supra-subduction setting.
2210 The potential for the slab to transport water beyond the zone of sub-arc melting is likely to be
2211 important for promoting small-degree hydrous melting in the mantle, which may have an impact
2212 on HSE through refertilisation processes.

2213

2214 ***The impact of subduction zone processes on HSE in convergent margin magmas and***
2215 ***recycled oceanic lithosphere.*** Fluxes into the mantle wedge produce two effects which have a
2216 bearing on HSE behavior and Os isotope composition. First, as discussed above, radiogenic Os
2217 may, in certain cases (Brandon et al., 1996; Becker et al., 2004), be transferred from the slab into
2218 the mantle wedge and then transferred by melts into arc crust and supra-subduction oceanic crust,
2219 sampled by ophiolites. Second, fluid addition will promote hydrous melting, allowing otherwise
2220 refractory mantle domains to partially melt and permitting melting of the mantle at temperatures
2221 below those of the normal geothermal regime.

2222

2223 The evidence for a radiogenic Os flux to arc magma sources is equivocal, due to the
2224 difficulty in knowing the precursor $^{187}\text{Os}/^{188}\text{Os}$ of the mantle source and other potential sources of
2225 radiogenic Os such as arc crust. Nevertheless, the ophiolite record provides a firmer basis for this
2226 contention. An additional HSE flux is the loss of Re from metabasic rocks during dehydration
2227 (~50-60%; Becker, 2000; Dale et al., 2007), and likely enrichment of Re in the mantle wedge
2228 (Sun et al., 2003a; Sun et al., 2003b). This flux could contribute, over time, to radiogenic ^{187}Os
2229 in the mantle wedge and also has implications for the composition of recycled crust which are
2230 outlined below. Other HSE may also be mobilised (McInnes et al., 1999; Kepezhinskis et al.,
2231 2002; Dale et al., 2009a), but whether the magnitude of flux is sufficient to produce a
2232 measureable effect in supra-subduction zone magmas is doubtful, given the relatively high
2233 concentrations of these elements in the mantle.

2234

2235 Melting of refractory domains increases the likelihood of sulfide exhaustion, which, under
2236 most circumstances, would reduce the compatibility of all HSE, resulting in less fractionated
2237 HSE patterns such as those seen in picrites and komatiites (e.g., Puchtel and Humayun, 2000). In
2238 the Tonga Arc, however, the relative proportions of the HSE are amongst the most fractionated
2239 for mantle melts (Dale et al., 2012b), with extreme Pt/Os approaching 15. This fractionation may
2240 be caused by increased HSE-rich phase stability during lower temperature hydrous melting (e.g.,
2241 laurite stable up to 1275°C; Brenan and Andrews, 2001) and/or the promotion of chromitite
2242 formation by interaction between hydrous melts and refractory mantle (Dale et al., 2012b).
2243 Chromitite formation during melt-rock reaction in the mantle is expected to fractionate HSE
2244 significantly, sequestering IPGE in PGM and producing a melt with high (Re+Au+PPGE)/IPGE
2245 (see ophiolite sections).

2246

2247 *The role of recycled lithosphere in producing HSE-Os signatures in convecting mantle*

2248 *melts.* Many previous attempts have been made to model the effects of recycling oceanic
2249 lithosphere, particularly the mafic crustal portion (e.g., Roy-Barman et al., 1996; Brandon et al.,
2250 1999; Becker, 2000; Brandon et al., 2007; Dale et al., 2009b; Day et al., 2009). While we
2251 recognize the importance of quantitatively assessing whether a particular process is possible or
2252 likely, given the numerous previous attempts and the dependency on the parameters chosen, here
2253 we direct the reader to those previous studies and we instead choose to focus on the record of
2254 pyroxenites in the mantle, as direct recorders of enriched, hybridized lithologies. Of course, it is
2255 important to bear in mind that the sampled pyroxenite database is still relatively small (62
2256 samples with HSE and/or Os isotope data collated in Fig. 25) and thus it is difficult to relate this
2257 to the mantle as a whole. That said, the processes identified are broadly applicable.

2258

2259 Both eclogitic and pyroxenitic enriched lithologies are present in the mantle. Eclogites
2260 represent unequivocal crustal materials, sampled as xenoliths in intraplate volcanic settings,
2261 which retain much of their crustal geochemical signature, albeit modified by subduction
2262 processing. The term ‘pyroxenite’ covers a complex array of lithologies and petrogenetic
2263 pathways that are beyond the scope of this chapter (see Lambart et al., 2013). In simple terms,
2264 pyroxenites are variably hybridized lithologies produced during reaction of peridotite with silica-
2265 saturated melt derived from an enriched lithology such as eclogite (or possibly also derived from
2266 small-degree melting of peridotite). Reaction with a silica-undersaturated, olivine-saturated melt
2267 would instead produce dunite, so depending on the exact mode of formation of particular dunites
2268 (some dunites might be cumulates), they may also carry an enriched Os signature, as seen in the
2269 ‘convergent margin ophiolite’ section. Unlike eclogites, pyroxenites form a significant part of
2270 mantle tectonites, constituting between 1 and 9% of the Beni Bousera mantle tectonite massif
2271 (Pearson and Nowell, 2004). These pyroxenites at Beni Bousera have been identified as having a
2272 recycled crust origin, on the basis of lithophile and stable isotopes. They typically have
2273 radiogenic $^{187}\text{Os}/^{188}\text{Os}$ ratios, even in samples that are Os-rich (>2 ng/g).

2274

2275 Pyroxenites and peridotites from the Totalp ultramafic massif, Swiss Alps, preserve a record
2276 of refertilisation of peridotites by both melt percolation from the pyroxenites and from
2277 mechanical stretching and thinning of websterite layers (van Acken et al., 2008). The
2278 pyroxenites are strongly enriched in ^{187}Os ($^{187}\text{Os}/^{188}\text{Os}$: 0.122 to 0.866; main range: 0.13-0.16)
2279 and in Re, whereas peridotites have a broadly chondritic average γOs value. It is noted, therefore,
2280 that refertilisation does not completely homogenise Os isotopes, at least not on a small scale, but
2281 isotopic differences are rapidly reduced due to assimilation of pyroxenite melt by peridotite.

2282

2283 A compilation of ultramafic mantle samples, in terms of Pt/Os and Re/Os ratios, is presented
2284 in Fig. 25. Pyroxenites form a distinct group at elevated Re/Os and Pt/Os ratio, relative to
2285 peridotites. The degree of this enrichment is, in itself, consistent with a partially pyroxenite
2286 source for some mantle melts with radiogenic Os over a period of ingrowth of 1 Ga or more.
2287 Actual measured $^{187}\text{Os}/^{188}\text{Os}$ for global pyroxenites, excluding the 10 highest and lowest values
2288 from a total of 94 samples, varies from 0.124 to 0.928. Obviously the ability for these
2289 pyroxenites to produce sufficiently radiogenic melts as part of a hybrid pyroxenite-peridotite
2290 mantle, depends on their Os contents. The Os concentrations also vary substantially, from 0.005
2291 ng/g to 4.6 ng/g, and this generally co-varies negatively with $^{187}\text{Os}/^{188}\text{Os}$ ratios. Thus, at some
2292 level, the effect of the pyroxenite in the mantle is self-limiting due to reduced Os content. As
2293 well as the strongly radiogenic signatures of the pyroxenites themselves, there is also evidence
2294 for radiogenic Os addition to peridotitic rocks (Becker et al. 2001; Büchl et al. 2002; van Acken
2295 et al., 2008; Marchesi et al., 2014), and this, combined with the radiogenic pyroxenites, will more
2296 easily produce radiogenic mantle melts.

2297

2298 One aspect of oceanic crust recycling that has commonly been overlooked is the
2299 geochemical distinction between the gabbroic and basaltic parts of the crust. This is now
2300 generally fully recognized for HSE – with gabbroic crust being, on average, significantly more
2301 Os- and Pt-rich and slightly poorer in Re than MORB – and this has been incorporated into
2302 models for crustal recycling (Peucker-Ehrenbrink and Jahn, 2001; Dale et al., 2007; Peucker-
2303 Ehrenbrink et al., 2012)

2304

2305 An alternative, but related, means by which recycled lithosphere may have an impact on
2306 the HSE composition of mantle melts is through the process of sulfide metasomatism. Sulfides
2307 with radiogenic ^{187}Os have been sampled in interstitial locations within peridotites (Alard et al.,
2308 2005; Harvey et al., 2006, 2010, 2011; Warren and Shirey, 2012). The ultimate source of those
2309 sulfides is unknown, but derivation from recycled crustal material, of at least some such sulfides,
2310 is plausible. Radiogenic, interstitial sulfides can then be readily incorporated into partial melts,
2311 whereas unradiogenic residual sulfides remain shielded from melt by the silicates that enclose
2312 them. The process of sulfide addition is a similar process to other forms of refertilisation, but in
2313 this case the lithophile and chalcophile element signatures may be decoupled. However, the
2314 overall broad coupling of $^{187}\text{Os}/^{188}\text{Os}$ with Al_2O_3 contents may suggest that this process is
2315 typically not large-scale and pervasive (cf. Fig. 2).

2316

2317 **^{186}Os - ^{187}Os coupled enrichments.** Over time, Pt/Os ratios greater than that of the primitive
2318 mantle (PM) will develop elevated $^{186}\text{Os}/^{188}\text{Os}$ ratios. A Pt/Os ratio of approximately greater than
2319 8 is required, over a 1.5 Ga period, to produce the most ^{186}Os -enriched mantle melt identified to
2320 date (cf. Pt/Os PM: 1.95; Becker et al., 2006). Of the current mantle database for peridotites,
2321 dunites, and some chromitites, approximately 11% have Pt/Os ratios greater than 4, while only
2322 4% have ratios greater than 8 (Fig. 25). Enriched pyroxenite lithologies, however, commonly
2323 have sufficiently high Pt/Os ratios; ~55% of the 62 pyroxenites compiled in Fig. 25 have Pt/Os
2324 >8. However, many rocks with elevated Pt/Os also possess elevated Re/Os which evolves to
2325 much higher $^{187}\text{Os}/^{188}\text{Os}$ ratios than observed in intraplate magmas with enriched ^{186}Os .
2326 Therefore, rocks with Pt/Os, Re/Os and Pt/Re all greater than the PM are of particular interest for
2327 the generation of coupled enrichments of ^{186}Os and ^{187}Os , but such rocks are a very minor
2328 proportion of the current mantle database (Fig. 25).

2329

2330 This difficulty in generating radiogenic ^{186}Os , without also producing enrichments in ^{187}Os
2331 beyond those observed, led Brandon et al. (1998), after Walker (1995), to propose a role for
2332 transfer of an outer core Os signature into the plume source of some high-degree melts in
2333 intraplate settings. Twenty years later, this remains a possible scenario, despite the alternative
2334 mechanisms proposed that are outlined here. The core-mantle interaction model does, however,
2335 require an early onset of inner core solidification (by 2.5 Ga, and earlier for 2.8 Ga Kostomuksha
2336 komatiites; Puchtel et al., 2005) in order to allow sufficient time for ingrowth to produce
2337 enrichments in ^{186}Os and ^{187}Os in the predicted high (Pt-Re)/Os outer core. A more complete
2338 discussion of the core-mantle interaction debate can be found in Brandon & Walker (2005) and
2339 Lassiter (2006).

2340

2341 Since the emergence of the core-mantle interaction theory, several other possible sources of
2342 radiogenic ^{186}Os have been proposed (e.g., Baker and Jensen, 2004; Luguet et al., 2008), though
2343 no proposed mechanism is completely convincing. The modification of pyroxenites,
2344 refertilisation of peridotites and accompanying sulfide removal and/or metasomatism is the most
2345 likely alternative to core-mantle interaction (Luguet et al., 2008; Marchesi et al., 2014), but
2346 suitable Pt/Os and Re/Os ratios in the current mantle database are the exception, rather than the
2347 rule (Fig. 25). One further, more complex, possibility is that signatures may be combined from
2348 separate mantle components each with either high Pt/Os or high Re/Os, but not both. As outlined
2349 in a previous section, extreme Pt/Os fractionation exists on a variety of scales in Earth's mantle,
2350 particularly during the formation of PGM. What is not yet clear is the fate of such PGM during

2351 mantle convection and whether there is sufficient separation and sampling of particular PGM
2352 compositions to produce specific signatures in mantle melts.

2353

2354 In summary, processes exist in Earth's mantle that can account for the ^{186}Os - ^{187}Os
2355 enrichments observed in intraplate magmas, but currently they appear to be rare.

2356

2357 **The relationship between abyssal peridotites and MORB: an osmium isotope perspective**

2358

2359 One major debate in the field of HSE chemistry, and a key issue for mantle geology as a
2360 whole, is the extent to which abyssal peridotites represent the mantle residues of partial melting
2361 at oceanic spreading centres. Osmium isotopes have been a key part of this debate, but the
2362 evidence is complex. Early analyses identified a large range of $^{187}\text{Os}/^{188}\text{Os}$ compositions in
2363 abyssal peridotites, ranging from sub-chondritic to significantly supra-chondritic (see Abyssal
2364 Peridotite section above). The elevated signatures were largely attributed to seawater interaction.
2365 After taking into account this process, the remaining abyssal peridotite data appeared to be far
2366 less radiogenic than data for mid-ocean ridge basalts, thus casting doubt on a genetic link
2367 between abyssal peridotites and MORB. Since that time, two important findings have been made
2368 which reduce this discrepancy.

2369

2370 Firstly, was the discovery of interstitial sulfides of magmatic origin possessing radiogenic,
2371 supra-chondritic $^{187}\text{Os}/^{188}\text{Os}$ ratios (Alard et al., 2005), together with non-chondritic PGE ratios
2372 (Alard et al., 2000). A preferential contribution from these interstitial sulfides to a partial melt,
2373 compared with the contribution from ancient, unradiogenic sulfides enclosed within silicates,
2374 could account for the more radiogenic signatures of MORB and other partial melts of the oceanic
2375 mantle, compared with those recorded in bulk-rock abyssal peridotites.

2376

2377 Secondly, but of at least equal importance, was the finding that the Os isotope
2378 compositions of MORB (see Gannoun et al., 2015, this volume) were less radiogenic than
2379 previous thought. In particular, the range of $^{187}\text{Os}/^{188}\text{Os}$ ratios in MORB glasses was found to be
2380 considerably less (0.126-0.148) than previous findings (e.g., Schiano et al., 1997), with a lower
2381 mean of 0.133 +/- 0.009, in part due to an analytical artefact in the original data (Gannoun et al.,
2382 2007). This mean value, while reduced, remains in excess of typical values for abyssal peridotites
2383 ($^{187}\text{Os}/^{188}\text{Os}$: 0.118-0.130). However, it was also found that the constituent phases of basalts had
2384 variable $^{187}\text{Os}/^{188}\text{Os}$ due to (i) ingrowth over poorly-constrained periods of time since
2385 emplacement (Gannoun et al., 2004), and (ii) the timing of crystallization of different phases with
2386 respect to the evolution of the melt and its interaction with seawater-modified crust (Gannoun et
2387 al., 2007). Most notably, the latter manifests itself in significantly less radiogenic Os isotope
2388 compositions in early-formed relatively Os-rich sulfides compared with their (Os-poor) host
2389 glasses. In some cases there is a difference of ~0.015 in the $^{187}\text{Os}/^{188}\text{Os}$ of glasses and
2390 corresponding sulfides (e.g., glasses: 0.1383 and 0.1479; sulfides: 0.1249 and 0.1308,
2391 respectively), with the sulfides falling in the range 0.1236 to 0.1310, largely equivalent to the
2392 range seen in abyssal peridotites. Moreover, a negative covariation of $^{187}\text{Os}/^{188}\text{Os}$ and Os content
2393 in MORB sulfides might indicate that sulfides are also affected by interaction with a radiogenic
2394 contaminant, casting doubt on the more radiogenic data for Os-poor sulfides.

2395

2396 Although sulfides included within silicates in abyssal peridotites (and other mantle
2397 tectonites) are known to possess even lower $^{187}\text{Os}/^{188}\text{Os}$ than bulk-rock samples (~0.114; Harvey
2398 et al., 2006) – and are therefore also lower than estimates of primitive MORB – such shielded
2399 sulfides likely contribute little to moderate degree partial melts relevant for MORB genesis.
2400 Therefore, in conclusion, not only has the 'gap' in composition between abyssal peridotites and
2401 MORB been largely bridged by radiogenic interstitial sulfides, but it seems likely that the gap is
2402 minor or non-existent when the most primitive parts of the MORB system are analysed.

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2404 **Reinterpretation of Re-Os model ages**

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Model ages, whereby the isotope ratio of a sample is compared to the evolution of a reference frame such as average chondrite compositions, have been extensively used in geochemistry to give melt depletion ages in systems where recent mobility of elements has obscured any isochronous isotope systematics. The Re-Os system has been of particular use in this regard, due to the contrasting behavior of Re and Os which can result in, for high degree melts, effective Re removal from the source, while Os remains present in high enough abundances (several ng/g) to provide a degree of robustness against alteration and contamination. For Os, the measured $^{187}\text{Os}/^{188}\text{Os}$ ratio of a sample (or, for xenoliths, the ratio calculated at the time of the host eruption) is compared to the evolution curve of the mantle (commonly either a chondrite reference or the primitive mantle estimate). For Re depletion ages (T_{RD}) it is assumed that the residue is completely depleted in Re after partial melting and, thus, there is no further ingrowth of ^{187}Os . The advantage of this method is that it provides a relatively robust guide to the long-term evolution of the sample, due to the generally conservative behavior of Os, without the difficulties induced by recent Re addition or loss. In reality, however, only in high degree melting events is complete Re removal attained and in many cases the T_{RD} age merely provides a minimum age. An alternative type of model age uses the measured Re/Os ratio to calculate the time when the $^{187}\text{Os}/^{188}\text{Os}$ of the sample intersected that of the reference frame (T_{MA} or $T_{\text{Re-Os}}$). In theory, this can provide a more accurate age, but it suffers the same sensitivity to Re mobility as do attempts to identify Re-Os isochron relationships.

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Numerous caveats and potential pitfalls of model age determinations have now been recognized and the reliability and interpretation of Re-Os model ages in peridotites was the subject of a comprehensive review by Rudnick and Walker (2009). Here, we summarise the main issues surrounding such model ages, in the context of the processes and tectonic settings discussed in this chapter.

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Perhaps the most obvious issue encountered has already been mentioned above – that of the degree of depletion of Re. Rudnick and Walker (2009) demonstrate that mantle melting at 3.5 Ga to form a basaltic melt would result in vastly different age estimates from T_{MA} and T_{RD} methods: the T_{MA} age for the residue would be 3.5 Ga, because the Re/Os ratio of the residue is used to back-calculate the isotope evolution of the sample, whereas the assumption of complete Re depletion in the case of a T_{RD} age would produce an age of just over 1 Ga. Clearly at this level of depletion T_{RD} ages are not useful and they only become more valuable when Re removal is close to complete (probably a boninitic or komatiitic melt depletion event).

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Alternatives to isochron ages and Re depletion ages have been used to gain age information for sample suites where, respectively, Re mobility is suspected or Re removal was not complete. An element of similar compatibility to Re, but less mobile, such as Al_2O_3 , can be used as a proxy for Re on an isochron diagram (Reisberg and Lorand, 1995; see earlier). Although there is sometimes much scatter on such plots, they appear to be broadly robust. For large datasets of >50 samples, but preferably more, probability density function plots provide a means to identify common apparent depletion ages, which lends weight to an argument for those ages having age significance. For instance, a range of $^{187}\text{Os}/^{188}\text{Os}$ ratios could be produced by variable degrees of depletion or by the same degree of depletion at different times. The identification of peaks on probability plots might indicate discrete times of melt depletion (perhaps partially obscured by variable depletion, preservation issues and/or inheritance) rather than a more continuous spectrum of compositions which might be expected from a suite of variably depleted samples.

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There is significant inherent uncertainty with any T_{RD} age, because they are based on a model evolution curve. There are two aspects to this issue: (i) it is known that Earth's mantle has broadly chondritic proportions of the HSE, but it is not known which chondrite group – if indeed

2457 any in the global collection – supplied Earth’s HSE or whether there was any fractionation of
2458 HSE during core formation. Models to account for the apparently supra-chondritic Ru/Ir and
2459 Pd/Ir ratios of the PM (Becker et al., 2006; Walker, 2009; Fischer-Gödde et al., 2011) may also
2460 have implications for the Re-Os isotope evolution of the PM. The choice of type of chondrite or
2461 PM estimate to use for the model evolution can result in an age variation of nearly 200 Ma for a
2462 $^{187}\text{Os}/^{188}\text{Os}$ of ~0.124, decreasing with increased age to an uncertainty of ~100 Ma at around 2 Ga
2463 ($^{187}\text{Os}/^{188}\text{Os} = 0.114$). (ii) As with lithophile isotope systems (e.g., Sm-Nd) a choice has to be
2464 made whether to use a primitive or depleted mantle reference frame. This can make an even
2465 more significant difference to the age given that the estimated $^{187}\text{Os}/^{188}\text{Os}$ of the primitive mantle
2466 is 0.1296, whereas an ‘average’ depleted mantle composition might be somewhere between
2467 0.1245 and 0.128, depending on whether the average for abyssal peridotites or a combination of
2468 chromitites, PGM and high-degree mantle melts is used (Walker et al., 2002b; Pearson et al.,
2469 2007; Dale et al., 2009b). This also illustrates the problem of inheritance, which relates to the
2470 large degree of Os isotope heterogeneity observed in the convecting mantle and is amongst the
2471 most important considerations. This effectively means that for small datasets without additional
2472 information there is little way of knowing whether an apparent old age reflects a significant
2473 ancient melt depletion event in the context of its tectonic setting, or whether the measured
2474 $^{187}\text{Os}/^{188}\text{Os}$ is a composite of that event superimposed on an already depleted (or enriched) Os
2475 signature. For this reason, larger datasets obviously produce more robust age estimates and plots
2476 displaying probability can be used to identify ‘significant’ common ages or ‘peaks’ (Pearson et
2477 al., 2007; Rudge, 2008).

2478

2479 So far, we have made no mention of potential petrological pitfalls for model ages. These
2480 encompass serpentinisation, sulfide breakdown, refertilisation and melt-rock reaction (Rudnick
2481 and Walker, 2009). Serpentinisation, as discussed in an earlier section, does not typically affect
2482 Os isotope systematics except in extreme cases, which can easily be avoided when selecting
2483 samples with which to gain age information. Sulfide breakdown is known to occur in mantle
2484 xenoliths, due to interaction with the host melt. This commonly results in Os loss which could
2485 potentially impact upon the model age if $^{187}\text{Os}/^{188}\text{Os}$ is variable between different host phases,
2486 and which also leaves the sample more susceptible to contamination and alteration.

2487

2488 Depending on the tectonic setting, some processes may or may not impact on model ages.
2489 For instance, melt-rock reaction in the convecting mantle is commonly associated with melting,
2490 and is therefore effectively zero age with respect to melting and won’t normally affect the model
2491 age recorded for that melting event. Such melt-rock reaction also usually produces discordant
2492 samples on an $^{187}\text{Os}/^{188}\text{Os}$ - Al_2O_3 diagram, and can thus be identified and avoided for the purposes
2493 of dating. Conversely, processes of melt percolation and reaction in the continental lithosphere
2494 may occur long after the melt depletion episode of interest and this has the potential to obscure
2495 the true age (Rudnick and Walker, 2009). These issues mean that samples with the lowest
2496 $^{187}\text{Os}/^{188}\text{Os}$ give the most reliable ages, but they too may still have experienced radiogenic Os
2497 input. The extent to which this process affects ages depends on the amount of addition of sulfide,
2498 and the Os isotope composition and concentrations of those sulfides. Such sulfides are typically
2499 poorer in Os than enclosed sulfides so significant additions of sulfide may be required to
2500 significantly affect the age.

2501

2502 Although the processes of metasomatism and refertilisation can have a significant effect on
2503 model ages, sometimes leading to recent T_{RD} ages or “future” T_{MA} ages, in some cases these
2504 processes can be traced using HSE behavior. For example, it has been recognized, in the cratonic
2505 setting, that the oldest T_{RD} ages for a suite of samples are associated with the lowest Pd/Ir ratios,
2506 reflecting the most pristine and severe melt depletion signatures (Pearson et al. 2004). Recently,
2507 the Se/Te ratio has also been combined with Pd/Ir, in order to further understand the effects of
2508 metasomatic sulfide addition on model ages and place limits on the levels of addition that can
2509 occur before the model age may no longer be reliable (Luguet et al., 2015).

2510

2511 In summary, there are numerous potential pitfalls and limitations for Re-Os model age
2512 determinations but, in the absence of isochron dating, the system remains amongst the most
2513 useful for providing the ages of melt depletion of the mantle.

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2517

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