

1 Termination of a continent-margin upwelling system
2 at the Permian-Triassic boundary (Opal Creek, Alberta, Canada)

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29 ABSTRACT

30

31 Models of mass extinctions caused by greenhouse warming depend on the ability of
32 warming to affect the oxygenation of the ocean, either through slowing circulation or
33 changes in biological productivity and the organic carbon budget. Opal Creek, Alberta,
34 Canada is a biostratigraphically continuous Permian-Triassic Boundary (PTB) section
35 deposited in deep water on an outer shelf setting in the vast and understudied Panthalassic
36 Ocean, along the western margin of Pangaea. The end-Permian extinction is here
37 represented as the disappearance of the previously dominant benthic fauna (siliceous
38 sponges). On the basis of nitrogen and reduced sulfur isotopes as well as productivity-
39 sensitive trace elements, the Middle Permian at Opal Creek is interpreted as a highly
40 productive coastal upwelling zone where vigorous denitrification and sulfate reduction
41 occurred in a mid-water oxygen minimum. Similar conditions appear to have continued into
42 the latest Permian until the onset of a euxinic episode represented by a discrete pyrite bed
43 and several trace element indicators of high productivity. This euxinic pulse is followed by
44 the extinction of benthic fauna and a shift in nitrogen and sulfur isotopes to more normal
45 marine values, suggesting the cessation of coastal upwelling and the consequent weakening
46 of the mid-water oxygen minimum. The Lower Triassic appears to be a dysoxic, relatively
47 unproductive environment with a bottom water oxygen minimum. Rhenium-Osmium
48 isotope systematics show a minimum of radiogenic Os near the main extinction event,
49 which may be due to volcanic input, and increasingly radiogenic values approaching the
50 PTB, possibly due to increased continental erosion. The Opal Creek system demonstrates

51 that, while the biogeochemical crisis in the latest Permian was capable of impacting the
52 coastal upwelling modality of ocean circulation, a transient increase in productivity likely
53 drove the system toward euxinia and, ultimately, extinction.

54

55 *Key words:* Permian; extinction; Panthalassic; marine productivity; upwelling; anoxia

56

57 **1. Introduction**

58

59 The latest Permian mass extinction (LPME), at ~252 Ma, was the greatest
60 biodiversity crisis of the Phanerozoic (Erwin et al., 2002), fundamentally and permanently
61 altering terrestrial and marine ecosystems (Benton et al., 2004; Bottjer et al., 2008). The
62 trigger for this crisis is thought to have been eruption of the Siberian Traps flood basalts
63 (Wignall, 2007; Reichow et al., 2009), which led to strong global warming as a consequence
64 of emissions of CO₂ and possibly methane (Payne and Kump, 2007; Retallack and Jahren,
65 2008). Climatic warming or related effects led to perturbations of terrestrial environments,
66 including changes in weathering patterns, aridification, vegetation loss, and fluvial drainage
67 alteration (Retallack, 1999; Looy et al., 1999, 2001; Ward et al., 2000). In the marine realm,
68 warming is thought to have been an important factor in reduced deepwater ventilation, a
69 buildup of hydrogen sulfide, hypercapnia, seawater acidification, and depletion of certain
70 nutrients such as nitrate (Knoll et al., 1996; Kump et al., 2005; Fraiser and Bottjer, 2007;
71 Luo et al., 2011). The development of global greenhouse conditions at this time is
72 paleontologically evident in the appearance of warm-climate floras at polar latitudes
73 (Taylor et al., 2000), the demise of high-latitude, cool-water siliceous sponge communities
74 (Henderson, 1997; Beauchamp and Baud, 2002), and a breakdown of temperature-based
75 conodont provincialism (Mei and Henderson, 2001).

76 Direct evaluation of changes in Permian-Triassic ocean circulation patterns has
77 proven difficult, owing to the relative paucity of preserved PTB sections from the
78 Panthalassic Ocean, which covered some 70% of the globe at the time of the extinction (Fig.
79 1) and hence is central to investigations of global change related to the LPME. . Extant
80 Panthalassic sections are limited to a few areas and depositional facies, including deep-sea
81 cherts (Isozaki, 1994, 1997; Algeo et al., 2010, 2011b; Sano et al., 2010) and shallow atoll
82 carbonates from Japan (Musashi et al., 2001), volcanic-arc slope sediments in New Zealand
83 (Krull et al., 2000), and a series of deep-shelf or upper slope siliciclastic successions in
84 western North America (Beauchamp and Baud, 2002; Wignall and Newton, 2003; Hays et
85 al., 2007). Anoxic facies have been documented at widespread Permian-Triassic boundary
86 (PTB) sections, leading some workers to infer that such redox changes were global in
87 extent, and due to strongly reduced overturning circulation (“ocean stagnation”) (Isozaki,
88 1994, 1997; Wignall and Twitchett, 1996, 2002). Paleooceanographic modeling studies have
89 demonstrated that, while strong climatic warming could slow the meridional overturning
90 circulation significantly, it would generally not be sufficient to generate the persistent,
91 highly reducing conditions necessary to drive surface euxinia without a substantial increase
92 in nutrient availability (Hotinski et al., 2001; Meyer et al. 2008). Modeling of
93 paleooceanographic conditions at the PTB generally supports the hypothesis that
94 overturning circulation persists despite extreme greenhouse conditions (Kiehl and Shields,
95 2005; Winguth and Maier-Reimer, 2005).

96 Recently, attention has focused on the role of increased nutrient inventories in
97 ocean-surface waters, and consequent increases in marine primary productivity, in driving
98 PTB oceans toward anoxia (Algeo et al., 2011a). This hypothesis has received support from
99 studies demonstrating a substantial expansion of the mid-ocean oxygen-minimum zone in

100 periequatorial regions (Algeo et al., 2010, 2011b) as well as a steepening of shallow-to-deep
101 carbon isotopic gradients in epicratonic seas (Meyer et al., 2011; Song et al., in review).

102 [Insert Figure 1]

103 In this study, we examine changes along the eastern margin of the subtropical
104 Panthalassic Ocean between the Early Permian and Early Triassic. Our primary goal is to
105 test models of anoxia-development in the Panthalassic ocean by collecting field geochemical
106 evidence for changes in circulation and organic productivity rates during the PTB transition
107 interval. The site for this study is the Opal Creek section in southwestern Alberta, which
108 represents an outer-shelf or upper slope setting on the western margin of the North
109 American plate at a paleolatitude of $\sim 30^{\circ}\text{N}$ (Henderson, 1989, Richards, 1989). As shown
110 by both field data and paleoceanographic modelling, this region was characterized by
111 strong upwelling during much of the Permian (Ziegler et al., 1998, Beauchamp and Baud,
112 2002, Winguth and Maier-Reimer, 2005), a mode of circulation potentially sensitive to
113 global climate changes. A shift in ocean circulation patterns is suggested by major changes
114 in boreal marine biotas during the Late Permian (Henderson and Baud, 1997; Beauchamp
115 and Baud, 2002; Gates et al., 2004; Beauchamp et al., 2009), culminating in a regional
116 extinction of sponges in the late Changhsingian (Algeo et al., 2012), as well as a shift from
117 the enriched nitrogen isotope values characteristic of coastal upwelling zones to values
118 characteristic of more nitrogen-limited environments (Schoepfer et al. 2012). Here, we
119 present sulfur isotope and trace-element concentration data in order to (1) document
120 changes in redox conditions, productivity, and euxinia within the upwelling system of the
121 eastern subtropical Panthalassic Ocean during Early Permian to Early Triassic, and (2)
122 evaluate the relationship of changes in this region during the PTB transition to broader
123 changes in the global environment based on detailed biostratigraphic correlations
124 (Henderson et al., 1994; Henderson, 1997).

125 **2. Geologic setting and biostratigraphy**

126 The Opal Creek section is located in the Kananaskis Valley, in the foothills of the
127 Canadian Rockies west of Calgary (Fig. 1). It comprises ~25 meters of Lower Permian
128 through Lower Triassic strata assigned to the Johnston Canyon, Ranger Canyon, and
129 Sulphur Mountain formations. Beds dip nearly vertically, with cherts and silicified
130 siltstones forming prominent cliffs while more fissile siltstones are exposed in hill slopes
131 (Fig. 2). The thermal maturity of the section is moderate, as shown by a conodont color
132 alteration index (CAI) of ~2.5 (Schoepfer et al., 2012), equivalent to a maximum burial
133 temperature of ~150°C (Tissot and Welte, 1984; Hunt, 1996). Similar maturity levels have
134 been inferred for other PTB sections in the Canadian Western Sedimentary Basin on the
135 basis of vitrinite reflectance (Radke et al., 1982), CAI (Utting et al., 2005), and biomarker
136 studies (Hays et al., 2007).

137 [Insert Figure 2]

138 The Early Permian is represented by the Johnson Canyon Formation (Fig. 3), which
139 is composed of silty dolostone containing abundant nodular phosphate layers. This unit
140 exhibits lithologic cyclicity, likely as a consequence of glacio-eustatic fluctuations during the
141 Early Permian, when ice is believed to have existed at the poles (Haq and Schutter, 2008).

142 The Middle Permian is represented by the Ranger Canyon Formation (RCF, Figs. 3-
143 4), which (together with its stratigraphic equivalent, the Fantasque Formation) extends
144 from southeastern British Columbia northward to the southern Mackenzie Fold Belt in the
145 Yukon (Henderson, 1989). The Ranger Canyon, which is just 2.35 m thick at Opal Creek,
146 consists of completely bioturbated, recrystallized spiculitic chert to cherty immature
147 siltstone containing microcrystalline carbonate cements and some reworked chert and
148 carbonate grains. The predominance of sponge spicules, abundance of phosphatic material

149 including abundant conodonts, and the absence of shallow-water sedimentary structures
150 (Henderson, 1997) suggest an outer shelf or slope environment at a few hundred meters
151 water depth, below storm wave base and probably oligophotic. The most common fossils
152 are large monaxon siliceous sponges spicules (up to 6 cm long), possibly of the class
153 Demospongia. Also abundant are conodonts of the genus *Mesogondolella*, which in the
154 Middle and Late Permian are indicative of cool-water environments, as inferred from their
155 boreal provincialism (Mei and Henderson 2001). The main species are *M. bitteri* and,
156 toward the top of the formation, *M. rosenkrantzi*, indicating a Capitanian age or, where *M.*
157 *rosenkrantzi* is found without *M. bitteri*, an early Wuchiapingian age (Henderson and Mei,
158 2007).

159 The upper contact of the RCF is defined by a major regional unconformity,
160 coinciding with the Guadalupian-Lopingian lowstand (Haq and Schutter, 2008). On the basis
161 of the conodont biostratigraphy, the duration of this unconformity has been estimated at 5-
162 7 million years. The Late Permian and Early Triassic are represented by the Phroso
163 Siltstone Member of the Sulphur Mountain Formation (SMF; Figs. 3-4), which is equivalent
164 to the Grayling Formation in northeastern British Columbia (Wignall and Newton, 2003).
165 The basal ~10 m of this unit represent a transgressive systems tract (TST) characterized by
166 an upward fining succession. Sediments consists of angular, immature silt to fine sand and
167 contains dolomite rhombs as well as reworked chert, micas, and microcrystalline carbonate.
168 Pyrite is also present, concentrated around organic fragments and, especially, in a cm-thick
169 bed that is ~10.5 cm above the basal unconformity. The lowermost beds of this unit are
170 bioturbated and contain abundant siliceous monaxon sponge spicules up to 2 cm in length,
171 but these features disappear abruptly ~40 cm above the basal unconformity. At the same
172 level, the conodont fauna dominated by *Mesogondolella* is abruptly replaced by a new fauna
173 dominated by the genus *Clarkina*, a transition that probably marks the latest Permian mass

174 extinction event (see below). Above this level, the SMF is composed of black to dark grey,
175 pyritic, thin-bedded shale and siltstone. Benthic biota and bioturbation are lost, suggesting
176 suboxic to anoxic bottom waters during deposition (Fig. 2) (Gibson, 1969, 1974; Gibson and
177 Barclay, 1989; Henderson, 1997).

178 Four conodont biozones are present in the Sulphur Mountain Formation (Figs. 3-4),
179 in ascending order:

180 1) *Mesogondolella sheni* Zone—Specimens of *M. sheni* from the basal TST of this
181 formation indicate a Changhsingian age, but the exact relationship between these
182 specimens and the type material from Selong, Tibet (Mei, 1996; Shen et al., 2006) remains
183 uncertain. Specimens of *M. rosenkrantzi* and *M. bitteri* are considered to have been
184 reworked from the underlying Ranger Canyon Formation during the transgression. The
185 extent of reworking is difficult to address since all specimens are fragmented in these
186 deformed shale units. At Ursula Creek in northeastern British Columbia, the uppermost
187 Fantasque Formation contains both *M. sheni* and *M. rosenkrantzi*, indicating a
188 Changhsingian age (Henderson, 1997; Wignall and Newton, 2003). We suggest a Late
189 Changhsingian age for this assemblage at Opal Creek given that indisputably latest
190 Changhsingian species are recovered only 40 cm higher in the section.

191 2) *Clarkina hauschkei*–*Clarkina meishanensis* Zone—This zone is defined by the
192 association of *C. hauschkei*, *C. meishanensis*, *C. cf. changxingensis*, and *C. cf. zhejiangensis*. A
193 comparable assemblage was recovered from the basal Blind Fiord Formation in the
194 Canadian Arctic (Algeo et al., 2012). *C. hauschkei* is similar to specimens described as *C.*
195 *hauschkei borealis* by Kozur (2005) from a zone immediately below the end-Permian
196 microbialite facies at Abadeh, Iran. *C. meishanensis sensu stricto* is restricted to Beds 25 to
197 28 at Meishan (Jiang et al., 2007), but a similar species *C. zhangji* originally described as a

198 subspecies of *C. meishanensis* (Mei et al., 1998) ranges from Bed 23 to 24e at Meishan.
199 These two species may be synonymous and therefore range from Late Permian (Bed 23) to
200 Lower Triassic (Bed 28). Kozur (2005) indicated that *C. meishanensis* migrated into Iran
201 during the late Changhsingian, correlative with the Bed 23-24e interval at Meishan. At Opal
202 Creek, the base of this zone is marked by a sharp reduction in *Mesogondolella* (which
203 survives to the PTB) and the appearance of the genus *Clarkina*. The latter genus is typical of
204 the equatorial warm-water province during the Late Permian (Mei and Henderson, 2001),
205 and its appearance in the Canadian Western Sedimentary Basin is thus indicative of a
206 breakdown of a temperature barrier to migration (Henderson and Mei, 2007). The faunal
207 turnover at Opal Creek is either equivalent in age to, or slightly precedes that of, the main
208 extinction horizon (Bed 25) at Meishan, China (Jin et al., 2000).

209 3) *Hindeodus parvus*–*Clarkina taylorae* Zone—*H. parvus* is the key marker for the
210 base of the Triassic at the GSSP in Meishan, China (Yin et al., 1996, 2001); its occurrence
211 elsewhere does not define the PTB, but according to chronostratigraphic principles
212 indicates only proximity to it within the *H. parvus* Zone (Henderson, 2006). At Meishan and
213 elsewhere, the lowermost Triassic is typically characterized by a short-term acme of
214 *Hindeodus* (Chen et al., 2009; Jiang et al., 2007; Zhang et al., 2007). Although this genus is
215 rare at Opal Creek, some small, poorly preserved specimens were found in the lower ~6 m
216 of the SMF. The local first occurrence of *H. parvus* is ~1.5 m above the base of the SMF,
217 where it is found in association with *C. taylorae*. The latter taxon has not been recovered
218 with certainty below the PTB (Orchard et al., 1994), so the co-occurrence of these two
219 species is probably a good marker for the stratigraphic position of the PTB at Opal Creek.

220 4) *Clarkina taylorae*–*Clarkina cf. carinata* Zone—*C. carinata* (= *Neogondolella* of
221 some authors) is the most commonly identified Lower Triassic conodont species (Clark,

222 1959; Orchard and Krystyn, 1998; Sweet, 1970). The co-occurrence of *C. taylorae* and *C. cf.*
223 *carinata* indicates a Griesbachian (Lower Induan) age.

224

225 **3. Materials and methods**

226 *3.1. Field protocol*

227 A total of 55 samples were collected over a 30-m-thick stratigraphic interval ranging
228 from the Lower Permian Johnston Canyon Formation through the Lower Triassic Sulphur
229 Mountain Formation for analysis of S isotopic compositions, S speciation, and trace-metal
230 concentrations, with collection concentrated in the basal Sulphur Mountain Formation,
231 around the extinction and boundary. An additional 14 samples were collected over a 1.4-m-
232 thick stratigraphic interval at the base of the Sulphur Mountain Formation for analysis of
233 Re-Os abundances and Os isotopic compositions. Samples were collected at 10 cm spacing
234 from 20 cm above the Ranger Canyon/Sulphur Mountain unconformity to a level 1.6 m
235 above the unconformity.

236 A separate set of 99 samples was collected at 40 cm intervals from the basal 44 m of the
237 SMF, and processed at the University of Cincinnati. Carbon and sulfur elemental
238 concentrations were determined using an Eltra 2000 C-S analyzer, and results calibrated
239 using USGS and internal laboratory standards. Analytical precision was better than $\pm 2.5\%$
240 for carbon and $\pm 5\%$ for sulfur. An aliquot of each sample was digested in HCl at 50°C for 12
241 hours, washed and filtered, and reanalyzed for C and S in order to determine concentrations
242 of total organic carbon (TOC) and non-acid volatile sulfur (NAVS). These data are presented
243 in Figure 5.

244 *3.2. Sulfur isotopes*

245 The ratio of sulfur stable isotopes ($^{34}\text{S}/^{32}\text{S}$) was measured by elemental analyzer
246 continuous-flow isotope-ratio mass spectrometry (EA-CF-IRMS), at the University of
247 Washington's IsoLab facility. Using a Eurovector EA and Thermo Finnigan ConFlo III and
248 MAT253 isotope-ratio mass spectrometer.

249 All stable isotope ratios for samples are reported in standard delta (δ) notation
250 indicating the per mille (‰) difference from a standard:

251

$$252 \quad \delta = (R_{\text{sample}}/R_{\text{std}} - 1) \times 1000$$

253

254 where R equals $^{34}\text{S}/^{32}\text{S}$ of the standard or sample. The standard used for sulfur isotopes is
255 Vienna Canyon Diablo Troilite (VCDT). The accuracy of ^{34}S measurements ($1\sigma = 0.2\text{‰}$, $n =$
256 20) was estimated through analysis of NIST NBS127 and NZ1 reference materials.

257 Analytical precision (1σ) was determined based on replicates of an internal laboratory
258 standard (Sigma Aldrich BaSO_4) during each run and was $\leq 0.2\text{‰}$ ($n = 5$ to 10). The average
259 precision (1σ) for sample duplicates was larger (0.6‰ , $n = 2$ -3), probably owing to sample
260 heterogeneity.

261 Sulfur concentrations were also calculated for each sample by calibrating the total S
262 signal for a range of masses of the internal standard during each isotopic analysis. In order
263 to identify the relative contributions of different sulfur species (sulfates including carbonate
264 associated sulfates, sulfides, organic sulfur, and barite) to total sulfur, different sample
265 fractions were obtained by chemical processing. Sulfates were removed from sample
266 aliquots by an acidification step, i.e., washing with an HCl solution (2N hydrochloric acid,
267 36-72 hours at 60°C), leaving only sulfide and organic S in the residue. An aliquot of the
268 residue was then bleached, i.e., treated with a 5.5% NaClO solution (12-24h, 60°C) to

269 remove organic sulfur, leaving only the sulfide component, although some of the sulfide
270 fraction may have been removed by this oxidation step. An aliquot of this second residue
271 was then treated with a nitric acid wash (6N HNO₃, 24-48h, 60°C) in order to oxidize
272 sulfides and isolate a possible barite component.

273 *3.3. Re-Os isotopes*

274 The Re-Os analytical protocol followed that described in detail by Selby and Creaser
275 (2003) and Selby et al. (2007) and was carried out in the TOTAL Laboratory for Source
276 Geochronology and Geochemistry at Durham University. Re was recovered by anion
277 chromatography and further purified using single anion bead chromatography. The Re and
278 Os isolates were analyzed for their isotopic compositions by negative thermal ionization
279 mass spectrometry. The data presented here were analyzed in two separate batch analyses.
280 Total procedural blanks for each batch were very similar, yielding [Re] = 15.1 and 17.2 pg/g,
281 [Os] = 150 and 193 fg/g, ¹⁸⁷Os/¹⁸⁸Os of 0.21 and 0.19 for the first and second batches,
282 respectively. In-house standard solutions of Re and Os were also run and were identical to
283 the long-term running average reported in Rooney et al. (2010 and references therein).

284 *3.4. Trace element concentrations*

285 The chemical composition of the samples was measured by the ALS Chemex
286 company of Vancouver, BC (Method ME-MS61). About 0.25 g of each sample was digested in
287 perchloric, nitric, hydrofluoric and hydrochloric acids. The residue was suspended in dilute
288 hydrochloric acid and analyzed by inductively coupled plasma-atomic emission
289 spectrometry (ICP-AES). Samples containing high concentrations of bismuth, mercury,
290 molybdenum, silver, or tungsten were diluted accordingly, then analyzed by inductively
291 coupled plasma-mass spectrometry (ICP-MS). Results were corrected for spectral

292 interelemental interferences. Detection limits were 0.01% by mass for most major
293 elements and <1 ppm for most trace elements.

294 To control for variations in sedimentation rate and terrigenous input, elemental
295 data were normalized to aluminum (Al), which was used as a tracer for the aluminosilicate
296 fraction of the sediment. For ease of comparison, elemental concentrations are reported as
297 enrichment factors (EFs), which were calculated as:

298

$$299 \quad EF_x = (\text{wt. \% X/wt. \% Al})_{\text{sample}} / (\text{wt. \% X/wt. \% Al})_{\text{AUCC}}$$

300

301 where X is the element in question and AUCC is Average Upper Continental Crust, as
302 determined by McLennan (2001)

303 *3.5. Petrographic analysis*

304 Petrographic study of a subset of samples was undertaken at the University of
305 Cincinnati and University of Kentucky. Standard petrographic thin sections were examined
306 at 20-400X magnification under reflected light using a Leitz Laborlux 12-Pol optical
307 microscope. Measurements were calibrated with a Zeiss micrometer. Polished sample
308 chips were photographed under oil-immersion lenses at varying magnifications.

309

310 **4. Results**

311 *4.1. Previous isotope results*

312 Organic carbon and nitrogen isotope datasets were generated for the Ranger Canyon
313 and Sulphur Mountain formations at the University of Washington's IsoLab facility. Both
314 records showed significant excursions in the lowermost SMF, with a positive peak in $\delta^{13}\text{C}_{\text{org}}$.

315 and a negative shift in $\delta^{15}\text{N}$. These data were previously published in Schoepfer et al., 2012
316 with a full description of methods and supplementary data table; the data are shown here in
317 Figs. 3-4 for easy comparison with sulfur and trace element results.

318 *4.2. Sulfur speciation*

319 Average S concentrations of bulk samples ($2.7\pm 1.1\%$, $n = 18$) and acidified samples
320 ($2.7\pm 1.2\%$, $n = 48$) (omitting sample OC24, the pyrite bed) are nearly identical. In addition,
321 the difference between $\delta^{34}\text{S}$ for bulk and acidified extracts of the same samples is small
322 ($0.8\pm 1.0\text{‰}$, $n = 18$). These observations suggest a negligible contribution of sulfates to the
323 total sulfur signal. All fully chemically processed (i.e., nitric-acid-treated) samples showed
324 sulfur levels below detection limits or too low (<0.03 wt%) to be significant, so little if any
325 barite is present. The bleaching step lowered sulfur levels significantly in all but one sample
326 (OC24, the pyrite bed), often to below detection limits. With such low S levels, $\delta^{34}\text{S}$ analysis
327 for bleached samples is much less precise, yet bleached $\delta^{34}\text{S}$ values are within $\pm 5\text{‰}$ of the
328 acidified-only $\delta^{34}\text{S}$ values, a much narrower spread than for the whole data set ($> 30\text{‰}$).
329 The relatively good agreement suggests the bleaching process was removing the majority of
330 a homogenous, labile sulfur pool that may be dominated by marcasite. This suggests the
331 acidified fraction represents an isotopically homogenous reduced sulfur pool (S_{reduced} , Fig. 3-
332 4), incorporating both sulfides and organic S.

333 [Insert Figure 3]

334 [Insert Figure 4]

335 *4.3. Sulfur concentration and isotopic profiles*

336 Sulfur isotopes vary markedly throughout the section, and several phases can be
337 distinguished (Figs. 3-4). In the Lower and Middle Permian, sediments are generally sulfur

338 poor (< 0.9 wt. %) and isotopically heavy, with $\delta^{34}\text{S}$ ranging from -5.4 to 7.2 ‰ (mean = 1.2
339 ‰, $\sigma = 4.0$), with the peak value 2.4 cm below the unconformity.

340 Heavy isotope values persist above the unconformity in the Late Permian lowermost
341 SMF, despite much higher total sulfur content. A continuous and rapid decline in $\delta^{34}\text{S}$ can be
342 observed between 1.5 and 29 cm (Fig. 4). This trend passes through the S wt. % maximum
343 in the pyrite bed at ~10 cm, which has an S content of 26.4% by weight. Bleached extract for
344 this sample yields S > 15wt% (IRMS signal saturated), and the S/Fe ratio is 1.13, nearly
345 identical to the S/Fe ratios in pyrite (1.14), which along with field observation confirms that
346 S in this layer is mostly in the form of pyrite. The declining isotope trend terminates in the
347 section's $\delta^{34}\text{S}$ minimum at 29 cm ($\delta^{34}\text{S} = -32.0$).

348 Above this minimum, sulfur concentrations and isotopes reach an equilibrium that
349 they maintain for the remainder of the section. Sulfur weight % is generally at a few percent
350 (mean = 2.3 %, $\sigma = 1.0$ %, Fig. 5) and $\delta^{34}\text{S}$ has a mean of -17.0‰ with a standard deviation
351 of 6.0 ‰, though there is a gradual and poorly defined trend toward less negative values.
352 There are also several minor positive excursions in $\delta^{34}\text{S}$. A small positive excursion in $\delta^{34}\text{S}$
353 (to -6 ‰) around 150 cm (sample OC145) is accompanied by a significant S concentration
354 peak (6.3%).

355 [Insert Table 1]

356 [Insert Figure 5]

357 *4.4. Petrographic data*

358 Petrographic study revealed characteristic relationships between pyrite, organic
359 matter, and calcite within the study samples. The basal beds (pre-LPME) of the SMF are
360 characterized by high concentrations of sulfides in the form of massive pyrite (Fig. 6A) and
361 large (5-50 μm), partially interlocking crystals of marcasite (Fig. 6B). These sulfides show

362 no relationship to primary sediment fabric; their quantity and form suggest that they may
363 be of hydrothermal origin. The beds immediately above the LPME are characterized by
364 abundant pyrite framboids, although these framboids tend to be larger (5-10 μm ; Fig. 6C)
365 than those formed in a euxinic water column ($<5 \mu\text{m}$; Wilkin et al., 1996; Wignall and
366 Newton, 1998). Many of the framboids show evidence of later diagenetic overgrowths (Fig.
367 6C-D), and blocky authigenic pyrite crystals are common (Fig. 6D-E). Higher in the section,
368 most pyrite occurs as clusters of framboidal or blocky crystals grown either within or on the
369 surface of large (50-500 μm) organic particles (e.g., Fig. 6G-H, L-N, P). Individual framboids
370 in these samples are often 10-40 μm in diameter (Fig. 6H, L, O) and hence distinctly larger
371 than syngenetic framboids (Wilkin et al., 1996). This form of pyrite is commonly associated
372 with large, blocky calcite crystals that have grown diagenetically within the sediment (e.g.,
373 Fig. 6G-J). Blocky to massive authigenic pyrite is sometimes found in the middle of large
374 carbonate-cemented areas without any obvious association with organic matter (e.g., Fig.
375 6F, K).

376 Several generations of carbonate are present in the sediments, including cements,
377 with isotopic measurements indicating a diagenetic origin (Schoepfer et al. 2012). This is
378 consistent with an increase in porewater alkalinity due to biological sulfate reduction and
379 the inference that most of pyrite in the section formed in the sediments after deposition.

380 [Insert Figure 6]

381 *4.5. Re-Os analysis*

382 The Re-Os data are presented in Table 2. Samples collected from 20 and 30 cm
383 above the Ranger Canyon/Sulphur Mountain unconformity are siltstone and did not contain
384 measureable Re and Os. Between 30 and 40 cm above the unconformity the siltstone grades
385 into black shale. The samples collected from this interval and to 1.6 m above the

386 unconformity are enriched in both Re and Os (~4 to 11 ppb; 78 to 157 ppt, respectively,
387 Table 2). The $^{187}\text{Re}/^{188}\text{Os}$ range between ~240 and 400, with the $^{187}\text{Os}/^{188}\text{Os}$ being
388 moderately radiogenic (~1.5 to 2.2). The initial $^{187}\text{Os}/^{188}\text{Os}$ (*IOs*) is calculated based on the
389 GTS 2008 age for the Permian-Triassic boundary, 252.2 Ma (Ogg et al., 2008; Shen S.Z. et al.,
390 2011) and the ^{187}Re decay constant of Smoliar et al. (1996). The calculated *IOs* values are
391 shown in Figure 4. Beginning 0.4 m above the unconformity (the LPME), the *IOs* values
392 become less radiogenic from 0.54 to 0.35. From the less radiogenic *IOs* at 0.6 m above the
393 unconformity to 1.5m (Permian-Triassic boundary) the *IOs* values become steadily more
394 radiogenic. At 1.6m the *IOs* trends to a less radiogenic value (~0.49).

395

396 [Insert Table 2]

397 4.6. Trace-element redox proxies

398 Certain trace elements tend to be authigenically enriched in sediments under
399 reducing conditions, including uranium (U), vanadium (V), and molybdenum (Mo) (Algeo
400 and Maynard, 2004; Tribovillard et al., 2006). The calculated Enrichment Factors of these
401 elements show similar patterns (Fig. 3), suggesting that they responded to a common redox
402 control. All three elements are highly enriched in the Johnston Canyon and Ranger Canyon
403 Formations, with Mo and U have EFs averaging >10x their concentrations in AUCC. These
404 values are probably indicative of sulfidic bottom water conditions throughout the
405 deposition of these units. All three elements show steep declines in the overlying SMF,
406 although varying somewhat in detail. V exhibits near-average values (EF ~1) throughout
407 the SMF. Mo and U remain modestly enriched (EF <10) in the SMF, with somewhat greater
408 enrichment in the basal 3 meters than higher in the formation. Relatively large EFs are
409 associated with the pyrite bed at ~10 cm, where Mo has an EF of 6.8 and U has an EF of 4.1.

410 As euxinic water conditions will often increase the particle reactivity of redox
411 sensitive metals, which particularly often from organo-metallic compounds (Tribovillard et
412 al., 2006), trace element concentrations were also normalized to TOC to control for the
413 effects of particle flux. While the actual ratios varied greatly between elements, all were
414 generally enriched in metals relative to TOC below the unconformity, but showed strong
415 peaks in the lowermost Sulphur Mountain Formation coincident with the pyrite bed, and
416 remained enriched throughout the latest Permian generally.

417 *4.7. Trace-element productivity proxies*

418 Several trace elements sensitive to changes in organic carbon export showed
419 significant fluctuations throughout the section. Cu, Ni, Cd, Zn, Ba, and P all show a nutrient-
420 like distribution in seawater and are mainly transferred to sediments through sinking
421 carbon flux. Of these, Cu, Cd, Ni, and Zn can be precipitated as sulfide minerals under euxinic
422 or sufficiently reducing conditions, and are likely to be preserved in sediments, whereas
423 barium is generally precipitated as barite, a sulfate, and may be released back into the water
424 column in a regime of sulfate reduction (Dehairs et al., 1980; Averyt and Paytan, 2004).
425 Phosphorous is mainly stored in sediments in organic matter or phosphate minerals. Under
426 reducing conditions, phosphate may be remineralized and re-enter the water column
427 (Meyer and Kump, 2008).

428 EFs were calculated for Ba, P, Cu, Ni, Cd and Zn. These elements all showed
429 significant enrichments above AUCC in the Johnston and Ranger Canyon Formations, with P
430 in particular being enriched above AUCC by at least a factor of 7.9 and by 2 orders of
431 magnitude in some samples. (2 samples collected 250 and 270 cm above the unconformity
432 saturated the ICP-AES Phosphorous peak. The maximum detectable concentration, 10^4
433 ppm, was used). Directly above the unconformity, P remains highly enriched with an EF of

434 7.6 while Zn is also extremely enriched, with a EF of 77.4, likely due to precipitation as a
435 sulfide in the pyrite bed. Cadmium, which also forms sulfide minerals, is also orders of
436 magnitude more enriched in this layer than the rest of the section (EF=581.4). The other
437 elements (Ba, Cu, Ni) drop off immediately above the unconformity to values similar to
438 AUCC, with Ba and Cu generally below an EF of 1, whereas Ni remains just above. After the
439 pyrite bed, Zn drops off to just below average EFs similar to Cu, Ni and Ba, while P remains
440 slightly more enriched with EFs ranging from 1 to 2.6 through the remainder of the section.
441 EFs are shown in logarithmic scale in Figure 3.

442

443 **5. Discussion**

444 *5.1. Effects of sediment mixing on geochemical signals*

445 Reworked conodonts above the unconformity indicate that some reworking of
446 subaerially exposed Middle Permian cherts or biological mixing of sediments has occurred.
447 This has the potential to exercise a controlling effect on geochemical signals, especially as
448 the resemblance of the Late Permian lowermost Sulphur Mountain to the Middle Permian
449 Ranger Canyon in many parameters (i.e. $\delta^{15}\text{N}$, $\delta^{34}\text{S}$, several trace elements) is central to
450 interpretation of the outcrop.

451 While the lowermost Sulphur Mountain shows elevated EFs in several trace elements
452 similar to or lower than the Ranger Canyon, this is after aluminum normalization, which
453 obscures the often higher absolute concentrations of many elements in the more
454 aluminosilicate rich Sulphur Mountain siltstones, for example the uppermost Ranger
455 Canyon has a measured Zn content of 19 ppm, whereas the lowermost Sulphur Mountain
456 has 3.85 ppt, In light of this orders of magnitude variation, it would be difficult for
457 reworking or upward mixing of Ranger Canyon material to exercise any major control on

458 the trace element content above the unconformity. The presence of discrete peaks in
459 geochemical signals above the unconformity, for example the presence of an ~1 cm thick
460 pyrite bed, the positive organic carbon isotope excursion reported in Schoepfer et al. 2012,
461 Cd content, which spikes to 39.9 ppm at ~10 cm above the unconformity despite an average
462 of 0.3 ppm in the Ranger Canyon Formation suggests that mixing over decimeter scales is
463 not a controlling factor on the inorganic geochemistry of this interval.

464 The major isotopic shifts observed in the basal Sulphur Mountain, however, occur
465 over smaller spatial scales, and without an independent tracer, control by mixing cannot be
466 completely ruled out. If mixing is predominantly responsible for the heavy $\delta^{15}\text{N}$ and $\delta^{34}\text{S}$
467 values seen in the basal Sulphur Mountain, then the environmental changes implied by
468 these isotopic shifts may have occurred at any time during the missing 5-7 My Lopingian
469 interval, and may have occurred more gradually than the preserved isotope signals would
470 suggest, however it is worth noting that Luo et al. (2011) observed abrupt nitrogen isotope
471 shifts of similar magnitude immediately preceding the main LPME at several Tethyan
472 sections in South China.

473 *5.2. Benthic redox conditions*

474 Sulfur isotopic compositions provide information about the processes of sulfide
475 formation and, hence, paleomarine redox conditions. Syngenetic framboidal pyrite, which
476 forms in the water column and thus from an unlimited sulfate reservoir, commonly exhibits
477 a large negative fractionation relative to source sulfate (Goldhaber and Kaplan, 1974;
478 Canfield and Thamdrup, 1994). In contrast, authigenic pyrite formed in the sediment
479 exhibits heavier $\delta^{34}\text{S}$ values; if all porewater sulfate is consumed within a closed diagenetic
480 system, the resulting sulfide will have an average $\delta^{34}\text{S}$ value equal to that of the source
481 sulfate. The degree of shift toward heavier $\delta^{34}\text{S}$ values depends on the timing of bacterial

482 sulfate reduction (BSR); early BSR in a relatively open system will produce more ^{34}S -
483 depleted sulfide than late BSR in a relatively closed system (Goldhaber and Kaplan, 1974;
484 Chambers and Trudinger, 1979).

485 Modern seawater sulfate has a $\delta^{34}\text{S}$ of +20‰, yielding sulfide $\delta^{34}\text{S}$ values of about -
486 30 to -40‰ in anoxic marine systems (Lyons, 1997; Wilkin and Arthur, 2001; Werne et al.,
487 2003). The isotopic composition of seawater sulfate varied between +10‰ and +30‰
488 during the PTB transition, possibly shifting toward heavier values in the Early Triassic
489 (Strauss, 1999; Newton and Wignall, 2004; Riccardi et al., 2006) although much variation
490 exists within and between sections (Luo et al., 2010; Song et al., in review). Given a BSR
491 fractionation of -30 to -60‰ (Habicht and Canfield, 2001), syngenetic pyrite in PTB
492 sections should yield isotopic compositions between -10 and -40‰ (cf. Nielsen and Shen,
493 2004; Algeo et al., 2008).

494 Reduced sulfur $\delta^{34}\text{S}$ in the Johnston Canyon and Ranger Canyon formations average
495 $1.0 \pm 4.2\text{‰}$ ($n = 8$). These relatively heavy values suggest precipitation in a closed system,
496 consistent with a diagenetic origin for pyrite in this interval. The evidence of quantitative
497 reduction of porewater sulfate in this interval despite the low organic content suggests poor
498 postdepositional preservation may be responsible for the lack of preserved carbon despite
499 other evidence of high productivity (see below).

500 The Johnston Canyon and Ranger Canyon formations contain high concentrations of
501 V, U, and Mo. Coeval enrichments in Mo, U, and V (Fig. 3), suggest the presence of anoxic
502 water in the Early and Middle Permian. That these enrichments are found in bioturbated
503 sediments supporting an abundant (if homogenous) benthic fauna suggests that bottom
504 water was oxygenated, and trace element enrichments may have resulted from a water
505 column oxygen minimum impinging on the bottom. Coastal upwelling zones often

506 dynamically maintain an oxygen minimum zone (OMZ) in the mid-water column through
507 the decomposition of sinking organic material. When the sinking flux of carbon exceeds the
508 supply of oxygen, other electron acceptors are often used in decomposition, including
509 nitrate and sulfate, with upwelling zones being among the primary zones of denitrification
510 in the modern ocean (Seitzinger et al., 2006).

511 Nitrogen isotopes (Schoepfer et al. 2012) from Opal Creek suggest the Middle
512 Permian upwelling system supported a mid-water oxygen minimum where denitrification
513 and sulfate reduction were occurring. Denitrification has a substantial fractionation effect,
514 and when it does not occur quantitatively, enriches the residual marine nitrate pool. $\delta^{15}\text{N}$ in
515 the Ranger Canyon is consistently high, spiking to values above 9 ‰, values typically seen
516 in modern environments where denitrification occurs in a strong water column oxygen
517 minimum (Algeo et al. 2008).

518 The lower ~4 m of the SMF contains the largest concentrations of small framboids,
519 and these beds exhibit some of the most ^{34}S -depleted values—a pattern consistent with
520 pyrite mainly of syngenetic origin. $\delta^{34}\text{S}_{\text{reduced}}$ in the SMF averages $-16.5 \pm 6.8\text{‰}$ ($n = 47$),
521 which is distinctly higher than the isotope compositions expected for syngenetic framboidal
522 pyrite. These compositions are consistent with pyrite formation in the diagenetic
523 environment under quasi-open conditions, or with diagenetic overgrowths on syngenetic
524 pyrite under somewhat more closed conditions.

525 TOC-S relationships are consistent with predominantly suboxic conditions during
526 deposition of the SMF. Most samples fall along the oxic-suboxic (“normal marine”) trend of
527 Berner and Raiswell (1983; Fig. 5). A relative handful of samples, mostly in the lower ~4 m
528 of the SMF, exhibit S concentrations considerably above those expected for oxic-suboxic
529 facies; other evidence supports this interval having been deposited under primarily euxinic

530 conditions (see below). Three horizons higher in the SMF (at 4.7, 12.8, and 20.4 m) also
531 exhibit significant S enrichment above the oxic-suboxic trendline (Fig. 5). Unlike samples
532 from the base of the formation, however, these samples are associated with markedly
533 heavier $\delta^{34}\text{S}$ values—an indication that the pyrite concentrations in these samples formed
534 in the sediment and not in the water column, which is supported by petrographic data. The
535 extinction of bioturbating benthic communities may have contribute to these intervals of
536 near-quantitative sulfate reduction in the sediments (Shen Y. et al, 2011).

537 Secondary pyrite overgrowths are common in many framboidal layers, and
538 framboid sizes are larger than those associated with syngenetic formation, especially above
539 the basal 4 m of the SMF (Fig. 5). The observation that most framboids grew in clusters
540 within or on the surface of large organic particles, rather than as isolated crystal aggregates,
541 indicates a strong association of BSR with such organic substrates. The distribution of
542 calcite within the samples, mostly as blocky diagenetic crystals adjacent to organic particles,
543 or as “halos” enclosing these particles, is an indication that BSR within the sediment
544 produced local increases in alkalinity that stimulated secondary carbonate precipitation
545 (Berner, 1984).

546 Trace element proxies support these redox interpretations for the SMF. Mo, U, and V
547 remain slightly enriched throughout the SMF, suggesting low oxygen conditions persisted,
548 which along with the dearth of benthic biota, lack of bioturbation, and increased organic
549 preservation indicate that the system transitioned to a bottom water oxygen minimum. Mo,
550 which forms highly particle reactive oxythiomolybdates under euxinic conditions
551 (Tribovillard et al., 2006), is particularly enriched in the lowermost 4 meters of the SMF,
552 suggesting free sulfide in the water column during this interval.

553 Except for the basal ~50 cm of this formation, $\delta^{15}\text{N}$ is low and stable (Schoepfer et
554 al. 2012, Figs. 3-4), providing no evidence of water-column denitrification as might be
555 expected if suboxic conditions were due to high sinking fluxes of organic matter (e.g.,
556 Jenkyns et al., 2001; Algeo et al., 2008). Although the SMF consists predominantly of black
557 shale (Munsell color), TOC values are only moderate (0.2-3.0%) yet locally higher than
558 expected for oxic-suboxic facies (Algeo and Maynard, 2004). One explanation for this
559 pattern may be a marked reduction in bioturbation intensity during the Early Triassic, as a
560 consequence of the loss of most benthic infauna during the LPME (Twitchett and Wignall,
561 1996) and transition from a mid-water to bottom water oxygen minimum. Reduced
562 bioturbation would severely constrain ventilation of the sediment and allow greater
563 quantities of organic matter to be preserved for a given flux of organic matter to the
564 sediment-water interface. Thus, multiple lines of evidence confirm that the basal ~4 m of
565 the SMF coincided with a euxinic event, and that the remainder of the formation was
566 deposited under oxic to suboxic conditions.

567

568 *5.3. Marine circulation and productivity*

569

570 The consistent enrichments in Ba, Cu, Ni, Zn, P, and Cd in the Johnson Canyon and
571 Ranger Canyon formations are here interpreted as representing a highly productive,
572 eutrophic marine system, despite the generally low organic carbon content of these units,
573 which may result from poor preservation in well-ventilated bottom water. The presence of
574 phosphate nodules in the Johnson Canyon and minor phosphatic material in the Ranger
575 Canyon is consistent with a high-productivity environment, and the dominant fauna of the
576 Ranger Canyon, rock-forming abundances of filter-feeding sponges, suggests a vigorous
577 export of organic detritus from the photic zone.

578 High-abundance but low-diversity conodont assemblages have been suggested to
579 indicate eutrophic conditions (Brasier, 1995), because fast-breeding species capable of
580 rapidly exploiting the available resources tend to dominate the environment. The conodont
581 assemblage of the Ranger Canyon Formation is dominated by high abundances of the genus
582 *Mesogondolella*, which had a circumboreal distribution in the Middle Permian and is
583 therefore also indicative of cold-water conditions (Henderson, 1997; Mei and Henderson,
584 2001).

585 The biota of the Johnston Canyon and Ranger Canyon formations is consistent with a
586 high-nutrient, cold-water environment. Replacement of this prolific carbonate factory by a
587 siliceous sponge-dominated biota occurred between the Artinskian (late Early Permian)
588 and Guadalupian (late Middle Permian) as a consequence of climatic cooling and an increase
589 in nutrient levels (Beauchamp and Henderson, 1994; Reid et al., 2007; Bensing et al., 2008).
590 The high concentration of sponge-derived chert in Upper Permian beds has been
591 interpreted as evidence of vigorous thermohaline circulation and nutrient upwelling in the
592 Panthalassic Ocean, and its absence in Lower Triassic beds as evidence of sluggish
593 circulation and nutrient-poor surface waters (Beauchamp, 1994; Beauchamp and
594 Desrochers, 1997; Beauchamp and Baud, 2002). Hyalosponge facies, indicative of cold,
595 deepwater conditions, are widespread along the western margin of Pangaea during the
596 Early to Middle Permian, and have been used to infer the presence of coastal upwelling
597 down to subtropical latitudes, as well as cold northern polar currents driven by sea ice in
598 contact with the ocean (Beauchamp and Baud, 2002). Upwelling at subtropical latitudes
599 along western Pangaea is also inferred by Ziegler et al. (1998) from sedimentological
600 evidence, and has been modeled under a range of atmospheric conditions by Winguth and
601 Maier-Reimer (2005).

602 The locality was erosive throughout most of the Lopinigan lowstand, but when

603 sedimentation resumed in the Changshingian the depositional environment resembled that
604 of the Ranger Canyon in many ways. Silica sponges were still the abundant, predominant
605 benthic fauna, although sponge spicules were increasingly diluted with siliciclastic
606 sediments. Nitrogen and sulfur isotopes remain highly enriched, suggesting that an anoxic
607 zone, fed by sinking organic matter and ultimately by upwelling nutrients, persisted into the
608 latest Permian. Sediments remain bioturbated in this interval, indicating that the sea floor
609 was relatively well oxygenated, with the oxygen minimum occurring in the water column.

610 ~ 10 cm above the unconformity is a continuous bed containing >26% sulfur by
611 weight, mostly in the form of pyrite, suggestive of highly euxinic water conditions, which is
612 supported by orders-of-magnitude enrichments in the S/TOC and Mo/TOC ratios. This ~1
613 cm thick pyrite bed corresponds with a discrete, ~ 8 ‰ spike in organic carbon isotopes
614 that has been interpreted as transient increase in productivity (Schoepfer et al. 2012) which
615 may have driven the system toward euxinia, Enrichments in P, Zn, and Cd are also
616 suggestive of enhanced productivity in this interval: while Zn and Cd are expected to
617 complex in the sulfide sediments, P should be unaffected by euxinia and record
618 productivity, whereas minor depletions in Barium relative to AUCC may reflect the difficulty
619 of preserving barite (BaSO₄) under conditions of euxinia.

620 The trace element enrichments suggest that this euxinic interval may have been
621 fueled by transient increased productivity. Numerous workers have suggested a substantial
622 increase in continental weathering and erosion in the latest Permian (Ward et al. 2000;
623 Sephton et al., 2005; Algeo and Twitchett, 2010; Algeo et al., 2011). This would have carried
624 an increased complement of terrigenous nutrients, such as phosphate and iron, and may
625 have pushed organic productivity above its already-high levels, driving the system toward
626 euxinia.

627 This hypothesis receives some support from the IOs data (Fig. 4). The primary
628 feature of this data set is a shift from relatively non-radiogenic values (0.35-0.40) around
629 the LPME boundary toward more radiogenic values (~ 0.45 to 0.62) between 60 and 150 cm
630 above the unconformity. This suggests that the primary control of the Os composition in
631 seawater during this interval (Os_{SW}) is the influx of radiogenic detritus from the cratonic
632 continental crust (Cohen et al., 1999; Peucker-Ehrenbrink and Ravizza, 2000), into the
633 ocean, beginning roughly with the onset of the euxinic pulse. Unradiogenic Os values seen
634 prior to this shift may be related to the input of Os associated with the eruption of the
635 Siberian Traps, which would be expected to possess mantle-like Os isotope signatures
636 (~ 0.13 -0.15; Horan et al., 1995). Values for Os_{SW} during the late Permian Arctic black shale
637 of the Mid-Norwegian shelf yield IOs values of ~ 0.6 (Georgiev et al., 2011), which is similar
638 to the IOs value of the extinction interval (~ 0.55) at Opal Creek, suggesting these
639 continental weathering effects may have been widespread in the global ocean. However, in
640 stark contrast to the Arctic black shales the $^{187}Re/^{188}Os$ values of the Opal Creek P-T section
641 are typical of most marine organic-rich sedimentary rocks (269-394; Table 2) and therefore
642 indicate that the conditions that permitted elevated $^{187}Re/^{188}Os$ values (~ 2000 to 6000) in
643 the Arctic is regional rather than global as suggested by Georgiev et al. (2011).

644 The end of this intensely euxinic interval corresponds with a drop off in the
645 previously enriched isotopes of nitrogen and sulfur. Nitrogen isotopes fall toward low
646 values of approximately 2-3 ‰, lower than average modern marine nitrate and likely
647 reflecting the influence of nitrogen fixation from the atmosphere in a nutrient limited
648 environment. These low nitrogen values persist with remarkable uniformity throughout the
649 remainder of the SMF, reflecting a new nutrient cycling regime in which water
650 denitrification, fueled by respiration of sinking organic matter, no longer occurred on a
651 significant scale. These changes probably reflect the collapse of vigorous coastal upwelling

652 and the concomitant high organic productivity that fueled anaerobic respiration in the
653 water column.

654 This is supported by the collapse of all productivity-sensitive trace elements to values
655 very near AUCC throughout the remainder of the section (excepting P and Cd, which remain
656 slightly enriched), and by the disappearance of silica-producing sponges 30 cm above the
657 pyrite bed, in what we interpret as corresponding to the main Permian-Triassic marine
658 extinction. Productivity in the subsequent SMF likely reflected typical values for
659 oligotrophic marine settings, rather than those of continental margin upwelling systems,
660 and may have been limited by nitrogen availability, with diazotrophic prokaryotes likely
661 playing an important role. The cessation of upwelling may have been the result of warming
662 following the eruption of the Siberian traps, with a consequent flattening of global thermal
663 gradients and slowing of ocean and atmospheric circulation, with both euxinia and
664 increasing nutrient limitation playing a role in bringing about benthic extinctions in this
665 ecosystem.

666

667 *5.4. Integrated model*

668

669 Here we present evidence that a coastal upwelling system, a major mode of
670 Panthalassic ocean circulation that had persisted for tens of millions of years, abruptly
671 terminated preceding the Permian-Triassic boundary, shifting to a relatively unproductive,
672 nutrient limited system with suboxic to dysoxic bottom water. However, this transition was
673 preceded by an interval of intense euxinia, corresponding to the pyrite bed and preceding
674 the main pulse of marine extinction. This likely resulted from a transient intensification of
675 marine productivity, which may have been caused by a combination of warming climate with
676 residual upwelling of nutrient rich waters, before the circulation slowed dramatically (Fig.

677 7). Higher temperatures could allow for faster algal growth and nutrient cycling, and
678 ultimately faster and more explosive productivity. Alternatively, this pulse of high
679 productivity may have been a global ocean scale event. The end-Permian has been linked to
680 a change in fluvial morphology patterns, and a die off of vegetation combined with a
681 warmer and more acidic hydrosphere following the eruption of the Siberian traps may have
682 led to faster continental weathering and increased nutrient input into the oceans. The
683 immaturity of siliciclastic sediments in the lowermost Sulphur Mountain Formation suggest
684 rapid weathering was occurring. Thin pyrite beds have been found in latest Permian marine
685 sediments along western Pangaea up into the Canadian arctic (Grasby and Beauchamp,
686 2009). Increased nutrient inputs may have led to one or several pulses of high productivity
687 in latest Permian coastal oceans, driving environments to extinction-causing levels of anoxia
688 and euxinia despite the long-term trend toward slower circulation and nutrient cycling into
689 the Triassic.

690

691 [Insert Figure 7]

692

693

694 **6. Conclusions**

695

696 The Permian at Opal Creek records an active cold-water coastal upwelling zone that
697 supported high productivity and a mid-water oxygen minimum zone where anaerobic
698 respiration occurred. This system persisted for tens of millions of years only to abruptly
699 terminate preceding the end-Permian extinction. Despite the Panthalassic Ocean's
700 potentially dominant role in Permian biogeochemical cycling, there have been few ways to
701 test the impact of the end-Permian crisis on circulation and ventilation. Here we present

702 isotopic and trace element evidence for the cessation of coastal upwelling, a modality of
703 oceanic circulation that leaves distinctive isotopic signatures. Despite the apparent
704 slowdown of circulation at the end Permian, a major, discrete euxinic pulse preceded the
705 extinction, and corresponds to a number of geochemical indicators of high productivity. A
706 transient pulse of increased carbon export may have been necessary to drive coastal oceans
707 toward the euxinia necessary for mass extinction.

708

709 **Acknowledgments**

710

711 The authors would like to acknowledge the generous support provided by The United
712 States National Science Foundation (NSF) and NASA Astrobiology Institute (NAI) University
713 of Washington node as well as the Natural Sciences and Engineering Research Council
714 (NSERC) of Canada, and the University of Washington Earth and Space Sciences
715 Department. Our gratitude goes out to the staff of the University of Washington IsoLab
716 facility for their extensive help with our isotopic analyses. TJA thanks the National Science
717 Foundation for support (EAR-0618003, EAR-0745574, and EAR-1053449). YS would like to
718 thank the National Natural Science Foundation of China for their support (41025011).

719

720

721 **References**

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1141 **Figure and Table captions**

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1143 **Figure 1.** (A) Late Permian paleogeography, showing the major ocean basins and location
1144 of the Opal Creek section. Map generated by PaleoDB.org, Alroy 2010. (B) Map of Canada
1145 showing the location of the Kananaskis Valley in southwestern Alberta. (C) Map showing
1146 the location of the Opal Creek section in the Kananaskis Valley.

1147 **Figure 2.** (A) Photograph of Opal Creek outcrop from approach. Note cliff-forming, silicified
1148 Ranger Canyon and less resistant Sulphur Mountain silts exposed in hillside. The cliff
1149 face corresponds to the unconformity. (B) Close up of Permian-Triassic boundary.
1150 Pyrite bed weathers to reddish in outcrop. RC = Ranger Canyon Formation, SM =
1151 Sulphur Mountain Formation, SB = unconformable Sequence Boundary, Py. = Pyrite
1152 layer, Ext. = main extinction horizon.

1153 **Figure 3.** Stratigraphic column with lithostratigraphy as well as isotopic and trace element
1154 records for the Opal Creek section. Heavy red line corresponds to the biostratigraphic
1155 PTB, the thin red line corresponds to the main marine extinction event. The number of
1156 samples analyzed for each proxy is as follows: $\delta^{13}\text{C}_{\text{org}}$, n = 191, $\delta^{15}\text{N}$, n = 58, $\delta^{34}\text{S}$, n = 56
1157 acidified (n = 7 acidified+bleached), %S, n = 55, TOC, n = 185, All trace element proxies, n
1158 = 29.

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1160 **Figure 4.** Expanded stratigraphic column for 4.5-m-thick interval of the Opal Creek section
1161 bracketing the unconformity and PTB, with isotope measurements and calculated initial
1162 osmium isotope ratios (IOs, see Re-Os Analysis in Results section). Solid red line
1163 indicates the biostratigraphic PTB, the dashed red line indicates the main extinction
1164 event. *M. bitteri* zone overlaps with *M. rosenkrantzi* in uppermost 5-10 cm of Ranger
1165 Canyon Formation. The number of measurements shown for each proxy is as follows:
1166 $\delta^{13}\text{C}_{\text{org}}$, n = 149, $\delta^{15}\text{N}$, n = 41, $\delta^{34}\text{S}$, n = 36 acidified (n = 2 acidified+bleached), IOs, n =
1167 14.

1168 **Figure 5.** (A) TOC versus total S. The majority of samples falls along the oxic-suboxic trend
1169 of Berner and Raiswell (1983). A subset of samples (n ~12) plots distinctly above the
1170 oxic-suboxic trend; they represent the basal 4 m of the SMF plus a few pyritic layers
1171 higher in the section. (B) Total S versus pyrite $\delta^{34}\text{S}$. Note that high-S samples are low in
1172 TOC and enriched in ^{34}S . Samples shown in this figure are a separate set processed at the
1173 U. of Cincinnati.

1174 **Figure 6.** Reflected light petrographic photomicrographs of SMF: (A) Massive pyrite in
1175 calcareous silicate matrix. (B) Massive marcasite in silicate matrix. (C) Medium-sized (8-
1176 12 μm) pyrite framboids with partial diagenetic overgrowths. (D) Coarse euhedral and

1177 irregular diagenetic pyrite masses. (E) Large (35 μm) pyrite framboid with diagenetic
1178 overgrowths. (F) Fine (3-10 μm) diagenetic pyrite crystals in a large organic particle. (G)
1179 Fine diagenetic pyrite within a diagenetic calcite crystal. (H) Variably sized (4-20 μm)
1180 pyrite framboids along an organic-rich lamina. (J) Coarse diagenetic pyrite associated
1181 with diagenetic carbonate crystals. (K) Coarse diagenetic pyrite within a carbonate
1182 nodule. (L) Variably sized (10-40 μm) pyrite framboids along an organic-rich lamina. (M)
1183 Organic particle with diagenetic pyrite overgrowths (lower left) and blocky diagenetic
1184 pyrite (upper right). (N) Fine (4-12 μm) pyrite framboids along an organic-rich lamina.
1185 (O) Coarse (20-40 μm) pyrite framboids with diagenetic overgrowths on larger
1186 framboid. (P) Fine (3-8 μm) diagenetic pyrite crystals in an organic particle.

1187 **Figure 7.** Integrated model of environmental conditions on the subtropical eastern
1188 Panthalassic margin (A) During the Early and Middle Permian (B) During the main latest
1189 Permian extinction interval (C) During the earliest Triassic

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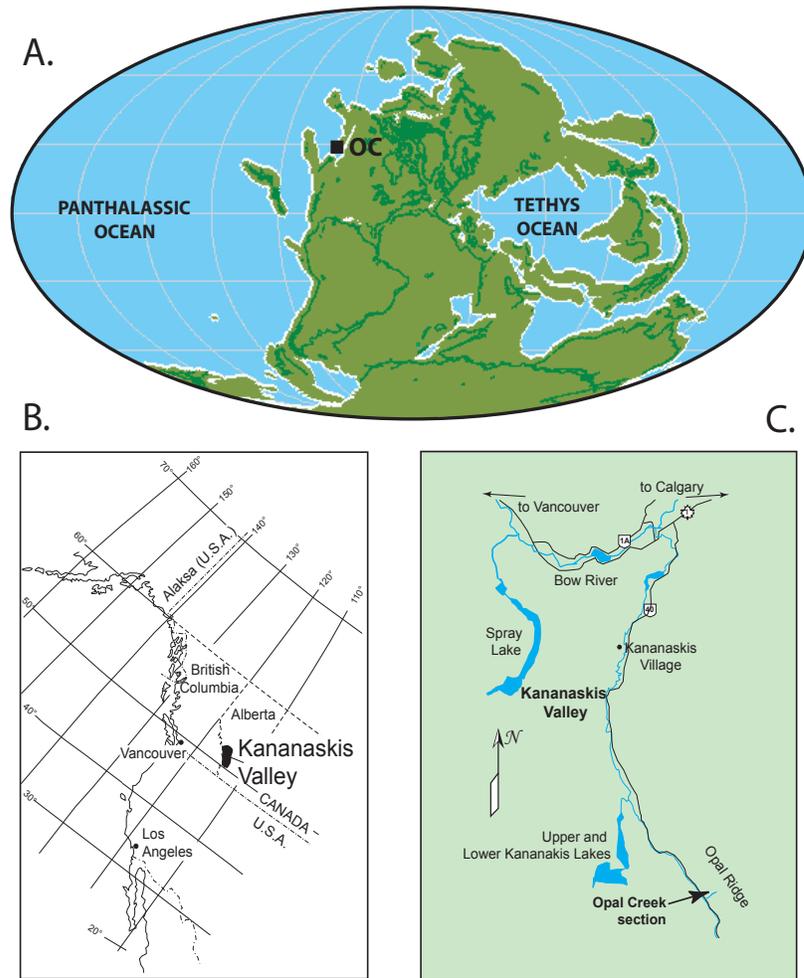
1191 **Table 1.** Concentrations and enrichment factors of some major and trace elements

1192 †AUGC = Average Upper Continental Crust (McLennan 2001)

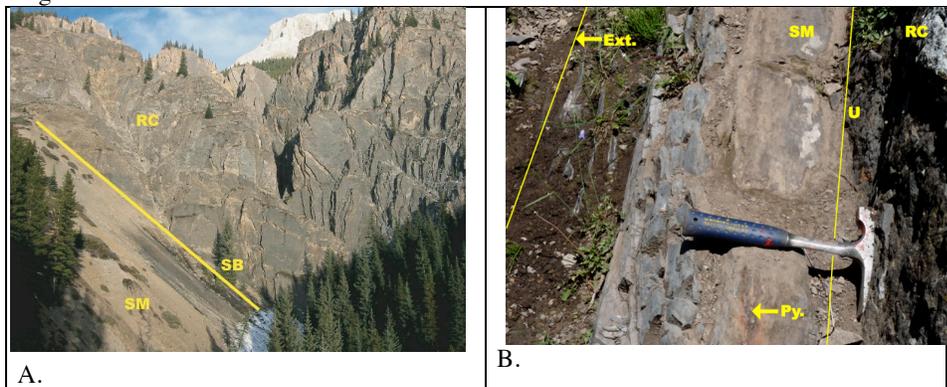
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1194 **Table 2.** Re and Os concentrations and isotopes. *IOs* uncertainties were less than or equal to
1195 .004.

1196 †Initial osmium compositions (*IOs*) calculated for an age of 252.2 Ma.

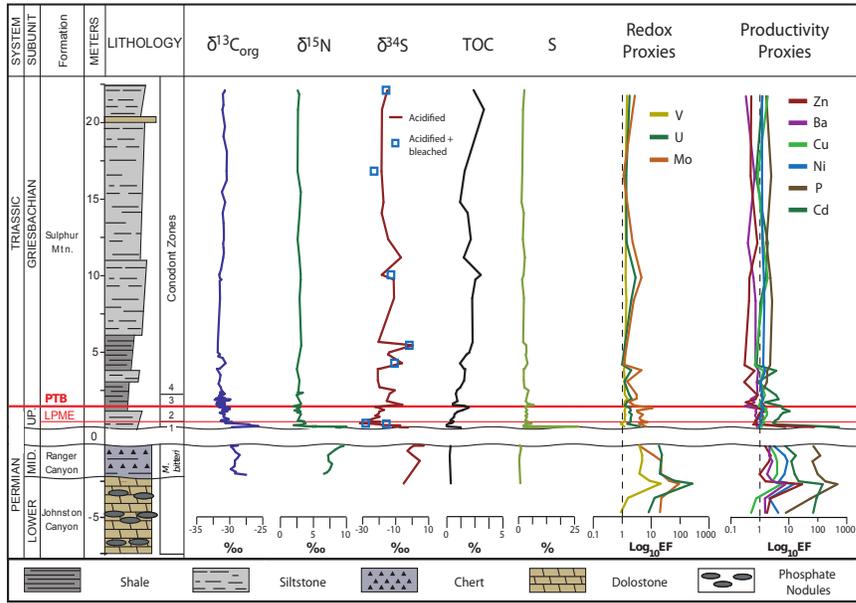


1186 Figure 2



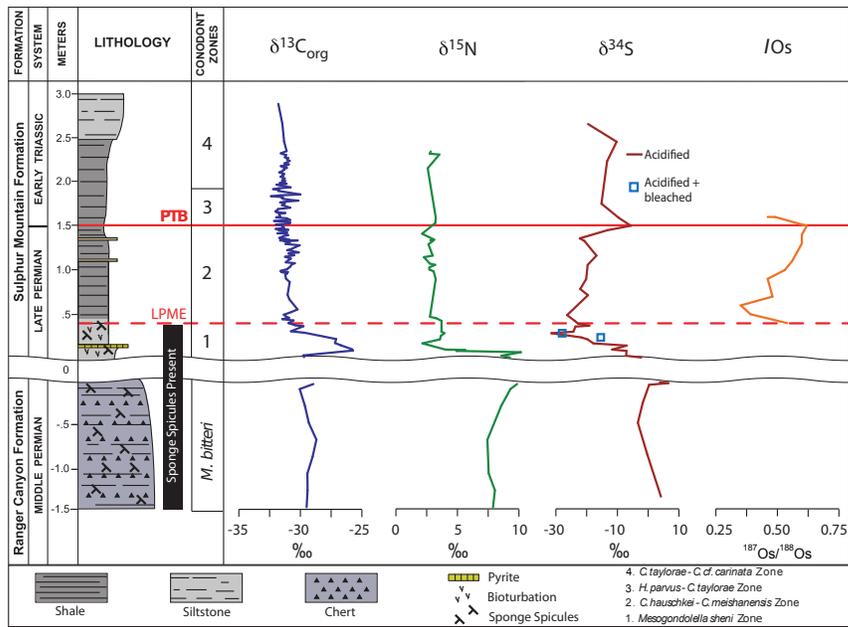
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1188 Figure 3



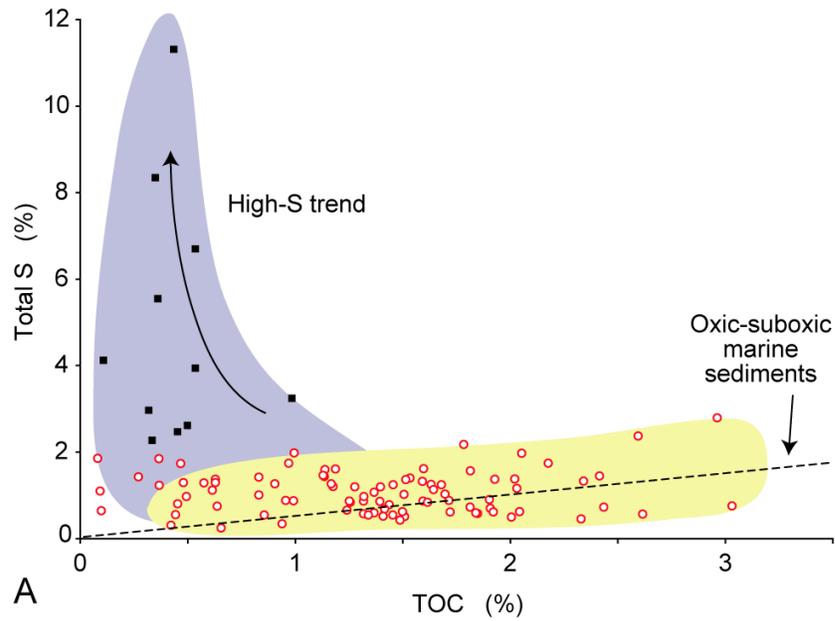
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1190 Figure 4.
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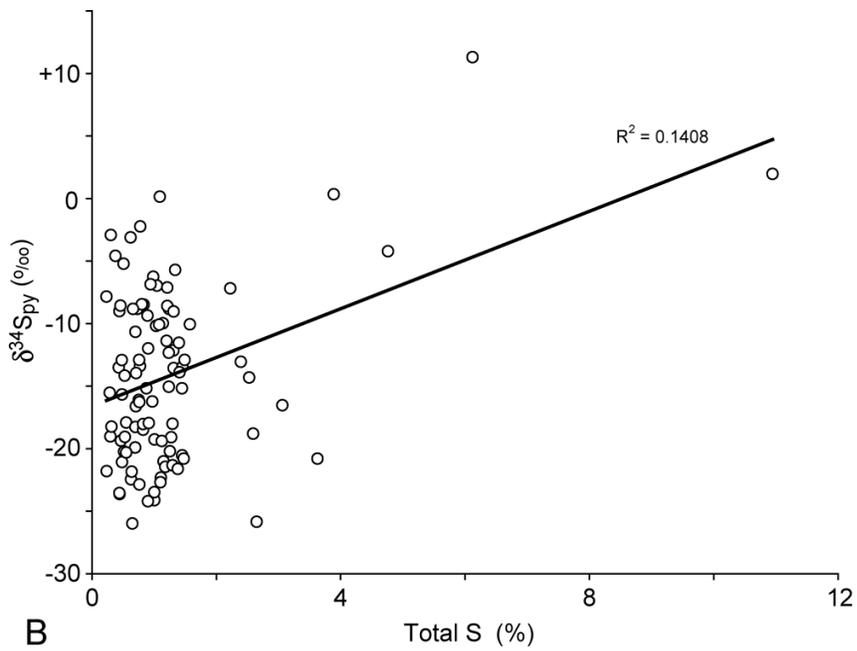


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1214 Figure 5.
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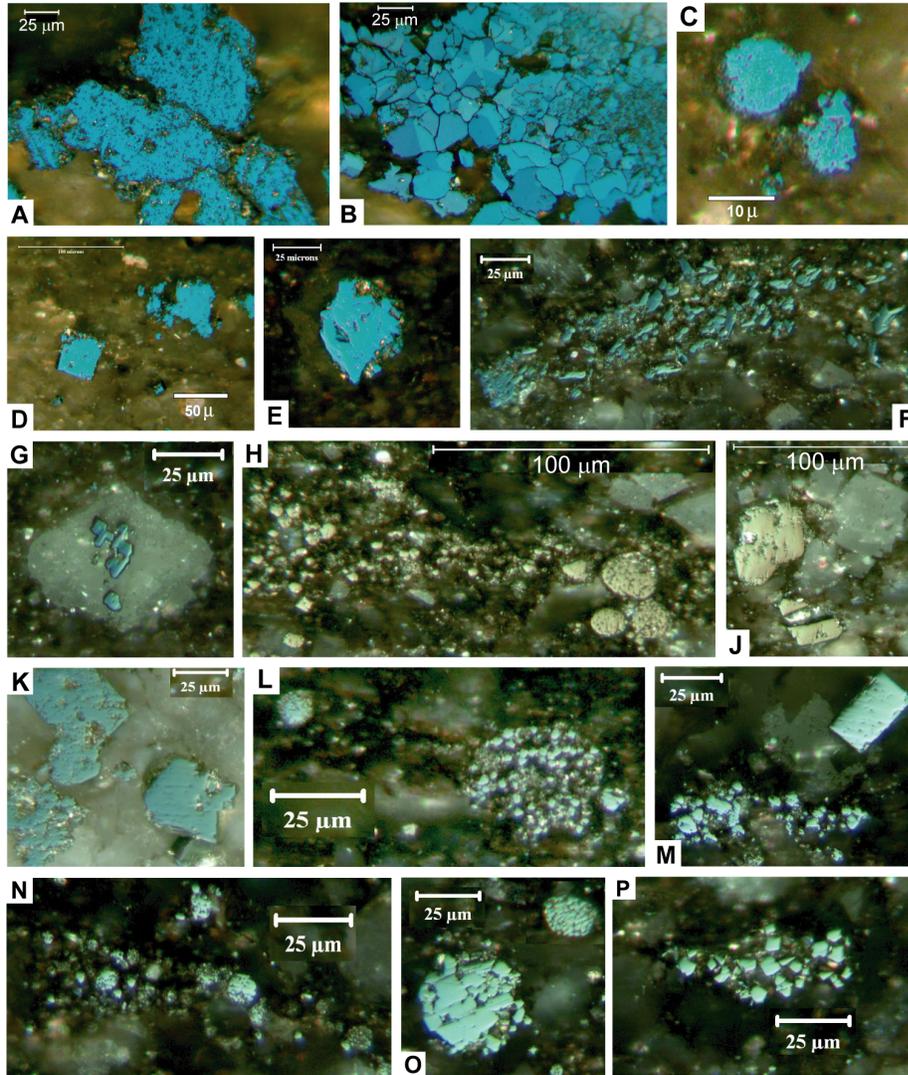
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1217 Figure 6.



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Figure 7.

