



www.chemeurj.org

# Two-photon Excitation of Bright Diaza[4]Helicenes for Isotropic and Circularly Polarized Emission

Bibiana Fabri,<sup>[a]</sup> Davide F. De Rosa,<sup>[b]</sup> Dominic J. Black,<sup>[b]</sup> Rebecca Mucci,<sup>[a]</sup> Artemijs Krimovs,<sup>[b]</sup> Robert Pal,\*<sup>[b]</sup> and Jérôme Lacour\*<sup>[a]</sup>

Helicenes are chiral organic dyes that are attracting growing attention due to the high tunability of their (chir)optical and electronic properties. In this work, a series of functionalized cationic diaza[4]helicenes based on dimethoxyquinacridinium (DMQA) scaffolds are presented. By merging branched N-alkyl side chains and triple para-functionalization with OMe groups, structures combine improved chiroptical responses (5x increase) and strong fluorescence quantum yields ( $\Phi_{\rm f}\approx70\%$  in acetonitrile). An overall improved efficiency of the emission of

circularly polarized light with  $B_{\text{CPL}}$  values reaching 3.4  $M^{-1}$  cm<sup>-1</sup> is obtained. Additionally, two-photon excitation (2PE) studies were performed, showing good values of cross section (CS) at 810 nm. Interestingly, 2PE circularly polarized luminescence (CPL) spectra were acquired for the most performant derivatives (N-isopropyl and N-cyclohexyl); this type of measurement being usually challenging for organic molecular species. Finally, the viability of these compounds in single-photon (1PE) and 2PE microscopy is also shown.

## 1. Introduction

Chiral organic dyes represent attractive research targets for the control of chiroptoelectronic properties, such as electronic circular dichroism (ECD) and circularly polarized luminescence (CPL), which in turn can be useful in various applications (e.g., photonics, optoelectronic devices). Among chiral dyes, helicenes generated an increasing interest due to the diversity of possible chemical structures and their strong interaction with polarized light. In this context, cationic diaza[4]helicenes of dimethoxyquinacridinium (DMQA) type (1, (Figure 1A) are a particularly versatile class of derivatives.

They are readily prepared and resolved into single enantiomers of (M) and (P) configurations, which present high configurational stability ( $\Delta G^{\dagger}_{\rm racem} \sim 42$  kcal mol<sup>-1</sup>).<sup>[3]</sup> Furthermore, these structures are suited for regioselective late-stage functionalization strategies introducing substituents either in *meta* to the formal positive charge (positions 6 and 2/12), or more recently, in *para* (positions 3, 7, and 11).<sup>[4]</sup> Diverse functional groups can then be placed on the outer rim of the helical core allowing an exten-

- [a] B. Fabri, R. Mucci, J. Lacour

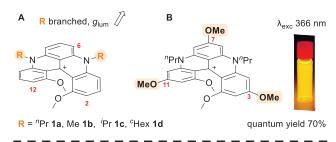
  Department of Organic Chemistry, University of Geneva, Quai Ernest

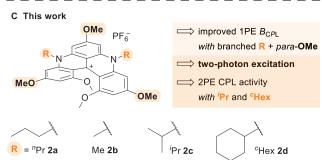
  Ansermet 30, Geneva 4 1211, Switzerland

  E-mail: jerome.lacour@uniqe.ch
- [b] D. F. De Rosa, D. J. Black, A. Krimovs, R. Pal Department of Chemistry, Durham University, Durham DH1 3LE, UK E-mail: robert.pal@durham.ac.uk
- Supporting information for this article is available on the WWW under https://doi.org/10.1002/chem.202501212
- © 2025 The Author(s). Chemistry A European Journal published by Wiley-VCH GmbH. This is an open access article under the terms of the Creative Commons Attribution License, which permits use, distribution and reproduction in any medium, provided the original work is properly cited.

sive tuning of key photophysical and electronic properties of the dyes. Nowadays, these small [4]helicenes have found applications as catalysts, electrolytes in organic redox flow batteries, spin filters, and bioprobes. In addition, they have been proven good candidates for obtaining persistent neutral radicals. Most commonly, helicenes based on scaffold 1 absorb and emit in the red domain of the visible spectrum (> 600 nm), with moderate to good quantum yields (5–50%). Recently, it was reported that, by adding three *para*-OMe groups on the helical scaffold (2a, (Figure 1B), a fluorescence quantum yield of 70% can be achieved, accompanied by a hypsochromic shift of the optical properties (1a  $\lambda_{max} = 616$  nm,  $\lambda_{em} = 667$  nm  $\rightarrow$  2a  $\lambda_{max} = 533$  nm,  $\lambda_{em} = 580$  nm). Aem = 580 nm).

For imaging purposes, the possibility to excite compounds at low energy, and ideally in the Near Infrared (NIR) domain, is an attractive possibility. In this regard, one solution is to excite the compound using two simultaneously absorbed photons of half the energy (double wavelength) via a multiphoton (MP) process.<sup>[7]</sup> Two-photon excitation (2PE) enables, at the same time, good tissue penetration of biological media and high spatial resolution, which make it widely employed in optical imaging.<sup>[8]</sup> In 2PE, the probability of two photons to be absorbed simultaneously is dependent on the light intensity and the cross section (CS, indicated by  $\sigma^2$ ) of the compound, with Goeppert-Mayer (GM) units, where  $1GM = 10^{-50} \text{ cm}^4 \text{ s}$ photon<sup>-1</sup>.<sup>[7b,9]</sup> Several organic derivatives able to undergo 2PE, even with very high values of CS (> 1000 GM), are known.[10] Even though 2PE of helical derivatives has been already reported, complemented in some cases by measurements of two-photon ECD spectra and one-photon excitation (1PE) CPL activity,[11] reports of 2PE CPL are seldom due to instrumental constrains and have only recently appeared including perovskites and liquid crystals.<sup>[12]</sup> In the context of molecular emitters, the work of Pal and co-workers on lanthanide complexes holds a pivotal role.[13] Additionally, it was demonstrated the importance





**Figure 1.** A) cationic diaza[4]helicenes 1 with various N-alkyl chains; B) previously reported triple *para*-OMe derivative **2a**; C) *para*-OMe derivatives with linear (**2a–2b**) and branched (**2c–2d**) alkyl chains.

of CPL confocal microscopy in extracting additional information, thanks to the polarization of the luminescent probe, making 2PE CPL-active probes particularly desirable. [13a,c] However, aside from few lanthanide complexes, molecular, and especially organic, systems able to perform such task are not yet readily available.

Herein, para-OMe diaza[4]helicenes 2a-2d (Figure 1C) are presented bearing primary and secondary alkyl chains on the N-atoms. While the introduction of methoxy (OMe) groups para to the formal positive charge of 1 affords fluorescent quantum yields ( $\Phi_f$ ) around 70%,<sup>[4d]</sup> higher chiroptical responses were looked-for by including N-isopropyl (iPr, 2c) and N-cyclohexyl ( ${}^{c}$ Hex, 2d) chains, known to give higher g factors compared to the more classical N-<sup>n</sup>Pr chains. [3b,14] In fact, the steric interactions between the branched N-alkyl chains and the helical skeleton provokes a local strain and a concomitant pyramidalization of the nitrogen bridging atoms. This distortion is then at the origin of the enhanced chiroptical properties, as supported by TD-DFT calculations.[14] Thanks to the combination of high  $\Phi_{\rm f}$  and increased  $g_{\rm lum}$  ( $\geq$  5  $\times$  10  $^{-4}$  ), singlephoton (1PE) CPL brightness  $(B_{\text{CPL}})^{[15]}$  values of 2.6 and 3.4  $\mathsf{M}^{-1}$  cm $^{-1}$  are reached with novel compounds  $\mathbf{2c}$  and  $\mathbf{2d}$ , respectively, which are among the highest obtained for this class of cationic [n]helicenes. [15] Additionally, derivatives 2a-2d are employed in 2PE studies, giving good values of CS at 810 nm ( $\sigma^2$  up to 137 GM). Again, branched N-alkyl derivatives are best, with the highest values of CS. This, combined with favorable q factors, allows for an efficient measurement of CPL spectra via 2PE. Noteworthy 2PE B<sub>CPL</sub> values of 0.03 and 0.05 GM are obtained with these small organic derivatives. [13a] Finally, microscopy studies using 2d (strongest CS) as probe are presented.

### 2. Results and Discussion

#### 2.1. Synthesis

Racemic compounds 1a-1d were synthesized following reported guidelines, [14,16] and used as substrates to obtain para-OMe 2a-2d, following synthetic steps previously developed for 2a (Scheme 1).[4d] Synthetically, pinacol boronic ester (Bpin) moieties were first introduced in para positions by Ir-catalyzed direct C—H borylations (step a) resulting in intermediates of type 3. Due to a lack of stability and fast protodeboronation of 3a-3d, these were subjected directly to hydroxylation conditions (H<sub>2</sub>O<sub>2</sub>/NaOH) affording tris(OH) derivatives 4 in 83-90% yields (step b).[17] Then, alkylation with methyl iodide (excess) under basic conditions was performed giving methylated analogues 2 (step c). While this step is effective for 2a and 2d (>90%,  $R = {}^{n}Pr$ , <sup>c</sup>Hex), it is less successful for compounds **2b** (R = Me) and **2c**  $(R = {}^{i}Pr)$ , which were obtained in only 40 and 51% yields, respectively (Scheme 1). The difference in reactivity is most probably linked to the variations in solubility of the tris(OH) species 4 and their deprotonated forms. In fact, with Me or <sup>i</sup>Pr chains, the structures possess a lower lipophilicity than with longer residues, <sup>n</sup>Pr or <sup>c</sup>Hex, making derivatives 2b and 2c less soluble in the reaction mixture. Moreover, upon deprotonation, precipitation of 4b and 4c occurs. Overall, a short series of para-functionalized OMe derivatives 2 was synthesized starting from the corresponding racemic 1. While (M) and (P) enantiomers of 2a were already available from a previous project, [4d] enantiopure 2b, 2c, and 2d were obtained by chiral stationary phase-HPLC resolution of the racemates using a semi-preparative CHIRALPAK IC column (see Supporting Information).

# 2.2. Chiroptical Properties

Absorption and emission of newly synthesized dyes 2b-2d were studied in acetonitrile at room temperature and compared to 2a (Figure S5). Overall, optical properties varied as expected from previous studies on unfunctionalized derivatives 1.[14,18] The presence of N-Me side chains, instead of N-nPr, had no major impact, with the absorption and emission spectra of 2b being almost superimposable to those of 2a. On the contrary, the strain induced by the N-Pr and N-Hex side chains brought about a slight red-shift in both absorption and emission (Table 1). All four para-OMe helicenes are equally efficient fluorophores, with  $\Phi_f$ ranging from 70 to 75% and  $\tau$  of 16 to 17 ns (Table 1), these differences being within the experimental error. Consequently, the rates of radiative  $(k_r)$  and nonradiative  $(k_{nr})$  decay are similar for all four derivatives. The para-electron donating group effect of the methoxy substituents, described previously for 2a, [4d] applies hence to 2b, 2c, and 2d. Molecular engineering of compounds  $1\rightarrow 2$  is therefore quite predictive irrespective of the different side chains.[14,18]

ECD and CPL spectra were also recorded in acetonitrile and showed, for all derivatives, a mirror image relationship for pairs of (M) and (P) enantiomers. Compound **2b** (R = Me) shows

32, Downloaded from https://chemistry-euope.onlinelibrary.wiley.com/doi/10.1002/chem.20201212 by Durham University - University - Wiley Online Library on [10.062025]. See the Terms and Conditions (https://onlinelibrary.wiley.com/rems-and-conditions) on Wiley Online Library for rules of use; OA articles are governed by the applicable Ceatwive Commons

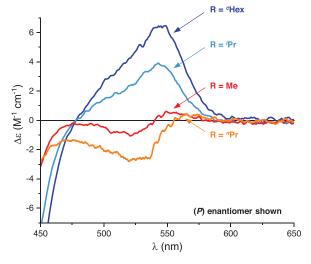
Scheme 1. Synthesis of triple para-OMe diaza[4]helicenes with differentiated nitrogen substituents. Conditions: a)  $[Ir(cod)OMe]_2$  (cod = 1,5-cyclooctadiene), 3,4,7,8-tetramethyl-1,10-phenanthroline, bis(pinacolato)diboron ( $B_2pin_2$ ), dry THF, 80 °C, 16 hours; b) (1)  $H_2O_2$  (aq), NaOH, THF, RT, 16 hours, (2) anion metathesis with KPF<sub>6</sub> (aq); c) Mel, Cs<sub>2</sub>CO<sub>3</sub>, DMF, RT, 16 hours.

Table 1. (Chir)optical properties of para-OMe diaza[4]helicenes 2 in acetonitrile.											
$Compound^{[a]}$	λ <sub>abs</sub> <sup>max</sup> (nm)	$\varepsilon$ (M $^{-1}$ cm $^{-1}$ )	λ <sub>em</sub> (nm)	Stokes shift (cm <sup>-1</sup> )	Φ <sub>f</sub> <sup>[b,c]</sup> (%)	$ au^{[d]}$ (ns)	$k_{\rm r}^{\rm [e]}$ (10 <sup>6</sup> s <sup>-1</sup> )	$k_{\rm nr}^{\rm [f]}$ (10 <sup>6</sup> s <sup>-1</sup> )	$g_{\rm abs} \ (\lambda) \ (10^{-4})$	Average $g_{lum}$ (10 <sup>-4</sup> )	1PE $B_{CPL}^{[g]}$ (M <sup>-1</sup> cm <sup>-1</sup> )
<sup>n</sup> Pr ( <b>2a</b> )	533	14 610	580	1521	70	15.7	45	19	+1.7/-1.4 (572)	+1.0/-1.1	0.5
Me ( <b>2b</b> )	531	14 280	580	1591	74	16.6	46	16	+0.8/-1.2 (554)	+2.1/-2.3	1.0
<sup>i</sup> Pr ( <b>2c</b> )	535	14 010	589	1714	75	17.0	44	15	+3.0/-2.9 (543)	+4.8/-4.9	2.6
<sup>c</sup> Hex ( <b>2d</b> )	539	15 280	593	1690	72	16.5	44	17	+4.6/-4.5 (548)	+5.8/-6.7	3.4

 $^{[a]}$  Concentrations 1  $\times$  10  $^{-5}$  to 5  $\times$  10  $^{-6}$  M.

<sup>[g]</sup> 
$$B_{\text{CPL}} = \varepsilon \times \Phi_{\text{f}} \times |g_{\text{lum}}|/2.$$
<sup>[15]</sup>

similar ECD signatures to 2a (R =  $^n$ Pr) across the whole UV-vis range, with pronounced Cotton effects in the UV domain and modest bands ( $|\Delta \varepsilon| <$  10  ${\rm M}^{-1}~{\rm cm}^{-1}$ ) above 400 nm (Figure S6 and S8), corresponding to  $|q_{abs}| \approx 1 \times 10^{-4}$  for the lowest-energy Cotton effect (Table 1). On the contrary, ECD features of derivatives bearing bulkier N-substituents, 2c and 2d, are globally different from those of 2a and 2b. Looking across the whole UV-vis range (Figures S10 and S12), aside from one strong peak below 300 nm ( $|\Delta \varepsilon| \approx$  75 M $^{-1}$  cm $^{-1}$ ), the ECD signals remain < 25 M<sup>-1</sup> cm<sup>-1</sup>. However, at low energies (> 500 nm), the branched N-substituents provoke a higher chiroptical response in comparison to linear N-alkyl chains. Similarly to what was observed from the unfunctionalized scaffolds, [14] the  $g_{\rm abs}$  is enhanced by a factor of 2 and 3 from 2a to 2c and 2d, respectively (Table 1). The effect of sterically demanding N-substituents ( $R = {}^{i}Pr$  and <sup>c</sup>Hex) on the most red-shifted Cotton effect is displayed clearly in Figure 2. The ECD features in this region are particularly sensitive to the nature of the substituents linked to the helicene core, as also evidenced in other helical systems.<sup>[19]</sup> Derivatives 2a and 2b show, above 500 nm, an inversion of sign (negative  $\rightarrow$  positive, for the (P) enantiomer) already reported for the introduction of para-OMe moieties onto 1a.[4d] However, for 2c and 2d, this inversion is not present. Looking at the ECD of the (P) enantiomer (Figure 2), it can be hypothesized that the stronger positive chiroptical response induced by the branched N-alkyl chains overrides the negative signature. Overall, all triple para-OMe (M)-2 and (P)-2 enantiomers maintain negative and positive signs for the lowest-energy ECD band, respectively.



**Figure 2.** Comparison of ECD signatures at low energies of (*P*)-2a (orange), (*P*)-2b (red), (*P*)-2c (light blue), and (*P*)-2d (dark blue) in air-equilibrated acetonitrile at RT between 450 and 650 nm and concentration  $5 \times 10^{-5}$  M.

As expected, the same trend is observed in the emission of polarized light (Figures S7, S9, S11, and S13). [20] Para-OMe 2b has low CPL similarly to 2a (Table 1), while both isopropyl and cyclohexyl derivatives (2c and 2d) exhibit higher average  $g_{\text{lum}}$  values, equal to  $+4.8/-4.9 \times 10^{-4}$  and  $+5.8/-6.7 \times 10^{-4}$ , respectively. In previous work, [4d] it was pointed out that, thanks to the strongly enhanced fluorescence of 2a, the overall efficiency of circularly polarized emission (1PE  $B_{\text{CPL}}$ , Table 1) is enhanced

 $<sup>^{[</sup>b]}$   $\Phi_{\text{f}}$  estimated error  $=\pm 10\%.$ 

 $<sup>^{[</sup>c]}$  rhodamine B ( $\Phi_{f}=70\%$  in MeOH).

 $<sup>^{[</sup>d]}~\lambda_{exc}=400$  nm.

 $<sup>^{[</sup>e]}$   $k_{\rm r}=\Phi_{\rm f}$  / au.

<sup>[</sup>f]  $k_{nr} = (1 - \Phi_f)/\tau$ .

Chemistry Europe

European Chemical Societies Publishing

with respect to the unfunctionalized helicene **1a** (0.2 M<sup>-1</sup> cm<sup>-1</sup>). Here, for derivatives **2c** and **2d**, in addition to the high  $\Phi_{\rm f}$ , the dissymmetry of emission is also improved giving interesting  $B_{\rm CPL}$  values of 2.6 and 3.4 M<sup>-1</sup> cm<sup>-1</sup>, respectively. These values are among the highest obtained for cationic [n]helicenes.<sup>[15]</sup>

#### 2.3. MP Excitation

Initial 2PE studies were conducted with compound 2a in acetonitrile. 2PE was carried out exciting the sample at 805 nm, using a tunable femtosecond pulsed laser (680-1300 nm, Coherent Discovery NX TPC, 100 fs, 80 MHz).[13a] In Figure 3a are reported the emission profiles of 2a obtained by 1PE and 2PE. Importantly, the same emission spectrum is obtained exciting in both the UV (365 nm) or in the NIR (805 nm). To confirm that a two-photon process takes place while exciting at 805 nm, the emission intensity-excitation power dependence was recorded and their logarithmic relationship was plotted. As anticipated, the Logl<sub>em</sub> versus LogP<sub>exc</sub> plot yielded a linear trend with slope 1.67 (Figure 3b), close to the theoretical value of 2.[7b] The 2PE spectrum was also recorded and compared to that obtained by 1PE (Figure 3c). The 2PE spectrum of 2a displays maxima at 715 and 805/810 nm, which follow the profile of the second and third one-photon absorption bands. Oppositely, two-photon absorption is not observed in correspondence to lowest-energy S<sub>0</sub>-S<sub>1</sub> transition. As expected because of the nonlinear effects of 2PE, the 2PE spectrum has not exactly double wavelengths compared to the 1PE spectrum.<sup>[7b,13b,21]</sup>

The ability of derivatives 2b, 2c, and 2d to undergo 2PE was also characterized in the same manner; all related spectra are reported in the Supporting Information (Figures S16-S21). Plotting the emission intensity versus excitation power  $(\lambda_{exc} = 810 \text{ nm})$  in logarithmic scale for compounds **2b** to **2d**, slopes of around 2 were found for all the three curves, confirming again that the excitation process occurs via two-photon absorption. 2PE spectra were also obtained. As for derivative 2a, the one-photon lowest energy absorption band is not active in 2PE, while the second and third bands appear in the 2PE spectra of all three compounds. The 2PE maxima are located at 750 and 810 nm (Figure S17), 780 and 810 nm, (Figure S19), and 770 and 810 nm (Figure S21) for 2b, 2c, and 2d, respectively. The absorption band centered at 400 nm is the one that finds the best correspondence in 2PE, with all four fluorophores under study being active in 2PE at 810 nm. Values of 2PE CSs were then calculated at 810 nm using the established procedure and rhodamine B ( $\sigma^2$  at 810 nm = 260 GM in MeOH) as a reference (Table 2). [10b] All triple para-OMe diaza[4]helicenes are good two-photon chromophores in the NIR, presenting CS values  $\geq$  100 GM. Overall, derivatives **2c** and **2d**, bearing N-<sup>i</sup>Pr and N-<sup>c</sup>Hex side chains, have the highest CS values.

At this point, for derivatives **2c** and **2d**, which combine favorable chiroptical response and higher values of CS, using the pulsed laser at 810 nm, 2PE CPL spectra were successfully acquired (Figure 4). The resulting 2PE CPL spectra are noisy due to the experimental challenges in recording the small signal deriving by the combination of 2PE and CPL emission ( $g_{lum}$  val-

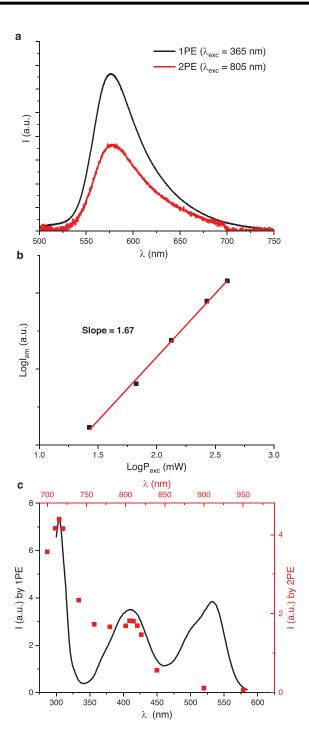


Figure 3. 2PE studies of 2a in acetonitrile with concentration  $1\times 10^{-5}$  M. a) Emission spectra under 1PE (black line) and 2PE (red line); b) log versus log plot of the excitation-power dependence of the emission intensity, the linear fit is displayed by a red line; c) 2PE (red dots, top x-axis) and 1PE (black line, bottom axis) spectra followed at  $\lambda_{em}=580$  nm (x and y axes are adjusted to emphasize the agreement of the spectral features).

ues <  $10^{-3}$ ). Nonetheless, the allied 2PE CPL spectra of **2c** and **2d** well reproduce the 1PE CPL profiles (Figures S11 and S13) both in sign and in intensity. In fact, 2PE average  $g_{\text{lum}}$  values of + 6.6/-6.1 ×  $10^{-4}$  and + 9.4/-9.2 ×  $10^{-4}$  were found for **2c** (R =  $^{i}$ Pr) and **2d** (R =  $^{c}$ Hex), respectively. These values are comparable to those recorded using single photon excitation at shorter wavelengths (Table 1). Finally, 2PE  $B_{\text{CPL}}$  values can be calculated in the

Chemistry Europe

European Chemical Societies Publishing

<b>Table 2.</b> 2PE chiroptical properties of <i>para</i> -OMe diaza[4]helicenes 2 in acetonitrile.									
Compound <sup>[a]</sup>	$\sigma^2$ at 810 nm (GM) <sup>[b]</sup>	Average $g_{lum}$ (10 <sup>-4</sup> )	2PE <i>B</i> <sub>CPL</sub> <sup>[c]</sup> (GM)						
<sup>n</sup> Pr ( <b>2a</b> )	99	n.d.	n.d.						
Me ( <b>2b</b> )	106	n.d.	n.d.						
<sup>i</sup> Pr ( <b>2c</b> )	117	+6.6/-6.1	0.03						
<sup>c</sup> Hex ( <b>2d</b> )	137	+9.4/-9.2	0.05						
[a] Concentrations 1 $\times$ 10 <sup>-5</sup> M.									

 $^{[b]}$  rhodamine B ( $\Phi_{\rm f}=70\%$  and  $\sigma^2=260$  GM in MeOH).  $^{[c]}$  2PE  ${\it B}_{\rm CPL}=\sigma^2\times\Phi_{\rm f}\times|g_{\rm lum}|/2.^{[13a]}$  n.d.: not determinable.

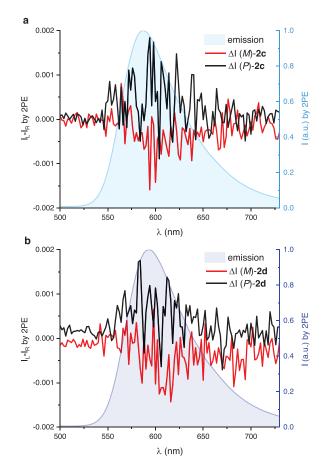
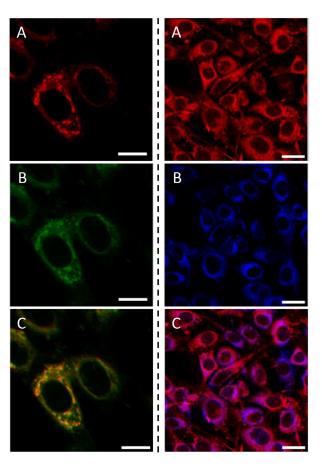


Figure 4. 2PE CPL spectra ( $\lambda_{exc}=810$  nm) of compounds a) 2c and b) 2d in acetonitrile with concentrations of  $1\times10^{-5}$  M respectively.

same fashion as the most classical 1PE  $B_{CPL}$ , replacing the molar absorption coefficient with the 2PE CS value: [13a]

$$2PEB_{CPL} = \sigma^2 \times \phi \times |g_{lum}|/2.$$

The calculated 2PE  $B_{\rm CPL}$  correspond to 0.03 and 0.05 GM for compounds **2c** and **2d**, respectively (Table 2). In comparison, recently reported chiral Eu complexes possess 2PE  $B_{\rm CPL}$  values which are only three times higher (0.16/0.17 GM), even though presenting  $g_{\rm lum}$  values of  $10^{-2}$ .[13b] This is possible due to the overall favorable 2PE and emission properties (high  $\Phi_{\rm f}$ ) of *para-*



**Figure 5.** Live cell LSCM of mouse embryonic fibroblasts (NIH-3T3). Left: A) compound **2d**,  $\lambda_{\text{exc}} = 543$  nm,  $\lambda_{\text{em}} = 560$ –700 nm; B) ER-Tracker<sup>TM</sup> Green,  $\lambda_{\text{exc}} = 488$  nm,  $\lambda_{\text{em}} = 500$ –550 nm; C) RGB merge of A) and B) showing strong localization to the ER. Scale bar = 10  $\mu$ m. Right: 2PE of live cell LSCM incubated with **2d** (500 nM) for 2 hours. A)  $\lambda_{\text{exc}} = 543$  nm,  $\lambda_{\text{em}} = 560$ –700 nm. B)  $\lambda_{\text{exc}} = 810$  nm,  $\lambda_{\text{em}} = 500$ –550 nm. C) RGB merge of A) and B) showing overlap of single and 2PE. Scale bar = 25  $\mu$ m.

OMe derivatives **2** that compensate for the modest emission dissymmetry.

## 2.4. Microscopy

Live cell imaging was performed using laser scanning confocal microscopy (LSCM) to determine the two-photon activation potential of the compounds and their localization. In practice, only compound 2d was selected as it possesses the highest experimental two-photon CS (137 GM) of all derivatives 2. NIH-3T3 (Swiss albino mouse embryonic fibroblasts) were incubated with 500 nM of 2d for 2 hours. In Figure 5 (left), localization of 2d in the endoplasmic reticulum (ER) is confirmed by co-staining with ER-Tracker Green (Invitrogen). The Pearson's coefficient for 2d and ER-Tracker Green is 0.89 and the Manders' coefficients are 0.998 and 0.835 showing a strong localization to the ER. Since 2d allies a favorable two-photon CS and a high concentration in a small area in the live imaging, both these factors make it a good candidate for 2PE microscopy.[13a,22] The local maximum of 810 nm was used to excite the compound rather than the overall maximum of 720 nm to allow for the full collection

Chemistry Europe

European Chemical Societies Publishing

of the emission. Additionally, the output power of the laser is significantly higher at 810 nm than at 720 nm ( $\sim$ 3510 mW and  $\sim$ 1530 mW, respectively), compensating the lower absorption intensity, whilst 810 nm wavelength is lower in energy and allows deeper tissue penetration in bioimaging. In addition, Figure 5 (right) shows single and two-photon excitation images for 2d and the merge of the two excitation modes. It is worth noting that an exact overlay is not achieved due to the slightly different focal (z) planes of the one and two-photon excitation voxels because of the inherent chromatic aberrations of the LSCM system.

### 3. Conclusion

In this work, a family of triple para-OMe cationic diaza[4]helicenes was prepared, by late-stage functionalization of DMQA derivatives bearing linear and branched alkyl chains. Expanding from previously reported results, [4d] para-OMe groups were successfully introduced on the different helical scaffolds, showing the generality of the tandem borylation / hydroxylation procedure. While all para-OMe derivatives **2** are strongly fluorescent with  $\Phi_{\rm f} \approx$  70%, higher  $B_{\rm CPL}$  values of 2.6 and 3.4  $\mathrm{M}^{-1}~\mathrm{cm}^{-1}$  are obtained with compounds 2c and 2d, respectively. These values, which are among the highest obtained for cationic [n]helicenes, [15] are possible, thanks to the combination of high  $\Phi_f$  and the enhanced chiroptical properties arising with branched N-substituents.[14] Additionally, all para-OMe derivatives revealed to be good two-photon chromophores with CS values ≥ 100 GM at 810 nm. For compounds bearing branched N-alkyl chains, 2PE CPL spectra were also successfully recorded giving emission profiles and  $g_{lum}$  values comparable to those of 1PE. Average 2PE  $g_{lum}$  values of + 6.6/- $6.1 \times 10^{-4}$  and  $+ 9.4/-9.2 \times 10^{-4}$  were found for **2c** (<sup>i</sup>Pr) and 2d ( $^{c}$ Hex), respectively, corresponding to 2PE  $B_{CPL}$  of 0.03 and 0.05 GM. Noteworthy, acquisition of 2PE CPL for small organic molecules is rather unique. Finally, test studies of 2d in cell imaging showed the viability of compound class 2 in 1PE and 2PE confocal microscopy with localization in the ER.

## **Supporting Information**

It contains general methods, experimental conditions, full characterization of new compounds, (chir)optical measurements, time-correlated single photon counting, additional 2PE data, CSP-HPLC traces and NMR spectra. The authors have cited additional references within the Supporting Information.<sup>[23]</sup>

## **Acknowledgments**

We thank the University of Geneva, the Swiss National Science Foundation (200020\_207539, JL), and the European Commission Research Executive Agency (Grant Agreement Number: 859752 HEL4CHIROLED H2020-MSCA-ITN-2019, BF, DFDR, RP, JL)

for financial support. AK, DJB and RP thank BBSRC BB/S017615/1 and BB/X001172/1 and EPSRC EP/X040259/1 for funding. Open access funding was provided by Durham University. Stéphane Grass (University of Geneva) for the help in preparing synthetic materials.

#### Conflict of Interests

The authors declare no conflict of interest.

# **Data Availability Statement**

The data that support the findings of this study are openly available in yareta.unige.ch at <a href="https://doi.org/10.26037/yareta:bcq6hrszpjgy3naore3vz2ohwi">https://doi.org/10.26037/yareta:bcq6hrszpjgy3naore3vz2ohwi</a>. It will be preserved for 10 years.

**Keywords:** bright fluorophores - chiroptical properties - circularly polarized luminescence - microscopy - two-photon excitation

- [1] a) C. Train, M. Gruselle, M. Verdaguer, Chem. Soc. Rev. 2011, 40, 3297; b) P. C. Mondal, C. Fontanesi, D. H. Waldeck, R. Naaman, Acc. Chem. Res. 2016, 49, 2560; c) H. Han, Y. J. Lee, J. Kyhm, J. S. Jeong, J.-H. Han, M. K. Yang, K. M. Lee, Y. Choi, T.-H. Yoon, H. Ju, S.-k. Ahn, J. A. Lim, Adv. Funct. Mater. 2020, 30, 2006236; d) F. Furlan, J. M. Moreno-Naranjo, N. Gasparini, S. Feldmann, J. Wade, M. J. Fuchter, Nat. Photonics 2024, 18, 658; e) A. Artigas, N. Ferdi, M. Rémond, F. Rigoulet, N. Vanthuyne, D. Hagebaum-Reignier, Y. Carissan, J.-V. Naubron, M. Giorgi, L. Favereau, Y. Coquerel, J. Org. Chem. 2024, 89, 498; f) W. C. Guo, W. L. Zhao, K. K. Tan, M. Li, C. F. Chen, Angew. Chem. 2024, 136, e202401835; g) F. Zhang, V. Brancaccio, F. Saal, U. Deori, K. Radacki, H. Braunschweig, P. Rajamalli, P. Ravat, J. Am. Chem. Soc. 2024, 146, 29782; h) S. Cadeddu, B. W. van den Bersselaar, B. de Waal, M. Cordier, N. Vanthuyne, S. C. Meskers, G. Vantomme, J. Crassous, Nanoscale 2024, 16, 21351; i) C. Yin, X. Ye, S. Tao, D. Zhao, Y. Zhi, D. Jiang, Angew. Chem., Int. Ed. 2024, 63, e202411558; j) P. He, J. Ye, J. Zhang, T. Lu, W. Cui, J. Liu, C. Shen, W. Hong, X. Liu, Angew. Chem., Int. Ed. 2025, 64, e202416319; k) Z. Zhou, M. A. Petrukhina, Chem. Sci. 2025, 16, 468.
- a) V. Hutskalova, A. Prescimone, C. Sparr, Helv. Chim. Acta 2021, 104, e2100182; b) J. Crassous, I. G. Stará, I. Stary, Helicenes: Synthesis, Properties, and Applications, John Wiley & Sons, 2022; c) X. Xiao, Q. Cheng, S. T. Bao, Z. Jin, S. Sun, H. Jiang, M. L. Steigerwald, C. Nuckolls, J. Am. Chem. Soc. 2022, 144, 20214; d) W. Fu, V. Pelliccioli, M. von Geyso, P. Redero, C. Böhmer, M. Simon, C. Golz, M. Alcarazo, Adv. Mater. 2023, 35, 2211279; e) P. Izquierdo-García, J. M. Fernández-García, S. M. Rivero, M. Šámal, J. Rybáček, L. Bednárová, S. Ramírez-Barroso, F. J. Ramírez, R. Rodríguez, J. Perles, D. García-Fresnadillo, J. Crassous, J. Casado, I. G. Stará, N. Martín, J. Am. Chem. Soc. 2023, 145, 11599; f) S.-M. Guo, S. Huh, M. Coehlo, L. Shen, G. Pieters, O. Baudoin, Nat. Chem. 2023, 15, 872; g) G. Zhang, J. Zhang, Y. Tao, F. Gan, G. Lin, J. Liang, C. Shen, Y. Zhang, H. Qiu, Nat. Commun. 2024, 15, 5469; h) Z. Sun, W. Xu, S. Qiu, Z. Ma, C. Li, S. Zhang, H. Wang, Chem. Sci. 2024, 15, 1077; i) K. Mutoh, J. Abe, Chem. Sci. 2024, 15, 13343; j) H. K. Singh, A. Bodzioch, A. Pietrzak, P. Kaszyński, Chem. Commun. 2025, 61, 496; k) S. Huang, H. Wen, Y. Li, W. Qin, P. Wang, Y. Lan, S. Jia, H. Yan, Nat. Commun. 2025, 16, 500; I) Y. Gisbert, M. Ovalle, C. N. Stindt, R. Costil, B. L. Feringa, Angew. Chem., Int. Ed. 2025, 64, e202416097; m) N. Hano, N. Zigon, B. Kuppan, L. Sturm, N. Vanthuyne, E. Pouget, S. Nlate, H. Bock, F. Durola, N. Avarvari, Nanoscale 2025, 17, 5081; n) T. Edlová, J. Rybáček, H. Cattey, J. Vacek, L. Bednárová, P. Le Gendre, A. T. Normand, I. G. Stará, I. Starý, Angew. Chem. 2025, 137, e202414698.
- [3] a) C. Herse, D. Bas, F. C. Krebs, T. Bürgi, J. Weber, T. Wesolowski, B. W. Laursen, J. Lacour, Angew. Chem., Int. Ed. 2003, 42, 3162; b) B. Laleu,

Chemistry Europe

European Chemical Societies Publishing

- P. Mobian, C. Herse, B. W. Laursen, G. Hopfgartner, G. Bernardinelli, J. Lacour, *Angew. Chem., Int. Ed.* **2005**, *44*, 1879.
- [4] a) I. H. Delgado, S. Pascal, A. Wallabregue, R. Duwald, C. Besnard, L. Guénée, C. Nançoz, E. Vauthey, R. C. Tovar, J. L. Lunkley, G. Muller, J. Lacour, *Chem. Sci.* 2016, 7, 4685; b) R. Duwald, S. Pascal, J. Bosson, S. Grass, C. Besnard, T. Bürgi, J. Lacour, *Chem.-Eur. J.* 2017, 23, 13596; c) L. Frédéric, B. Fabri, L. Guénée, F. Zinna, L. Di Bari, J. Lacour, *Chem.-Eur. J.* 2022, 28, e202201853; d) B. Fabri, T. Funaioli, L. Frédéric, C. Elsner, E. Bordignon, F. Zinna, L. Di Bari, G. Pescitelli, J. Lacour, *J. Am. Chem. Soc.* 2024, 146, 8308.
- [5] a) V. Kiran, S. P. Mathew, S. R. Cohen, I. Hernández Delgado, J. Lacour, R. Naaman, Adv. Mater. 2016, 28, 1957; b) A. Babič, S. Pascal, R. Duwald, D. Moreau, J. Lacour, E. Allémann, Adv. Funct. Mater. 2017, 27, 1701839; c) G. N. Hargenrader, R. B. Weerasooriya, S. Ilic, J. Niklas, O. G. Poluektov, K. D. Glusac, ACS Appl. Energy Mater. 2019, 2, 80; d) L. Mei, J. M. Veleta, T. L. Gianetti, J. Am. Chem. Soc. 2020, 142, 12056; e) J. Moutet, D. Mills, M. M. Hossain, T. L. Gianetti, Mater. Adv. 2022, 3, 216; f) F. Calogero, G. Magagnano, S. Potenti, F. Pasca, A. Fermi, A. Gualandi, P. Ceroni, G. Bergamini, P. G. Cozzi, Chem. Sci. 2022, 13, 5973; g) A. C. Shaikh, M. M. Hossain, J. Moutet, A. Kumar, B. Thompson, V. M. Huxter, T. L. Gianetti, Angew. Chem., Int. Ed. 2025, e202420483; h) N. Lal, Deepshikha, P. Singh, A. C. Shaikh, Chem. Commun. 2025, 61, 3005.
- [6] a) A. Ueda, H. Wasa, S. Suzuki, K. Okada, K. Sato, T. Takui, Y. Morita, Angew. Chem., Int. Ed. 2012, 51, 6691; b) T. J. Sørensen, M. F. Nielsen, B. W. Laursen, ChemPlusChem 2014, 79, 1030; c) A. C. Shaikh, J. Moutet, J. M. Veleta, M. M. Hossain, J. Bloch, A. V. Astashkin, T. L. Gianetti, Chem. Sci. 2020, 11, 11060.
- [7] a) D. M. Friedrich, W. M. McClain, Annu. Rev. Phys. Chem. 1980, 31, 559;
   b) J. R. Lakowicz, Principles of fluorescence spectroscopy, 3 ed., Springer, Berlin. 2006.
- [8] a) W. Denk, J. H. Strickler, W. W. Webb, Science 1990, 248, 73; b) L. Kubínová, J. Janáček, I. Krekule, A. Diaspro, Confocal and Two-Photon Microscopy: Foundations, Applications and Advances, Wiley-Liss Inc., New York, 2002; c) W. Denk, D. W. Piston, W. W. Webb, in Handbook Of Biological Confocal Microscopy (Ed.: J. B. Pawley), Springer US, Boston, MA, 2006, pp. 535.
- [9] M. Göppert-Mayer, Ann. Phys. 1931, 401, 273.
- [10] a) S. R. Wecksler, A. Mikhailovsky, D. Korystov, F. Buller, R. Kannan, L.-S. Tan, P. C. Ford, *Inorg. Chem.* 2007, 46, 395; b) N. S. Makarov, M. Drobizhev, A. Rebane, *Opt. Express* 2008, 16, 4029; c) M. Pawlicki, H. A. Collins, R. G. Denning, H. L. Anderson, *Angew. Chem., Int. Ed.* 2009, 48, 3244.
- [11] a) R. Bouvier, R. Durand, L. Favereau, M. Srebro-Hooper, V. Dorcet, T. Roisnel, N. Vanthuyne, Y. Vesga, J. Donnelly, F. Hernandez, J. Autschbach, Y. Trolez, J. Crassous, Chem.-Eur. J. 2018, 24, 14484; b) Y. Vesga, C. Diaz, J. Crassous, F. E. Hernandez, J. Phys. Chem. A 2018, 122, 3365; c) C. M. Cruz, I. R. Márquez, I. F. A. Mariz, V. Blanco, C. Sánchez-Sánchez, J. M. Sobrado, J. A. Martín-Gago, J. M. Cuerva, E. Maçôas, A. G. Campaña, Chem. Sci. 2018, 9, 3917; d) W.-T. Deng, H. Qu, Z.-Y. Huang, L. Shi, Z.-Y. Tang, X.-Y. Cao, J. Tao, Chem. Commun. 2019, 55, 2210; e) Z.

- Hu, L. Li, Z. Liu, Z. Liu, D. Zhang, K. Li, ChemPlusChem 2020, 85, 742; f) P. Reine, A. M. Ortuño, I. F. A. Mariz, M. Ribagorda, J. M. Cuerva, A. G. Campaña, E. Maçôas, D. Miguel, Front. Chem. 2020, 8, 306; g) S. Míguez-Lago, I. F. A. Mariz, M. A. Medel, J. M. Cuerva, E. Maçôas, C. M. Cruz, A. G. Campaña, Chem. Sci. 2022, 13, 10267.
- [12] a) W. Chen, S. Zhang, M. Zhou, T. Zhao, X. Qin, X. Liu, M. Liu, P. Duan, J. Phys. Chem. Lett. 2019, 10, 3290; b) S. Mizzoni, S. Ruggieri, A. Sickinger, F. Riobé, L. Guy, M. Roux, G. Micouin, A. Banyasz, O. Maury, B. Baguenard, A. Bensalah-Ledoux, S. Guy, A. Grichine, X.-N. Nguyen, A. Cimarelli, M. Sanadar, A. Melchior, F. Piccinelli, J. Mater. Chem. C 2023, 11, 4188; c) S. Morishita, M. Kunihiro, M. Funahashi, N. Tsurumachi, J. Lumin 2023, 263, 120068.
- [13] a) P. Stachelek, L. MacKenzie, D. Parker, R. Pal, Nat. Commun. 2022, 13, 553; b) O. G. Willis, F. Petri, D. F. De Rosa, A. Mandoli, R. Pal, F. Zinna, L. Di Bari, J. Am. Chem. Soc. 2023, 145, 25170; c) D. F. De Rosa, P. Stachelek, D. J. Black, R. Pal, Nat. Commun. 2023, 14, 1537.
- [14] S. G. Stenspil, A. H. Olsson, R. Mucci, M. Pink, C. Besnard, G. Pescitelli, J. Lacour, A. H. Flood, B. W. Laursen, *Angew. Chem., Int. Ed.* 2024, 63, e202412320.
- [15] L. Arrico, L. Di Bari, F. Zinna, Chem.-Eur. J. 2021, 27, 2920.
- [16] N. Mehanna, S. Grass, J. Lacour, Chirality 2012, 24, 928.
- [17] Compounds 4 should be considered as synthetic intermediates necessary for preparation of OMe derivatives 2; they were not used in photophysical or biological studies.
- [18] O. Kel, P. Sherin, N. Mehanna, B. Laleu, J. Lacour, E. Vauthey, Photochem. Photobiol. Sci. 2012, 11, 623.
- [19] S. Abbate, G. Longhi, F. Lebon, E. Castiglioni, S. Superchi, L. Pisani, F. Fontana, F. Torricelli, T. Caronna, C. Villani, R. Sabia, M. Tommasini, A. Lucotti, D. Mendola, A. Mele, D. A. Lightner, J. Phys. Chem. C 2014, 118, 1682
- [20] a) H. Tanaka, Y. Inoue, T. Mori, ChemPhotoChem 2018, 2, 386; b) M. Cei,L. Di Bari, F. Zinna, Chirality 2023, 35, 192.
- [21] L.-O. Pålsson, R. Pal, B. S. Murray, D. Parker, A. Beeby, *Dalton Trans.* 2007, 2007, 5726.
- [22] D. Liu, V. García-López, R. S. Gunasekera, L. Greer Nilewski, L. B. Alemany, A. Aliyan, T. Jin, G. Wang, J. M. Tour, R. Pal, ACS Nano 2019, 13 6813
- [23] a) R. Carr, R. Puckrin, B. K. McMahon, R. Pal, D. Parker, L.-O. Pålsson, MAF
  2014, 2, 024007; b) C. Xu, W. W. Webb, J. Opt. Soc. Am. B 1996, 13, 481; c)
  L. Young, J. Sung, G. Stacey, J. R. Masters, Nat. Protoc. 2010, 5, 929; d)
  R. Pal, Faraday Discuss. 2015, 177, 507; e) C. A. Schneider, W. S. Rasband,
  K. W. Eliceiri, Nat. Methods 2012, 9, 671; f) S. Bolte, F. P. Cordelières, J. Microsc. 2006, 224, 213.

Manuscript received: March 27, 2025 Revised manuscript received: April 15, 2025 Version of record online: May 2, 2025