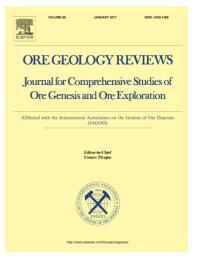
#### Accepted Manuscript

Multiphase magma intrusion, ore-enhancement and hydrothermal carbonatisation in the Siah-Kamar porphyry Mo deposit, Urumieh-Dokhtar magmatic zone, NW Iran

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#### 35

#### 36 Abstract

37 The Siah-Kamar Mo deposit (SKMD) is located at the northwestern termination of the 38 Urumieh-Dokhtar magmatic zone and it is the only porphyry Mo ore reserve in Iran. The exploration program documented 39.2 Mt proved reserves @ 539 ppm Mo and 66.4 Mt 39 probable reserves @ 266 ppm Mo. In this study, field and petrographic investigations, 40 41 integrated with geochemical (fluid inclusion and quartz chemistry) and geochronological (U-42 Pb zircon, Re-Os molybdenite, and Rb-Sr multimineral isochron) studies are used to propose 43 a metallogenic model for the Mo mineralisation in the SKMD. The geology of the SKMD is 44 characterized by the emplacement of a multiphase Oligocene basic/intermediate (at ca. 33-30 45 Ma) to acidic (29-28 Ma) magmatic suite, which intruded the Eocene volcanic country rocks. The alteration zone, about  $4 \times 3$  km in size and with a general NW-SE trend, is centered 46 47 within the main basic porphyry stock, grading from an inner potassic-sodic zone to peripheral phyllic/propylitic halos. The late acidic magmatic products (stocks and dykes) intruded and 48 49 post-dated the main alteration zone. Two-stage Mo mineralisation is recognised, including: 50 (i) stage-1, disseminated molybdenite, coeval with the formation of potassic-sodic alteration 51 and minor, microscale Fsp, Bt, Qz+Po veinlets; and (ii) stage-2, high-grade molybdenite+carbonate (± sericite), structurally-controlled stockwork veining. Fluid 52 53 inclusion systematics combined with TitaniQ thermometry documents a mineralising fluid 54 system compatible with a transition from high-temperature (up to ca. 600 °C) magmatic to 55 epithermal (250 °C) conditions during progressive cooling, exhumation and mixing with 56 meteoric sources at shallow crustal conditions (ca. 7-3 km). The Re-Os molybdenite dating 57 constrains the high-grade Mo ore formation at ca. 29-28 Ma, attesting for the intimate linkage 58 between the main Mo mineralisation and the acidic magmatic phase in the area. The Rb-Sr 59 geochronology of the potassic-sodic alteration zones confirms the two-stage 60 magmatic/mineralisation scenarios, overlapping within errors with the results obtained from 61 the U-Pb zircon geochronology and constraining the formation of the potassic-sodic and phyllic alteration at ca. 33 and 28 Ma, respectively. Our results document an uncommon 62 63 scenario of two-stage porphyry Mo mineralisation associated with intensive late stage 64 carbonate precipitation and achieved during a long-lasting and multiphase magmatic pulses 65 of Oligocene age. We highlight the dominant role of acidic fluid neutralisation for further ore 66 enrichment during polyphase magma intrusion as the dominant factor controlling the Mo 67 mineralisation in the SKMD. Comparison at a regional-scale indicates that parameters such 68 as longevity of magma supply, progressive magma crystallization/differentiation, and the

69 presence of a possible pre-enriched crustal material should be considered responsible for the70 Mo endowment in the UDMZ.

71

72 Key words: porphyry Mo, ore deposits, magmatism, fluids, Urumieh-Dokhtar zone, Iran

73

#### 74 1. Introduction

75 Molybdenum (Mo) porphyry-type deposits supply the main Mo demands worldwide (> 76 95%) (John and Taylor, 2016). The Mo deposits are mainly associated with subduction-77 related settings, such as volcano-plutonic magmatic arcs, back-arcs and continental collision 78 orogens (Westra and Keith, 1981; Wallace, 1995; Seedorff et al., 2005; Taylor et al., 2012; 79 Wu et al., 2017). They are mainly located in the western Cordillera of North America (Ayres et al., 1982; Duke, 2007; Kerr et al., 2009), in Eastern Europe (Janković, 1982), Australia 80 (Whitcher, 1975), China (Mao et al., 2008; Zeng et al., 2012; Wu et al., 2017), Russia 81 82 (Sutulov, 1973), and elsewhere in Asia (Heinhorst et al., 2000). Porphyry molybdenum deposits have been classified based on the Mo modal abundance, F 83

84 content, geochemical fingerprint of associated magmatism, and geodynamic setting of

formation (Westra and Keith, 1981; Ludington et al., 2009; Ludington and Plumlee, 2009;

86 Taylor et al., 2012; Wu et al., 2017). In particular, two end-member porphyry Mo

87 mineralisation types have been proposed in the literature: the Endako- and the Climax-type

88 (Selby and Creaser, 2001). The Endako-type deposits are characterised by low-grade

89 (Mo<0.15 wt.%), but large-tonnage Mo ores and low F (<0.05 wt.%) in a continental arc

90 setting associated with calc-alkaline magmatism. Tungsten may be enriched as a by-product

91 metal, but copper occurs as in minor quantities or is absent (Doebrich et al., 1996; Taylor et

al., 2012). The Climax-type deposits are instead characterised by high-grade (Mo>0.15 wt.%)

Mo ores and high F (>0.05 wt.%) in an intraplate setting and associated with alkali-calcic and
alkali magmatism (Westra and Keith, 1981).

95 The duration of the mineralisation process in porphyry systems commonly ranges from

96 100 k.y. to 1.5 m.y. (Sillitoe, 2010). Based on numerical modelling, Cathles et al. (1997)

97 proposed a lifetime no longer than 800,000 years for a geothermal system of temperatures

98 greater than 200 °C. The long lifetime (> 1 m.y.) of magmatic-hydrothermal activity mainly

99 depends upon volume, emplacement depth, and temperature of magma, conductive and

100 convective cooling, and permeability of the host rocks (Taylor et al., 2012). It is also

101 demonstrated that multiple events of magma intrusion can increase the lifetime of porphyry

copper deposits in excess of 1 m.y. (Chiaradia et al., 2009; John et al., 2010). The intermineral intrusive pulses can be either mineralizing or barren and, consequently, can enrich or
destroy the former mineralized zones (Sillitoe, 2010). In particular, long-lived magmatism
and fractionation, together with fluid exsolution and a change toward more reducing and acid
conditions in the mineralising environment are recognised as the key factors for the Mo
endowment in the residual melts (e.g., Candela and Holland, 1986; Keith et al., 1986; Blevin
and Chappell, 1992; Candela, 1992; Audetat et al., 2008; Seo et al., 2012; Wilkinson, 2013;

109 Zhang et al., 2018).

The majority of the porphyry Cu–(Mo, Au) deposits in Iran occur along the Cenozoic Urumieh-Dokhtar magmatic zone (UDMZ) and the Arasbaran magmatic zone, including the world-class SarCheshmeh, Sungun, and Meiduk deposits (Asadi et al., 2014; Aghazadeh et al., 2015; Richards and Sholeh, 2016) (Fig. 1 and supplementary material #1). The Siah-Kamar Mo deposit (SKMD) (Nabatian et al., 2017b; Simmonds et al., 2019), located at the northwestern termination of the UDMZ, from Mianeh to Hashtroud cities (Figs. 1 and 2), is

the only Mo ore reserve in Iran. Despite the presence of some alteration zones, no porphyry

117 deposits or significant mineralisation has been reported in the area before the exploratory

118 studies carried out by National Iranian Copper Industries Company since 2005 (NICICO,

119 2010). Preliminary exploration on the altered zones has shown a significant Mo-W

120 enrichment, whereas the deposit is barren with respect to Cu and Au. The exploration

121 program held by the owner within the SKMD documented the in-depth mineralisation down

to 600 m, with a 39.2 Mt proved reserves @ 539 ppm Mo and 66.4 Mt probable reserves @

123 266 ppm Mo (Aria Kansar Samin Co., 2013).

124 Early research studies focused on the magmatism and mineralisation characteristics of the 125 ore deposit. These studies defined the metaluminous, high-K calc-alkaline to shoshonitic 126 geochemical fingerprint of the magmatism referred to a final stage of Neo-Tethys subduction 127 beneath the Iranian plate (Khaleghi et al., 2013; Nabatian et al., 2017b). These studies also 128 reported low salinity and low-to-moderate temperature magmatic fluids as responsible for the 129 mineralisation (Nabatian et al., 2017b). Contrasting age estimates exist for the timing of the 130 Mo mineralisation at the SKMD and still uncertain is the timing and geological scenario of 131 the Mo mineralisation. Nabatian et al. (2017b) presented molybdenite weighted mean Re-Os age of  $41.9 \pm 3.6$  Ma, whereas Simmonds et al. (2019) reported ages between  $28.10 \pm 0.15$ 132 133 and  $29.06 \pm 0.20$  Ma using the same geochronological technique. Simmonds et al. (2019) also 134 reported early Oligocene U-Pb zircon ages  $(32.70 \pm 0.40 \text{ Ma to } 30.90 \pm 0.40 \text{ Ma})$  from the

mineralized body, suggesting mineralization has potentially occurred ~1.5 Ma after the
crystallization of the porphyry stock. Moreover, key aspects, such as the longevity of the
magmatic porphyry system, the hydrothermal-alteration processes, and the structural control
on fluid flow and ore mineralisation were poorly considered.

139 Based on detailed field and petrographic investigations integrated with analytical data 140 (fluid inclusion microthermometry, U-Pb zircon, Re-Os Molybdenite, and Rb-Sr 141 geochronology, quartz geochemistry), this contribution is aimed at (i) defining a genetic 142 model for the Mo mineralisation at the SKMD, and (ii) improving our knowledge of the Cenozoic metallogeny within the UDMZ (Fig 1). Our results document an Oligocene, long-143 lasting history of magma crystallisation/emplacement/alteration (33-28 Ma), which 144 culminated with Mo stockwork mineralisation at ca. 29-28 Ma. The Mo mineralisation was 145 structurally controlled and caused by ore enhancement during renewed magmatism, which 146 induced metal leaching and deposition in the transition from ductile-to-brittle environments 147 above a cooling and degassing magma. These data are used to propose a genetic 148 149 mineralisation model for the Mo ore and to discuss implications for the regional ore 150 distribution, and for the Mo mineralisation in porphyry systems in general.

151

#### 152 2. Geological Background

The UDMZ (Fig. 1) records prolonged Cenozoic magmatic activity, spanning from the Paleocene to the Quaternary (e.g., Richards et al., 2006; Chiu et al., 2013; Yeganehfar et al., 2013; Heidari et al., 2015; Ahmadian et al., 2016; Heidarzadeh et al., 2017; Hosseini et al., 2017; Sarjoughian and Kananian, 2017; Shahsavari Alavijeh et al., 2017). A main phase of magmatism occurred during the Eocene-Oligocene, synchronous with the magmatic flare-up at regional scale (Stocklin, 1968; Verdel et al., 2011; Chiu et al., 2013; Moghadam et al., 2016).

160 Based on the mineralisation distribution and types along the UDMZ, a southern and a 161 northern region can be recognised. The southern region, also known as the Kerman Belt, 162 extending from the Naeen city to Bazman volcano (Naeen-Bazman volcanic zone, NBMZ in 163 Fig. 1), hosts the dominant and largest porphyry Cu-Mo deposits of Iran, which are mostly Miocene in age (Aghazadeh et al., 2015). Magmatic products show a general calc-alkaline 164 165 (moderate to high-K) and metaluminous fingerprint, with a pronounced adakitic signature during the Miocene (Shahabpour and Kramers, 1987; Hassanzadeh, 1993; Shafiei et al., 166 167 2008; Taghipour et al., 2008; Shafiei et al., 2009; Mirnejad et al., 2011; Asadi et al., 2014).

168 Three metallogenic stages have been proposed by Aghazadeh et al. (2015) within this segment, including late Oligocene (29-27 Ma), middle Miocene (14-11 Ma), and late 169 170 Miocene (9–6 Ma). The northern region, extending from Urumieh/BostanAbad to 171 Naeen/Anarak cities (Fig. 1; supplementary material #1), instead hosts a few relatively 172 smaller deposits. Within this region, two magmatic zones can be recognised: (i) a volcano-173 plutonic zone from BostanAbad to Anarak cities (BAMZ in Fig. 1; supplementary material 174 #1), Eocene to Oligocene in age, typified by high-K calc-alkaline and shoshonitic signatures 175 (e.g., granitoid batholiths of Zanjan, Hassanzadeh et al., 2008; Kal-e kafi, Ahmadian et al., 2009; Abhar, Castro et al., 2013; Kuh-e Dom, Sarjoughian and Kananian, 2017) and 176 associated with some small, intrusion-related, polymetallic deposits (Ahmadian et al., 2009; 177 178 Fazel et al., 2015; Ahmadian et al., 2016); and (ii) an Oligocene-Miocene volcano-plutonic 179 zone from Urumieh to Naeen (UNMZ; in Fig. 1), which is characterized by post-collisional 180 calc-alkaline and high-K calc-alkaline to adakite products (e.g., Richards et al., 2006; Chiu et al., 2013; Yeganehfar et al., 2013; Babazadeh et al., 2017; Ballato et al., 2017b; Sarjoughian 181 and Kananian, 2017) and associated with minor Au-rich Cu (Mo) porphyry (e.g., Dalli, Ayati 182 et al., 2013) and epithermal Au deposits (e.g., Zarshuran, Mehrabi et al., 1999; Muteh, Moritz 183 184 et al., 2006; SariGunay, Richards et al., 2006). Regarding the Miocene magmatism, Shafiei et 185 al. (2008) suggested syn- and post-collisional thickening and melting of Cu- and sulfur-rich peridotite, near the crust-mantle boundary, as responsible for fertile adakitic magmatism in 186 187 the southern Urumieh-Dokhtar magmatic zone. In contrast, Haschke et al. (2010) suggested that a syn- or post-collisional Miocene delamination in the northern UDMZ had removed the 188 189 Cu- and S-enriched metasomatised lithospheric arc-root and, subsequently, prevented the Cu 190 mineralisation during collisional magmatism.

191 The AMZ is adjacent and sub-parallel to the UDMZ, running from the Sabalan 192 volcano to the Jolfa city, probably continuing to Armenia into the lesser Caucasian Magmatic 193 zone (Innocenti et al., 1982; Hassanpour et al., 2014; Simmonds et al., 2017) (Fig. 1). The 194 AMZ boundaries have not been defined yet, although it is normally considered as a segment 195 of the UDMZ. Nonetheless, it has been also considered as a part of western Alborz zone 196 (Nabavi, 1976; Aghanabati, 2004; Moritz et al., 2016; Rolland, 2017). The porphyry Cu (Mo-197 Au) mineralisation span from Oligocene to early Miocene (e.g., Shahabpour and Kramers, 198 1987; McInnes et al., 2003; Aghazadeh et al., 2015) and it is more intense with respect to the 199 northern region of the UDMZ.

200

#### 201 **3.** Materials and Methods

202 A multidisciplinary research approach is adopted in this study that includes field work 203 and laboratory analyses. Field work was based on the existing 1:250,000 cartography (Amidi 204 et al., 1987) and addressed to map the main magmatic bodies and the alteration types (Fig. 2). A detailed map at the scale 1: 5000 is produced for describing the alteration zoning and the 205 206 main ore zones (Fig. 3). Field work and petrographical observation form the basis for the 207 sampling strategy for the laboratory work that includes: (i) fluid inclusion microthermometry, 208 (ii) Raman spectroscopy; (iii) LA-ICPMS trace and rare earth (REE) elements analysis; (iv) 209 zircon U-Pb geochronology; (v) molybdenite Re-Os geochronology; and (vi) multi-mineral 210 Rb-Sr geochronology of the alteration assemblages.

The studied samples are shown in Figure 3 and listed in Table 1, where their location, petrography and analytical methods adopted are detailed. The analytical protocols are described in the Appendix. In the following, mineral abbreviations are after Whitney and Evans (2010).

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- 216

#### 4. Geology of the study area and the ore deposit

217 The SKMD is located within the Mianeh intermountain basin (Heidarzadeh et al., 2017), in-between the Alborz and the AMZ, at the southern margin of the Azerbaijan Block at the 218 219 tip of the active dextral Tabriz Fault (Figs. 1, 2). The stratigraphy of the area is dominated by 220 a wide exposure of Eocene volcanic rocks, which make-up the country rocks of the SKMD. 221 The Eocene volcanic rocks consist of andesite-basalt and trachy-andesite lava beds and vitric 222 dacitic to andesitic tuff (Amidi et al., 1987) (Fig. 2). These volcanic products are intruded or 223 covered by a composite Cenozoic (Oligocene-Miocene) volcano-plutonic complex (Amidi et al., 1987; Khodabandeh et al., 1999). A variety of intrusive and sub-volcanic rocks with 224 225 variable geometry and composition occur in the study area, including diorite, monzonite to 226 leuco-granite, and rhyodacite porphyry (Amidi et al., 1987; Khodabandeh et al., 1999). The granular intrusive bodies crop out in the northwest and centre of the study area, generally 227 228 aligned along a NW-SE orientation (Fig. 2). To the southeast (from Khatoon Abad to Siah-Kamar), an array of E-W to NE-SW striking microgranular and porphyritic felsic stocks and 229 230 dykes crop out (Fig. 2). Miocene volcanic country rocks consist of tuff and ash layers with 231 interlayered andesitic lava beds. This volcanic succession is covered by upper Miocene 232 continental sedimentary deposits (Upper Red Fm.) that crop out in the northwestern and southwestern sectors of the study area. These deposits are unconformably covered by 233

Pliocene-Quaternary continental sedimentary successions (Amidi et al., 1987; Ballato et al.,
2017a) (Fig. 2).

236 The ore deposit geology is characterised by a major alteration zone centered on a 237 porphyry-type, composite magmatic complex that, intruded within the Eocene volcanic rocks, 238 hosts the main Mol ore (Figs. 2 to 4). Despite the different alteration types, the magmatic 239 bodies share common textural features consisting of a porphyritic microgranular and/or 240 glassy groundmass, with various proportions of AFs, Pl, and mafic phenocrysts (Cpx and/or 241 Amp) in the primary igneous assemblage (see Table 1 for a detailed description). Based on 242 the estimated modal abundance of the primary igneous mineral components, the rock compositions are andesitoid/basaltoid (Le Maitre et al., 2005). The mineralized porphyry 243 244 body (samples MN02 and MN03 in Table 1) is the largest stock in deposit area with a NW-245 trending ellipsoid shape (1800×1300 m). The body experienced intense alteration and in 246 some places the primary texture has been destroyed. Where preserved, the primary rock composition ranges from svenite to monzonite-quartzmonzonite (Nabatian et al., 2017b). 247 248 Also barren magmatic bodies can be recognised in the area, to the south (sample MN31 in 249 Table 1) and to the north (sample MN65 in Table 1) of the mineralised body, in a continuum 250 from syn-to-post ore stage as documented by the alteration patterns and distribution (see inset 251 in Fig. 3). Based on the mapped outcrops of alteration zones and assuming lateral continuity below 252

the Miocene and Pliocene cover rocks, the reconstruction of the alteration zone shows an ellipsoid shape, with a general NW-SE trend and about 4×3 km of areal extent (Fig. 3). The Mol mineralisation occurs both as minor disseminated within the inner potassic-sodic alteration zone and major focused in Qz-Afs-Cb veins.

A suite of acidic subvolcanic products (rhyolitic and dacitoid in composition; samples Mn01 and Mn19 in Table 1) are observed to intrude and post-date the alteration zone. These magmatic rocks show only minor evidence of rock alteration (Fig. 3; Table 1).

#### 261 **5.** Alteration types

Alteration zoning grades from potassic-sodic alteration in the centre to adjacent phyllic and peripheral propylitic halos (Fig. 3). In the following, we described the main alteration zones, mineral assemblages and their distribution.

265 5.1. Potassic-sodic alteration

266 The potassic-sodic alteration is observed in the core of the altered zone, with an areal extent of ca. 2.4 km<sup>2</sup> (Figs. 3, 4). The alteration is typified by the mineral assemblage Bt+ Kfs 267 268  $+ Ab + Mag + Anh \pm Ilm$ , Ap, Mol. Typical is the occurrence of a secondary spotted texture 269 made of Bt-Mag aggregates, likely overprinting early magmatic layering (Fig. 5a). Secondary 270 anhedral Kfs and Ab aggregates overprint early magmatic Pl (Fig. 5b). Molybdenite typically 271 occurs as disseminated flakes within the alteration matrix (Fig. 5c). Biotite occur as very fine 272 grained, dominantly as disseminated grains but also as irregular curvy vein-like aggregates 273 (Fig. 5d). Magnetite grains occur both as fine-grained disseminated crystals in the 274 groundmass and as inclusion in Kfs (Figs. 5b, e). Minor amount of Ilm is observed as inclusions in Bt (Fig. 5e). Late formed Po veins also occur (Figs. 5d). Stockwork 275 276 development and partial brecciation are ubiquitous and resulted in diverse vein types (Fig. 5a;

see also below).

#### 278 5.2. Propylitic alteration

Propylitic alteration is the largest alteration zone that mainly affects the Eocene volcanic host rocks and the porphyry-type magmatic complex (Figs. 3, 4). The width of the propylitic alteration is at least 1 km, but it suddenly decreases with distance from the porphyry-type magmatic complex. This alteration is typified by the assemblage  $Ep \pm Act \pm Cb \pm Py$  with stockwork, disseminated and locally brecciated textures close to the porphyry-type magmatic complex (Fig. 5f).

#### 285 5.3. Phyllic alteration

The phyllic alteration rims the core of the potassic-sodic and the marginal alteration zone. Locally it overprints the inner part of the potassic alteration zone (Figs. 3, 4). The phyllic alteration consists of the assemblage Qz + Mus + Py, which is variable distributed within the alteration zone, usually associated with veining and rock fractures, seldom associated with intensive pyritization (ca. 5- 10 % vol.). Pyrite is typically altered to secondary iron oxide and hydroxide minerals in the uppermost portions of the alteration zone. Intermediate argillic alteration (Sillitoe, 2010) is observed to decorate the damage zone of major faults (Fig. 5g-j).

#### 293 **6.** Veining

We documented multistage veining, locally associated with pronounced alteration halos. The vein density is highly variable and commonly decreases toward margins of the alteration zones, where the alteration distribution is mainly influenced by vein distribution and persistence. In particular, the vein volume abruptly interrupts at the boundary between the mineralised porphyry complex and the Eocene volcanic host rocks (Fig. 6a).

Azimuth rose diagram generated from the measured cumulative vein strikes reveals a

300 mean E-W orientation, associated with minor NW-SE and N-S ones (Fig. 6b). In the

301 following, a description of the vein texture and assemblages is described for each alteration

302 zone.

#### 303 6.1. Veins in the potassic-sodic alteration zone

A conceptual scheme of the vein generation at the microscale within the inner potassicsodic zone is presented in Figure 7. The early veins generation (V1) consists of up to mmsize Bt and minor Qz+Po, Ab+Kfs and Kfs veins, sharing a discontinuous and irregular curvy shape. The Fsp veins are commonly associated with Fsp halos; no evidence of Mol mineralisation is observed. Locally, small (up to 0.5 mm) curvy shapes  $Qz \pm Po$  veins are also observed (Fig. 5d).

310 The second generation (V2a) consists of a voluminous, cm-sized,  $Qz-Kfs \pm Mol \pm Ab$  non-

311 systematic (stockwork) vein array (Figs. 5a and 8a-c). These veins are characterized by a

312 blocky texture, mainly formed by Qz-Kfs aggregates, commonly associated with minor Mol,

313 Sch, and Rt (Figs. 7 and 8a,c). The overall vein texture suggests syntaxial growth and Mol

314 occurs in equilibrium with Qz (Figs. 7 and 8a,b). Carbonate mineralisation is typified by the

early appearance of paraspurite  $(Ca_5(SiO_4)_2CO_3)$  (Colville and Colville, 1977), and Ank and

316 Sid/Ank in the stage-2 veins, which occurs in the vein suture, likely filling remaining vugs

317 (Fig. 8).

The third generation (V2b) corresponds to the main ore stage and is characterised by Cb-Mol veins, associated with minor sericite and pyrite (Fig. 9). The Cb-Mol veins mostly

320 rework the pre-existing Qz-Fsp veins/host rocks contacts, often developed as cross-cutting

321 and inter-granular cracks (Fig. 9 a,b). Locally, brecciation of the early Qz-Fsp veins is

322 observed, associated with massive Mol precipitation that often show rosette texture (Figs.

323 9a). A ribbon appearance is commonly observed in the V2b veins due to alternating variation

324 of Cb minerals (Fig. 9c). It is worth nothing that Cb compositions vary from Cal to

325 Sid/Rds/Ank when in equilibrium with the Mol precipitation (Fig. 9b,c). Muscovite mostly

developed at the expenses of pristine Kfs (Fig. 9d).

In terms of vein distribution, the overall array of Mol mineralised veins (V2a-V2b) is nonsystematic (Fig. 6c). When taking into account the V2b vein array, a more systematic
distribution is observed, which cluster along the E-W (dominant) and NS strikes (Fig. 6d).

#### 330 **6.2.** Veins in the propylitic alteration zone

The propylitic zone is associated with development of brecciation and veins, close to the porphyry mineralised body and in the distal country rocks, respectively. Brecciation causes

- fragmentation of the country rocks, cemented by Ep+Act+Cb aggregates. Veins occur as
- high-angle systematic arrays, striking roughly E-W and N-S (Fig. 6e), and are filled with the
- 335 same mineralogical assemblage (Ep+Act+Cb).

#### 336 **6.3.** Veins in the phyllic alteration zone

Within the phyllic alteration zone, veins are made of mm-size Qz + Py, associated with sub-mm-size Ms (sericite) halos (Fig. 5g-j). They form non-systematic arrays. Within the higher topographic levels of phyllic alteration, the Py grains are converted to iron oxide and hydroxide minerals.

341

#### 342 **7.** Mo ore mineralisation

Two main stages of Mol mineralisation were recognized either as disseminated or 343 344 stockwork (Figs. 9 and 10). Stage-1 (Mol-I) occurs as disseminated Mol flakes within the 345 porphyry body, in association with the potassic-sodic alteration (Fig. 5c). Stage-2 can be 346 subdivided in two Mol formation events (stage-2a and -2b in Figs. 9 and 10), Mol-II and 347 Mol-III, respectively. Mol-II occurs during V2a formation, in association with minor Sch and Rt. Mol-III is the high-grade ore stage during the Cb V2b formation, beginning with 348 349 paraspurite and continuing with Fe-Mn Cb mineralisation. Minor amount of Py and 350 secondary Ms (sericite) are associated with the Mol-III precipitation.

351

#### 352 8. Quartz chemistry and Ti-in quartz thermometry

Ouartz crystals from V2a veins from the potassic-sodic alteration zone were analyzed by 353 in situ LA-ICP-MS for Ti content (16 grains) and trace elements (7 grains). Results are 354 355 presented in Table 2 and 3, respectively. Analytical details are provided in the Appendix. 356 Analyzed quartz show high Ti (17.4-54.7 ppm) and Al (35.9-160.2 ppm), with significant 357 enrichment in Li (1.9-28.0 ppm) and Ge (0.9-2.0 ppm). In the Al vs. Li diagram, the analysed 358 quartz grains show positive correlation and fall near the 1:1 Al/Li molar ratio line, which is 359 indicative of magmatic and hydrothermal quartz (Dennen, 1966; Müller et al., 2018). In the 360 Al vs. Ge diagram, the quartz compositions are compatible with those from hydrothermal 361 quartz and the Ti contents are typical of quartz veins from (Mo-Cu-Au)-porphyry-type 362 deposits (e.g., Müller et al., 2018) (Fig. 11). 363 The TitaniQ thermometry (Wark and Watson, 2006; revised by Huang and Audétat, 2012),

assuming a  $aTiO_2 = 1$  for rutile-present conditions provides temperature estimates of 479-575  $\pm 50$  °C and 568-675  $\pm 50$  °C at 0.5 and 2.5 kbar, respectively.

366

#### 367 9. Fluid-Inclusion Study

368 9.1. Petrography

369 Samples from (i) Qz-Fsp V2a and Cb-Mol V2b veins from the potassic-sodic alteration 370 zone (Fig 9a) and (ii) Qz-Py-Ms veins from the phyllic alteration zone (Fig. 5i) were studied 371 for fluid inclusions (FI) analysis in order to assess the chemical-physical properties of the 372 mineralising fluid(s). FI were measured both in guartz and paraspurite from the same vein 373 assemblage, considered as representative of the main Mol mineralisation stages. 374 Based on the observed phases at room temperature, two main types of aqueous FI were recognised following the classification proposed by Roedder (1984): (i) dominant type-A (L-375 rich + V); and (ii) minor, solid-bearing, type-B (L-rich + V + solid). Type-A is subdivided 376 377 into four sub-groups based on the alteration zone, host mineral and the V to L volume ratio, 378 from A1 to A4. A summary of FI characteristics and results are given in Table 4. The various 379 identified FI types in this study are briefly described below. Type-A1 and A2 are L-rich + V FI, found within V2a guartz grains in the potassic-sodic alteration zone with the V to L ratios 380 381 from 0.5 to 1 and from 0.1 to 0.4, respectively. Type-A3 are L-rich + V FI. They are the only 382 detected FI inside quartz crystals from Qz+Py+Ser veins within phyllic alteration zone. Type-383 A4 are L-rich + V FI from V2b paraspurite crystals within the potassic-sodic alteration zone. 384 Type-A1 are mostly found as solitary or randomly distributed in well-protected Qz grains form post entrapment shearing, and have regular to subhedral geometry, large size (ranging 385 386 15-100  $\mu$ m). They occasionally show negative crystal shape, and possess the highest V to L 387 ratios (Fig 12a). Type-A2 FIs are characterized by small sizes (5 to 20 µm), regular with 388 negative crystal shape to irregular geometry (Fig 12b). Type-A3 are very similar to type-A2 389 FI, but with smaller sizes (ranging 5 to  $10 \,\mu$ m). They have regular to irregular shapes, some 390 showing negative crystal shapes and their V to L ratios range from 0.2 to 0.4 (Fig 12c). Type-391 A4 occur along the cleavages of the host crystal, showing irregular and elongated geometry 392 with sizes range 5-20 µm. The V to L ratios range from 0.1 to 0.25, but commonly are below 393 0.15 (Fig 12d). 394 Type-B are solid-bearing L-rich + V FI, rarely found in the cores of V2a quartz grains (Fig. 395 12e). They occur mainly isolated and in confined trails that are interpreted to be quartz

396 growth lines. Their sizes range 15-30  $\mu$ m and mostly have irregular geometry. Their V to L

ratios range from 0.1 to 0.2.

398 9.2. Raman Spectroscopy

399	Raman spectroscopy was utilized to define the composition of V and L phases of the type-
400	A FI. For the types-A1 and A4 FI, the obtained spectra for the L and V phase revealed water
401	and water vapor as the major components, respectively (Fig 13). In addition, the L phase of
402	type-A1 FI also showed a clear peak for $CO_2$ (at 1384 cm <sup>-1</sup> ) in solution (Fig 13a). Other
403	peaks were too weak and not identified. In the V phase of type-A1 FI, apart from water
404	vapor, the only detected gas component is CO <sub>2</sub> (Fermi doublet at 1285 and 1388 cm <sup>-1</sup> ). A
405	weak peak belonging to $CO_3^{2-}$ (1064 cm <sup>-1</sup> ) was also detected (Fig 13b).
406	The results from paraspurite-hosted type-A4 FI were not so clear due to the strong
407	fluorescence of the hosting mineral. No evidence of gases were detected.
408	9.3. Microthermometry
409	A summary of the FI micro thermometric data is reported in Table 4 and histograms of
410	salinity and $T_h$ are illustrated in Figure 14. All the FI types homogenized to L. The eutectic
411	temperatures ( $T_e$ ) for the type-A1 FI range -30 to -22.5 °C (mean of -25.3 °C), which are
412	higher than eutectic temperature for the H2O-NaCl system (Bodnar, 1993). The minimum
413	recorded Te could be considered as the ternary eutectic point temperature (-23 °C) of the
414	$H_2O$ -NaCl-KCl system (Hall et al., 1988). Since minor amounts of $CO_3^{2-}$ was detected by
415	Raman spectroscopy studies inside type-A1 FI, the higher recorded $T_e$ values could be
416	explained by the presence of variable amounts of other components, such as Na <sub>2</sub> CO <sub>3</sub>
417	(K <sub>2</sub> CO <sub>3</sub> ?) in addition to NaCl and KCl. The final ice-melting ( $T_{m-ice}$ ) temperatures range from
418	-3.7 to -0.6 °C ( $n = 44$ ), which corresponds to salinities between 6.85 and 1 wt.% NaCl eq.
419	(with a mode of 3.85 wt.% NaCl eq). Type-A1 shows the highest homogenisation
420	temperature ( $T_h$ ) estimates, ranging 298-365 °C and a well-defined mode at 340 °C (mean
421	value of 333 °C). These values are similar to those obtained by Nabatian et al. (2017b) on the
422	same FI type.
423	The $T_e$ for the type-A2 FI range -31.5 to -23°C, with a mode at -23 °C, compatible with
424	the ternary eutectic point temperature of the H <sub>2</sub> O-NaCl-KCl system (Hall et al., 1988). The
425	higher recorded values suggest the presence of other components as for type-A1. The $T_{m-ice}$
426	ranges from -3.7 to -0.1 °C ( $n = 195$ ), which correspond to salinities between 6.0 and 0.18
427	wt.% NaCl eq., with a mean of 2.6 wt.% NaCl eq The $T_h$ values distribute almost
428	homogeneously in a wide range from 170 to 352 °C, with a mean at 272°C.
429	Due to the small size of type-A3 FI, the $T_e$ were measured only in two cases, which
430	provided -22 and -21 °C, respectively. The obtained values are close to the eutectic point

- 431 temperature of H<sub>2</sub>O-NaCl system. The  $T_{m-ice}$  ranges from -2.6 to -1 °C (n = 52), which
- 432 correspond to salinities between 4.3 and 1.7 wt.% NaCl eq., with a mean of 1.8 wt.% NaCl

433 eq. The  $T_h$  values range 232-365 °C, but the data are skewed to higher temperatures, with a mode at 310 °C. 434 The  $T_e$  for the type-A4 FI range -33 and -24 °C, with a mean of -27 °C. Since these FI are 435 hosted in carbonate minerals, these  $T_e$  values can be attributed to the presence of CO<sub>3</sub><sup>2-</sup> and/or 436 ions such as Ca<sup>2+</sup>, Mg<sup>2+</sup>, Fe<sup>2+</sup> in addition to H<sub>2</sub>O+NaCl in the fluid. The  $T_{m-ice}$  ranges from -437 1.2 to -0.1 (n = 91), which correspond to salinities between 2.1 and 0.2 wt.% NaCl eq., with a 438 439 mean of 0.6 wt.% NaCl eq. The  $T_h$  values are distributed in a very wide range, from 145 to 440 300 °C, with a well-defined mode at 225 °C. The salinity estimates for type-B FI were calculated only in four FI from the melting 441 temperature of halite in the liquid phase upon heating. Halite crystals disappeared prior to the 442 bubble in all FI. The calculated salinity estimates range from 30.5 to 32 wt.% NaCl eq. The 443 average  $T_h$  spreads between 203 and 320 °C. 444 445 U-Pb zircon dating of magmatic rocks 446 **10.** 447 The zircon U-Pb geochronological study was carried out on zircon separates as obtained 448 both from the mineralised (sample MN03) and barren magmatic rocks (samples MN01, MN19 and MN31; Fig. 3 and Table 1). Few zircon grains were recovered from the 449 450 andesitoid/basaltoid rocks (MN03 and MN31), whilst the acidic rocks (MN01 and MN19) 451 provided a large number of grains. Zircons were first investigated through cathodoluminescence (CL), and back scattered electron (BSE) imaging techniques and then 452 453 analysed in situ LA-ICP-MS system at the university of Nagoya, Japan (see the Appendix for 454 the methodology and). Analytical results are summarized in Table 5. 455 *MN03* Zircons are generally euhedral to subhedral. Zircon grains show short to long prismatic 456 457 habits (usually >100  $\mu$ m up to 500  $\mu$ m), with length/width ratios 2:1 and up to 6:1. Most of 458 the grains appear as nearly homogeneous in BSE images with rare brighter cores. Most of 459 them possess a few continuous longitudinal fractures and apatite and feldspar inclusions are 460 common. The CL images show oscillatory to sector growth zoning (Fig. 15a). The Th/U 461 values are in the range 0.38-6.3 (n = 21; Table 5), compatible with an igneous origin (e.g., 462 Rubatto, 2002; Kirkland et al., 2015). Despite discordance of some of the spot results due to 463 Pb loss, the majority of the data define a nearly concordant cluster at about 33 Ma. In a Tera-Wasserburg Concordia diagram, an intercept age of  $32.8 \pm 1.1$  Ma is obtained ( $2\sigma$ ; MSWD = 464 465 3.5, n= 13) (Fig. 15b).

#### 466 *MN31*

467 Most of the zircon grains are of small size (50 to 150  $\mu$ m) and can be classified as type-1

468 subhedral to anhedral and type-2 euhedral crystals. Type-1 shows complex growth zoning

469 with marginal resorption. Type-2 show oscillatory growth zoning (Fig. 15c). The Th/U values

470 range between 0.3 and 2.57, attesting a magmatic origin. Age analyses of type-1 zircons

471 reveals a large spread in apparent  ${}^{206}Pb/{}^{238}U$  ages from  $41.1 \pm 1.6$  to  $2611 \pm 70$  Ma (Table 5).

472 Type-2 zircons yielded the youngest apparent <sup>206</sup>Pb/<sup>238</sup>U ages at ca. 30 Ma, with a Concordia

473 age of  $30.4 \pm 1.2$  Ma ( $2\sigma$ ; MSWD = 0.12, n= 2) (Fig. 15d).

#### 474 *MN01*

475 Zircon grains are generally euhedral to subhedral and show short to medium prismatic 476 habits (usually >100  $\mu$ m up to 250  $\mu$ m), with length/width ratios of around 2:1. Most of the 477 grains host abundant inclusions of Ap, Fsp, and Qz. The grains show heterogeneous zoning in 478 BSE images with oscillatory and sector growth zoning in CL images. Xenocrystic cores are 479 rarely preserved, and the cores are overgrown by oscillatory and sector zoned rims (Fig. 15e). Th/U values are systematically above 0.4. The majority of the apparent zircon  $^{206}$ Pb/ $^{238}$ U ages 480 481 from the growth zoning domains cluster at about 29 Ma, with a few grains providing early 482 Cambrian (ca. 529 Ma) to the Upper Triassic (ca. 220 Ma) ages. A Concordia age of  $28.9 \pm$ 0.4 Ma ( $2\sigma$ , MSWD = 1.2; n=15) is obtained from the Oligocene zircon population (Fig. 483 15f). 484

485 **MN19** 

Zircon grains are generally euhedral to subhedral and show short to long prismatic habits (usually >100  $\mu$ m up to 300  $\mu$ m), with length/width ration of 2:1 to 3:1. Most of the grains are heterogeneous and inclusion bearing in BSE images and some of them contain xenocrystic cores. In CL images, most of the grains show oscillatory and sector zoning in rim domains (Fig. 15g). The measured Th/U values are higher than 0.4. Most of the spots performed on the growth domains provided concordant ages, with a Concordia age of 28.3 ± 0.4 Ma (2 $\sigma$ , MSWD = 0.68; n=21) (Fig. 15h).

493

#### 494 11. Molybdenite Re-Os geochronology

Two molybdenite-bearing samples were collected from V2b high-grade Mol veins (stage-2b mineralisation) from the core of the potassic alteration zone (Fig. 3 and Table 6) for Re-Os Mol geochronology (see Appendix). The two samples possess similar Re abundances (~19.3 and 22.1 ppm), and as they yield very similar Re-Os dates (below) also contain similar <sup>187</sup>Os

499 abundances (~12.2 and 13.9 ppb). The two samples yielded similar Re-Os dates of  $28.3 \pm 0.2$ 

and  $28.7 \pm 0.2$  Ma (Table 6). These ages are in excellent agreement with those previously

501 presented for the SKMD (ca. 29 – 28 Ma) by Simmonds et al. (2019), but are in contrast with

502 much older reported Re-Os Mol dates (ca. 42 Ma) by Nabatian et al. (2017b). The Re-Os Mol

503 dating results of this study and those of Simmonds et al. (2019) also overlap with the U-Pb

504 zircon ages (ca. 29-28 Ma) of the acidic magmatic bodies (samples MN01 and MN19) that

- 505 intrude the main alteration zone (Fig. 3).
- 506

#### 507 12. **Rb** -Sr geochronology of the alteration zone(s)

508 Several samples were selected for Rb-Sr geochronology in order to constrain the age of 509 the fluid-driven alteration processes within the SKMD. Only sample MN02 (rock matrix) 510 coming from the inner potassic-sodic alteration zone provided reliable results (see Table 7, 511 Fig. 16). In other samples either Rb/Sr values of minerals like sericite or Kfs were too low to 512 facilitate derivation of useful age information, or samples were affected by supergene 513 alteration. Results for these samples will be reported elsewhere.

514 For sample MN02, the mineral assemblage in the matrix consists of the early-stage 515 alteration products made of Kfs + Bt + Ap with minor amounts of Mol and Qz. Lately formed 516 Ms (sericite) is also documented. We analyzed Bt, two grain size fractions of Kfs, Ap, and 517 two grain size fractions of sericite. The age obtained for the primary assemblage Bt+ 2 Kfs+Ap is  $33.3 \pm 0.9$  Ma with a MSWD (mean square weighted deviation) of 0.16, 518 documenting that this sub-assemblage was in perfect Sr-isotopic equilibrium at that time. The 519 520 late-stage Ms was formed by fluid-rock interaction and most probably grew at the expense of 521 primary Kfs. Linear regression of the Rb-Sr data for 2 Kfs + Ap +2 Ms results in an age of 522  $28.5 \pm 1.8$  Ma (MSWD = 6.3). It appears that there is slight Sr-isotopic disequilibrium 523 between the two different grain size fractions of sericite, which slightly deteriorates the 524 precision of the age but does not affect its accuracy.

525

#### 526 13. Discussion

527 The zircon U-Pb geochronology from the ore-related intrusive bodies points to a

528 multistage magma emplacement in the study area that, based on the petrographic

529 characteristics, can be grouped in two main episodes of magmatism, at 33-30 Ma (basic to

530 intermediate in composition) and at ca. 29-28 Ma (acidic in composition), respectively. The

early episode of magmatism is compatible with the early Oligocene U-Pb zircon ages

reported in Simmonds et al. (2019) on the same mineralised body.

533 These magmatic episodes are associated with chronologically distinct

534 alteration/mineralisation types, as discriminated by the Rb-Sr multimineral and Re-Os Mol. 535 geochronology. In particular, the Rb-Sr geochronology documented an early Bt-Kfs-Ap 536 assemblage at ca. 33 Ma and a subsequent Ms-bearing one at ca. 28 Ma, thus coeval with the 537 early basic and late acidic magmatic episodes, respectively. We can thus infer the two-stage 538 Mo mineralisation as defined by the textures and mineralogical assemblages in the SKMD (Fig. 10) was polyphase and the consequence of : (i) an early low-grade, disseminated-type 539 mineralisation (stage-1) at ca. 33 Ma, associated with the pervasive potassic-sodic alteration 540 541 during basic to intermediate magmatism; and (ii) a subsequent, high-grade focused stockwork mineralisation (stage-2) at ca. 29-28 Ma, associated with Qz-Fsp veining and subsequent Cb 542 precipitation during acidic magmatism. This evidence suggests a long-lived (at least 4-5 Ma) 543 period of magma production, differentiation and emplacement to sustain the Mo 544

545 mineralisation episode(s) in the study area.

To form an economic porphyry Mo mineralisation, a significant and multiple magma 546 supply and differentiation within a large magma chamber, is needed to endow reasonable 547 548 quantities of Mo (e.g., Candela and Holland, 1986; Candela, 1992; Robb, 2004; Audétat, 2010; Zhang et al., 2018). The lack of Cu mineralisation in the SKMD, which is at odds with 549 550 the all known porphyry systems along the Urumieh-Dokhtar magmatic zone (supplementary 551 material #1), also supports a scenario of a long-lasting magma crystallisation and 552 differentiation in the chamber(s), which efficiently led to Cu extraction from the melt to the 553 crystallizing marginal zones into sulfide minerals (e.g., Candela and Holland, 1986; Blevin and Chappell, 1992; Candela, 1992; Klemm et al., 2008; Audétat, 2010). The residual melt 554 magma from such a chamber is then expected to show depletion in Cu and relative 555 556 enrichment in Mo. With progressive crystallization, a more acidic, fluid-rich magma is 557 produced which would be more fertile with respect to Mo (e.g., Candela and Holland, 1986; 558 Candela, 1992; Robb, 2004; Audétat, 2010; Seo et al., 2012).

The longevity of magma chamber(s) could be simply related to the amount of primitive magma nesting deep in the crust (Hawkesworth et al., 2000), the water content of the magma that prevented magma to rise to the same crustal levels unlike its drier equivalent (Robb, 2004), or input of numerous batches of magmas from the deep crustal melt reservoir (Glazner et al., 2004).

564

#### 565

#### 566 13.1. Source of the Mo-bearing magma

567 Apparently, there is a good correlation between Re concentration values in Mol and source 568 of the parental magma. In particular, high (hundred ppm) and low (few ppm) Re contents 569 have been reported in many studies and were referred to as mantle- or to crust-derived 570 magmas, respectively (e.g., Stein et al., 2001; Berzina et al., 2005; Mao et al., 2006; Wu et 571 al., 2017). However temperature of Mol deposition (Ishihara, 1988), the alteration type within 572 the same deposit (Newberry, 1979), and the availability of molybdenite could affect the amount of Re concentrations. Furthermore, based on Nd and Sr isotopic studies, Farmer and 573 DePaolo (1984) proposed that the Cu source is in the mantle and the Mo source in preexisting 574 575 crust. Many studies have also documented that the most fertile sources for syn- and postcollisional porphyry deposits is a subduction-modified lithosphere (Pettke et al., 2010; 576 577 Richards, 2015). Molybdenite samples from Tertiary porphyry deposits in Iran show a general positive 578 579 correlation between higher Re concentration and younger ages (Taghipour et al., 2008; Mirnejad et al., 2013; Aghazadeh et al., 2015; Simmonds and Moazzen, 2015; Nabatian et al., 580 581 2017a; 2017b). It was therefore proposed that the Neogene magmas along UDMZ were less 582 contaminated by crustal materials (Nabatian et al., 2017b). The Re contents in the molybdenite samples analysed in this study (19-22 ppm), together 583

584 with those presented in Simmonds et al. (2019), ranging 10-41 ppm, are remarkably lower 585 than all other porphyry Cu-Mo (Au) systems along the Urumieh-Dokhtar magmatic zone, 586 which are typically above 100 ppm (e.g., Aghazadeh et al., 2015). Significant in this regard is 587 the presence of inherited zircons within both the early (dated at 33-30 Ma) mafic to 588 intermediate and late (dated at 29-28 Ma) acidic magmatic rocks cropping out in the 589 mineralised area (Table 5). These inherited zircons also show resorbed and heterogeneous 590 textures (Fig. 15d), compatible with renewed melting and corrosion (Corfu et al., 2003). In 591 this regard, various papers have outlined the dominant role of juvenile lower crust in metal 592 endowment in porphyry Mo (e.g., Berzina et al., 2015; Liu et al., 2015; Zeng et al., 2015; 593 Zhang et al., 2018) and Cu deposits along southern UDMZ (Asadi et al., 2014). We therefore 594 suggest that the molybdenite Re contents from the SKMD most likely derive from a crustally-595 contaminated mantle sourced magma. This is in line with the calculated initial isotopic Sr 596 composition of ~0.70497 as derived from sample MN02 (Fig. 16) that is indicative of a 597 mantle-derived magma with minor crustal contaminations (Zindler and Hart, 1986).

598

#### 599 13.2. Structural control on fluid flow and mineralisation

600 The Mo mineralisation SKMD is structurally controlled as documented by the 601 development of a network of polyphase veining. Overall, the vein arrays are non-systematic, 602 as indicated by a nearly radial pattern of jointing within the main alteration zones (Fig. 6b-c). 603 and the Mol abundance in veins (V2a and V2b) decrease significantly when moving from the 604 inner potassic-sodic alteration zone focused on the main intrusive body to the outer Eocene 605 volcanic host rocks (Fig. 3). Similarly, the intensity of the phyllic alteration changes from 606 intense to low when crossing the contact between the mineralised porphyry and the volcanic host rocks. This evidence suggests that the original intrusive contacts constituted a major 607 608 structural/rheological boundary that likely acted as a permeability barrier for the fluids 609 exsolved from the crystallizing intrusion (e.g., Oliver and Bons, 2001; Rossetti et al., 2007). 610 Significantly, within the ore body the vein array is more systematic in the transition 611 from V2a to V2b mineralisation stages and in the marginal propylitic alteration zone (Fig. 6). In particular, the occurrence of a nearly orthogonal vein set in the potassic-sodic and 612 613 propylitic alteration zones points to a geological setting dominated by a horizontal biaxial tensile stress field in an extensional regime (vertical maximum principal stress  $\sigma_1$ ) and 614 615 characterized by reversal of the intermediate ( $\sigma_2$ ) and minimum ( $\sigma_3$ ) principal stress (Rives et 616 al., 1994; Caputo, 1995) between the NS and E-W paleo-directions. Since the main vein 617 strike distribution cluster around the E-W strike (Fig. 6), the regional  $\sigma_3$  paleo-direction 618 during the main stage of Mol mineralisation is inferred to trend NS. Therefore, a transition 619 from a nearly isotropic stress field to a differential stress regime dominated by a N-S directed  $\sigma_3$  direction is envisaged. This also suggests a different style of structural control on ore 620 621 mineralisation, likely controlled by the fluid pressure changes in the mineralising 622 environment: from early hydrofracturing and stockwork vein arrays under supralithostatic 623 fluid pressures conditions to late planar vein arrays formation under sub-lithostatic 624 conditions.

#### 625 13.3. Fluid evolution and the mineralising environment

No information is provided in this study regarding the fluid(s) geochemistry responsible for the main potassic-sodic alteration and associated disseminated stage-1 Mol precipitation. The main products of this stage fluid(s) (as disseminated and micro-vein types) are Kfs+Ab+Bt+Mt+Po, which is the common high-temperature mineral assemblage in the early stages (orthomagmatic) of porphyry systems (e.g., Shinohara et al., 1995; Seedorff et

631 al., 2005; Sinclair, 2007; Sillitoe, 2010), including Mo deposits (e.g., Questa, Bloom, 1981; and Hudson Bay Mountain, Klemm et al., 2008). This is in line with the compositional 632 633 layering in the porphyry stock (Fig. 5a), which suggests magma degassing close to the 634 carapace of the rising intrusive body (Sinclair, 2007; Lawley et al., 2010). A magmatic origin 635 for this stage-1 fluids is thus tentatively proposed indicated by the ubiquitous unidirectional 636 solidification textures observed in potassic-sodic alteration zone as it proposed at MAX 637 (Lawley et al., 2010) and Henderson (Shannon et al., 1982) porphyry Mo for the early stage 638 fluids (Lawley et al., 2010). The transition from an oxidising to a reducing mineralising environment is attested by the sequential formation of magnetite, to ilmenite and pyrrhotite 639 (Fig. 10), which can be associated to the Mol-I disseminated precipitation. 640 641 The early ore-bearing fluids (type-A1 and A2 FI) in V2a quartz veins (Mol-II mineralisation) show  $T_e$  values (-22.5°C to -33°C) consistent with a H<sub>2</sub>O-NaCl-KCl system 642 (Hall et al., 1988), with variable amounts of other cations such as Ca<sup>2+</sup>, Mg<sup>2+</sup>. Quartz 643 geochemistry has demonstrated a hydrothermal (magmatic) origin for the mineralising 644 645 fluid(s) (Fig. 11). The presence of CO<sub>2</sub> in the type-A1 FI may thus indicate an originally low-646 salinity and low CO<sub>2</sub>-bearing magma-derived fluid(s). The association of low saline fluids, 647 usually containing low amounts of CO<sub>2</sub>, is common in porphyry Mo deposits, in particular 648 within arc-related Mo deposits, such as Endako (Selby et al., 2000), Thompson Creek (e.g., Thompson Creek, Hall et al., 1984; Cannivan Gulch, Darling, 1994; Geumeum, Kim et al., 649 650 2016), and, apparently, it has no effect on the size or the ore-grade of the deposit (Taylor et 651 al., 2012). Cline and Bodnar (1991, 1994) have demonstrated that when magma 652 crystallization progresses, the released fluid(s) will have less than 10% chlorine 653 concentration. Since we do not have firm evidence of fluid boiling from the FI assemblage, 654 the rare type-B, high saline (up to 32 wt.% NaCl eq.) fluids in FIs can hence be interpreted as magma-sourced saline brines, segregated in the early stages of magma crystallization (Cline 655 656 and Bodnar, 1991; 1994). Alternatively, fluid boiling could have occurred at deeper levels, 657 predating the Mol mineralization. The vaporization of CO<sub>2</sub> during boiling could have 658 increased the pH, leading to precipitation of Cu from the fluids (Drummond and Ohmoto, 659 1985), whereas Mo remained dissolved in the residual fluids with lower salinity and CO<sub>2</sub> 660 content as hydroxyl (H<sub>2</sub>MoO<sub>4</sub>) and alkali molybdate complexes (Ulrich and Mavrogenes, 2008). In addition, boiling and the associated early metal deposition could have consumed the 661 662 sulphur, increasing further the solubility of Mo in the residual fluids (Xiaoyun, 1989; Ulrich 663 and Mavrogenes, 2008).

664 Fluids from the phyllic alteration zone (type-A3 FI) show no evidence of CO<sub>2</sub> but similar  $T_h$  (240-350 °C) and salinity (1.7-4.3 wt.% NaCl eq.). This is comparable with the Endako 665 666 porphyry Mo deposit, where the fluid chemistry associated with K-feldspar and sericite 667 alteration assemblages show no distinct differences (Selby et al., 2000). A significant 668 contribution of meteoric fluids is inferred due to the absence of CO<sub>2</sub> and the observed lower  $T_h$  and salinities. The paragenetic occurrence of Ser (Ms) with Mol+Cb mineralisation (Mol-669 670 III mineralisation) highlights the dominant role of neutralisation and cooling of acidic fluids 671 (Hemley and Hunt, 1992; Giggenbach, 1997) caused by percolating meteoric fluids and mixing, which also caused the sulphide precipitation. Moreover, the coincidence of high-672 grade Mo precipitation with very low salinity type-A4 FI could be explained by the 673 674 importance of hydroxyl complex (H<sub>2</sub>MoO<sub>4</sub>, Ulrich and Mavrogenes, 2008) for molybdenum 675 transportation (Candela and Holland, 1986) in temperatures below 300 °C (Xiaoyun, 1989). Since the Mo solubility decreases with increasing salinity (Keppler and Wyllie, 1991), Mo 676 677 precipitation should have been facilitated by continuous fluid exsolution and mixing, temperature/pressure fluctuations and/or pH changes (Audétat, 2010; Seo et al., 2012). 678

679 In order to better constrain the thermo-baric environment associated with Mo 680 mineralisation, FI isochores (Fig. 17a) have been constructed using the BULK and ISOC 681 softwares (Bakker, 2003). Isochores were chosen considering representative fluid inclusions for type-A1 (bulk density of 0.599 g/cc) and A-4 (bulk density of 0.826 g/cc) FI. Calculations 682 683 were based on the equations of state of Zhang and Frantz (1987) for density and Bodnar (1993), Bodnar and Vityk (1994), and Knight and Bodnar (1989) for the bulk composition. 684 685 The isochores are then combined with the T estimates as derived from the TitaniQ thermometry for pressure correction. For the type-A1, the area in P-T space defined by the 686 687 intersections of isochores with T estimates is centered at  $1.5 \pm 0.5$  kbar and  $580^{\circ} \pm 75^{\circ}$ C. Assuming lithostatic pressure conditions, this conforms to paleogeothermal gradient of ca. 688 689 100 °C/km and a depth of 7.4 km (based on a lithostatic pressure gradient of 27 MPa km<sup>-1</sup>) 690 for the early (V2a stage) mineralisation. This pressure/depth estimates for the mineralising 691 environment is consistent with those reported in many porphyry Mo deposits (Selby et al., 692 2000; Taylor et al., 2012; Ni et al., 2015). This is also compatible with the marginal 693 magmatic textures (compositional layering) preserved at the roof of the main mineralised 694 body (Fig. 5a). Using the same geothermal gradient and assuming a progressive transition 695 from lithostatic to hydrostatic pressure conditions, the intersection with the type-A4 isochore 696 provides pressure estimates ranging from 0.25 (based on a hydrostatic pressure gradient of 10 697 MPa km<sup>-1</sup>) to 0.75 kbar in a narrow temperature range of ca. 250 °C. These P-T estimates

constrain the high-grade Mo mineralisation during the V2b stage at shallower crustal levels
(ca. 3 km) during progressive exhumation of the mineralising system (Fig. 17a).

700 When the different FI types are plotted in a Th vs. salinity diagram, there is a 701 significant overlap in the populations of L-rich+V FI. Nonetheless, a general trend of 702 decreasing Th and salinities values from Type-A1, A2 and A3 (hosted by quartz) to type-A4 703 (hosted by paraspurite) FI can be observed (Fig. 17b). Three main groups of FI can be 704 discriminated in the Th vs. salinity diagram: (i) group-1, with the highest mean of Th, that 705 includes all the type-A1 and part of the type-A2 FI; (ii) group-2 that includes the type-A2 FI, 706 with a general trend to lower Th and salinity values; and (iii) group-3, formed by the type-A4 707 FI. Taking into consideration the fluid evolution processes typical of hydrothermal ore 708 deposits depicted in Wilkinson (2001), group-1 is consistent with isothermal mixing and group-2 with meteoric fluid mixing/dilution, respectively. Group-3 is instead consistent with 709 710 cooling and pressurisation (Fig. 17b-c). This evidence suggests a mineralising system recording the progressive evolution from a closed to an open fluid circulation system, 711 712 controlled by the progressive transition from a ductile to brittle environment, where rock fracturing has enhanced permeability and meteoric fluids percolation and mixing. 713

#### 714 13.4. A model for the Siah Kamar Mo mineralisation

715 Collectively, our results from the SKMD confirm a Mo mineralisation scenario 716 compatible with an Endako-type porphyry Mo deposit (see also Nabatian et al., 2017b). 717 Nonetheless, our results also point to peculiar characteristics for the SKMD. The main 718 peculiarity of the SKMD is a prolonged history of magma production and emplacement (ca. 719 4-5 Ma). In fact, despite that multiple magma intrusion processes are common in the 720 evolution of porphyry systems, the time lapse from early- to late-stage mineralisation is 721 usually less than 1 m.y. (Sillitoe, 2010; Mercer et al., 2015). Therefore the SKMD can be 722 classified as an uncommon porphyry Mo mineralization, which operated in a time scale of at 723 least 4-5 Ma. Such mineralisation time scales have been reported from giant and large porphyry systems (e.g., Chuquicamata, Correa et al., 2016; and Halasu, Xue et al., 2016). 724 725 The concurrence of carbonatization and Mol mineralisation is also a peculiar 726 characteristic of the SKMD. Carbonate precipitation is uncommon in porphyry systems 727 (Taylor et al., 2012) and carbonate veins commonly, which occurred in the latest stages, are 728 either barren (e.g., Selby et al., 2000; Seedorff and Einaudi, 2004; Mao et al., 2011; Ni et al., 729 2015) or with only minor Mo enrichment (e.g., Yang, 2007; Wang et al., 2017; Zhai et al., 730 2017).

731 Therefore, any model for the formation of the Mo Siah Kamar mineralisation should take 732 into consideration the following constraints: (i) the U-Pb zircon magmatic ages and the Rb-Sr 733 ages of the alteration mineralogy support a two-stage mineralisation process: early (at ca. 33 734 Ma) low-grade disseminated mineralization (stage-1, Mol-I) during potassic-sodic alteration 735 in a reducing environment and subsequent (at ca 28-29 Ma) high-grade stockwork Mo 736 mineralisation (stage-II, Mol-II and Mol-III, as constrained by the Re-Os Mol 737 geochronology) during phyllic alteration; (ii) the two-stage mineralisation is associated with 738 multiphase intrusions and transition from basic to acidic magmatism, in a time lapse of ca. 4-739 5 Ma; (iii) the mineralising fluid(s) records transition from a magmatic/hydrothermal- to a 740 meteoric-dominated system across a progressive ductile-to-brittle environment (temperature 741 range from ~600 to 250 °C); and (iv) the dominant role of carbonate precipitation and 742 neutralisation of acidic fluids in the Mo mineralisation (Mol+Ser+Cb assemblage) during fluid cooling. 743 Based on the synchronicity (in the time lapse 28-29 Ma) existing between the 744 crystallisation of the acidic dyke and stock at the rims of the mineralised zone (Fig. 3), 745 746 phyllic alteration and Mo-mineralisation, it is postulated that mineralising fluids were sourced 747 from these acidic intrusives. Nonetheless, these acidic magmatic rocks are Mo-barren. We 748 then propose a model of Mo mineralisation as primarily caused by the re-fertilisation (ore enhancement) induced by the renewed magma supply at 28-29 Ma. The exposed acidic 749 750 intrusions are considered as the apophyses of a magmatic system likely supported by a 751 voluminous magma chamber at depth, able to sustain highly perturbed geothermal conditions 752 and renewed fluid supply within the early formed potassic-sodic alteration zone formed at ca. 753 33 Ma (Fig. 18). In such a scenario, the mode of fluid exsolution was controlled and 754 modulated by the rheological evolution of the solidifying shell surrounding the cooling 755 magma. At high-temperature conditions (T>600 °C), fluid-rock equilibrium was likely 756 attained and fluids were entrapped in solidifying magma at depth (ca. 7 km). When 757 temperature decreased and brittle deformation prevailed, more fluids were released into the system and interacted with the ground waters (e.g., Giggenbach, 1997). At this stage (T< 758 759  $300^{\circ}$ C), the CO<sub>2</sub> dissolved in the fluids became reactive and the release of CO<sub>2</sub> caused 760 acidification of the fluids (Giggenbach and Soto, 1992; Kaszuba et al., 2005; Little and 761 Jackson, 2010) and significant interaction with the country rocks, leading to mineral 762 dissolution and metal leaching (Bandstra and Brantley, 2008; Liu et al., 2012; Kirsch et al., 763 2014; Lions et al., 2014) from the early-formed potassic-sodic alteration zone. The deposition 764 zone is thus interpreted as zone of fluid mixing with meteoric ground water, when fluid

- neutralization and temperature drop down leading to the Ms-Cb precipitation and metal (Mol)
   mineralisation in the system, focused by structurally-controlled pathways assisted by
   continuous fracturing of the intrusive carapace.
- 768

#### 769 13.5. Correlation at regional scale

770 Similar Eocene-Oligocene high-K calc-alkaline to shoshonitic magmatism as reported 771 from the SKMD (Khaleghi et al., 2013; Nabatian et al., 2017b) has been also reported in 772 many studies along the UDMZ, AMZ, BAMZ, Lesser Caucasus, and the western Alborz (e.g., Aftabi and Atapour, 2000; Ahmadian et al., 2009; Aghazadeh et al., 2010; Sarjoughian 773 et al., 2012; Kananian et al., 2014; Jamali and Mehrabi, 2015) (Fig. 1). Within this context, 774 775 the SKMD shows the oldest reported ages among the porphyry deposits within the UDMZ (except the Kal-e-Kafi porphyry Cu-Mo deposit, with zircon U-Pb age of 52-50 Ma, 776 777 Ahmadian et al., 2016), and only two porphyry Cu-Mo-Au deposits have been reported within the same time window including Bondar-e-Hanza (with Mol Re-Os age of  $28.4 \pm 0.5$ 778 779 Ma, Aghazadeh et al., 2015) and Reagan (with zircon U-Pb age of  $29.3 \pm 0.2$  Ma, Hassanzadeh, 1993) (Fig. 1 and supplementary material #1). 780 781 By contrast, many porphyry Cu-Mo deposits with older (Eocene) and similar (Oligocene) ages have been reported in the AMZ (Haftcheshmeh Cu-Mo, Mol Re-Os age of 28.2-27.1 782 Ma, Aghazadeh et al., 2015) and Lesser Caucasian magmatic zones (Kadjaran Cu-Mo; 27.3-783

- 26.4 Ma, Rezeau et al., 2016). Nonetheless, apart from the SKMD, none of these porphyry
- systems shows Mo as the first commodity. In particular, whereas the Mo/Cu ratio within

786 SKMD is about 1, the coeval porphyry Cu-Mo deposits of Kadjaran, Haftcheshmeh, and Kal-

- e Kafi show values of about 0.1, which are significantly higher than the Miocene ones
- 788 (supplementary material #1). The Mo enrichment in the Eocene-Oligocene porphyry systems
- could be attributed to either to a relatively deeper emplacement level of the metal-bearing
- 790 magma (Candela, 1986) or to the contribution and re-fertilisation of a preexisting Mo-
- enriched crustal source (Farmer and DePaolo, 1984; Wan et al., 2018).
- 792

#### 793 14. Conclusions

The main results of this study can be summarised as follows:

(1) The SKMD developed during a long-lived magmatic and alteration history, spanning

- from ca. 33 Ma to 29-28 Ma, associated with a diachronous, low- and high-grade Mo
- 797 mineralisation.

- (2) Re-fertilisation of an early formed disseminated Mo ore by renewed magmatism and
- associated magma degassing and fluid exsolution is proposed as the scenario for ore
- 800 enhancement and, consequently, for high-grade Mo mineralisation.
- 801 (3) Leaching of metals from the early formed disseminated ore bodies operated by acidic
- 802 fluids exsolved from the cooling magmas is recognised as the cause of Mo-enrichment in the 803 mineralising fluids.
- (4) Mixing of magmatic and meteoric fluids during progressive embrittlement (and cooling)
  of the mineralising environment is considered as the key factor leading to the high-grade Mo
  ore formation, through progressive neutralisation of the acidic fluids via structurallycontrolled carbonation and sericitisation.
- 808 Comparison at a regional scale indicates that parameters such as longevity of magma 809 supply, progressive magma crystallization/differentiation and the presence of a possible pre-
- 810 enriched crustal material should be considered for the Mo endowment in the UMDZ.
- 811

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827	
828	APPENDIX:
829	Analytical techniques
830	LA-ICP-MS quartz trace element geochemistry
831	Based on petrographic observations (optical and BSE) and mineral chemistry on polished
832	thin-section samples, 16 representative quartz crystals were selected for in situ trace and
833	REEs element analyses, using laser ablation (LA, CETAC LSX-213 laser system) combined
834	with an inductively coupled plasma mass spectrometer (ICP-MS, AGILENT 7700) at Institut
835	für Mineralogie, Universität Stuttgart. The following laser ablation parameters were used:
836	spot diameter 50 $\mu$ m, 60% laser power (100% = 4 mJ at a spot diameter of 150 $\mu$ m), 10 Hz
837	laser pulse repetition rate, and a mixed helium and argon gas flow with 500 ml/min and 800
838	ml/min, respectively, serving as carrier of the ablated material.
839	The data acquisition was performed using the Agilent Mass Hunter software (version
840	B.01.01). Each analysis comprised the acquisition of the individual background ion
841	intensities (gas-blank) for roughly 23 s followed by the acquisition of the ion intensities on
842	laser irradiation of the sample spot for roughly 57 s.
843	The method involved monitoring of 18 isotopes (7Li, 9Be, 11B, 27Al, 28,29Si, 31P, 44Ca,
844	47,49Ti, 55Mn, 56,57Fe, 71Ga, 72,73Ge, 75As, 85Rb, 88Sr; this selection excludes isobaric
845	interferences of isotopes from other elements) using individual integration times of 10 ms.
846	Reference materials were DLH7 and DLH8 glasses from P&H Developments Ltd., and
847	NIST (National Institute of Standards and Technology, USA) 612 and 610 glasses. Certified,
848	recommended and proposed values for these reference materials are from Jochum et al.
849	(2011). The validity of the calibration, data evaluation, and reproducibility were checked
850	with the reference materials Diorite (DRN) and Zinnwaldite (ZW-C) from Service
851	d'Analyses des Roches et des Mineraux du CNRS. Lithium-borate glass disks of both were
852	prepared using 0.6 g of powder of these reference materials and 3.6 g of lithium borate. All
853	elemental concentrations were calculated relative to the silicon elemental concentration as
854	internal standard which was determined from the results of the 29Si isotope. The absolute
855	silicon concentration was previously measured with the EMP. The same spot target was
856	applied for both EMP and LA-ICP-MS. Precision is estimated better than 5%. The software
857	used for calibration and data evaluation was developed by J. Opitz (Massonne et al., 2013;
858	Baggio et al., 2018).
0.50	

#### 860 Fluid inclusion microthermometry

861 Samples representative of the different mineralised quartz-carbonate vein sets were prepared as 150- µm-thick doubly polished wafers and petrography studies were carried out with a 862 863 polarizing microscope to define the types of fluid inclusions, the populations and their 864 genetic relationships. Microthermometric measurements were made utilizing a Linkam 865 THMSG600 freezing-heating stage at the Istituto di Geologia Ambientale e Geoingegneria 866 of CNR at the University of Roma "La Sapienza". The systems were calibrated using pure 867 natural CO2 inclusions for low temperatures and synthetic pure water fluid inclusions for high temperatures, respectively. Reproducibility of the triple point of CO<sub>2</sub> at -56.6 °C and 868 pure H<sub>2</sub>O critical point at 374 °C were  $\pm 0.3$  °C and  $\pm 0.5$  °C respectively. Low temperature 869 (Te, Tm) data were determined with a 1–5 °C/min rate, whereas high temperature (Th, Tms) 870 871 data with a heating rate of 5–20°C/min, respectively. Bulk composition and Isochores for various FI types were calculated by BULK and ISOC softwares provided by (Bakker, 2003). 872 873 **Raman spectroscopy** 874

A HORIBA LabRAM HR Raman spectrometer at the Dipartimento di Scienze, Università degli Studi di Roma Tre, calibrated using a silicon standard, was employed to acquire the 875 876 Raman spectra. The micro-Raman spectrometer is equipped with a green argon ion laser (532 nm), focused through a 50× and 100× objectives. Optical filters were employed in order to 877 878 achieve ~2.5 mW at the surface of the glassy interface. The Raman systems were set with 600 879 T and 1200 T gratings, exposure time 20 s (3 times), confocal hole of 200–500 µm, and slit of 880 200  $\mu$ m, and the spatial resolution of the sample surface is ~1  $\mu$ m (theoretical spatial resolution = 0.61  $\lambda$ /NA where  $\lambda$  is the wavelength of the laser (532 nm), and NA is the 881 882 numerical aperture of the microscope objective being used (0.9 for 100×). Raman spectra were acquired over the range from 90 to 2000 cm<sup>-1</sup>. Reference values from Frezzotti et al. 883 884 (2012) for common gases inside fluid inclusions and some minerals were used to interpret the 885 obtained spectra. Furthermore, the CrystalSleuth software (http://rruff.info/) was employed to 886 define the accessory minerals.

#### 887 Zircon U-Pb Geochronology

888 The zircon U-Pb geochronology study was carried out at the Department of Earth and

889 Environmental Sciences of Nagoya University. Four samples were selected for zircon grains

890 separation. More than 10 kg for each sample were collected and crushed. Only few zircon

- grains (ca. 30) were found. Some of them were useless due to the small grain size ( $< 30 \square m$ )
- 892 or to the strongly fractured crystal structure. Cathodoluminescence (CL) and Back Scatter

893 (BSE) imaging was used first to gather information on the grain texture and internal growth and/or alteration zoning. Zircon grains with intense fracturing and inclusions were avoided. 894 895 The zircon grains were analyzed by LA-ICP-MS (Agilent 7700XICPMS machine connected 896 with NWR213 (Electro Scientific Industries) laser ablation system (Kouchi et al., 2015). A 897 standard glass (NIST SRM 610) and two zircon standards, named 91500 (1059 Ma, 898 Wiedenbeck et al., 1995) and OD-3 (33.1 Ma, Iwano et al., 2013) were used. Blanks, the 899 zircon standards, and the standard glass were measured at the beginning and ending of each 900 measurement cycle. Eight points were measured in each cycle. The ISOPLOT V4.15 901 software (Ludwig, 2011) was utilized to calculate the Concordia, statistics and to prepare the 902 age plots. Correction for the common Pb was performed using <sup>204</sup>Pb intensity (Cox and 903 Wilton, 2006). Most of the results with high common Pb (values more than 20 %) were excluded from calculations. 904

#### 905 Molybdenite Re-Os geochronology

906 Molybdenite grains were isolated from samples and then gently were milled, washed and 907 dried (Selby and Creaser, 2004). Impurities were separated under microscope and about 100 908 mg of pure molybdenite was prepared for each sample with the analytical protocol following 909 that of (Selby and Creaser, 2004). The Re-Os molybdenite analysis were performed in the 910 Laboratory for Sulfide and Source Rock Geochronology and Geochemistry and the Arthur 911 Holmes Laboratory in the Durham Geochemistry Centre at Durham University (UK). In brief, molybdenite samples were dissolved and equilibrated with a known amount of <sup>185</sup>Re 912 and isotopically normal Os in inverse aqua regia (2:1 16 N HNO3 and 12 N HCl, 3 mL) at 913 914 240 °C for 24 h in a Carius-tube. Rhenium and Os were isolated and purified by solvent 915 extraction, microdistillation, and anion exchange chromatography, and analyzed by negative 916 thermal ionization mass spectrometry on a ThermoScientific TRITON mass spectrometer 917 using Faraday collectors. Total procedural blanks for Re and Os 2.5 pg and 0.1 pg, 918 respectively, with an Os blank composition of 0.21 0.02 (n 6), which are insignificant to the 919 Re and Os levels in the analysed samples. Rhenium and Os concentrations, and Re-Os 920 molybdenite date uncertainties are presented at the  $2\sigma$  level, which includes the uncertainties 921 in Re and Os mass spectrometer measurement, spike and standard Re and Os isotopic 922 compositions, and calibration uncertainties. Because a mixed tracer solution is used, 923 uncertainties in weights of sample and tracer solution do not affect the calculated age, and are 924 not considered. However, sample and tracer solution weight uncertainties are considered in

- determining the uncertainty in the Re and <sup>187</sup>Os concentrations. Uncertainty in the <sup>187</sup>Re
  decay constant (0.35%, Smoliar et al., 1996)) is also considered.
- 927

#### 928 **Rb-Sr multimineral geochronology**

929 To determine the ages of mineral assemblage crystallization, we employed the Rb-Sr 930 internal mineral isochron approach. Age information mainly is based on the high-Rb/Sr 931 phases muscovite/sericite, and biotite. Mineral separates were prepared following Glodny et 932 al. (2008). Crushing was performed by hand using pestle and mortar combined with repeated 933 sieving, to preserve the original grain sizes of minerals. Rb–Sr isotopic data were acquired at 934 GFZ Potsdam using a Thermo Scientific TRITON thermal ionization mass spectrometer. Sr isotopic compositions were measured in dynamic multi-collection mode. Rb isotope dilution 935 936 analysis was done in static multi-collection mode. The value obtained for <sup>87</sup>Sr/<sup>86</sup>Sr of the NIST SRM 987 reference material was  $0.710255 \pm 0.000024$  ( $2\sigma$ , n = 23). For age 937 calculation, uncertainties of  $\pm 0.005\%$  for  ${}^{87}$ Sr/ ${}^{86}$ Sr and of  $\pm 1.5\%$  for  ${}^{87}$ Rb/ ${}^{86}$ Sr ratios were 938 939 assigned to the results. Uncertainties of isotope and age data are quoted at  $2\sigma$ . The program 940 ISOPLOT/EX 4.15 (Ludwig, 2011) was used for calculating regression lines. Decay .d b. constants are those recommended by Villa et al. (2015). 941 942

#### 944 References

- 945 Aftabi, A., Atapour, H., 2000. Regional aspects of shoshonitic volcanism in Iran.
- 946 Episodes 23, 119-125.
- Aghanabati, A., 2004. Geology of Iran, GSI Publication, Tehran. 947
- 948 Aghazadeh, M., Castro, A., Omran, N.R., Emami, M.H., Moinvaziri, H., Badrzadeh, Z.,
- 949 2010. The gabbro (shoshonitic)-monzonite-granodiorite association of Khankandi
- 950 pluton, Alborz Mountains, NW Iran. Journal of Asian Earth Sciences 38, 199-219.
- 951 Aghazadeh, M., Hou, Z., Badrzadeh, Z., Zhou, L., 2015. Temporal-spatial distribution
- 952 and tectonic setting of porphyry copper deposits in Iran: Constraints from zircon U-Pb 953 and molybdenite Re-Os geochronology. Ore Geology Reviews 70, 385-406.
- 954 Ahmadian, J., Haschke, M., McDonald, I., Regelous, M., RezaGhorbani, M., Emami,
- 955 M.H., Murata, M., 2009. High magmatic flux during Alpine-Himalayan collision:
- 956 Constraints from the Kal-e-Kafi complex, central Iran. Geological Society of America
- 957 Bulletin 121, 857-868.
- Ahmadian, J., Sarjoughian, F., Lentz, D., Esna-Ashari, A., Murata, M., Ozawa, H., 2016. 958
- 959 Eocene K-rich adakitic rocks in the Central Iran: implications for evaluating its Cu-Au-960 Mo metallogenic potential. Ore Geology Reviews 72, 323-342.
- 961 Amidi, M., Alavi Tehrani, M., Lotfi, P., Haghipour, A., Sabzehei, M., Behroozi, A.,
- 962 Lescuyer, J., Riou, R., 1987. Geological map of Mianeh. Geological Survey of Iran, 963 Tehran.
- 964 Aria Kansar Samin Co., u.r., 2013. Report on exploratory operations of Siah-Kamar 965 molybdenum occurrence.
- Asadi, S., Moore, F., Zarasvandi, A., 2014. Discriminating productive and barren 966
- 967 porphyry copper deposits in the southeastern part of the central Iranian volcano-plutonic 968 belt, Kerman region, Iran: a review. Earth-Science Reviews 138, 25-46.
- 969 Audétat, A., 2010. Source and evolution of molybdenum in the porphyry Mo (-Nb) 970 deposit at Cave Peak, Texas. Journal of Petrology 51, 1739-1760.
- 971 Audetat, A., Pettke, T., Heinrich, C., Bodnar, R., 2008. The composition of magmatic-972 hydrothermal fluids in barren and mineralized intrusions: ecoNomic geology, v. 103.
- 973
- Ayati, F., Yavuz, F., Asadi, H.H., Richards, J.P., Jourdan, F., 2013. Petrology and 974 geochemistry of calc-alkaline volcanic and subvolcanic rocks, Dalli porphyry copper-
- 975 gold deposit. Markazi Province, Iran. International Geology Review 55, 158-184.
- 976 Ayres, L., Averill, S., Wolfe, W., 1982. An Archean molybdenite occurrence of possible 977 porphyry type at Setting Net Lake, northwestern Ontario, Canada. Economic Geology 77, 1105-1119. 978
- 979 Babazadeh, S., Ghorbani, M.R., Bröcker, M., D'Antonio, M., Cottle, J., Gebbing, T.,
- 980 Carmine Mazzeo, F., Ahmadi, P., 2017. Late Oligocene-Miocene mantle upwelling and
- 981 interaction inferred from mantle signatures in gabbroic to granitic rocks from the
- 982 Urumieh-Dokhtar arc, south Ardestan, Iran. International Geology Review 59, 1590-
- 983 1608.
- 984 Baggio, S.B., Hartmann, L.A., Lazarov, M., Massonne, H.-J., Opitz, J., Theye, T.,
- 985 Viefhaus, T., 2018. Origin of native copper in the Paraná volcanic province, Brazil,

- 986 integrating Cu stable isotopes in a multi-analytical approach. Mineralium Deposita 53,987 417-434.
- 988 Bakker, R.J., 2003. Package FLUIDS 1. Computer programs for analysis of fluid
- inclusion data and for modelling bulk fluid properties. Chemical Geology 194, 3-23.
- 990 Ballato, P., Cifelli, F., Heidarzadeh, G., Ghassemi, M.R., Wickert, A.D., Hassanzadeh, J.,
- Dupont-Nivet, G., Balling, P., Sudo, M., Zeilinger, G., Schmitt, A.K., Mattei, M.,
- 992 Strecker, M.R., 2017a. Tectono-sedimentary evolution of the northern Iranian Plateau:
- insights from middle-late Miocene foreland-basin deposits. Basin Research 29, 417-446.
- 994 Ballato, P., Cifelli, F., Heidarzadeh, G., Ghassemi, M.R., Wickert, A.D., Hassanzadeh, J.,
- 995 Dupont Nivet, G., Balling, P., Sudo, M., Zeilinger, G., 2017b. Tectono sedimentary
- 996 evolution of the northern Iranian Plateau: insights from middle-late Miocene
- 997 foreland basin deposits. Basin Research 29, 417-446.
- Bandstra, J.Z., Brantley, S.L., 2008. Surface evolution of dissolving minerals
- 999 investigated with a kinetic Ising model. Geochimica et Cosmochimica Acta 72, 2587-1000 2600.
- Berberian, M., King, G., 1981. Towards a paleogeography and tectonic evolution of Iran.
  Canadian Journal of Earth Sciences 18, 210-265.
- 1003 Berzina, A., Berzina, A., Gimon, V., Bayanova, T., Kiseleva, V.Y., Krymskii, R.S.,
- 1004 Lepekhina, E., Palesskii, S., 2015. The Zhireken porphyry Mo ore-magmatic system
- 1005 (eastern Transbaikalia): U–Pb age, sources, and geodynamic setting. Russian Geology1006 and Geophysics 56, 446-465.
- 1007 Berzina, A.N., Sotnikov, V.I., Economou-Eliopoulos, M., Eliopoulos, D.G., 2005.
- 1008 Distribution of rhenium in molybdenite from porphyry Cu–Mo and Mo–Cu deposits of 1009 Russia (Siberia) and Mongolia. Ore Geology Reviews 26, 91-113.
- 1010 Blevin, P.L., Chappell, B.W., 1992. The role of magma sources, oxidation states and
- 1011 fractionation in determining the granite metallogeny of eastern Australia. Earth and
- 1012 Environmental Science Transactions of The Royal Society of Edinburgh 83, 305-316.
- 1013 Bloom, M.S., 1981. Chemistry of inclusion fluids; stockwork molybdenum deposits from
- 1014 Questa, New Mexico, Hudson Bay Mountain and Endako, British Columbia. Economic1015 Geology 76, 1906-1920.
- 1016 Bodnar, R., 1993. Revised equation and table for determining the freezing point
- 1017 depression of H2O-NaCl solutions. Geochimica et Cosmochimica Acta 57, 683-684.
- 1018 Bodnar, R., Vityk, R., 1994. Interpretation of microthermometric data for H<sub>2</sub>O-NaCl
- 1019 fluid inclusions. Fluid inclusions in minerals: Methods and applications. . Short course of
- 1020 the working group, Inclusions in Minerals, 117-130.
- 1021 Candela, P.A., 1992. Controls on ore metal ratios in granite-related ore systems: an
- 1022 experimental and computational approach. Earth and Environmental Science
- 1023 Transactions of The Royal Society of Edinburgh 83, 317-326.
- 1024 Candela, P.A., Holland, H.D., 1986. A mass transfer model for copper and molybdenum
- in magmatic hydrothermal systems; the origin of porphyry-type ore deposits. EconomicGeology 81, 1-19.
- 1027 Caputo, R., 1995. Evolution of orthogonal sets of coeval extension joints. Terra Nova 7,1028 479-490.

- 1029 Castro, A., Aghazadeh, M., Badrzadeh, Z., Chichorro, M., 2013. Late Eocene-Oligocene
- 1030 post-collisional monzonitic intrusions from the Alborz magmatic belt, NW Iran. An
- 1031 example of monzonite magma generation from a metasomatized mantle source. Lithos
- 1032 180-181, 109-127.
- 1033 Cathles, L.M., Erendi, A., Barrie, T., 1997. How long can a hydrothermal system be 1034 sustained by a single intrusive event? Economic Geology 92, 766-771.
- 1035 Chiaradia, M., Merino, D., Spikings, R., 2009. Rapid transition to long-lived deep crustal
- 1036 magmatic maturation and the formation of giant porphyry-related mineralization
- 1037 (Yanacocha, Peru). Earth and Planetary Science Letters 288, 505-515.
- 1038 Chiu, H.-Y., Chung, S.-L., Zarrinkoub, M.H., Mohammadi, S.S., Khatib, M.M., Iizuka,
- 1039 Y., 2013. Zircon U–Pb age constraints from Iran on the magmatic evolution related to
- 1040 Neotethyan subduction and Zagros orogeny. Lithos 162, 70-87.
- 1041 Cline, J.S., Bodnar, R.J., 1991. Can economic porphyry copper mineralization be
- generated by a typical calc alkaline melt? Journal of Geophysical Research: Solid Earth96, 8113-8126.
- 1044 Cline, J.S., Bodnar, R.J., 1994. Direct evolution of brine from a crystallizing silicic melt 1045 at the Questa, New Mexico, molybdenum deposit. Economic Geology 89, 1780-1802.
- 1046 Colville, A.A., Colville, P.A., 1977. Paraspurrite, a new polymorph of spurrite from Inyo1047 County, California. American Mineralogist 62, 1003-1005.
- 1048 Corfu, F., Hanchar, J.M., Hoskin, P.W.O., Kinny, P., 2003. Atlas of Zircon Textures.
  1049 Reviews in Mineralogy and Geochemistry 53, 469-500.
- 1050 Correa, K.J., Rabbia, O.M., Hernández, L.B., Selby, D., Astengo, M., 2016. The timing
- 1051 of magmatism and ore formation in the El Abra porphyry copper deposit, northern Chile:
- 1052 Implications for long-lived multiple-event magmatic-hydrothermal porphyry systems.
- 1053 Economic Geology 111, 1-28.
- Cox, R.A., Wilton, D.H., 2006. U–Pb dating of perovskite by LA-ICP-MS: an example
  from the Oka carbonatite, Quebec, Canada. Chemical Geology 235, 21-32.
- 1056 Darling, R.S., 1994. Fluid inclusion and phase equilibrium studies at the Cannivan Gulch
- 1057 molybdenum deposit, Montana, USA: Effect of  $\overline{CO}_2$  on molybdenite-powellite stability.
- 1058 Geochimica et Cosmochimica Acta 58, 749-760.
- Dennen, W., 1966. Stoichiometric substitution in natural quartz. Geochimica et
   Cosmochimica Acta 30, 1235-1241.
- 1061 Doebrich, J.L., Garside, L.J., Shawe, D.R., 1996. Characterization of mineral deposits in
- 1062 rocks of the Triassic to Jurassic magmatic arc of western Nevada and eastern California.1063 Citeseer.
- Drummond, S., Ohmoto, H., 1985. Chemical evolution and mineral deposition in boilinghydrothermal systems. Economic Geology 80, 126-147.
- Duke, C.P., 2007. Pidgeon Molybdenum project: Toronto, Ontario, Wardrop Engineering
   Inc., NI 34–101 Compliant Report for MPH Ventures Corporation. 92.
- 1068 Farmer, G.L., DePaolo, D.J., 1984. Origin of Mesozoic and Tertiary granite in the
- 1069 western United States and implications for Pre-Mesozoic crustal structure: 2. Nd and Sr
- 1070 isotopic studies of unmineralized and Cu- and Mo-mineralized granite in the Precambrian
- 1071 Craton. Journal of Geophysical Research: Solid Earth 89, 10141-10160.

- 1072 Fazel, E.T., Mehrabi, B., Shabani, A.A.T., 2015. Kuh-e Dom Fe-Cu-Au prospect,
- 1073 Anarak metallogenic complex, Central Iran: a geological, mineralogical and fluid
- 1074 inclusion study. Mineralogy and Petrology 109, 115-141.
- Frezzotti, M.L., Tecce, F., Casagli, A., 2012. Raman spectroscopy for fluid inclusionanalysis. Journal of Geochemical Exploration 112, 1-20.
- 1077 Giggenbach, W., 1997. The origin and evolution of fluids in magmatic-hydrothermal1078 systems. Geochemistry of hydrothermal ore deposits.
- 1079 Giggenbach, W.F., Soto, R.C., 1992. Isotopic and chemical composition of water and
- 1080 steam discharges from volcanic-magmatic-hydrothermal systems of the Guanacaste
- 1081 Geothermal Province, Costa Rica. Applied Geochemistry 7, 309-332.
- 1082 Glazner, A.F., Bartley, J.M., Coleman, D.S., Gray, W., Taylor, R.Z., 2004. Are plutons
- assembled over millions of years by amalgamation from small magma chambers? GSA
  today 14, 4-12.
- 1085 Glodny, J., Kühn, A., Austrheim, H., 2008. Geochronology of fluid-induced eclogite and
- 1086 amphibolite facies metamorphic reactions in a subduction-collision system, Bergen Arcs,
- 1087 Norway. Contributions to Mineralogy and Petrology 156, 27-48.
- Hall, D.L., Sterner, S.M., Bodnar, R.J., 1988. Freezing point depression of NaCl-KCl-H
  2 O solutions. Economic Geology 83, 197-202.
- 1090 Hall, W., Schmidt, E., Howe, S., Broch, M., 1984. The Thompson Creek, Idaho,
- 1091 porphyry molybdenum deposit—an example of a fluorine-deficient molybdenum
- 1092 granodiorite system, International Association on the Genesis of Ore Deposits 6th
- 1093 Symposium, Tbilisi 1982, Proceedings, pp. 349-357.
- 1094 Haschke, M., Ahmadian, J., Murata, M., McDonald, I., 2010. Copper Mineralization
- Prevented by Arc-Root Delamination during Alpine-Himalayan Collision in Central Iran.
  Economic Geology 105, 855-865.
- 1097 Hassanpour, S., Alirezaei, S., Selby, D., Sergeev, S., 2014. SHRIMP zircon U-Pb and
- 1098 biotite and hornblende Ar–Ar geochronology of Sungun, Haftcheshmeh, Kighal, and
- 1099 Niaz porphyry Cu–Mo systems: evidence for an early Miocene porphyry-style
- 1100 mineralization in northwest Iran. International Journal of Earth Sciences 104, 45-59.
- 1101 Hassanzadeh, J., 1993. Metallogenic and tectono-magmatic events in the SE sector of the
- 1102 Cenozoic active continental margin of Iran (Shahr-e-Babak area, Kerman province). (Ph.
- 1103 D. Thesis). University of California.
- 1104 Hassanzadeh, J., Stockli, D.F., Horton, B.K., Axen, G.J., Stockli, L.D., Grove, M.,
- 1105 Schmitt, A.K., Walker, J.D., 2008. U-Pb zircon geochronology of late Neoproterozoic-
- 1106 Early Cambrian granitoids in Iran: Implications for paleogeography, magmatism, and
- 1107 exhumation history of Iranian basement. Tectonophysics 451, 71-96.
- 1108 Hawkesworth, C., Blake, S., Evans, P., Hughes, R., Macdonald, R., Thomas, L., Turner,
- 1109 S., Zellmer, G., 2000. Time scales of crystal fractionation in magma chambers—
- 1110 integrating physical, isotopic and geochemical perspectives. Journal of Petrology 41,
- 1111
   991-1006.
- 1112 Heidari, S.M., Daliran, F., Paquette, J.-L., Gasquet, D., 2015. Geology, timing, and
- 1113 genesis of the high sulfidation Au (–Cu) deposit of Touzlar, NW Iran. Ore Geology
- 1114 Reviews 65, 460-486.

- 1115 Heidarzadeh, G., Ballato, P., Hassanzadeh, J., Ghassemi, M.R., Strecker, M.R., 2017.
- 1116 Lake overspill and onset of fluvial incision in the Iranian Plateau: Insights from the
- 1117 Mianeh Basin. Earth and Planetary Science Letters 469, 135-147.
- 1118 Heinhorst, J., Lehmann, B., Ermolov, P., Serykh, V., Zhurutin, S., 2000. Paleozoic
- 1119 crustal growth and metallogeny of Central Asia: evidence from magmatic-hydrothermal
- 1120 ore systems of Central Kazakhstan. Tectonophysics 328, 69-87.
- 1121 Hemley, J., Hunt, J., 1992. Hydrothermal ore-forming processes in the light of studies in
- rock-buffered systems; II, Some general geologic applications. Economic Geology 87,23-43.
- 1124 Hosseini, M.R., Hassanzadeh, J., Alirezaei, S., Sun, W., Li, C.-Y., 2017. Age revision of
- 1125 the Neotethyan arc migration into the southeast Urumieh-Dokhtar belt of Iran:
- 1126 Geochemistry and U–Pb zircon geochronology. Lithos 284-285, 296-309.
- Huang, R., Audétat, A., 2012. The titanium-in-quartz (TitaniQ) thermobarometer: A
  critical examination and re-calibration. Geochimica et Cosmochimica Acta 84, 75-89.
- Huber, H., 1977. Geological map of Iran at scale of 1: 1,000,000, National Iranian OilCo., Tehran, Iran.
- 1131 Innocenti, F., Mazzuoli, R., Pasquare, G., Radicati di Brozolo, F., Villari, L., 1982.
- 1132 Tertiary and quaternary volcanism of the Erzurumkars area (Eastern Turkey):
- 1133 geochronological data and geodynamic evolution. Journal of Volcanology and
- 1134 Geothermal Research 13, 223-240.
- 1135 Ishihara, S., 1988. Rhenium contents of molybdenites in granitoid-series rocks in Japan.
- 1136 Economic Geology 83, 1047-1051.
- 1137 Iwano, H., Orihashi, Y., Hirata, T., Ogasawara, M., Danhara, T., Horie, K., Hasebe, N.,
- Sueoka, S., Tamura, A., Hayasaka, Y., 2013. An inter laboratory evaluation of OD 3
  zircon for use as a secondary U–Pb dating standard. Island Arc 22, 382-394.
- Jamali, H., Mehrabi, B., 2015. Relationships between arc maturity and Cu–Mo–Au
- 1141 porphyry and related epithermal mineralization at the Cenozoic Arasbaran magmatic belt.
- 1142 Ore Geology Reviews 65, 487-501.
- 1143 Janković, S., 1982. Yugoslavia, in Dunning, F.W., Mykura, W., and Slater, D., eds.,
- 1144 Mineral deposits of Europe, v. 2, Southeast Europe: Institution of Mining and
- 1145 MetallurgyMineralogical Society, London, p. 143–202.
- 1146 Jochum, K.P., Weis, U., Stoll, B., Kuzmin, D., Yang, Q., Raczek, I., Jacob, D.E., Stracke,
- 1147 A., Birbaum, K., Frick, D.A., 2011. Determination of reference values for NIST SRM
- 610–617 glasses following ISO guidelines. Geostandards and Geoanalytical Research 35,397-429.
- John, D., Ayuso, R., Barton, M., Blakely, R., Bodnar, R., Dilles, J., Gray, F., Graybeal,
- 1151 F., Mars, J., McPhee, D., 2010. Porphyry copper deposit model, chap. B of Mineral
- 1152 deposit models for resource assessment. US Geological Survey Scientific Investigations
- 1153 Report 2010, 1-169.
- John, D.A., Taylor, R.D., 2016. By-products of porphyry copper and molybdenum
- 1155 deposits: Chapter 7.
- 1156 Kananian, A., Sarjoughian, F., Nadimi, A., Ahmadian, J., Ling, W., 2014. Geochemical
- 1157 characteristics of the Kuh-e Dom intrusion, Urumieh–Dokhtar Magmatic Arc (Iran):

- 1158 Implications for source regions and magmatic evolution. Journal of Asian Earth Sciences1159 90, 137-148.
- 1160 Kaszuba, J.P., Janecky, D.R., Snow, M.G., 2005. Experimental evaluation of mixed fluid
- reactions between supercritical carbon dioxide and NaCl brine: Relevance to the integrity of a geologic carbon repository. Chemical Geology 217, 277-293.
- 1163 Keith, J.D., Shanks, W.C., Archibald, D.A., Farrar, E., 1986. Volcanic and intrusive
- 1164 history of the Pine Grove porphyry molybdenum system, southwestern Utah. Economic
- 1165 Geology 81, 553-577.
- 1166 Keppler, H., Wyllie, P.J., 1991. Partitioning of Cu, Sn, Mo, W, U, and Th between melt
- and aqueous fluid in the systems haplogranite-H 2 O- HCl and haplogranite-H 2 O- HF.
- 1168 Contributions to Mineralogy and Petrology 109, 139-150.
- 1169 Kerr, A., van Nostrand, T., Dickson, W., Lynch, E., 2009. Molybdenum and tungsten in
- 1170 Newfoundland: a geological overview and a summary of recent exploration
- 1171 developments. Current Research, Newfoundland and Labrador Department of Natural
- 1172 Resources, Geological Survey, Report, 09-01.
- 1173 Khaleghi, F., Hosseinzadeh, G., Rasa, I., Moayyed, M., 2013. Geological and
- Geochemical Characteristics of the Syah Kamar Porphyry Molybdenum Deposit, Westof Mianeh, NW Iran. ULUM-I ZAMIN 88, 187-196.
- 1176 Khodabandeh, A.A., Amini Azar, R., Faridi, M., 1999. Geological map of the Mianeh
- 1177 quadrangle. Geological Survey of Iran and Mining Exploration, Tehran.
- 1178 Kim, H., Yang, K., Lentz, D., Jeong, H.-Y., Kil, Y., Hwang, J., Park, S.-G., 2016. Low-
- salinity Liquid-rich or Vapor-like fluids in a porphyry-type Mo deposit, South Korea.The Canadian Mineralogist 54, 715-736.
- Kirkland, C., Smithies, R., Taylor, R., Evans, N., McDonald, B., 2015. Zircon Th/U
  ratios in magmatic environs. Lithos 212, 397-414.
- 1183 Kirsch, K., Navarre-Sitchler, A.K., Wunsch, A., McCray, J.E., 2014. Metal release from
- 1184 sandstones under experimentally and numerically simulated  $CO_2$  leakage conditions.
- 1185 Environmental science & technology 48, 1436-1442.
- 1186 Klemm, L.M., Pettke, T., Heinrich, C.A., 2008. Fluid and source magma evolution of the 1187 Questa porphyry Mo deposit, New Mexico, USA. Mineralium Deposita 43, 533.
- Knight, C., Bodnar, R., 1989. Synthetic fluid inclusions: IX. Critical PVTX properties of
   NaCl-H2O solutions. Geochimica et Cosmochimica Acta 53, 3-8.
- 1190 Lawley, C., Richards, J., Anderson, R., Creaser, R., Heaman, L., 2010. Geochronology
- 1191 and geochemistry of the MAX porphyry Mo deposit and its relationship to Pb-Zn-Ag
- mineralization, Kootenay arc, southeastern British Columbia, Canada. Economic
  Geology 105, 1113-1142.
- Le Maitre, R.W., Streckeisen, A., Zanettin, B., Le Bas, M., Bonin, B., Bateman, P., 2005.
- 1195 Igneous rocks: a classification and glossary of terms: recommendations of the
- 1196 International Union of Geological Sciences Subcommission on the Systematics of
- 1197 Igneous Rocks. Cambridge University Press.
- 1198 Lions, J., Devau, N., De Lary, L., Dupraz, S., Parmentier, M., Gombert, P., Dictor, M.-
- 1199 C., 2014. Potential impacts of leakage from CO<sub>2</sub> geological storage on geochemical
- 1200 processes controlling fresh groundwater quality: a review. International Journal of
- 1201 Greenhouse Gas Control 22, 165-175.

- 1202 Little, M.G., Jackson, R.B., 2010. Potential impacts of leakage from deep CO<sub>2</sub>
- geosequestration on overlying freshwater aquifers. Environmental science & technology44, 9225-9232.
- 1205 Liu, F., Lu, P., Griffith, C., Hedges, S.W., Soong, Y., Hellevang, H., Zhu, C., 2012. CO<sub>2</sub>-
- 1206 brine–caprock interaction: Reactivity experiments on Eau Claire shale and a review of
- relevant literature. International Journal of Greenhouse Gas Control 7, 153-167.
- 1208 Liu, J., Mao, J.-W., Wu, G., Wang, F., Luo, D.-F., Hu, Y.-Q., 2015. Geochemical
- signature of the granitoids in the Chalukou giant porphyry Mo deposit in the
- 1210 Heilongjiang Province, NE China. Ore Geology Reviews 64, 35-52.
- 1211 Ludington, S., Hammarstrom, J., Piatak, N., 2009. Low-fluorine stockwork molybdenite
- 1212 deposits. US Geological Survey Open- File Report 1211.
- Ludington, S., Plumlee, G.S., 2009. Climax-type porphyry molybdenum deposits. USGeological Survey.
- 1215 Ludwig, K., 2011. Isoplot v. 4.15: A Geochronological Toolkit for Microsoft Excel.
- 1216 Berkeley Geochronology Center Special Publication 4, 75.
- 1217 Mao, J., Wang, Y., Lehmann, B., Yu, J., Du, A., Mei, Y., Li, Y., Zang, W., Stein, H.J.,
- 1218 Zhou, T., 2006. Molybdenite Re–Os and albite 40Ar/39Ar dating of Cu–Au–Mo and
- magnetite porphyry systems in the Yangtze River valley and metallogenic implications.
  Ore Geology Reviews 29, 307-324.
- 1221 Mao, J.W., Pirajno, F., Xiang, J.F., Gao, J.J., Ye, H.S., Li, Y.F., Guo, B.J., 2011.
- 1222 Mesozoic molybdenum deposits in the east Qinling–Dabie orogenic belt: Characteristics 1223 and tectonic settings. Ore Geology Reviews 43, 264-293.
- 1224 Mao, J.W., Xie, G.Q., Bierlein, F., Qü, W.J., Du, A.D., Ye, H.S., Pirajno, F., Li, H.M.,
- 1225 Guo, B.J., Li, Y.F., Yang, Z.Q., 2008. Tectonic implications from Re–Os dating of
- 1226 Mesozoic molybdenum deposits in the East Qinling–Dabie orogenic belt. Geochimica et 1227 Cosmochimica Acta 72, 4607-4626.
- 1228 Massonne, H.-J., Opitz, J., Theye, T., Nasir, S., 2013. Evolution of a very deeply
- subducted metasediment from As Sifah, northeastern coast of Oman. Lithos 156, 171185.
- 1231 McInnes, B., Evans, N., Belousova, E., Griffin, W., 2003. Porphyry copper deposits of
- the Kerman belt, Iran: timing of mineralization and exhumation processes. Sci. Res. Rep.
  Australia. CSIRO (41 pp).
- Mehrabi, B., Yardley, B., Cann, J., 1999. Sediment-hosted disseminated gold
  mineralisation at Zarshuran, NW Iran. Mineralium Deposita 34, 673-696.
- 1236 Mercer, C.N., Reed, M.H., Mercer, C.M., 2015. Time scales of porphyry Cu deposit
- 1237 formation: Insights from titanium diffusion in quartz. Economic Geology 110, 587-602.
- 1238 Mirnejad, H., Mathur, R., Hassanzadeh, J., Shafie, B., Nourali, S., 2013. Linking Cu
- 1239 mineralization to host porphyry emplacement: Re-Os ages of molybdenites versus U-Pb
- ages of zircons and sulfur isotope compositions of pyrite and chalcopyrite from the Iju
- and Sarkuh porphyry deposits in Southeast Iran. Economic Geology 108, 861-870.
- 1242 Mirnejad, H., Simonetti, A., Molasalehi, F., 2011. Pb isotopic compositions of some Zn-
- 1243 Pb deposits and occurrences from Urumieh–Dokhtar and Sanandaj–Sirjan zones in Iran.
- 1244 Ore Geology Reviews 39, 181-187.

- 1245 Moghadam, H.S., Rossetti, F., Lucci, F., Chiaradia, M., Gerdes, A., Martinez, M.L.,
- 1246 Ghorbani, G., Nasrabady, M., 2016. The calc-alkaline and adakitic volcanism of the
- 1247 Sabzevar structural zone (NE Iran): implications for the Eocene magmatic flare-up in
- 1248 Central Iran. Lithos 248, 517-535.
- 1249 Moritz, R., Ghazban, F., Singer, B.S., 2006. Eocene gold ore formation at Muteh,
- 1250 Sanandaj-Sirjan tectonic zone, Western Iran: a result of late-stage extension and
- 1251 exhumation of metamorphic basement rocks within the Zagros Orogen. Economic
- 1252 Geology 101, 1497-1524.
- 1253 Moritz, R., Rezeau, H., Ovtcharova, M., Tayan, R., Melkonyan, R., Hovakimyan, S.,
- 1254 Ramazanov, V., Selby, D., Ulianov, A., Chiaradia, M., Putlitz, B., 2016. Long-lived,
- 1255 stationary magmatism and pulsed porphyry systems during Tethyan subduction to post-
- 1256 collision evolution in the southernmost Lesser Caucasus, Armenia and Nakhitchevan.1257 Gondwana Research 37, 465-503.
- 1258 Müller, A., Herklotz, G., Giegling, H., 2018. Chemistry of quartz related to the
- 1259 Zinnwald/Cínovec Sn-W-Li greisen-type deposit, Eastern Erzgebirge, Germany. Journal1260 of Geochemical Exploration 190, 357-373.
- 1261 Nabatian, G., Li, X.-H., Wan, B., Honarmand, M., 2017a. The genesis of Mo-Cu deposits
- and mafic igneous rocks in the Senj area, Alborz magmatic belt, Iran. Mineralogy andPetrology, 1-20.
- 1264 Nabatian, G., Wan, B., Honarmand, M., 2017b. Whole rock geochemistry, molybdenite
- 1265 Re-Os geochronology, stable isotope and fluid inclusion investigations of the Siah-
- 1266 Kamar deposit, western Alborz-Azarbayjan: New constrains on the porphyry Mo deposit1267 in Iran. Ore Geology Reviews 91, 638-659.
- Nabavi, M.H., 1976. An introduction to the geology of Iran. Geological survey of Iran109.
- Newberry, R., 1979. Polytypism in molybdenite (I); A non-equilibrium impurity-inducedphenomenon. American Mineralogist 64, 758-767.
- 1272 Ni, P., Wang, G.-G., Yu, W., Chen, H., Jiang, L.-L., Wang, B.-H., Zhang, H.-D., Xu, Y.-
- 1273 F., 2015. Evidence of fluid inclusions for two stages of fluid boiling in the formation of
- 1274 the giant Shapinggou porphyry Mo deposit, Dabie Orogen, Central China. Ore Geology
- 1275 Reviews 65, 1078-1094.
- NICICO, 2009a. Geological report and map of Goydaraq area, at scale of 1:25000,Internal report.
- NICICO, 2009b. Geological report and map of Khatoon-Abad area, at scale of 1:25000,Internal report.
- NICICO, 2009c. Geological report and map of Siah-Kamar area, at scale of 1:25000,
  Internal report.
- 1282 NICICO, 2010. Geological report and map of Siah-Kamar area, at scale of 1:5000,
- 1283 Internal report.
- 1284 Oliver, N.H., Bons, P.D., 2001. Mechanisms of fluid flow and fluid-rock interaction in
- 1285 fossil metamorphic hydrothermal systems inferred from vein–wallrock patterns,
- 1286 geometry and microstructure. Geofluids 1, 137-162.

- 1287 Pettke, T., Oberli, F., Heinrich, C.A., 2010. The magma and metal source of giant
- 1288 porphyry-type ore deposits, based on lead isotope microanalysis of individual fluid
- inclusions. Earth and Planetary Science Letters 296, 267-277.
- 1290 Rezeau, H., Moritz, R., Wotzlaw, J.-F., Tayan, R., Melkonyan, R., Ulianov, A., Selby,
- 1291 D., d'Abzac, F.-X., Stern, R.A., 2016. Temporal and genetic link between incremental
- pluton assembly and pulsed porphyry Cu-Mo formation in accretionary orogens. Geology44, 627-630.
- Richards, J.P., 2015. Tectonic, magmatic, and metallogenic evolution of the Tethyan orogen: From subduction to collision. Ore Geology Reviews 70, 323-345.
- 1296 Richards, J.P., Sholeh, A., 2016. The Tethyan tectonic history and Cu-Au metallogeny of
- 1297 Iran, Tectonics and Metallogeny of the Tethyan Orogenic Belt, Society of Economic1298 Geologists Special Publication, pp. 193-212.
- Richards, J.P., Wilkinson, D., Ullrich, T., 2006. Geology of the Sari Gunay epithermal
  gold deposit, northwest Iran. Economic Geology 101, 1455-1496.
- Rives, T., Rawnsley, K., Petit, J.-P., 1994. Analogue simulation of natural orthogonal
  joint set formation in brittle varnish. Journal of Structural Geology 16, 419-429.
- 1303 Robb, L., 2004. Introduction to ore-forming processes. Blackwell publishing.
- Rolland, Y., 2017. Caucasus collisional history: Review of data from East Anatolia toWest Iran. Gondwana Research 49, 130-146.
- 1306 Rossetti, F., Tecce, F., Billi, A., Brilli, M., 2007. Patterns of fluid flow in the contact
- 1307 aureole of the Late Miocene Monte Capanne pluton (Elba Island, Italy): the role of
- 1308 structures and rheology. Contributions to Mineralogy and Petrology 153, 743-760.
- Rubatto, D., 2002. Zircon trace element geochemistry: partitioning with garnet and the
  link between U–Pb ages and metamorphism. Chemical Geology 184, 123-138.
- 1311 Sarjoughian, F., Kananian, A., 2017. Zircon U-Pb geochronology and emplacement
- history of intrusive rocks in the Ardestan section, central Iran. Geologica acta 15, 0025-0036.
- 1314 Sarjoughian, F., Kananian, A., Haschke, M., Ahmadian, J., Ling, W., Zong, K., 2012.
- Magma mingling and hybridization in the Kuh-e Dom pluton, Central Iran. Journal ofAsian Earth Sciences 54-55, 49-63.
- 1317 Seedorff, E., Dilles, J.H., Proffett, J.J.M., Einaudi, M.T., Zurcher, L., Stavast, W.J.A.,
- Johnson, D.A., Barton, M.D., 2005. Porphyry Deposits: Characteristics and Origin of
- 1319 Hypogene Features, In: Hedenquist, J.W., Thompson, J.F.H., Goldfarb, R.J., Richards,
- 1320 J.P. (Eds.), One Hundredth Anniversary Volume. Society of Economic Geologists.
- 1321 Seedorff, E., Einaudi, M.T., 2004. Henderson porphyry molybdenum system, Colorado:
- 1322 II. Decoupling of introduction and deposition of metals during geochemical evolution of
- 1323 hydrothermal fluids. Economic Geology 99, 39-72.
- 1324 Selby, D., Creaser, R.A., 2001. Re-Os geochronology and systematics in molybdenite
- 1325 from the Endako porphyry molybdenum deposit, British Columbia, Canada. Economic
- 1326 Geology 96, 197-204.
- 1327 Selby, D., Creaser, R.A., 2004. Macroscale NTIMS and microscale LA-MC-ICP-MS Re-
- 1328 Os isotopic analysis of molybdenite: Testing spatial restrictions for reliable Re-Os age
- 1329 determinations, and implications for the decoupling of Re and Os within molybdenite.
- 1330 Geochimica et Cosmochimica Acta 68, 3897-3908.

- 1331 Selby, D., Nesbitt, B.E., Muehlenbachs, K., Prochaska, W., 2000. Hydrothermal
- alteration and fluid chemistry of the Endako porphyry molybdenum deposit, BritishColumbia. Economic Geology 95, 183-202.
- 1334 Seo, J.H., Guillong, M., Heinrich, C.A., 2012. Separation of molybdenum and copper in
- porphyry deposits: The roles of sulfur, redox, and pH in ore mineral deposition at
- 1336 Bingham Canyon. Economic Geology 107, 333-356.
- 1337 Shafiei, B., Haschke, M., Shahabpour, J., 2009. Recycling of orogenic arc crust triggers
- 1338 porphyry Cu mineralization in Kerman Cenozoic arc rocks, southeastern Iran.
- 1339 Mineralium Deposita 44, 265.
- 1340 Shafiei, B., Shahabpour, J., Haschke, M., 2008. Transition from Paleogene normal calc-
- alkaline to Neogene adakitic-like plutonism and Cu-metallogeny in the Kerman porphyry
   copper belt: response to Neogene crustal thickening. Journal of Sciences, Islamic
- 1343 Republic of Iran 19, 67-84.
- 1344 Shahabpour, J., Kramers, J., 1987. Lead isotope data from the Sar-Cheshmeh porphyry 1345 copper deposit, Iran. Mineralium Deposita 22, 278-281.
- 1346 Shahsavari Alavijeh, B., Rashidnejad-Omran, N., Corfu, F., 2017. Zircon U-Pb ages and
- 1347 emplacement history of the Nodoushan plutonic complex in the central Urumieh-Dokhtar
- 1348 magmatic belt, Central Iran: Product of Neotethyan subduction during the Paleogene.
- 1349 Journal of Asian Earth Sciences 143, 283-295.
- 1350 Shannon, J., Walker, B., Carten, R., Geraghty, E., 1982. Unidirectional solidification
- 1351 textures and their significance in determining relative ages of intrusions at the Henderson
- 1352 Mine, Colorado. Geology 10, 293-297.
- 1353 Shinohara, H., Kazahaya, K., Lowenstern, J.B., 1995. Volatile transport in a convecting 1354 magma column: Implications for porphyry Mo mineralization. Geology 23, 1091-1094.
- 1355 Sillitoe, R.H., 2010. Porphyry copper systems. Economic Geology 105, 3-41.
- 1356 Simmonds, V., Moazzen, M., 2015. Re-Os dating of molybdenites from Oligocene Cu-
- 1357 Mo-Au mineralized veins in the Qarachilar area, Qaradagh batholith (northwest Iran):
- 1358 implications for understanding Cenozoic mineralization in South Armenia, Nakhchivan,
- and Iran. International Geology Review 57, 290-304.
- 1360 Simmonds, V., Moazzen, M., Selby, D., 2017. Re-Os dating of mineralization in Siah
- 1361 Kamar porphyry Mo deposit (NW Iran) and investigating on its temporal relationship
- 1362 with porphyry Cu-Mo deposits in the southern Lesser Caucasus, NW and central Iran,
- 1363 EGU General Assembly Conference Abstracts, p. 665.
- 1364 Simmonds, V., Moazzen, M., Selby, D., 2019. U-Pb zircon and Re-Os molybdenite age
- 1365 of the Siah Kamar porphyry molybdenum deposit, NW Iran. International Geology1366 Review, 1-17.
- 1367 Sinclair, W., 2007. Porphyry deposits. Mineral deposits of Canada: A synthesis of major
- deposit-types, district metallogeny, the evolution of geological provinces, and exploration
- 1369 methods: Geological Association of Canada, Mineral Deposits Division, Special
- 1370 Publication 5, 223-243.
- 1371 Smoliar, M.I., Walker, R.J., Morgan, J.W., 1996. Re-Os ages of group IIA, IIIA, IVA,1372 and IVB iron meteorites. Science 271, 1099-1102.
- 1373 Stein, H., Markey, R., Morgan, J., Hannah, J., Scherstén, A., 2001. The remarkable Re-
- 1374 Os chronometer in molybdenite: how and why it works. Terra Nova 13, 479-486.

- 1375 Stocklin, J., 1968. Structural history and tectonics of Iran: a review. AAPG Bulletin 52,1376 1229-1258.
- 1377 Sutulov, A., 1973. Mineral resources and the economy of the USSR. Backbeat Books.
- 1378 Taghipour, N., Aftabi, A., Mathur, R., 2008. Geology and Re-Os Geochronology of
- Mineralization of the Miduk Porphyry Copper Deposit, Iran. Resource Geology 58, 143-160.
- 1381 Taylor, R.D., Hammarstrom, J.M., Piatak, N.M., Seal II, R.R., 2012. Arc-related
- porphyry molybdenum deposit model: Chapter D in Mineral deposit models for resourceassessment. US Geological Survey.
- 1384 Ulrich, T., Mavrogenes, J., 2008. An experimental study of the solubility of molybdenum
- in  $H_2O$  and  $KCl-H_2O$  solutions from 500 °C to 800 °C, and 150 to 300 MPa. Geochimica et Cosmochimica Acta 72, 2316-2330.
- 1387 Verdel, C., Wernicke, B.P., Hassanzadeh, J., Guest, B., 2011. A Paleogene extensional1388 arc flare-up in Iran. Tectonics 30.
- Villa, I.M., De Bièvre, P., Holden, N., Renne, P., 2015. IUPAC-IUGS recommendationon the half life of 87Rb. Geochimica et Cosmochimica Acta 164, 382-385.
- 1391 Wallace, S., 1995. The Climax-type molybdenite deposits: what they are, where they are 1392 and why they are. Economic Geology 90, 1359-1380.
- 1393 Wan, B., Deng, C., Najafi, A., Hezareh, M.R., Talebian, M., Dong, L., Chen, L., Xiao,
- W., 2018. Fertilizing porphyry Cu deposits through deep crustal hot zone melting.Gondwana Research 60, 179-185.
- 1396 Wang, G., Wu, G., Xu, L., Li, X., Zhang, T., Quan, Z., Wu, H., Li, T., Liu, J., Chen, Y.,
- 1397 2017. Molybdenite Re–Os age, H–O–C–S–Pb isotopes, and fluid inclusion study of the
- 1398 Caosiyao porphyry Mo deposit in Inner Mongolia, China. Ore Geology Reviews 81, 728-1399 744.
- 1400 Wark, D.A., Watson, E.B., 2006. TitaniQ: a titanium-in-quartz geothermometer.
- 1401 Contributions to Mineralogy and Petrology 152, 743-754.
- Westra, G., Keith, S.B., 1981. Classification and genesis of stockwork molybdenumdeposits. Economic Geology 76, 844-873.
- 1404 Whitcher, I., 1975. Anduramba molybdenum prospect, Economic Geology of Australia 1405 and Papua New Guinea. 1—Metals, pp. 793-794.
- Whitney, D.L., Evans, B.W., 2010. Abbreviations for names of rock-forming minerals.American Mineralogist 95, 185-187.
- 1408 Wiedenbeck, M., Alle, P., Corfu, F., Griffin, W., Meier, M., Oberli, F.v., Quadt, A.v.,
- 1409 Roddick, J., Spiegel, W., 1995. Three natural zircon standards for U-Th-Pb, Lu-Hf,
- 1410 trace element and REE analyses. Geostandards Newsletter 19, 1-23.
- 1411 Wilkinson, J., 2001. Fluid inclusions in hydrothermal ore deposits. Lithos 55, 229-272.
- 1412 Wilkinson, J.J., 2013. Triggers for the formation of porphyry ore deposits in magmatic
- 1413 arcs. Nature Geoscience 6, 917-925.
- 1414 Wu, Y.-S., Chen, Y.-J., Zhou, K.-F., 2017. Mo deposits in Northwest China: Geology,
- 1415 geochemistry, geochronology and tectonic setting. Ore Geology Reviews 81, 641-671.

- 1416 Xiaoyun, C., 1989. Solubility of molybdenite and the transport of molybdenum in
- 1417 hydrothermal solutions. Ph. D. thesis. Iowa State University, 1~111.
- 1418 Xue, C., Chi, G., Zhao, X., Wu, G., Zhao, Z., Dong, L., 2016. Multiple and prolonged
- 1419 porphyry Cu–Au mineralization and alteration events in the Halasu deposit, Chinese
- 1420 Altai, Xinjiang, northwestern China. Geoscience Frontiers 7, 799-809.
- 1421 Yang, Z., 2007. Re-Os isotopic ages of Tangjiaping molybdenum deposit in Shangcheng
- 1422 County, Henan and their geological significance. Kuangchuang Dizhi(Mineral Deposits)1423 26, 289-295.
- 1424 Yeganehfar, H., Ghorbani, M.R., Shinjo, R., Ghaderi, M., 2013. Magmatic and
- 1425 geodynamic evolution of Urumieh–Dokhtar basic volcanism, Central Iran: major, trace 1426 element, isotopic, and geochronologic implications. International Geology Review 55,
- 1427 767-786.
- 1428 Zeng, Q.-D., Sun, Y., Chu, S.-X., Duan, X.-X., Liu, J., 2015. Geochemistry and
- 1429 geochronology of the Dongshanwan porphyry Mo–W deposit, Northeast China:
- implications for the Late Jurassic tectonic setting. Journal of Asian Earth Sciences 97,472-485.
- 1432 Zeng, Q., Liu, J., Chu, S., Wang, Y., Sun, Y., Duan, X., Zhou, L., 2012. Mesozoic
- 1433 molybdenum deposits in the East Xingmeng orogenic belt, northeast China:
- 1434 characteristics and tectonic setting. International Geology Review 54, 1843-1869.
- 1435 Zhai, D., Liu, J., Tombros, S., Williams-Jones, A.E., 2017. The genesis of the Hashitu
- 1436 porphyry molybdenum deposit, Inner Mongolia, NE China: constraints from
- 1437 mineralogical, fluid inclusion, and multiple isotope (H, O, S, Mo, Pb) studies.
- 1438 Mineralium Deposita 53, 377-397.
- 1439 Zhang, X., Lentz, D.R., Yao, C., Liu, R., Yang, Z., Mei, Y., Fan, X., Huang, F., Qin, Y.,
- 1440 Zhang, K., 2018. Geochronology, geochemistry, and Sr–Nd–Pb–Hf isotopes of the
- 1441 Zhunsujihua granitoid intrusions associated with the molybdenum deposit, northern Inner
- 1442 Mongolia, China: implications for petrogenesis and tectonic setting. International Journal
- 1443 of Earth Sciences 107, 687-710.
- 1444 Zhang, Y.-G., Frantz, J.D., 1987. Determination of the homogenization temperatures and
- 1445 densities of supercritical fluids in the system NaClKClCaCl<sub>2</sub>H<sub>2</sub>O using synthetic fluid 1446  $(A_{225}, A_{25})$
- 1446 inclusions. Chemical Geology 64, 335-350.
- 1447 Zindler, A., Hart, S., 1986. Chemical geodynamics. Annual review of earth and planetary1448 sciences 14, 493-571.
- 1449

### 1450 Figures captions

- 1451 Figure 1: Simplified tectono-geological map of Iran with the main Cu, Au and Mo ore
- 1452 deposits and their ages. Based on maps from the (Huber, 1977), with additional information
- 1453 from (Berberian and King, 1981) and Richards et al. (2012). See also Table 1 for the
- 1454 references for each mineralisation site.

Figure 2: Simplified geological map of Mianeh-Hashtroud region, as derived from field and remote sensing investigations (modified after Amidi et al., 1987; NICICO, 2009c, b, a).

Figure 3: Top: Geological-alteration map of the Siah-Kamar Mo deposit (see Fig. 3 for
map location). The ellipse indicates the estimated alteration zone. Bottom: Interpretative
geological cross section across the Mo ore deposit, also showing distribution of the main
alteration types. Abbreviations: Ab, Albite; Act, Actinolite; Ap, Apatite; Bt, Biotite, Cal,
Calcite; Chl, Chlorite; Ep, Epidote; ill, Illite. Kal, Kaolinite; Kfs, K-feldspar; Mag,
Magnetite; Or, Orthose, Psp, Paraspurite; Po, Pyrrhotite; Py, Pyrite; Qtz, Quartz; Sch,
Scheelite; Ser, Sericite.

Figure 4: General field view of the Siah-Kamar Mo deposit area (field of view 2.5 km).
The line drawing illustrates the main rock and alteration types. Abbreviations: alt., alteration;
Cb, carbonate; K, Potassic; mp, mineralized basic/intermediate porphyry rocks.

1467 Figure 5: Rock textures and alterations. (a) The textural appearance of the potassic-sodic alteration within the main mineralised porphyry body in the outcrop. Alteration mineralogy 1468 1469 consists of Bt-Fsp+ Po, Mg assemblage overprinting early igneous phases. (b) potassic-sodic 1470 alteration at the thin section scale (plane polarised light). (c) Back-scattered electron (BSE) image showing disseminated molybdenite flakes within the secondary feldspar-biotite 1471 assemblage. (d) Biotite veinlets in the potassic-sodic alteration zone overprinted by Kfs-Qz 1472 1473 veins (plane polarised light). (e) BSE image showing early magnetite inclusions in secondary 1474 feldspar and ilmenite inclusions in texturally-late biotite. (f) Brecciated andesite country 1475 rocks and epidote veining within the propylitic alteration zone. (g) General appearance of the 1476 phyllic and argillic alteration zones. (h) Rock slab from phyllic/silicified zones showing the development of Oz + Py veins with sericitic halos; (i-j) Microtexture of volcanic country rock 1477 within pronounced phyllic alteration made of Qz + Ser + Py assemblage (crossed polars). 1478

1479

Figure 6: (a) The contact between the mineralised porphyry body and the country rocks. Note the presence of potassic alteration and intense veining within the porphyry body and how the veining dies out abruptly across the contact (see also the interpretative line drawing in the inset). (b) Rose diagram of the cumulative vein strikes as measured within the different alteration zones. (c) Contour diagram of the cumulative poles to Mo-mineralised veins (V2a and V2b) within the potassic-sodic alteration zone (Equal area Schmidt net, lower

1486 hemisphere projection). (d) Contour diagram of the poles to (V2b) Cb-Mo veins within the

1487 potassic-sodic alteration zone (Equal area Schmidt net, lower hemisphere projection). (e)

1488 Contour diagram of the poles to epidote (Ep) veins within the propylitic alteration zone

1489 (Equal area Schmidt net, lower hemisphere projection).

Figure 7. Conceptual model for veining and Mo mineralisation within the potassic-sodicalteration zone (see text for further details).

Figure 8. (a) Rock slab of an early Qz-Kfs-Ab V2a vein with V2b Mol-bearing carbonate mineral assemblages growing in pre-existing vugs. Note the marginal K-feldspar + albite selvage. (b) BSE image showing the textural appearance of a V2a Qz-Kfs vein overprinted by a late, post-ore Fe-oxide vein. Molybdenite occurs in the centre of the vein. (c) BSE image showing coexisting scheelite (Sch) and Mol within V2a Qz + Kfs vein.

Figure 9. (a) Rock slabs of composite V2a-V2b veins showing different degree of the textural overprinting operated by the Cb-Mol mineralisation. (b) BSE image showing V2b veining overprinting early V2a Qz, sericite halos and compositional zoning in Cb. (c) BSE image showing a ribbon texture along margins of a V2b vein, attesting episodic Cb-Mol precipitation. (d) BSE image showing the textural overprint operated by the V2b Cb-Mo veining onto pre-existing Qz-Kfs V2a one. Sericite formed after alteration of Kfs. Note the reworking of the Qz grain boundary during V2b veining.

1504 Figure 10: Mineralization stages and their associated mineral assemblages in the SKMD.

Figure 11. Representative discrimination diagrams for quartz chemistry (after Müller et al., 2018) as derived from in situ LA-ICPMS analyses of quartz grains from V2a veins. (a) Li vs Al. (b) Al vs Ge. (c) Al vs Li.

1508 Figure 12: Microphotographs of the fluid inclusions types hosted in V2a-V2b veins. (a)

1509 Type A1 (two-phase L-rich) fluid inclusions in Qz (V2a vein). (b) Type A2 (two-phase L-

rich) fluid inclusions in Qz (V2a vein). (c) Type A3 (two-phase L-rich) fluid inclusions in

1511 Qz-Py vein from the phyllic alteration zone. (d) Type A4 (two-phase L-rich) fluid inclusions

1512 in Psp (V2b Vein). Type B solid-bearing fluid inclusions in Qz (V2a vein).

1513 Figure 13. (a) and (b) Representative Raman spectra from fluid and gas phases from1514 types A1 fluid inclusions in Qz from V2a veins.

Figure 14. Histogram of homogenization temperature (Th) (a) and salinity (b) for the analysed fluid inclusion types.

1517 Figure 15. (a, c, e and g) Cathodoluminescence images from the analysed zircon grains,

1518 with location of the laser spots (circles and relative id numbers in parenthesis) and obtained

1519 <sup>206</sup>Pb/<sup>238</sup>U ages (2 $\sigma$  error; Ma). Spots that provided inherited zircon ages are denoted by

1520 dashed circles. (b) Tera-Wasserburg concordia diagram for sample MN03 (d, f, h).

1521 Conventional Concordia diagrams with Concordia ages indicated (2 $\sigma$  error; Ma) for the

1522 MN01, MN19, and MN31 samples. See also Table 5 for the corresponding analytical results.

Fig 16: Mineral data and Rb–Sr age results for sample MN02. Analytical data are given in Table 7. Grain size is indicated when different grain size fractions were analyzed. Kfs, Kfeldspar.

Figure 17: (a) Calculated Isochores for the representative type-A1 (quartz) and -A4 (paraspurrite) fluid inclusions (BULK and ISOC software; Bakker, 2003), combined with isopleths as obtained from Ti-in-quartz thermometry. (b) Salinity versus homogenization temperature (Th) diagram for all the FI types. Dashed arrows indicate the possible fluid evolution. (c) Schematic diagram showing typical trends in a Th-salinity space for fluid evolution in hydrothermal ore deposits (after Wilkinson, 2001).

Figure 18: (a) A model for the Mo mineralisation at Siah-kamar framed within a scenario 1532 1533 of long-lived thermal anomaly and magmatism. (a) Stage-1 mineralisation (33Ma). Emplacement of the basic/intermediate porphyry body with pervasive potassic-sodic 1534 1535 alteration and low-grade disseminated Mo mineralization. (b) Emplacement of a new batch of 1536 acidic magmas (at 29-28 Ma) caused re-fertilisation of the intrusive/cap rock system, with 1537 formation of V2a veining (stage-2a). (c) The main mineralisation stage (29-28 Ma). The 1538 continuous exsolution of acidic, CO2-bearing fluid from the new magma products caused 1539 leaching of the early deposited Mo and, by neutralisation due to the interaction with meteoric 1540 fluids, the high-grade Mo mineralization during structurally-controlled stockwork Cb veining (stage-2b). 1541

1542

1543Table 1. List of the studied samples with their location, petrographical characteristic and1544analytical methods adopted.

	S	5	Ro	Rock texture and	Alteration	Locati	Method
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amp	ck	petrography	type	on	EPMA	A-	Zircon	Mol.	Rb-Sr
le	Туре			(decimal degree)	EF	Γ	Zir	Σ	Rt
M N01	Rh yolitoi d	Porphyritic with microgranular groundmass. Qz (5-10 %), Pl+Kfs (5-10 %) phenocrysts. Groundmass: Qz (20-30 %), KFsp (40-50 %), Pl (10-15 %), Biotite (<5 %).	Weakly phyllic alteration. Qz veins, rare Ser, Py and Fe- oxides	N: 37.41297 E: 47.59745	0	2	R		
M NO2 MN O3	A ndesit oid/Ba saltoid (Diorit e to monzo diorite porphy ry)	Porphyritic with microgranular groundmass. Magmatic texture mostly overprinted. Relic Pl (20-30 %) phenocrysts and pseudomorphous secondary aggregates after Cpx (5-10 %).	Strong sodic-potassic alteration. Disseminated secondary fine grained Ab+Kfs+Qtz+ Bt+Mag (40- 50%), Qz-Afs- Mol-Cb-Rt- Sch veins (5- 10 %)	N: 37.39916 E: 47.59519		·	*		
M N19	Da citoid	Porphyritic with glassy groundmass. Glass (50-60 %), euhedral to subhedral Pl (20-25 %), Bt (5-7 %), Amph (2-3 %), Qz (5-7 %) phenocrysts.	Almost unaltered; secondary Chl on Amph	N: 37.39925 E: 47.58888					

M N31	A ndesit oid/Ba saltoid (Diorit e to monzo diorite	Porphyritic with microgranular groundmass. Pl (30-35 %), Cpx (15-20 %) phenocrysts. Groudnmass: Pl (25-30 %), Cpx (15-20 %)+	Propylitic alteration. Ep (10-15 %)+ Act/Tr (5-10 %), Py	N: 37.38710 E: 47.60161	
M N65	porphy ry) A ndesit oid/Ba saltoid (monz odiorit e)	Mag (4-6 %), Porphyritic with microgranular groundmass. Pl (20-30 %), Cpx (7-10 %), Opx (5 %) phenocrysts. Groundmass: Pl+KFs+Px+Mag (40-50 %)	Weakly propylitic alteration. Cb (disseminated over mafic minerals and Pl) and Cb veining (2-3 %), secondary Chl on Cpx	N: 37.406533 E: 47.589648	
M N55	A ndesit oid/Ba saltoid (monz onite porphy ry)	Porphyritic with microgranular groundmass. Pl (5-7 %) phenocrysts. Groudnmass: Pl (60-70), KFs (5-10? %), Mag (5- 7 %).	and Opx. Weakly altered to Cb	N: 37.40281 E: 47.59006	

1545

15.

Sp	Li	В	Al	M n		G e	G		
ot ·				ppm	a			3	
 v4	3.2	3.	35.	0		0	0		0
q	6	90	97	.15			)	.01	
v4	24.	5.	15	1		0	1		2
a2	52	18	6.97	.05	.11	.41		.02	
v4	3.3	3.	98.	0		0	1		1
b2	4	57	49	.70	.13	.20	)	.77	
v4	6.5	8.	89.	0		0	1		0
c2	0	35	96	.27	.01	.98	;	.94	
v4	28.	7.	16	0		0	1		0
d2	03	63	0.22	.20	.06	.87	,	.80	
v4	1.8	4.	45.	0		b	b		0
e2	6	58	10	.24	dl	dl		.14	
v4	17.	4.	10	0		0	1		b
f2	07	00	2.47	.66	.17	.33		dl	
bdl:	Below detecti	on limit							
	16.								
		$\mathcal{N}$	, 						

#### Table 2. Trace element chemistry of quartz from V2 veins 1547

1548

		Te	mperatu	re (°C) c	alculated	for P
S	Ti			(kbar)		
pot	(ppm)	0.	1	1.	2	2.
		5	1	5	2	5
v	21.1	4	5	5	5	5
4b	7	94	27	50	69	85
v	21.5	4	5	5	5	5
4c	2	95	28	51	70	86
v	31.8	5	5	5	6	6
4d	8	27	61	85	05	21
v	29.7	5	5	5	5	6
4e	8	21	55	79	99	15
v	17.4	4	5	5	5	5
4f	2	79	11	34	53	68
v	44.2	5	5	6	6	6
4g	0	55	91	16	36	53
v	45.0	5	5	6	6	6
4i	0	57	92	18	38	55
v	20.9	4	5	5	5	5
4j			26			
v	44.6	5	5	6	6	6
4n			92			
v	17.3					
4q			11			
v	54.6					
4a2			11			
	41.6					
	41.0 6					
.02	v		00	10	20	• /

### Table 3. Ti-in-Quartz content and results of TitaniQ thermometry (V2 veins)

	v	24.5	5	5	5	5	5	
	4c2	7	05	39	63	81	98	
	v	48.0	5	5	6	6	6	
	4d2	2	63	98	24	44	61	
	V	23.6	5	5	5	5	5	
	4e2	1	02	36	59	78	94	
	V	24.6	5	5	5	5	5	
	4f2	7	06	39	63	82	98	
1552								2
							2	
					$\bigcirc$			
				X				
	P		Q					
		$\mathbf{G}$						
	C							

T ype	V to L ratio	H ost minera l	A lterat ion zone	M easur ed numb er	T freezing (°C)	T eutectic (°C)	T homogen ization (°C)	T melting (°C)	S alinit y (wt.% NaCl eq.)
T ype A1 (L- rich+ V)	0 .5-1	Q uartz	P otass ic	4	- 44.4 to - 29.6 (Mean= -38.5)	-30 to -22.5 (Mean= -25.3)	298 to 365 (Mean= 333)	4,3 to - 0,6 (Mean= -2)	1 to 6,9 (Mod e= 3.85)
T ype A2 (L- rich+ V)	0 .1- 0.4	Q uartz	P otass ic	1 95	- 47.2 to - 25.3 (Mean= -39)	- 31.5 to -23 (Mode= -23)	170 to 352 (Mean= 272)	- 3.7 to - 0.1 (Mean= -1.5)	0 2 to 6 (Mea n= 2.6)
T ype A3 (L- rich+ V)	0 .2- 0.4	Q uartz	P hylli c	5 2	- 41.4 to - 31 (Mean= -33.6)	-22 to -21 (Mean= -21.5)	232 to 365 (Mode= 320)	- 2,6 to - 1 (Mean= -1.8)	1 7 to 4.3 (Mea n= 3.1)
T ype A4 (L- rich+ V)	0 .1- .25	Pa raspurr ite	P otass ic	9 1	-30 to -42 (Mean= -39)	-24 to -33 (Mean= -27)	145 to 300 (Mode= 225)	- 1,2 to - 0,1 (Mean= -0.35)	0 2 to 2.1 (Mea n= 0.6)

1554 Tabl 1 0 f th inti d +1 htai **\**+h .+-11+ FLak 1 •



#### - • MN01; Northing: 37.41297, Easting: 47.59745 -----••• ----•• ~ ~~ ~ 2 $\langle \langle \langle \rangle \rangle$ • ~ ., ~ ~ . . . . . . . . . . . . -------• • ^ ~ ^ ^ •• \_ ^ ^ ~ ~ •• ~ ~ • ~ ^ ^ • • ~ ~ ----~ ~ ·· -~ -----20.0 • • ^ ^ ^ ^ ^ ^ ----~ ~ ~ ~ ~ • • • ., ., • ~ ~ ^ ^ AO 1 ..... ^ 40.1 ~ ... ... --------• ^ ~ ~ • • ----^ ~ --------MN03; Northing: 37.39916, Easting: 47.59519 ••• ~ ~ ~ -----••• 2 ^ ~ ~ • • • • ^ ~ •••• ~ ~ ... ... ----~ ^ ---~ ~ ~ 20.5 • ----.... ^ • • • • ~ --------.... ~ ----. ----~ ~ ~ .... . . . •••• -- -----~ ~ 20.7 ----^ .... ~ 2 ----~ ~ \_ •••• MN19; Northing: 37.39925, Easting: 47.58888 • • ~ ----•••• · · · · · · · · · · · · · ~ 42.1 -----. . . . . . . . . . . . . • ^ ---------• ~ -----~ ^ ~ ^ ^ --------~ ~ -- ----------------~ ~ ~ ----• ~ ^ ^ ^ ^ ••• ----• ~ ~ · · · ----^ -----.... ^ ^ ~ ~ ^ ~ -----20.2 $\langle \cdot \cdot \rangle$ ----. . *.* ----^ ~ ^ .... ~ · ~ -----~ ^ ~ ~ ~ ------------• ----~ --------2 • • ~ ~ .... .... -----~ ~ • ^ ^ ^ ^ ^ ... ^ ~ ~ ~ ----MN31; Northing: 37.38710, Easting: 47.60161 -------• • • 470 ----• • • • ~ ^ ~ ~ ~ ~ ~ ~ ~ ~ ~ • ~ . . . . . . . . ^

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#### 1557 Table 5. Results of LA-ICP-MS U-(Th)-Pb zircon data.

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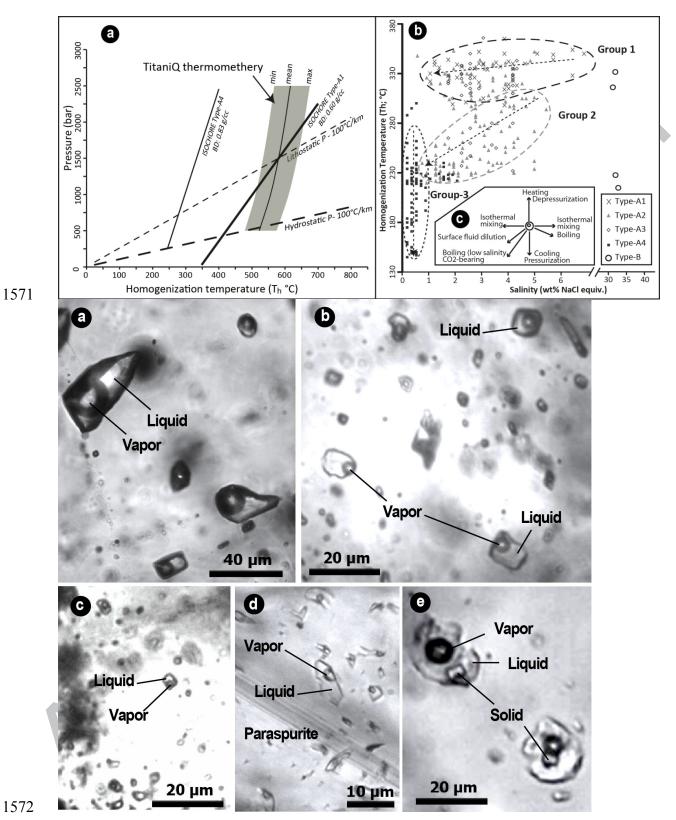
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									_	200	222
			<u>^</u>	<u>^</u>	^	^	<u>^</u>		2	202	222
		0	0	^	^	^	^		~	20.0	24.0
	••	~	2	^	^	^	^		~		
		-	2	<u>^</u>	~	_			~	- ···	
		0	0	^	^	^	^		~	10.0	10.0
		•	^	^	^	^	^		-	20.4	
	••	-	<u>_</u>	-		<u>^</u>	_		-		
			^	^	^	^	^		~	~ • •	
		^	^	^	^	^	^		~	250	255
1558	* 206 Persontage	of Dh. contributed by	• • • • • • • • • • • • • • • • • • •	a basis of 204 Pb	Value of some	~ 		V rom ora (1075)		latastian of 204Ph	
	* 206 Percentage	of Pb contributed by	common Pb on th	te basis of 204 Pb.	Value of comm	ion Pb was assume	d by Stacey and	Kramers (1975)	model; n.d.: no o	letection of 204Pb	
1559	** Degree of disc	ordance (%); negativ	ve numbers and bla	anks show normal	discordant and o	concordanct withir	$12\sigma$ of the analy	tical error, respec	tively.		
									5		

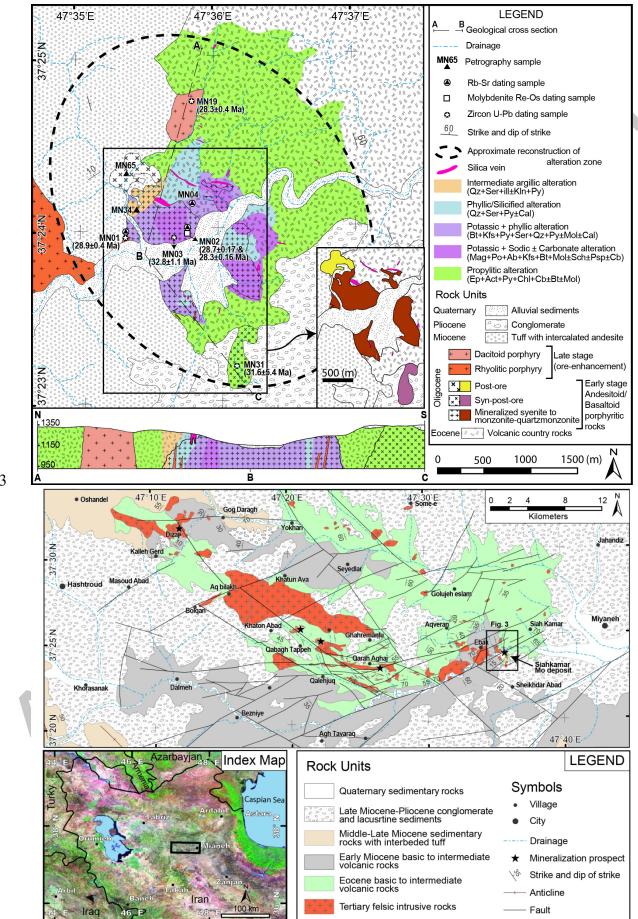
Sample         t (g)         ppm)         ppm)         (ppb)         ge         *         #           RO780-         0         19.         12.15         .05         5.81         .02         8.70         .14         .17           RO780-         0         22.         13.92         .06         6.56         .03         8.29         .14         .16           *uncertainty including all sources of analytical uncertainty         "uncertainty including all sources of analytical uncertainty plus decay constant         .05         .03         8.29         .14         .16
13_Mo1B       .016       33       .09       12.15       .05       5.81       .02       8.70       .14       .17         RO780-       0       22.       13.92       .06       6.56       .03       8.29       .14       .16         *uncertainty including all sources of analytical uncertainty       .06       .03       8.29       .14       .16
13_Mo1B       .016       33       .09       .05       .02       8.70       .14       .17         RO780-       0       22.       13.92       6.56       .03       8.29       .14       .16         *uncertainty including all sources of analytical uncertainty       .06       .03       8.29       .14       .16
14_Mo2B       .016       14       .10       .06       6.56       .03       8.29       .14       .16         *uncertainty including all sources of analytical uncertainty
14_Mo2B       .016       14       .10       .06       .03       8.29       .14       .16         *uncertainty including all sources of analytical uncertainty
<sup>#</sup> uncertainty including all sources of analytical uncertainty plus decay constant

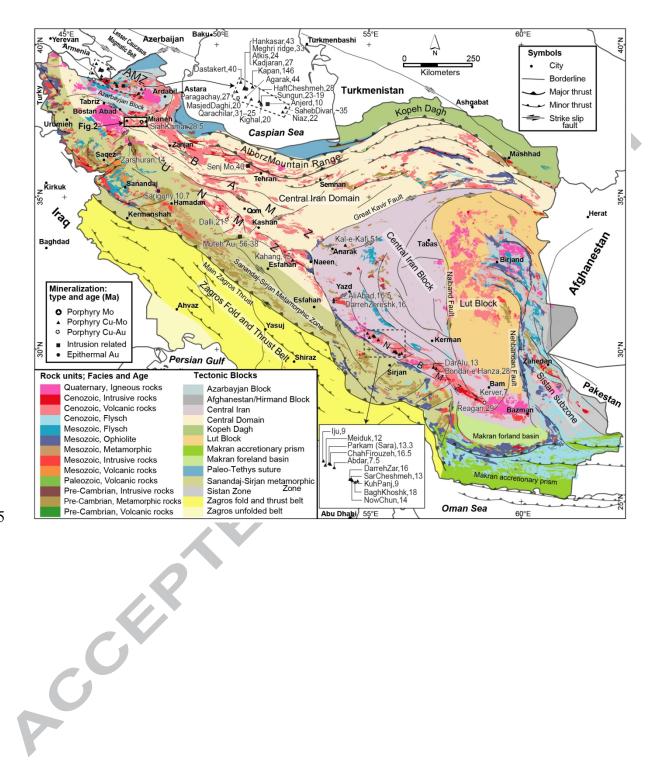
	Ana lysis	material	Rb ppm	Sr ppm	<sup>87</sup> Rb / <sup>86</sup> Sr	<sup>87</sup> Sr/ <sup>86</sup> Sr	<sup>87</sup> Sr/ <sup>86</sup> Sr 2σ <sub>m</sub> (%)
-	PS3 104	biotite	306	19 4	4.56	0.70 7085	0.0019
	PS3 105	kfsp >160 μm	138	87 5	0.45 5	0.70 5176	0.0011
	PS3 106	sericite >125 μm	397	11 8	9.73 2	0.70 8913	0.0015
	PS3 107	kfsp <160 μm	145	89 3	0.47 1	0.70 5179	0.0008
	PS3 108	sericite <125 μm	375	19 0	5.72	0.70 7161	0.0022
	PS3 109	apatite conc.	91. 2	65 2	0.40 5	0.70 5162	0.0021
	6			r			

1565Table 7. Rb-Sr analytical data for the mineral separated from sample MN02

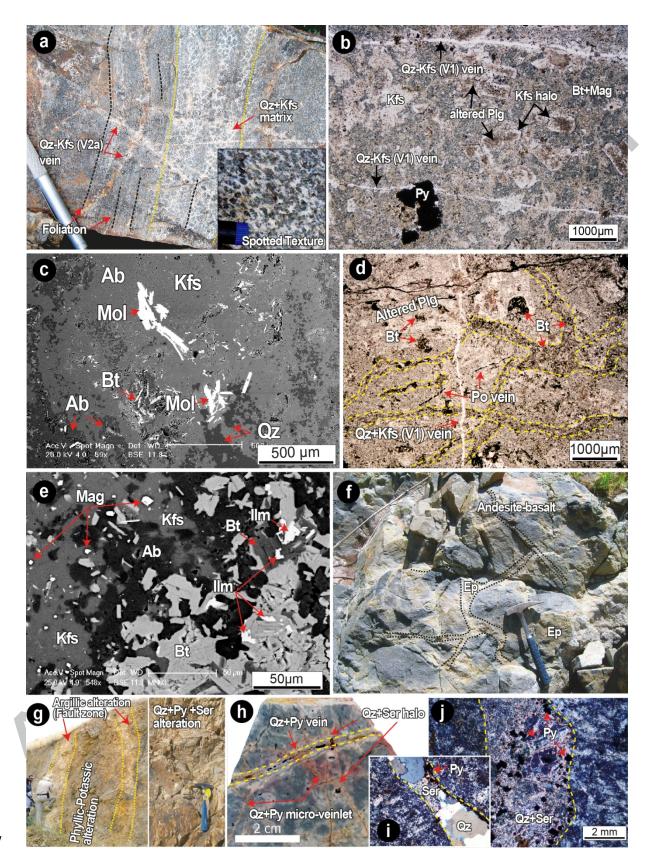


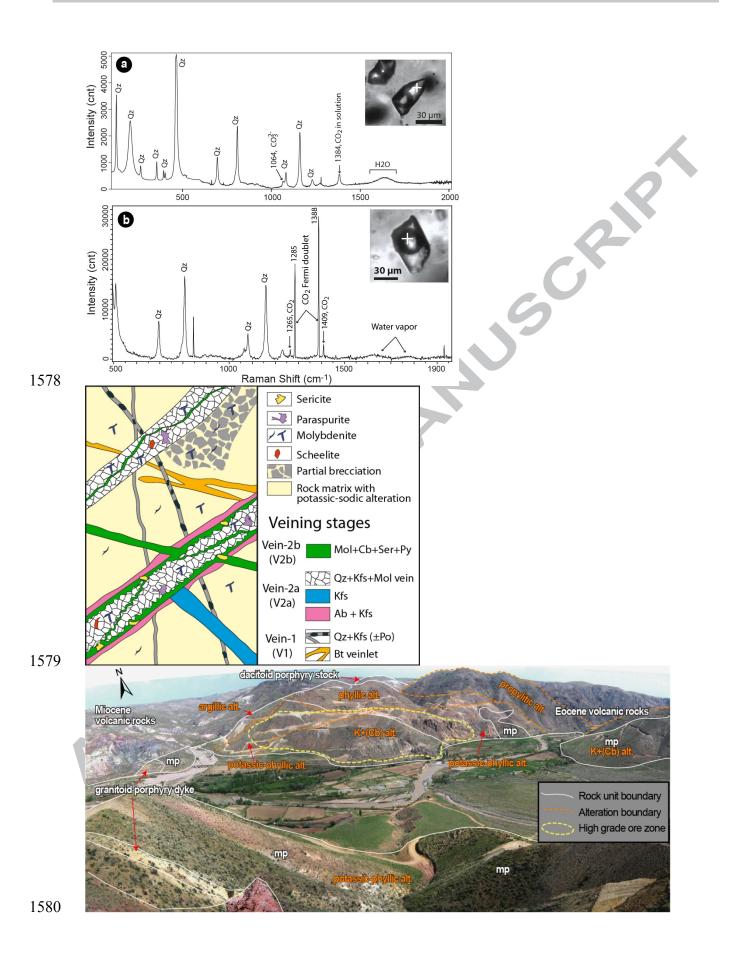


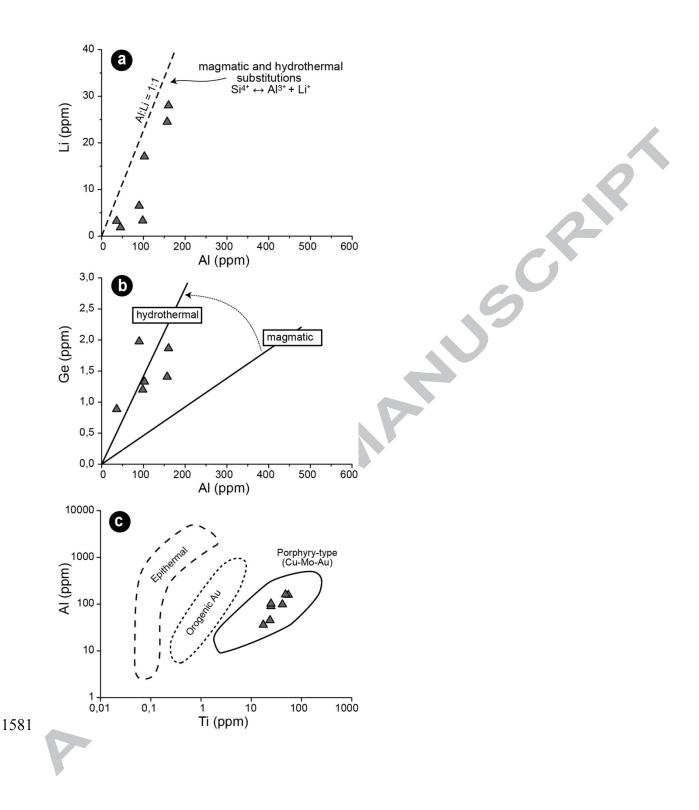


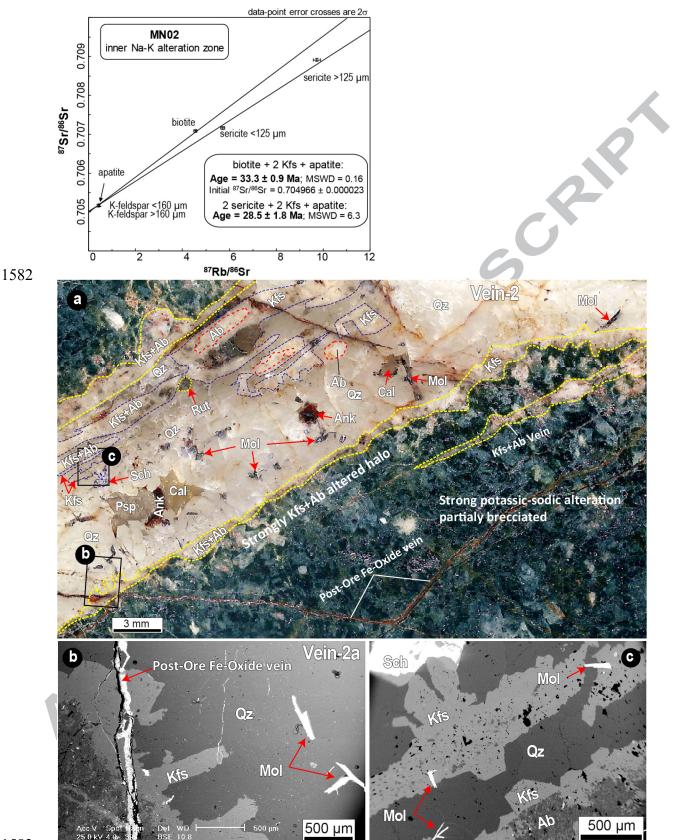


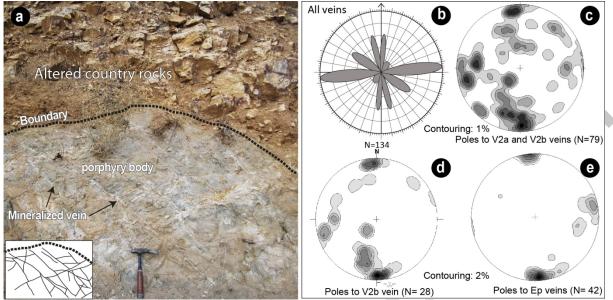
	Sty	le	Stage-1(V1)	Stage-2a (V2a)		Stage-2b (V2b)
Minerals	Major	Minor	Na alteration K alteration	Phyllic-Silicic	Siliceous- carbonatization	Molybdenite enrichment and carbonatization
Magnetite		鬰				
Albite	辙	×		1		Mineralization styles
K-Feldspar I	徽					Ø Disseminated
Biotite	繳	×		1		× Veining
Ilmenite		躑				Brecciated
Quartz I		X 淡		1		
Pyrrhotite	Li	×				
Molybdenite I		癬		1		
Apatite		X 欲		1		
K-Feldspar II	X	†				
Quartz II						
Anhydrite		$\times$				
Scheelite	1 1	×				
Rutile		$\mathbf{x}$				
Molybdenite II	l i	\$® ×				
Pyrite	\$\$					
Quartz III	×					
Sericite	靈	×				
Paraspurite		×				
Ankerite	×			1		
Rhodochrosite	X					
Calcite	X			1		
Molybdenite III	X					
Hemimorphite		×				
Kaolinite		×		1		

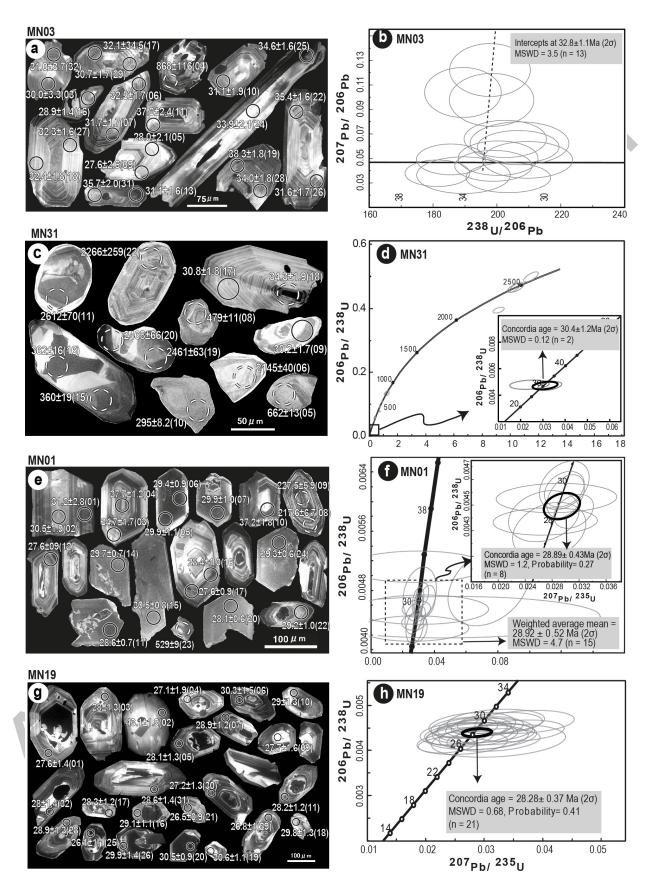


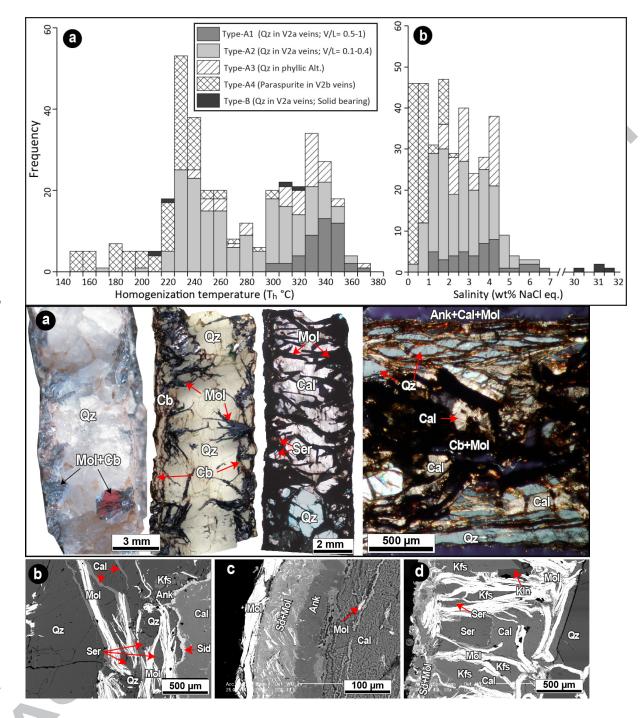


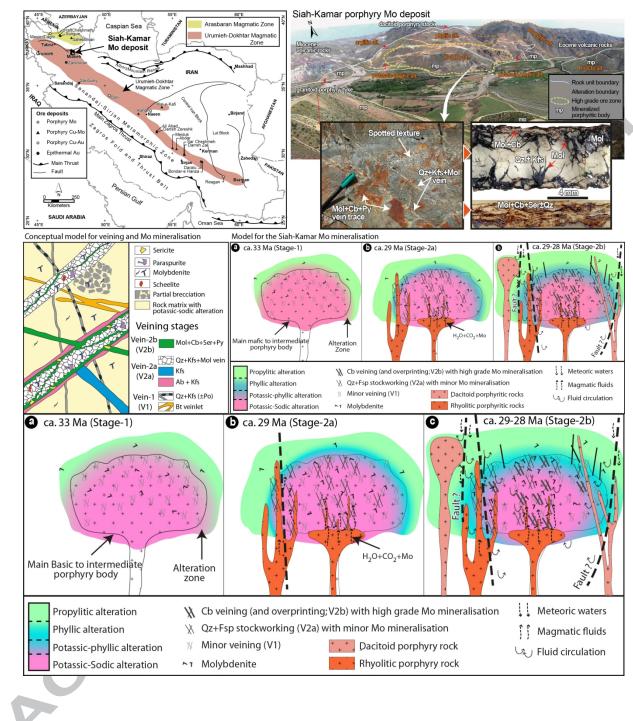














MAN

### 1595

### 1596 Highlights

- 1597 Long-lived magmatism and differentiation control Mo ore mineralisation
- 1598 Mo ore-enhancement caused by renewed magmatism

- 1599 Mo ore associated with carbonatisation, acidic fluid neutralisation and
- 1600 cooling
- 1601 Refertilised crust leading to Mo ore formation in the Urumieh-Dokhtar zone
- 1602