Femtosecond Photoelectron Imaging of Aligned Polyanions: Probing Molecular Dynamics through the Electron-Anion Coulomb Repulsion.

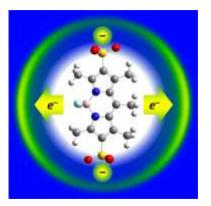
Daniel. A. Horke, Adam S. Chatterley and Jan R.R. Verlet*

University of Durham, Department of Chemistry, DH1 3LE Durham, United Kingdom

*j.r.r.verlet@durham.ac.uk

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The first time-resolved photoelectron imaging study of a polyanion is presented. Using the alignment induced through resonance excitation, the photoelectron angular distributions can be qualitatively understood in terms of the position of localised excess charges on the molecular skeleton, which influence the photoemission dynamics. Pump-probe experiments are used to demonstrate that the photoelectron angular distribution is also sensitive to molecular dynamics. This is shown here for the rotational dynamics of a polyanion, in which the photoelectron anisotropy tracks the rotational coherence as it dephases. The methodology can in principle be applied to general molecular dynamics in large polyanions, providing a new route to studying ultrafast structural dynamics in complex gas-phase systems.



KEYWORDS: Multiply Charged Anions; Polyanions; Time-Resolved Photoelectron Spectroscopy; Repulsive Coulomb Barrier; Velocity-map Imaging; Alignment; Photoelectron Angular Distribution

Polyanions are common in condensed phases, where the interaction with neighbouring species stabilises the inherent intra-molecular Coulombic repulsion. When isolated in the gas-phase, this repulsion renders many polyanions unstable, either with respect to electron loss or fragmentation. ¹⁻⁴ Despite this instability, polyanions also possess a dynamic stability due to the repulsive Coulomb barrier (RCB) that arises from a balance between the short-range attraction and the long-range repulsion between an electron and anion (see Fig. 1(a)). ⁵⁻¹⁰ The RCB is generally not isotropic and depends on the location of the excess charges within the molecular framework. While for anions or neutrals the photoelectron angular distribution (PAD) is determined by the interference of the partial photoelectron waves, in polyanions it has been shown through photoelectron (PE) imaging¹¹⁻¹³ that the shape of the RCB strongly influences the outgoing electrons.¹⁴⁻¹⁶ In this letter, we report the first time-resolved PE imaging study of a polyanion. It is shown that the PAD can be qualitatively understood based on the photo-induced alignment of the polyanion. Moreover, the PADs are shown to be sensitive to molecular rotational dynamics. These observations suggest that time-resolved PE imaging of polyanions may provide an elegant route to discerning large-amplitude structural dynamics in the gas-phase, with femtosecond resolution.

In PE imaging, the outgoing electron is captured in the laboratory frame. Relating the photoemission process to the molecular frame through molecular alignment has been demonstrated using a number of

methods. ¹⁷⁻²⁰ Here, we align a dianion in the laboratory frame using a well-defined molecular transition, from an initial state *m* to a final state *n*. If the direction of the transition dipole moment, \mathbf{d}_{nm} , is known, excitation with linearly polarised light, $\boldsymbol{\varepsilon}$, will produce an ensemble of molecules in an excited state, aligned in a $\cos^2\theta$ distribution (where θ is the angle between \mathbf{d}_{nm} and the laser polarisation, $\boldsymbol{\varepsilon}$). Because $\boldsymbol{\varepsilon}$ defines the laboratory frame and \mathbf{d}_{nm} the molecular frame, a measurement in the laboratory frame will provide molecular frame information.

As a model system, the laser dye pyrromethene 556 (PM²⁻) is used. ^{21,22} Its structure, shown in Fig. 1(b), comprises of a central boron-dipyrromethene (BODIPY) chromophore and two terminal SO₃⁻ groups, where the excess charges are localised. The BODIPY chromophore has been extensively studied. ^{23,24} It has a strong S₁ \leftarrow S₀ transition centered around 2.49 eV (498 nm) in methanol²⁵ and **d**₁₀ is aligned along the *z*_{MF}-axis, defined in Fig. 1(b).

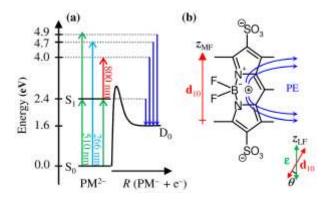


Figure 1. (a) Energy level diagram of PM^{2-} , showing the repulsive Coulomb barrier (RCB) to photodetachment and the excitation and photodetachment schemes used. (b) Structure of PM^{2-} , with the transition dipole moment for $S_1 \leftarrow S_0$ excitation, \mathbf{d}_{10} , shown which defines the molecular frame *z*-axis, z_{MF} . Also shown is the polarization vector of the excitation field, $\boldsymbol{\varepsilon}$, which defines the laboratory frame *z*-axis, z_{LF} . Excitation of PM^{2-} results in the relative alignment of z_{MF} to z_{LF} with an angle θ , according to a $\cos^2\theta$ distribution. A qualitative sketch indicating the direction of the outgoing photoelectrons (PE) is shown as blue arrows, which is determined by the Coulombic repulsion from the remaining charges on the molecule.

Experimentally, isolated PM²⁻ is generated from a 1mM solution of the PM sodium salt (Exciton) in methanol, using electrospray ionisation. ^{26,27} An ion bunch of mass-selected PM²⁻ is produced by trapping the ions in a radio-frequency ion trap. Following trapping, the ion packet is injected into a timeof-flight mass-spectrometer at 500 Hz. The ion-bunch is then intersected perpendicularly by femtosecond laser pulses at the centre of a velocity-map-imaging PE spectrometer.²⁸ Electrons are guided in the direction mutually perpendicular to the ion and laser beams, towards a position sensitive detector. The position of the PE striking the detector is monitored using a charge-coupled device. The resultant PE images are reconstructed using polar onion peeling, to yield the central slice through the 3D PE cloud, providing the PE spectrum and PAD.²⁹ Femtosecond pulses are derived from a commercial amplified Ti:Sapphire laser system, with a fundamental output at 1.55 eV (800 nm). Femtosecond pulses at 4.66 eV (266 nm) are generated by third harmonic generation. Light at 2.43 eV (510 nm) is produced using an optical parametric amplifier and subsequent frequency mixing of the signal with the fundamental. Pump and probe pulses are delayed relative to each other using a motorised optical delay line and the temporal resolution, as determined by a cross-correlation, is on the order of 140 fs. The intensity of all pulses in the interaction region is on the order of 1×10^{10} W cm⁻² and ϵ is set to be parallel to the detector and defines the z_{LF} -axis (see Fig. 1(b)). The polarisation purity of pump and probe pulses is ~10:1. PE spectra have been calibrated to the known spectrum of Γ and the spectrometer has a spectral resolution of $\Delta E/E = 5\%$.

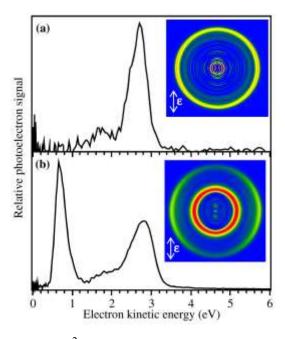


Figure 2. Photoelectron spectrum of PM^{2-} collected at (a) 4.66 eV (266 nm) and (b) 2.43 eV (510 nm). Shown inset is the central slice of the corresponding photoelectron image.

Fig. 2(a) shows the PE spectrum taken at 4.66 eV (266 nm). It shows a single PE feature at an electron kinetic energy $eKE \sim 2.7$ eV. From this feature, the adiabatic binding energy of PM²⁻ can be determined to be 1.6 ± 0.2 eV. This is in agreement with *ab initio* calculations^{30,31} that also show that the highest-occupied molecular orbital (HOMO) of PM²⁻ is that of the π -system of the BODIPY chromophore. PE spectroscopy will therefore form a zwitterionic system, in which a hole is left on the BODIPY system and two negative charges on the terminal SO₃⁻ groups. The outgoing PE will consequently be influenced by the resultant electrostatic RCB potential. The RCB will be anisotropic with respect to the molecular axis, exhibiting large potential maxima around the SO₃⁻ groups, while the lowest saddle point in the RCB will be along the plane perpendicular to the molecular axis. There will be influences on the exact shape of the RCB, particularly due to the out-of-plane F atoms and the fact that the system is not symmetric about z_{MF} . These will however be relatively minor compared to the presence of the charged regions of the system.

The inset in Fig. 2(a) shows the central slice through the PE Newton sphere. The PE anisotropy can be quantified by fitting the PAD to the well-known expression:

$$I(\theta) = \frac{\sigma}{4\pi} \left(1 + \sum_{2n} \beta_n P_n(\cos \theta) \right)$$

where σ is the cross-section for detachment, $P_n(\cos\theta)$ are the n^{th} -order Legendre polynomials, and β_n is a measure of the anisotropy. ³² The sum runs over 2n, where *n* is the number of photons absorbed. For the one-photon PE image in Fig. 2(a), $\beta_2 = -0.15 \pm 0.10$, indicating that the PAD is almost isotropic. At first glance, this observation would suggest that the RCB has little or no effect on the outgoing photoelectrons. However, the PEs are collected in the laboratory frame and if there is no correlation between laboratory and the molecular frames, then experimentally one cannot determine the effect of the RCB on the outgoing electron. In the experiments by the Wang group,^{14,16} this correlation is provided by the dependence of the integral photodetachment cross section on the molecular orientation. The polyanions in their study¹⁴ have a similar overall RCB to PM²⁻ (i.e. a central chromophore and terminal negatively charged groups), but the photodetachment cross section peaks when the molecular axis is aligned with ε . This leads to the observed negative anisotropy parameter as the PE leaves preferentially perpendicular to the molecular axis to avoid the large RCB near the negative charges. For PM²⁻ the integral cross section appears to be significantly less sensitive to the molecular orientation such that the influence of the RCB cannot be clearly observed. This is reflected in the one-photon detachment from PM²⁻, where the observed anisotropy following detachment at 266 nm is near isotropic (Fig. 2(a)). Therefore in order to reliably observe the influence of the RCB on an outgoing photoelectron in PM²⁻, a connection between the laboratory and molecular frames must be established. This can be achieved through alignment of the molecule in space. Such a methodology is generally applicable to any polyanion, irrespective of variations in the integral photodetachment cross-section with respect to the molecular frame and this is demonstrated here using alignment through a resonant electronic transition.

Fig. 2(b) shows the PE spectrum taken at hv = 2.43 eV (510 nm), resonant with the S₁ \leftarrow S₀ transition. The PE spectrum shows features centered at eKE = 0.7 eV and eKE = 2.9 eV. The PE feature at eKE = 0.7 eV arises from absorption of a single photon. We can crudely estimate the lowest RCB by calculating the potential energy of an electron interacting with a system in which we assume a negative point-charge on the two SO₃⁻ groups and a point-charge hole on the BODIPY. This yields an outer RCB of 1.3 eV. ^{33,34} Given the adiabatic binding energy of 1.6 eV, at least 2.9 eV is required to directly detach an electron. Hence, the feature observed at eKE = 0.7 eV must arise from tunneling through the RCB.⁵ In solution, the S₁ is highly fluorescent with a quantum yield = 0.98^{22} and a nanosecond lifetime,³⁵ and hence, it is not unreasonable to anticipated that tunneling through the RCB will be a competitive decay pathway for the S₁ state in the gas-phase.^{5,36}

The feature at eKE = 2.9 eV is commensurate with the absorption of two-photons. This would impart a total of 4.86 eV, extending the PE spectrum by an additional 0.2 eV relative to the PE spectrum taken at 4.66 eV (266 nm) shown in Fig. 2(a). Importantly, this is a resonance-enhanced 2-photon detachment (R2PD) feature, as the first photon is resonant with the S₁ \leftarrow S₀ transition. This accounts for the large PE signal observed for this feature. Hence, the feature at eKE = 2.9 eV arises from the photodetachment of PM²⁻ in its S₁ state, not the ground S₀ state.

Analysis of the PE image shown inset in Fig. 2(b) reveals that the two PE features have very different PADs. Specifically, the tunneling feature is almost isotropic ($\beta_2 = -0.16 \pm 0.08$), while the R2PD feature is strongly anisotropic ($\beta_2 = -0.60 \pm 0.09$, $\beta_4 = 0.06 \pm 0.10$) in the direction perpendicular to ε . As shown by the Wang group, an outgoing PE can be strongly affected by the RCB.^{14,16} For PM²⁻, the R2PD feature arises from the photodetachment from the S₁ state. As this state is prepared by excitation with linearly polarised light, PM²⁻ molecules in the S₁ state will be aligned relative to the z_{LF} with a cos² θ distribution. Hence, any PE generated by absorption of an additional photon will experience an RCB that is *aligned* in the lab-frame. In Fig. 1(b), **d**₁₀ is aligned with z_{LF} such that, in a simplistic classical picture, a PE will experience an RCB that forces it in a direction perpendicular to z_{LF} , which leads to the observed $\beta_2 < 0$ anisotropy. The fact that the observed β_4 is close to zero for this feature suggests that the cross-section for photodetachment from the S₁ state is also relatively invariant to the molecular alignment.

The tunneling feature at eKE = 0.7 eV shows an almost isotropic PAD, despite the fact that this feature also arises from an aligned distribution of PM^{2–} molecules in the S₁ excited state. However, as this peak is formed by electrons that tunnel and this is a delayed process, molecular dynamics in the excited state may compete with tunneling.

We probe the relaxation dynamics of the S₁ state using time-resolved PE spectroscopy. Population in the S₁ state is prepared via S₁ \leftarrow S₀ excitation with a 2.43 eV (510 nm) photon, and probed at various delays later using a 1.55 eV (800 nm) photon. The probe is sufficiently energetic to directly detach an electron from the S₁ excited state, but insufficient for direct detachment from the S₀ ground state of the dianion. A PE spectrum near the temporal overlap between the two pulses is shown in Fig. 3(a) and shows the additional PE feature centered at eKE = 1.9 eV. This feature is a direct probe of the S₁ state population, and in Fig. 3(b) we show the integrated PE signal (between 1.6 < eKE < 2.1 eV) as a function of pump-probe delay. The S₁ state population shows first order decay kinetics⁵ and we fit these data with a single exponential decay, yielding a tunneling lifetime of 120 ± 30 ps. The tunneling lifetime is relatively long and rotational dephasing dynamics are expected to occur on timescales that are significantly faster than this, suggesting that the alignment may be lost before the majority of electrons are ejected from the excited state, thus leading to an isotropic distribution.

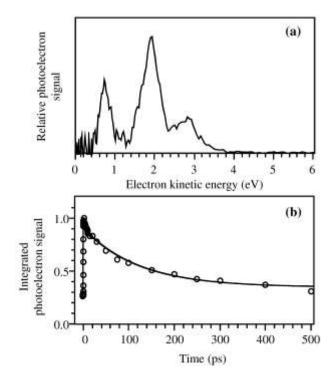


Figure 3. (a) Photoelectron spectrum taken with pump (2.43 eV, 510 nm) and probe (1.55 eV, 800 nm) pulses overlapping, t = 0.5 ps. A large 1+1' detachment feature is observed at eKE = 1.9 eV. (b) Total integrated photoelectron signal of the 1+1' detachment feature, fitted with a single exponential decay, yielding an excited state lifetime of 120 ps.

The observed PAD for detachment from the S₁ is a measure of the degree of alignment within the ensemble of S₁ excited PM²⁻ and therefore, the time-resolved PADs can be used to track the rotational dephasing following excitation of the S₁ state. To exclusively monitor the anisotropy of the S₁ state, we have subtracted from each time-resolved PE image an image taken when the probe arrives before the pump ($t < t_0$). Representative central slices through the PE images following such a subtraction are shown in Fig. 4(a)-(c) for a number of different pump-probe delays. These indicate changes in the PADs with pump-probe delay. These changes can be quantified through the changes in anisotropy parameter $\Delta\beta_2$, as shown in Fig. 4(d), that have been referenced to the long-time anisotropy which is assumed to be isotropic. When pump and probe pulse are temporally overlapped (t = 0), the anisotropy is maximal at $\beta_2 = -0.53 \pm 0.17$. Within error, this value agrees with the value obtained from the R2PD feature in the PE spectrum taken at 2.43 eV (510 nm). We note that, although our determination of the absolute β_2 value has a relatively large error, we have shown that we are significantly more sensitive to changes in β_2 .²⁶ As the pump-probe delay increases, the images become more isotropic, reflected as a decrease in the absolute β_2 value. After a few picoseconds, the PADs are almost isotropic and show no further changes at longer delays where only the intensity of the image decreases.

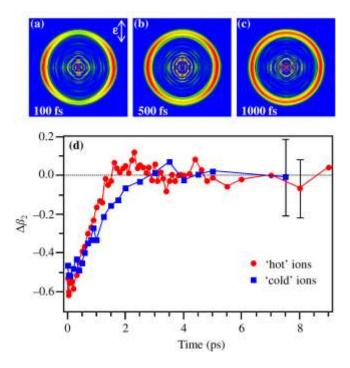


Figure 4. (a)-(c) Background subtracted deconvoluted photoelectron images, showing the varying anisotropy as a function of pump-probe delay, for ions prepared hot. (d) Observed change in photoelectron anisotropy parameter, $\Delta\beta_2$, as a function of pump-probe delay for ions prepared with different amounts of internal energy.

The observed changes in PADs can be rationalised in terms of the dephasing of the rotational coherence initially excited in the S₁ state. Following excitation, a coherent distribution of rotational states is excited in the S₁ state as shown by Felker and Zewail. ^{37,38} This leads to an aligned distribution at t = 0 by virtue of the alignment between \mathbf{d}_{10} and $\boldsymbol{\varepsilon}$. As the rotational coherences evolve, their phases rapidly become scrambled and the ensemble of S₁ excited PM²⁻ appears isotropically distributed, leading to the observed changes in β_2 . As the superposition is coherent, rotational revivals are expected to occur at times determined by the rotational constants of the system.³⁹ For PM²⁻, the earliest possible half revival occurs on a timescale that greatly exceeds the observed S₁ lifetime and so these revivals are not observed experimentally.

The fast dephasing time of the initial ensemble arises from the fact that the ground state PM^{2-} is intentionally prepared hot and hence a large number of rotational states contribute to the dynamics. The temperature is crudely controlled by the potential used to empty the ion trap and by the buffer gas in the trap. For the current experiments air was used as a buffer gas, which leads to a substantial heating of ions. Changing the buffer gas to He is expected to reduce the internal energy transferred to the trapped ions by a factor of ~ 6, bringing the temperature close to 300 K. At this lower temperature the rotational dynamics slow appreciably, as shown in Fig. 4(d). This provides strong support for our interpretation of the observed dynamics arising from rotational dephasing, which is determined by the initial rotational distribution of the molecules in the ground state. The observed picosecond timescale for rotational dephasing is in agreement with studies in systems of a similar size. ⁴⁰

The key observations here are that the RCB directs the photoelectron emission in polyanions and that PE imaging can be used to qualitatively predict the shape of the RCB. In general, photodetachment is governed by the quantum nature of the outgoing electron and the PADs are governed by the interference between outgoing partial waves as well as partial wave cross sections. This remains the correct description for understanding the PE emission from polyanions. However, it is also evident that the RCB has a determining influence on the possible directions in which the PE can be ejected: certain regions of space are simply inaccessible as the Coulomb barriers are too high relative to the energy of the ejected electron. Despite its shortcomings, this then provides a simple classical framework in which to qualitatively interpret experimentally observed PADs. The observation of temporal changes in the anisotropy as the molecular alignment in PM²⁻ is lost provides additional support for this conceptual picture. More importantly, the temporal evolution of the PAD suggests that time-dependent changes of the RCB can be used as a probe for the molecular dynamics. This has been demonstrated here for changes in the RCB within the laboratory frame, but it can in principle also be extended to changes in the molecular frame RCB. Because the RCB depends critically on where the excess charges are localised on the molecular frame, the time-resolved PAD provides a crude measure of the structure of the polyanion at a given time following excitation. Time-resolved PE imaging therefore provides a route to measuring structural dynamics on a femtosecond timescale. For example, isomerisation dynamics will be reflected in dynamic changes in PADs as there can be large changes in the relative location of the

excess negative charges during such structural changes. These experiments are currently underway in

our laboratory.

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