

1 **Understanding the role of selenium in defect passivation for highly efficient selenium-alloyed cadmium**
2 **telluride solar cells**

3 Thomas A. M. Fiducia¹, Budhika G. Mendis², Kexue Li³, Chris R. M. Grovenor³, Amit H. Munshi⁴, Kurt
4 Barth⁴, Walajabad S. Sampath⁴, Lewis D. Wright¹, Ali Abbas¹, Jake W. Bowers¹, and John M. Walls¹

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6 ¹Centre for Renewable Energy Systems Technology (CREST), Wolfson School of Mechanical, Electrical and
7 Manufacturing Engineering, Loughborough University, Loughborough, Leicestershire, LE11 3TU, United
8 Kingdom.

9 ²Department of Physics, Durham University, South Road, Durham DH1 3LE, United Kingdom.

10 ³Materials Department, Oxford University, Oxford, OX1 3PH, United Kingdom.

11 ⁴Colorado State University, Fort Collins, Colorado, 80523, USA.

12
13 **Abstract**

14 Electricity produced by cadmium telluride (CdTe) photovoltaic modules is the lowest cost in the solar industry, and
15 now undercuts fossil fuel-based sources in many regions of the world. This is due to recent efficiency gains
16 brought about by alloying selenium into the CdTe absorber, which has taken cell efficiency from 19.5% to its
17 current record of 22.1%. While the addition of selenium is known to reduce the bandgap of the absorber material
18 and hence increase cell short-circuit current, this effect alone does not explain the performance improvement. Here,
19 by means of cathodoluminescence (CL) and secondary ion mass spectrometry (SIMS), we show that selenium
20 enables higher luminescence efficiency and longer diffusion lengths in the alloyed material, indicating that
21 selenium passivates critical defects in the bulk of the absorber layer. This passivation effect explains the record-
22 breaking performance of selenium-alloyed CdTe devices, and provides a route for further efficiency improvement
23 that can result in even lower costs for solar generated electricity.

24
25 **Introduction**

26 In the last 7 years the efficiency of cadmium telluride (CdTe) solar cells has improved from 16.7% to 22.1% [1], [2].
27 This has enabled the cost of CdTe photovoltaic electricity to decrease to the point where it is lower than for silicon
28 photovoltaics, and lower than for conventional fossil fuel sources in many regions of the world [3], [4]. The most
29 recent efficiency improvements, from 19.5% to 22.1%, have been achieved by the addition of selenium to the front
30 of the CdTe absorber layer [5]. This creates a CdSe(x)Te(1-x) (CdSeTe) alloy that lowers the material bandgap,
31 increases absorption in the long wavelength part of the spectrum, and increases device short-circuit current density
32 [6]–[8]. However, in addition to improved current generation, selenium alloying maintains or improves open circuit
33 voltages – despite the lower bandgap [9]. Recent studies have shown that this is associated with improved minority
34 carrier lifetimes in the absorber, but it is not known why the lifetimes increase [9]–[11].

35 One reason that has been proposed for the higher carrier lifetimes is that the selenium improves band alignments
36 between the absorber and buffer layers at the front interface of the device, reducing interfacial recombination [9].
37 Another is that the wider-bandgap, non-alloyed CdTe at the back of the device acts as an electron reflector and
38 reduces back surface recombination [12]. These explanations both relate to the effects of selenium at the device
39 level, i.e. its effects on the one-dimensional band structure through the depth of the cell. However the effects of
40 selenium on the basic optoelectronic properties of the absorber material have not been investigated.

41 Here we use high resolution cathodoluminescence (CL) microscopy to map nanoscale variations in electronic
42 properties in a high efficiency selenium-graded CdTe device, including variations in luminescence efficiency,
43 diffusion length, and effective bandgap. We then compare these with secondary ion mass spectrometry (SIMS)
44 elemental maps of the selenium distribution from the same area of the film, so that the electronic property variations
45 can be directly related to nanoscale changes in the selenium concentration.

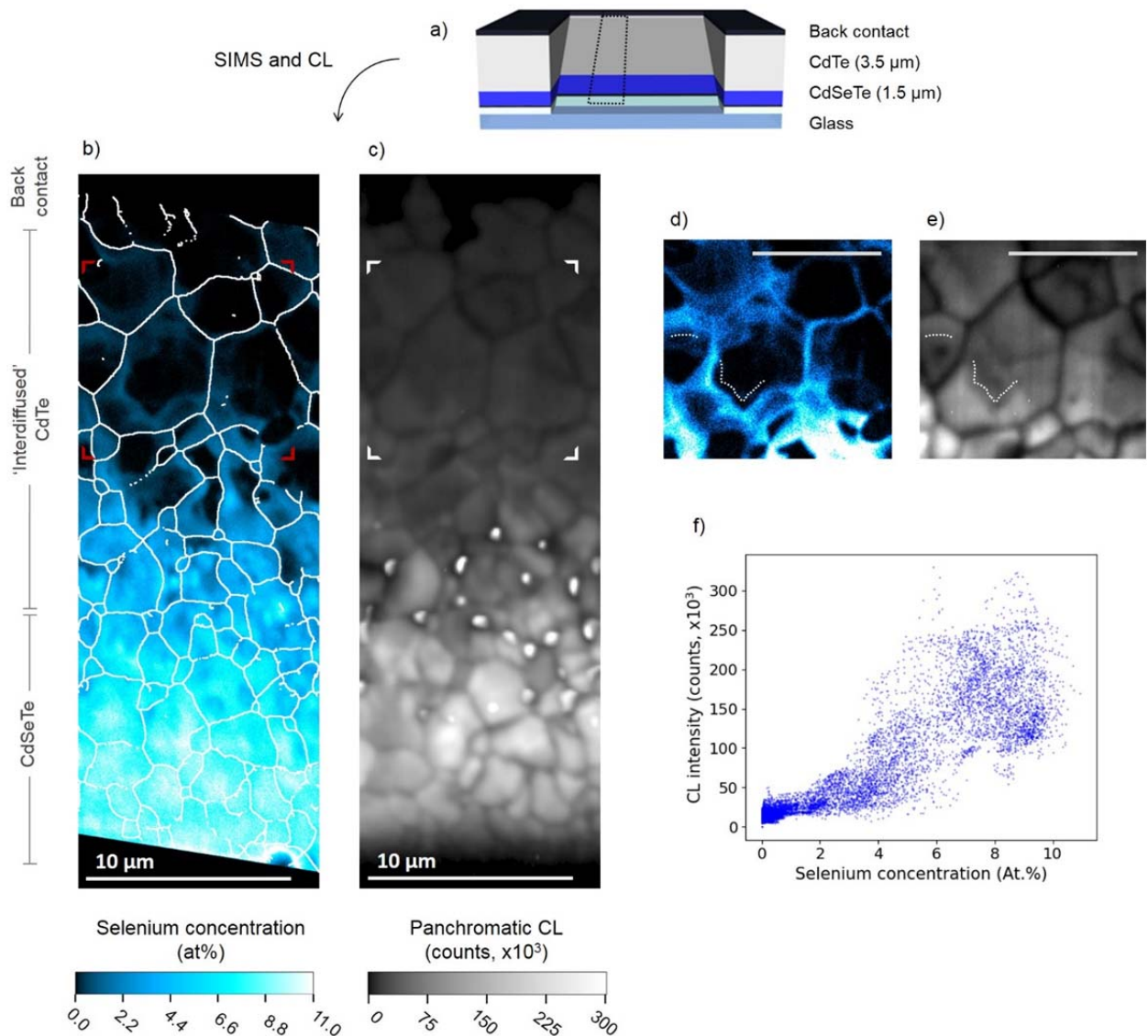
46 The results reveal that selenium causes clear and dramatic improvements in the local luminescence efficiency of
47 the absorber material, indicating that it passivates deep-level defects in CdTe. This provides an explanation for the
48 superior voltage and performance of selenium-alloyed CdTe devices and may enable further gains in efficiency
49 beyond the current record level of 22.1%.

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51 **Selenium diffusion into the CdTe layer**

52 In order to perform the investigation, a cell stack incorporating a ~ 1.5 μm layer of CdSeTe at the front of the
53 absorber layer was fabricated as shown schematically in Fig 1a and described in the Methods section. Following a

54 cadmium chloride (CdCl_2) activation treatment, which is used universally to produce high efficiency CdTe-based
55 devices [13]–[16], the



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58 **Fig 1: CdTe luminescence efficiency improved by selenium alloying.** (a) Schematic of the CdSeTe/CdTe

59 device structure and 7° bevel, with an outline of the SIMS and CL measurement area (dotted black line).

60 (b) SIMS elemental map of the selenium distribution on the bevelled surface of the CdCl₂ treated device, with the

61 'skeletonised' chlorine signal overlaid in white to help delineate grain boundaries (see Methods section for

62 calibration of the selenium concentration). Darker blue corresponds to lower selenium concentrations and brighter

63 cyan/white corresponds to higher concentrations, towards 10 at% (see colour bar). (c) Cathodoluminescence (CL)

64 map of the panchromatic CL intensity, taken on the same area as the selenium map. Black and dark grey
65 corresponds to lower CL signal intensity and brighter white to higher signal intensity (see colour bar). (d) High
66 magnification selenium map of a region at the top of the bevel, shown by the red annotations in (b) (the image has
67 no skeletonised chlorine overlay, so that selenium signal at the grain boundaries can be seen). The scale bar is 5
68 μm). (e) Corresponding high magnification CL map to the SIMS map in (d). The dashed lines in (d) and (e)
69 highlight similarity in positioning of the selenium and CL signals. (f) Scatter plot of selenium concentration versus
70 CL counts for equivalent regions (i.e. pixels) in (b) and (c), with grain boundary regions and voids omitted from the
71 analysis (voids can be seen as bright white spots in the panchromatic CL in (c)).
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73 cell was measured to have an efficiency of 16.8%. This value is high amongst CdTe photovoltaics research
74 laboratories, and devices made using this exact deposition system and method have achieved certified conversion
75 efficiencies of up to 18.3% [17] (only exceeded by the 22.1% champion cell fabricated by First Solar Inc [2]). After
76 the efficiency measurement, a shallow 7° bevel was milled through the device stack using a Focussed Ion Beam
77 (FIB). This provided direct access to the CdSeTe layer and presented an extended, smooth cross-section for the CL
78 and elemental mapping. Cathodoluminescence measurements were performed on the bevel surface on an area as
79 outlined on the schematic in Fig 1a. Correlative SIMS elemental mapping at high resolution was then performed on
80 exactly the same region as the CL.

81 A NanoSIMS map of the selenium concentration on the bevel surface in the CdCl₂-treated cell is shown in Fig 1b.
82 At the bottom of the bevel, near the front interface of the absorber layer, selenium concentrations are in the range
83 8-10.5 at% (cyan/white). This is similar to the as-deposited device. However, in contrast to the as-deposited device,
84 which has a sharp interface between the CdSeTe and CdTe layers (see [18], and Supplementary Figure 1), the
85 CdCl₂-treated cell shows a gradual decrease in the selenium concentration with distance away from the front
86 interface. This means that selenium signal is detected towards the back of the cell, in the CdTe layer (for clarity now
87 referred to as the ‘interdiffused CdTe’ region).

88 In addition to this grading through the absorber depth, there is also some non-uniformity in the selenium
89 concentration around the grain structure of the absorber. In the interdiffused CdTe region at the top of the bevel,
90 close to the back contact, there is an enhanced selenium signal at the grain boundaries and around the fringes of the
91 grains (this can be seen clearly in the high magnification image in Fig 1d). However at the bottom of the bevel, in
92 the CdSeTe region, the selenium concentration is lower around the grain boundaries than in the grain interiors.

93 This distribution of selenium around grain boundaries indicates that during the high-temperature cadmium chloride
94 activation treatment the grain boundaries provide channels for fast diffusion of selenium from the CdSeTe layer into
95 the CdTe above, which then slowly out-diffuses from the grain boundaries into the grain interiors. This is a mixed
96 grain boundary and lattice diffusion regime of type B in the Harrison classification system, and is also observed with
97 sulphur interdiffusion in conventional CdS/CdTe devices [15], [19].

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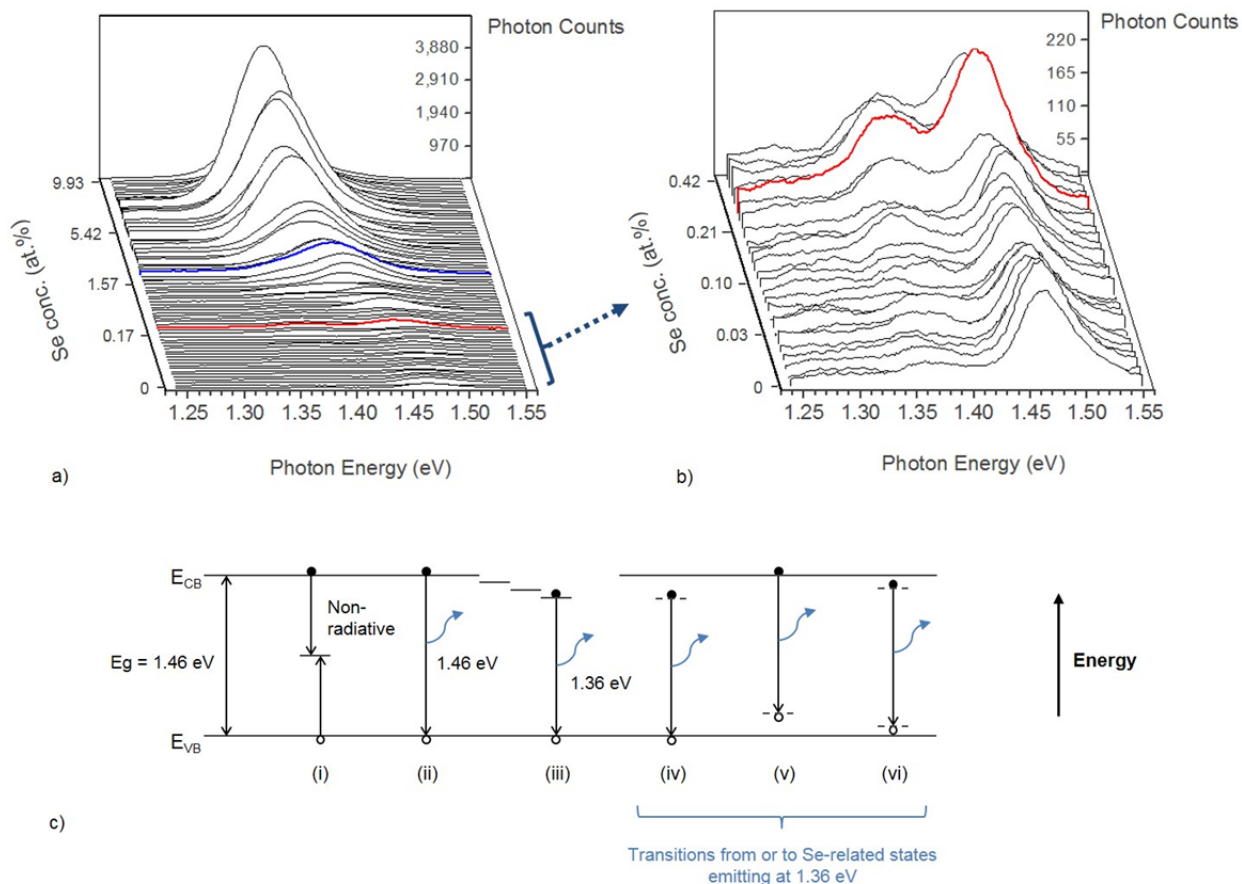
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101 **Selenium-induced defect passivation**

102 To assess the electronic effects of selenium alloying in CdTe we have performed correlative
103 cathodoluminescence imaging on the bevel, on the same area as the selenium map (see Fig 1c). The map
104 shows that at the top of the bevel, in the interdiffused CdTe region, the luminescence intensity is relatively
105 low. However at the bottom of the bevel, in the CdSeTe region, the luminescence intensity is much brighter,
106 with counts 10 to 20 times higher than at the top (150,000 - 300,000 counts versus ~ 15,000). This steep
107 increase in CL signal through the depth of the absorber layer mirrors the increase in selenium concentration.

108 In addition to this general trend down the bevel, there are variations in CL signal around the grain structure
109 of the absorber. Firstly, it can be seen that luminescence is lower at grain boundaries than in grain interiors.
110 This is in line with previous CL measurements on non-alloyed CdTe, and shows that grain boundaries act as
111 non-radiative recombination centres [20]–[23]. Secondly, brighter cathodoluminescence signals can be seen
112 around the fringes of grains in the interdiffused CdTe region. This closely matches the distribution of selenium
113 observed around grains in the SIMS maps (see the higher magnification images in Fig 1d and Fig 1e)).

114 To quantify this selenium concentration vs luminescence relationship we have exactly aligned and
115 repixelated the SIMS and CL images so that the intensity values in each can be compared pixel-for-pixel.
116 This alignment is relatively straightforward because of the obvious features that delineate grain boundaries in
117 both maps (see Supplementary Figure 2b, where the SIMS and CL images have been superimposed on top of
118 one another). The results of this comparison are shown in the scatter plot in Fig 1f. The plot shows that in
119 regions containing < 2% selenium the CL intensity is generally less than 30,000 counts, and in regions
120 containing > 9% selenium the CL intensity is up to 300,000 counts. Since high luminescence efficiency is
121 indicative of low levels of defect-mediated non-radiative recombination in a semiconductor, this clear positive
122 correlation suggests that selenium passivates a non-radiative recombination centre in the alloyed material.
123 Moreover, we have observed this effect in multiple SEM systems and on multiple samples, including on
124 untreated CdSeTe/CdTe devices (see Supplementary Figure 3). Together these results



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126 **Fig 2: Hyperspectral CL imaging shows that selenium is associated with a sub-bandgap emission peak.**

127 (a) A waterfall plot of a random sample of CL spectra taken from the bevel measurement area. Curves are sorted

128 by the concentration of selenium present in the corresponding pixel in the SIMS map. The curves highlighted in

129 blue and red are deconvoluted in Supplementary Figures 4a and 4b respectively. (b) A waterfall plot of the low-

130 selenium CL spectra in (a), representing a selenium concentration range from 0 to 0.42%. The curve highlighted

131 in red is the same as that highlighted in (a) and is deconvoluted in Supplementary Figure 4b. (c) Energy band

132 diagram showing a selection of possible electron and hole recombination channels in the material: (i) defect-

133 mediated non-radiative recombination, (ii) radiative band-to-band recombination with emission at 1.46 eV, (iii)

134 radiative band-to-band recombination with emission at 1.36 eV, (iv)-(vi) radiative recombination with emission at

135 1.36 eV, either from or to selenium-related states in the band gap.

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137 suggest that selenium alloying creates a passivation effect in CdTe that explains the higher open circuit voltage
138 and improved performance of selenium-graded CdTe solar cells.

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140 **Selenium related sub-bandgap states**

141 The cathodoluminescence image shown in Fig 1 is of the panchromatic CL signal. This means that the
142 intensity value in each pixel is the total number of photons collected by the CL system over all wavelengths
143 (i.e 700 – 999 nm) when the electron beam is in that region of the sample. However, important information is
144 contained in the specific wavelengths of light that are emitted from the material, which is not displayed in the
145 panchromatic image since it is a simple photon count. As such, in this section we build on the general picture
146 provided by the panchromatic image by analysing the underlying spectra of wavelengths that make it up. This
147 is possible because the CL data we collected on the bevel is hyperspectral, meaning that a full spectrum is
148 collected at each pixel.

149 Figure 2a shows a waterfall plot of a representative sample of CL spectra from the panchromatic image of the
150 bevel. To make the plot, a sample of 73 of the spectra in the hyperspectral map was selected at random for
151 display. The curves from these pixels were then sorted by the concentration of selenium present in the
152 corresponding pixel in the SIMS map, with the curves with the highest selenium content at the back of the
153 plot and the lowest at the front.

154 The data shows a steep increase in CL intensity as the selenium concentration in the corresponding pixel
155 increases. In addition, it shows a shift of the emission peak towards lower energies with increasing selenium
156 content. This shift is expected, since it is known that increasing the selenium concentration initially decreases
157 the bandgap of CdSeTe alloys due to a bandgap ‘bowing’ effect [7], [24] (this effect is explored in more
158 detail in the “Selenium-induced bandgap gradients” section, where we map the effective bandgap of CdSeTe
159 material over the bevel measurement area).

160 In the curves with a low selenium concentration (0-0.42 at%) plotted in Fig 2b we observe two distinct
161 emission peaks: one at ~1.46 eV, which we attribute to band-to-band recombination, and a sub-bandgap peak
162 at ~1.36 eV. The intensity of the sub-gap peak increases with selenium content, indicating that selenium
163 creates states within the bandgap when present in the CdTe in low concentrations. Possible transitions from
164 or to such selenium-related states are depicted in Fig 2c. While the intensity of the sub-gap peak increases

165 with selenium content, it is noted that that this does not affect the intensity and position of the band-to-band
166 recombination peak. This is more clearly shown in the deconvoluted CL spectra in Supplementary Figure 4a
167 and 4b, where the highlighted curves in the waterfall plots are analysed). It is also noted that the emission tail
168 of the band-to-band recombination peak can make a contribution to the intensity of the 1.36 eV peak (see
169 Supplementary Figure 4).

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171 **Impact of selenium concentration on diffusion lengths**

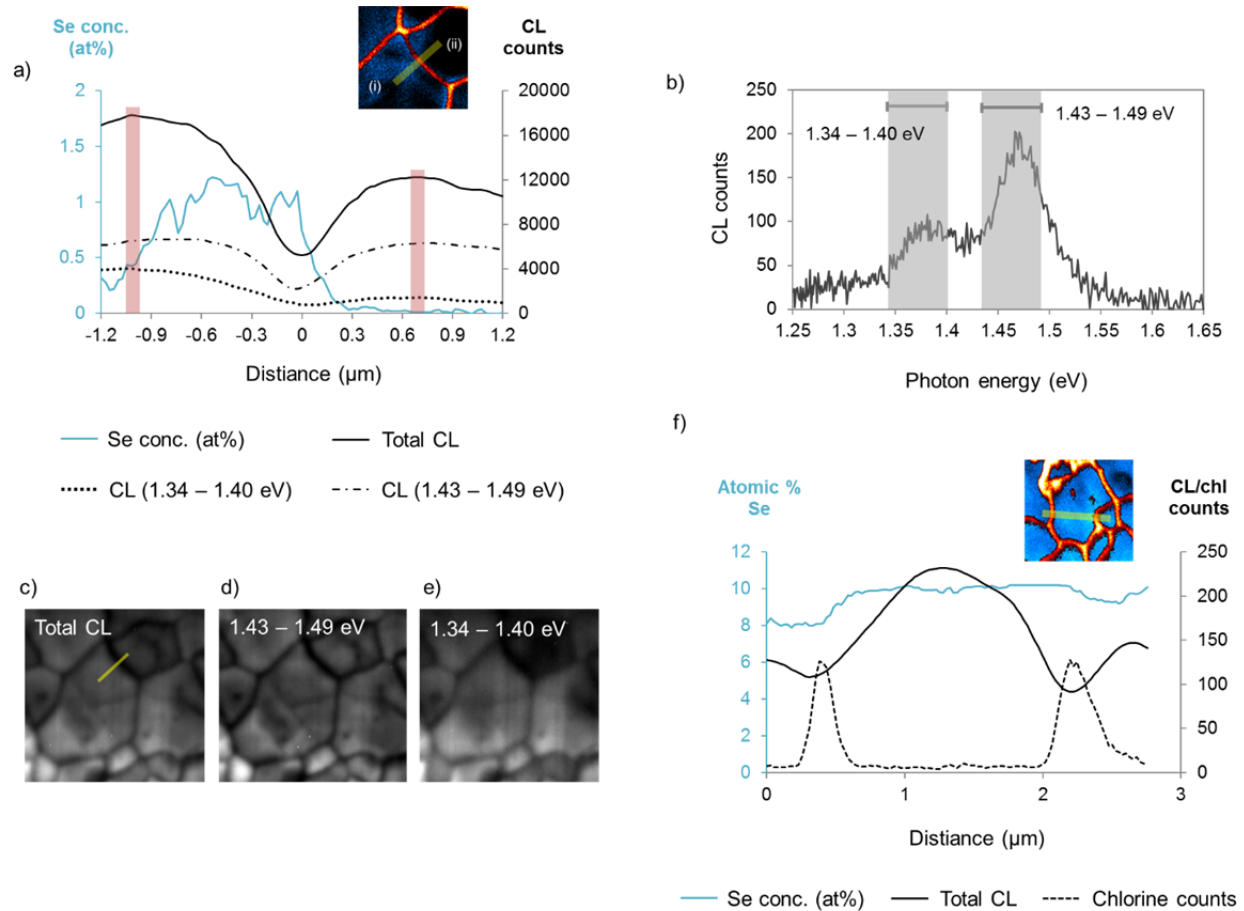
172 The luminescence data presented here shows that higher levels of selenium in the CdTe material lead to an
173 increased CL signal intensity. This suggests that selenium passivates a defect in bulk CdTe, decreasing non-
174 radiative recombination and increasing cell performance. To confirm this, we can use CL to estimate local
175 diffusion lengths in the material. This is done by analysing how the CL intensity varies in proximity to grain
176 boundaries. An example of this analysis is shown in figure 3a. Here we take a line profile of the selenium and
177 CL signals across a grain boundary in the interdiffused CdTe region of the cell, as shown in the inset. The
178 profile shows that the selenium levels in grain (i) are higher than grain (ii), at 1 at% compared to ~0.03 at%
179 respectively, resulting in a step in selenium concentration across the grain boundary. While there is some
180 variation in the selenium concentration in grain (i) the profile is uniform on the length scale of the CL
181 analytical volume, which is ~1 μm .

182 Along with the change in selenium concentrations, the figure shows that there is clear asymmetry in the
183 intensity of the CL profiles either side of the boundary. In the grain with higher selenium concentration, the
184 total (panchromatic) CL signal plateaus at a distance $> 1 \mu\text{m}$ from the grain boundary (referred to from now
185 on as the 'plateau distance', and shown by red shading in the Fig 3a). This compares to $\sim 0.7 \mu\text{m}$ in the low-
186 selenium grain. This is indicative of a greater diffusion length in the high-selenium grain because here, even
187 when the electron beam is up to a micron away from the grain boundary, generated carriers experience the
188 effects of the boundary (i.e. they are able to travel the $\sim 1 \mu\text{m}$ distance to the boundary and recombine there,
189 quenching the CL signal compared to its maximum level at the plateau). Calculated diffusion lengths from
190 the profiles back this up, with the diffusion length in the ~ 1 at% selenium-containing grain being $0.25 \mu\text{m}$,
191 compared to $0.14 \mu\text{m}$ in the low-selenium grain (it should be noted that calculations of diffusion lengths are
192 approximate since the model assumes that the grain boundary is a free surface where carriers generated in

193 one grain cannot diffuse past the grain boundary into the adjacent grain [25], [26]. In addition, states on the
194 bevelled surface may artificially decrease minority carrier diffusion lengths).

195 This kind of analysis was performed in two other regions with a similar step in selenium concentration across grain
196 boundaries, with the results shown in table 1. The data shows that diffusion lengths in the grains with higher
197 selenium concentration are consistently longer than in the low-selenium grains (note that grain boundaries were only
198 chosen for this analysis if they had a step-change in selenium concentration across the boundary, as well as no other
199 grain boundaries nearby that could influence the CL profile).

200 To gain further insight into why the diffusion lengths differ, we can assess the relative contributions to total CL
201 signal variations of the band-to-band peak at ~ 1.46 eV and the selenium-related sub-gap peak at ~ 1.36 eV (Fig 3
202 a). The 1.43 – 1.49 eV line in the figure plots the variation in CL counts within just that spectral window (shown
203 by the right-hand grey region in Fig 3b). This captures intensity variations in just the band-to-band peak, and shows
204 that variations in band-to-band transitions are symmetric either side of the grain boundary, despite the differences
205 in selenium content (the plateau distance is ~ 0.7 μm on both sides of the GB). However, variations in the CL
206 intensity in the 1.34 to 1.40 eV window, where the 1.36eV selenium-related peak falls, are not symmetric across
207 the boundary. The plateau distance for this energy range in the high-selenium grain is > 1 μm , compared to ~ 0.65
208 μm in the low- selenium grain. The 1- μm long plateau distance in the sub-gap peak is therefore the reason for the
209 similarly long plateau distance observed in the total panchromatic CL curve. This demonstrates that the long
210 lifetime of selenium-related transitions enables the longer diffusion lengths in regions with higher selenium
211 content. In addition, the symmetry of the band-edge luminescence profile (1.43 – 1.49 eV) shows that any field-
212 effects due to the stepped selenium profile across the grain boundary have not influenced the CL profiles, since the
213 effect of an electric field is to weaken CL emission at all wavelengths. Therefore, any asymmetry in the field due to
214 the selenium step would have caused asymmetry in the 1.43 – 1.49 eV profile.



215

216 **Fig 3: CdTe diffusion lengths improved by selenium alloying.** (a) Profile plot of the selenium concentration
 217 (blue line, plotted on the left y-axis) and CL intensities (black lines, plotted on the right y-axis) across a grain
 218 boundary in the interdiffused CdTe region at the top of the bevel. The exact profile area is shown by the yellow
 219 line in the inset (red in the inset is the chlorine SIMS signal, blue is the selenium SIMS signal, and (i) and (ii)
 220 indicate the grains with high- and low- selenium concentrations respectively). The solid, dashed, and dot-dashed
 221 black lines show the profile of the total, 1.43 - 1.49 eV, and 1.34 - 1.40 eV CL signals respectively (see legend).
 222 Red shading indicates the 'plateau distance' for the total CL signal on either side of the grain boundary i.e. the
 223 distance from the grain boundary, at 0 μm , to where the CL signal plateaus. (b) Plot of the average CL spectrum
 224 for the area analysed in panels (c - e), with shading showing the spectral windows used in the analysis. (c) Map of
 225 total panchromatic CL signal in a region at the top of the bevel, with the yellow line showing the region where the
 226 profile in (a) was taken. (d) Map of CL counts in the 1.43 - 1.49 eV spectral window shown by the shading in (b).
 227 (e) Map of CL counts in the 1.34 - 1.40 eV spectral window shown in (b). (f) Profile plot of the selenium

228 concentration (blue line, plotted on the left y-axis), chlorine SIMS counts (dashed black line, plotted on the right y-
229 axis), and total CL intensity (solid black line, plotted on the right y-axis) across the high-selenium CdSeTe region
230 shown in the inset (red in the inset is the SIMS chlorine signal and cyan is the SIMS selenium signal). Chlorine
231 signal intensity is included in the plot to show the position of the grain boundaries. Note that the profile is from a
232 region with a relatively uniform selenium concentration so as not to influence the shape of the CL profile.
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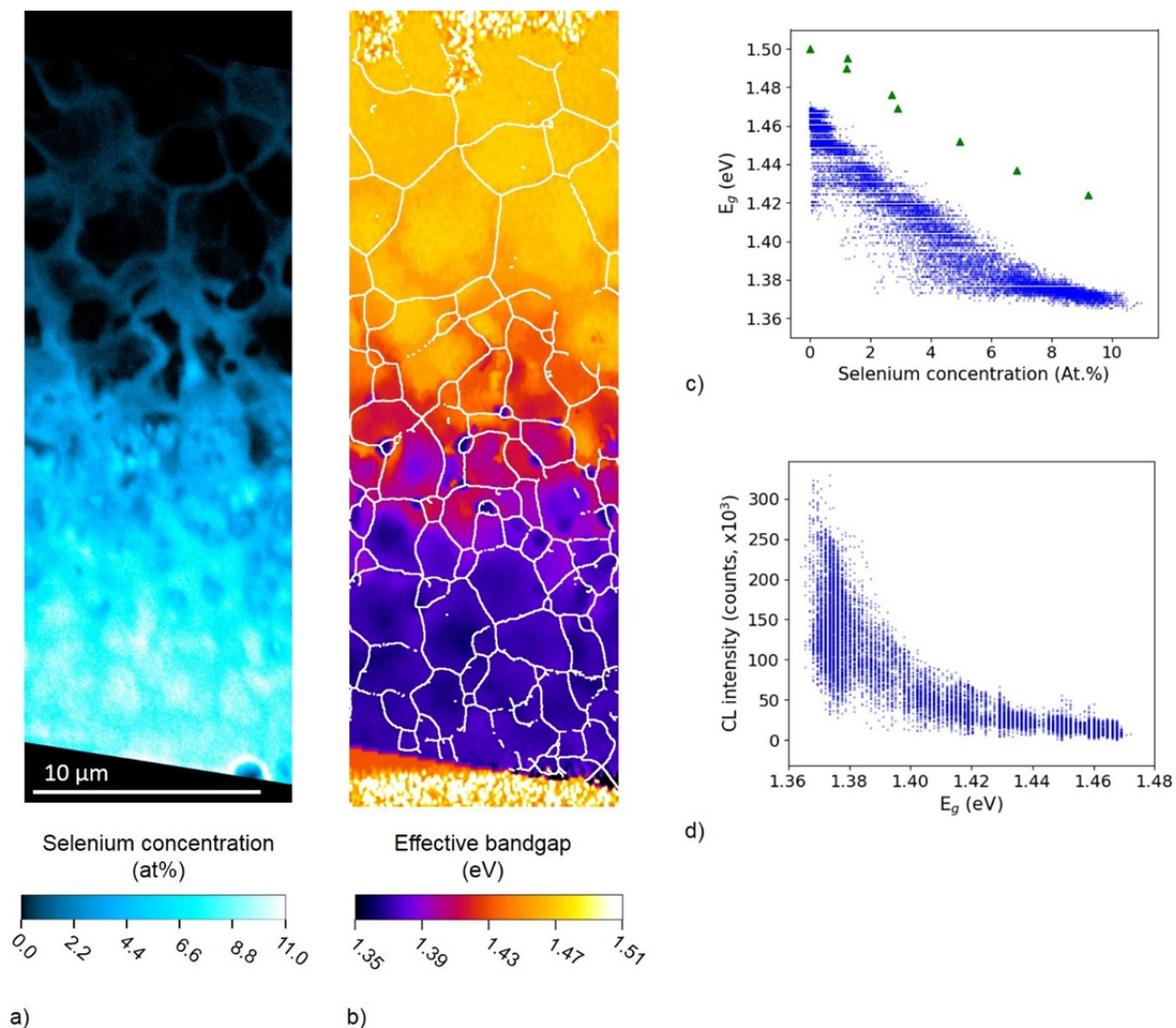
234 Mapping variations in CL counts in these different energy windows shows that the trend for longer plateau distances
235 in the 1.34 – 1.40 eV energy window vs the 1.43 – 1.49 eV window is present across large regions of the bevel
236 measurement area (see Figs 3c - 3e). For instance, sharp image contrast is seen in the band-to-band transitions map
237 in Fig 3d, indicating steep V-shaped drops in CL signal across grain boundaries and therefore short plateau
238 distances. However, when the sub-gap energy window is mapped (Fig 3e), the contrast in the image is low. This
239 means that signal variations across grain boundaries in the image have a shallower V-shape, indicating larger plateau
240 distances, as demonstrated by the 1.34 – 1.40 eV line profile in Fig 3a.

241 The diffusion length analysis has only been performed on the upper part of the bevel where the concentration of
242 selenium is low compared to the CdSeTe region at the bottom of the bevel. This is because in the CdSeTe region
243 there are no obvious plateaus in the CL signal within the grain interiors, making it difficult to perform either a rough
244 ‘plateau distance’ analysis as done in Fig 3a, or to calculate diffusion lengths. However, this in itself is an indication
245 of long diffusion lengths in the region, since the signal does not plateau even in large grains, such as the 2-micron
246 diameter grain shown in Fig 3f. These results provide further evidence that selenium passivates bulk defects in
247 CdSeTe alloys, even at very low alloying fractions.

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249 **Selenium-induced bandgap gradients**

250 The initial purpose of alloying selenium into the front of CdTe solar cell absorbers was to decrease the material
251 bandgap and therefore increase absorption in the low-energy part of the visible spectrum. However, very little
252 research has been published quantifying the effects of selenium on the bandgap



253

254 **Fig 4: Mapping selenium-induced bandgap changes in the absorber.** (a) SIMS map of the selenium

255 concentration on the bevel surface, as seen in figure 1 but without the skeletonised chlorine signal overlay.

256 Darker blue corresponds to lower selenium concentrations and brighter cyan/white corresponds to higher

257 concentrations, towards 10 at% (see colour bar). (b) Map of the peak emission energy of the CL (i.e. the

258 effective band gap of the material) on the same area as (a), with a skeletonised chlorine signal overlay and with

259 pixel dimensions matched to the SIMS map. This enables pixel-for-pixel comparison of the selenium

260 concentration and the effective band gap at each point on the bevel surface, as shown in the scatter plot in (c).

261 Yellow/orange corresponds to a higher effective bandgap of ~ 1.47 eV and purple corresponds to a lower effective

262 bandgap of ~ 1.37 eV (see colour bar). (c) (Blue) scatter plot of the effective bandgap vs selenium concentration at

263 each point/pixel on the bevel, and (green) plot of the band gap vs selenium concentration reported in [7]. (d)
264 Scatter plot of the effective bandgap vs panchromatic CL signal at each point on the bevel. Regions containing
265 voids were omitted from the analysis in producing the scatter plot (voids can be seen as white spots mid-way
266 down the bevel in the CL map in Fig 1c).
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of CdTe [7]. In this section we map the effective bandgap of CdSeTe material at high resolution over the bevel measurement area by tracking the photon energy of the dominant emission in each pixel in the CL. Fig 4a shows a SIMS map of the selenium distribution on the bevel surface, as seen in Fig 1b. Alongside this in fig 4b is a map of the peak CL photon emission energy (i.e. effective bandgap), taken on the same area as the SIMS. The map shows that at the top of the bevel, towards the back contact of the device, the effective bandgap of the absorber is ~ 1.46 eV (yellow). This is typical for non-alloyed CdTe [7]. Further down the bevel, there is a steady decrease in the effective bandgap down to ~ 1.36 eV in the CdSeTe region (purple). This mirrors the grading of the selenium concentration and will form a built-in field through the depth of the device.

In Fig 4c we have plotted a scatter graph of selenium concentration against effective bandgap for each pixel in the two images, in the same way as performed previously for the total CL intensity scatter plot (Fig 1f). The plot has a classic bowing shape with an initial, steeper decrease in bandgap followed by a slight levelling off towards 10 at% selenium. This is consistent with the data points plotted in green (replotted from [7]), which show bandgap vs composition for untreated CdSeTe films deposited on glass using the same deposition system as used in this study. For these films, composition was obtained using energy-dispersive x-ray spectroscopy (EDX) measurements and the bandgap was extracted from absorption edge Tauc plots. Although the two curves are well-matched, the curve produced with the CL data is shifted down by ~ 0.03 eV. This may be due to the influence of excitonic peaks in the emission spectra, which would lower the energy of the luminescence peak compared to the true bandgap [27]. In addition, the CL is measured in a CdCl₂ treated film incorporated into a working device, as opposed to untreated CdSeTe films on a glass substrate, so intentional absorber impurities such as chlorine may slightly alter the material bandgap.

Table 1: Diffusion lengths. Calculated diffusion lengths either side of three grain boundaries in the interdiffused CdTe region. Across each boundary there is a step in the selenium concentration. Grain boundary 1 is shown in Fig 3a.

Grain Boundary	Side	Se Content (at%)	Diffusion Length (μm)
1	Se-poor side	0.03%	0.143
	Se-rich side	0.94%	0.246
2	Se-poor side	0.14%	0.163
	Se-rich side	0.90%	0.197
3	Se-poor side	0.02%	0.118
	Se-rich side	0.58%	0.203

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298 As well as a grading through the depth of the absorber the map in Fig 4b shows variations in the effective
 299 bandgap around grain boundaries in the material. At the top of the bevel, in the interdiffused CdTe region, there is a
 300 decrease in the effective bandgap around grain boundaries and grain fringes (darker orange) vs the grain interiors
 301 (lighter yellow). This matches the positioning of selenium seen in the SIMS maps, and corresponds to a decrease in
 302 bandgap of ~ 0.01 eV in these regions. However, at the bottom of the bevel, in the CdSeTe region, there is an
 303 increase in effective bandgap around grain boundaries (lighter purple). This tracks the decrease in the selenium
 304 content seen at grain boundaries in the SIMS maps in this region.

305 Fig 4d shows a scatter plot of effective bandgap versus total CL signal at each point on the bevel. The plot
 306 shows an initially shallow, and then steep rise in material luminescence efficiency with decreasing bandgap. This
 307 means that, for instance, decreasing the bandgap from 1.46 eV to 1.42 eV gives a much smaller increase in defect
 308 passivation than an equal step from 1.42 eV to 1.38 eV. This information will be useful to device designers and
 309 fabricators when deciding on, or modelling the optimum selenium grading profile.

310

311 **Discussion**

312 Remarkable progress has been made in cadmium telluride photovoltaics despite the lack of a full, fundamental
 313 understanding of absorber layer material properties. This is especially true of the record-breaking selenium-
 314 graded devices. In this work, by means of correlative cathodoluminescence and SIMS measurements, we have
 315 shown that selenium alloying enables high luminescence efficiency in alloyed CdTe, suggesting that it passivates

316 defects in both CdCl₂-treated and untreated bulk CdTe. Further evidence for this passivation effect is provided by
317 an analysis of CL signal variations across grain boundaries, which shows longer carrier diffusion lengths in
318 alloyed regions with higher selenium content. Hyperspectral CL imaging shows that selenium not only shifts the
319 band-to-band emission peak from ~1.46 eV to ~1.36 eV, creating a bandgap gradient and built-in field across the
320 absorber, but it also creates radiative sub-gap states at ~1.36 eV, which is the same energy as the lowest point on
321 the bandgap bowing curve. The high-resolution SIMS measurements show that during the CdCl₂ heat treatment
322 process selenium diffuses from the CdSeTe into the CdTe, primarily along grain boundaries but with some out-
323 diffusion from grain boundaries into grain interiors. This causes an excess of selenium at grain boundaries in the
324 CdTe layer that decreases the material bandgap in those regions. In addition, it causes a selenium deficiency at
325 grain boundaries in the CdSeTe layer, resulting in higher band gaps in these regions relative to the grain interiors.
326 This will have unknown effects on carrier transport around grain boundaries in selenium-graded devices.
327 These results show that in non-alloyed CdTe there are deep-level defects in the bulk material that act as
328 recombination centres and limit device efficiency, even following the CdCl₂ treatment. Potential candidates for
329 these defects are cadmium vacancies (VCd) and tellurium-on-cadmium antisites (TeCd), which Density
330 Functional Theory (DFT) modelling has shown act as harmful recombination centres [28], [29]. We suggest that
331 these defects can be passivated or made less likely to form when selenium is present, with an increasing
332 passivation effect for concentrations of selenium up to 10 – 11 at%. A slight levelling out of the passivation effect
333 towards 10-11% suggests this may be close to the optimum alloying concentration in terms of absorber material
334 quality.

335 This defect passivation explains the remarkable performance of CdSeTe devices. However, current high
336 efficiency devices only have selenium in significant concentrations at the very front of the device, leaving much of
337 the back of the absorber layer unpassivated. This means that extending the selenium profile further towards the
338 back of the device, whilst maintaining a concentration gradient and built-in field, will passivate more defects at the
339 back of the absorber and might improve efficiencies further. In addition, we note that the energy of the sub-gap
340 selenium-related emission (~ 1.36 eV) is similar to the lowest achievable effective bandgap energy in the selenium-
341 rich layer, which also shows the highest luminescence intensity. This indicates that all these effects can be related
342 to a common underlying mechanism.

343 In summary, in this work we have correlated variations in the electronic properties of selenium-alloyed CdTe

344 with local variations in the selenium concentration. We find a strong correlation between selenium concentration,
345 high material luminescence efficiency, sub-gap transitions at 1.36 eV, and longer diffusion lengths, all at the sub-
346 micron scale. This indicates that selenium passivates defects present in bulk CdTe, and provides an explanation for
347 the remarkable performance of selenium-alloyed CdTe devices. In addition, the results provide crucial insights into
348 the fundamental electronic behaviour of selenium alloyed CdTe, which could unlock further improvements in the
349 photovoltaic performance of CdSeTe solar cells.

350

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359

360 **Methods**

361 **Cell Fabrication and Electrical Testing.** The two types of cells used in this study were deposited on TEC 10
362 glass substrates supplied by NSG Pilkington Ltd. The substrates consist of 3 mm of soda lime glass coated with a
363 ~400 nm layer of fluorine-doped tin oxide, which acts as the transparent conducting oxide (TCO). The other
364 layers of the cells were then fabricated as follows: 100 nm of MgZnO was deposited on the TCOs by magnetron
365 sputtering, forming the buffer/window layer of the devices. This was followed by 1.5 – 2 μm of CdSeTe,
366 deposited with the substrates held at ~ 420 °C and from a source containing 40% CdSe at 575 °C. A ~ 3.5 μm
367 layer of CdTe was then deposited on top of the CdSeTe with the substrates held at 500 °C and the CdTe source
368 at 555 °C. A cadmium chloride (CdCl_2) activation treatment was then performed on one of the substrates. This
369 involved sublimation of a CdCl_2 vapour onto the back surface of the substrate whilst it was maintained at 430 °C
370 for 600 seconds, followed by a 180 °C cooling step with the substrate removed from the vapour. Both device
371 stacks then received a copper doping treatment whereby copper chloride was deposited on the back surface of the

372 CdTe for 110 seconds whilst the substrate was held at
373 ~ 140 °C. This was followed by an anneal in vacuum at 220 °C for 220 seconds to drive copper into the device.
374 Finally, a ~ 30 nm Te film was evaporated on to the back of the CdTe to improve the back contact. At this stage
375 the two substrates were split in half, with one half of each substrate undergoing contacting and performance
376 testing, and the other half left bare for materials characterisation.

377 For cell contacting a layer of carbon and nickel paint in a polymer binder was sprayed on the back of the
378 device stack, forming the back electrode of the device. This was then masked and sand- blasted to delineate 10
379 separate cells of area 0.55 cm^2 . The devices were then tested for electrical performance using current density vs
380 voltage measurements using an AM1.5 spectrum. An ABET Technologies 10,500 solar simulator
381 with uniform illumination accessory was used to illuminate the devices for measurements. The lamp used for
382 illumination is an ozone free DC xenon arc lamp that produces 1 Sun power output over a 35mm diameter field
383 and meets ASTM, IEC and JIS Class A AM1.5G output requirements. Current density-voltage curves were
384 generated based on electrical measurements performed using a Keithley 2420 SourceMeter controlled by a
385 LabView program. Short-circuit current density was calibrated to CdTe cells measured by NREL. Device areas
386 were measured using a webcam that took an image of a backlit solar cell and counted the pixels below certain
387 brightness. Both the light intensity and area were calibrated for each set of measurements. The cells were
388 contacted by a fixture of spring-loaded gold pins that provided a 4-point connection and collect current from all
389 around the front contact of the device. The setup accurately measures the J- V parameters and the agreement of
390 these measurements with an externally certified photovoltaic device has been shown in previous work [18].

391
392 **TEM.** The specimen foil for scanning transmission electron microscopy (STEM) was prepared using an FEI
393 focused ion beam (FIB) dual beam system using a standard in-situ lift out method [30]. STEM imaging was
394 performed using a FEI Tecnai F20 S/TEM equipped with Gatan Bright and Dark field STEM detectors, Fischione
395 High Angle Annular Dark Field (HAADF) STEM detector and an Oxford Instruments X-Max 80mm^2
396 windowless energy-dispersive spectrometer (EDX). STEM imaging was performed at 200 kV with a camera
397 length of 100mm and condenser aperture size of $70\mu\text{m}$.

398
399 **Cathodoluminescence.** To present an extended cross-section for cathodoluminescence (CL) and SIMS

400 characterisation, a bevel was milled through the CdCl₂ treated device stack at an angle of 7 degrees to the
401 horizontal using a 30 keV gallium focussed ion beam in an FEI Nova 600 Nanolab Dual Beam scanning electron
402 microscope(SEM). A platinum over-layer was deposited on the back surface of the CdTe to homogenize the FIB
403 milling of the bevel. Hyperspectral CL characterisation was carried-out at room temperature in a Hitachi SU-70
404 scanning electron microscope (SEM). During the measurement a 7 keV beam was rastered across an area on the
405 bevel surface. Luminescence emitted from the sample surface was collected by a parabolic mirror and fed through
406 a diffraction grating to a Gatan MonoCL system for CL detection. This resulted in a 102 x 282 pixel image, with
407 each pixel containing a full CL spectrum. At 7 keV, CASINO Monte Carlo simulations show that most of the
408 carriers excited by the beam in CdTe (75%) are generated within a tear-drop shaped volume that extends 200 nm
409 below the sample surface and has a diameter of ~ 200 nm (note that results are for simulations performed at 7.5
410 keV) [21].

411

412 **NanoSIMS.** Following the CL, high resolution elemental mapping was performed on the same area of the bevel
413 surface using a Cameca secondary ion mass spectrometer (NanoSIMS 50) with a 16 keV Cs⁺ primary beam. The
414 diameter of the D1 aperture was set to 100 μm (D1-4). Entrance and Aperture slits are 50 x 220 μm (ES-1) and
415 open, respectively. During the measurement, a 0.5 – 1 pA Cs⁺ primary beam with a nominal diameter of 60 nm
416 was rastered over the measurement area and sputtered secondary ions analysed with a double-focused mass
417 spectrometer. The raster size was 25 x 25 μm (512 x 512 pixels) and the dwell time was 500 μs per pixel. Masses
418 analysed were 35Cl⁻ and 80Se⁻, giving high resolution images of distributions of chlorine and selenium in the
419 measured area. The scan was repeated 20 times giving 20 stacked images of the distributions of each element and
420 sputtering a total depth of ~ 200 nm below the bevel surface. Images used in the figures are a sum of each stack of
421 20 images. The ‘auto-track’ feature in ImageJ was used to correct a small amount of image drift before the images
422 were summed. Summing the images ensures that the information depth of the SIMS, at ~ 200 nm, is similar to the
423 CL information depth (100- 200 nm excluding carrier drift/diffusion). EDX measurements were taken on the bevel
424 at to calibrate the selenium counts obtained in the NanoSIMS measurements (beam energy 20 keV). They showed
425 that the average selenium concentrations in the CdSeTe region were 8.9%. This value could then be used to
426 calibrate the average selenium counts over the same region/area given by the NanoSIMS.

427

428 **Image processing.** Some image manipulations were required to allow exact alignment of the pixels of the CL
429 and SIMS images on the same area of the bevel. First, the SIMS maps were rotated to match the orientation of
430 the CL map. The CL map was then scaled by +13.5% in the Y-direction, i.e. height (this was necessary
431 because of some drift in the raster of the electron beam in the CL measurements, creating a slight distortion that
432 shortened the image). The two sets of images could now be superimposed on top of one another with exact
433 matching of grain boundary features in each (grain boundaries are delineated clearly by the chlorine signal in the
434 SIMS and darker valleys in the CL). At this point the images are matched in everything but the pixel size (the
435 SIMS images are higher resolution). For the pixel-for-pixel comparisons given in the scatter graphs, the pixel size
436 of the SIMS was increased to match the CL, giving images of 102 x 282 pixels that could be directly compared.
437 For the maps used in the figures, the resolution of the SIMS image was maintained (389 x 1152 pixels). Image
438 rotation, scaling, and repixelation were performed using ImageJ.

439

440 **Additional information**

441 Correspondence and requests for materials should be addressed to J.M.W

442

443 **Data Availability Statement**

444 The data that supports the plots within this paper and other findings of this study are available in the repository at
445 Loughborough University or from the corresponding author upon reasonable request.

446

447 **Author contributions**

448 This work was conceived of by TF, and planned with JW, KL, CG, and BM. The cells were made by AM, with
449 the assistance of KB and WS. TEM was performed by AA. Samples were prepared for CL and SIMS
450 characterisation by TF. CL measurements were carried out by BM. SIMS characterisation was performed by KL,
451 with assistance from CG. Data analysis was performed by TF and LW. The manuscript was written by TF, with
452 help from JB and JW.

453

454 **Competing interests**

455 The authors declare no competing financial or non-financial interests.

456 **References**

457

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