In situ oxygen-isotope, major-, and trace-element constraints on the metasomatic modification and crustal origin of a diamondiferous eclogite

from Roberts Victor, Kaapvaal Craton

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1 Abstract (430 words)

2 A subducted oceanic crustal origin for most eclogite xenoliths in kimberlites has long been a cornerstone of tectonic models for craton development. However, eclogite xenoliths often have 3 protracted and complex histories involving multiple metasomatic events that could overprint 4 5 some of the key geochemical indicators typically taken as evidence of a subducted origin (e.g. garnet δ^{18} O-values and mineral 87 Sr/ 86 Sr compositions). To assess the potential for disturbance 6 of oxygen isotopic compositions in mantle eclogites via diamond-forming and other possible 7 metasomatic fluids, we have conducted a multi-technique *in situ* study of a diamondiferous 8 eclogite xenolith from the Roberts Victor kimberlite, S. Africa. Using SIMS we provide the first 9 texturally-controlled *in situ* measurements of δ^{18} O-values in eclogitic garnet in close proximity 10 to diamond. 11

Garnet and clinopyroxene modal proportions are heterogeneous in the xenolith and garnet 12 compositions vary from Mg# = 75.8-79.2; grossular proportions = 8.05-10.14 mol. %, and 13 omphacitic pyroxene has Jd_{13-24} and Mg# = 86.6-90.0. Rare earth element patterns of minerals 14 across the xenolith, including grains close to diamond, are typical LREE-depleted garnets and 15 markedly LREE-enriched pyroxenes. These silicate minerals also record detectable intra- and 16 inter-grain LREE abundance variations. Clinopyroxenes of the studied xenoliths show HFSE 17 and Sr abundance variations that are decoupled from LREE contents and major-element 18 variations. 19

20 Mineralogical constraints and bulk-rock reconstructions indicate that the studied sample
21 likely experienced selective incompatible element enrichment during small-volume (<<0.03 wt.
22 %) infiltration of metasomatic fluid(s) potentially linked to ancient diamond evolution. Intra-

grain major-element, LREE and HFSE variations in clinopyroxene resulted from late-stage metasomatism. Oxygen isotope compositions in garnet are decoupled from all major- and traceelement variations, with garnet δ^{18} O-values being uniform across the xenolith in a wide variety of textural settings. Garnet δ^{18} O-values of 6.5 ± 0.2 ‰ are higher than the mean (5.19 ± 0.26 ‰) of the mantle garnet range (4.8-5.5 ‰).

Modelling of the buffering effect of mantle peridotite on CO₂-rich and H₂O-rich 28 metasomatic fluids at temperatures within the diamond stability field indicates that the likelihood 29 of a metasomatic fluid with exotic oxygen isotopic composition arriving at a mantle eclogite 30 body with its isotopic composition unmodified, after percolative flow through dominantly 31 peridotitic mantle at great depth, is very low. As we find no evidence of metasomatically induced 32 garnet oxygen isotope variations in the studied diamondiferous eclogite xenolith we conclude 33 that the most likely origin for the elevated garnet δ^{18} O-values is via inheritance from a crustal 34 35 protolith altered at relatively low temperatures. These results have broader relevance and support the hypothesis of a low-pressure protolith for mantle eclogite xenoliths, demonstrating 36 the robust nature of garnet oxygen isotope compositions – even in diamond-bearing eclogites. 37

Keywords: Eclogite, Achaean Crust/Mantle, Subduction, Oxygen Isotopes, Diamond.

39 Main Text (5,727)

40 **1. Introduction**

41 The origins of kimberlite-borne eclogite xenoliths that are erupted through cratons have been debated for the last 30 years or more (e.g., MacGregor and Manton, 1986; Hatton and Gurney, 42 1987; Jacob et al., 1994; Beard et al., 1996; Jacob and Foley, 1999; Schmidberger et al., 2007; 43 Viljoen et al., 2005; Aulbach et al., 2007; Gurney et al., 2010; Huang et al., 2012; Shu et al., 44 2014). Eclogites are volumetrically dominated by garnet and omphacitic pyroxene, and are 45 generally considered as meta-igneous rocks having broadly basaltic bulk-rock compositions. 46 Eclogites sampled as xenoliths preserve mineralogical and cryptic geochemical records of 47 complex and protracted lithospheric mantle histories (e.g., Heaman et al., 2002, 2006). Broad 48 49 analogies have been drawn between mantle eclogite xenoliths sampled from depth and crustal eclogites that generally occur in orogenic settings, but which have distinct metamorphic histories 50 (e.g., Coleman et al., 1965; Nadaeu et al., 1993; Baker et al., 1997; Zack et al., 2002; Zheng et 51 al., 2003; Konrad-Schmolke et al., 2008). Critically, the distribution of oxygen isotope 52 compositions of eclogite xenolith garnets, with δ^{18} O-values ranging significantly above and 53 below canonical mantle values, have drawn analogies with bulk-rock compositions reported for 54 shallow-level ophiolite and mid-ocean ridge basalt (MORB) sequences altered by fluids at 55 relatively low-temperatures. As such, these variations are often cited as evidence supporting a 56 recycled crustal origin (e.g., Jagoutz et al., 1984; Neal and Taylor, 1990; Neal et al., 1990; 57 Snyder et al., 1995, 1997; Schulze et al., 2000; Barth et al., 2001; Jacob et al., 2003; Spetsius et 58 al., 2008; Riches et al., 2010; Tappe et al., 2011; Carmody et al., 2013; Pernet-Fisher et al., 59 60 2014). Prior studies of oxygen isotope compositions of eclogite garnets have, however, largely employed multi-grain conventional-fluorination methods and laser-fluorination approaches on 61

mg-sized garnet fragments, which will result in compositional averages of larger garnet volumes. 62 In addition, garnet-clinopyroxene modal banding and wide-spread evidence for cryptic 63 metasomatism linked to the passage of incompatible-element-rich fluids \pm diamond occurrences 64 in eclogite xenoliths (e.g., Taylor et al., 1996, 2000; Ishikawa et al., 2008a-b; Spetsius and 65 Taylor, 2008; Liu et al., 2009; Smart et al., 2009) lead some scientists to question the primary 66 nature of garnet δ^{18} O-compositions (e.g., Huang *et al.*, 2014). The potential for eclogite garnet 67 oxygen isotope compositions being of metasomatic origin has profound implications for the 68 hypothesis of a subducted crustal origin for eclogite xenolith protoliths (e.g., Helmstaedt and 69 Doig; 1975; Jacob, 2004) and warrants further study. 70

To assess intra-sample garnet oxygen isotope homogeneity and thereby test the validity of 71 conclusions drawn from conventional- and laser-fluorination studies, which are considered 72 fundamental evidence supporting the subduction origin of mantle eclogites, we selected a 73 74 diamondiferous eclogite xenolith from the Roberts Victor kimberlite (Fig. 1) and utilised new generation secondary-ion mass spectrometry (SIMS) to obtain highly precise in situ garnet 75 oxygen isotope data for texturally constrained grains at small spatial scales and low total 76 volumes (15 µm spot diameter of 1-2 µm depth; << 5 ng; e.g., Page *et al.*, 2010; Ickert and Stern, 77 2013). This sample exhibited significant variation in diamond content, silicate mineral texture 78 and mineral chemistry, enabling inter- and intra-grain garnet oxygen isotope variation to be 79 assessed in the context of macro- and grain-scale textural and mineral chemical variations. 80 Significantly, our study includes the first report of *in situ* δ^{18} O-compositions of garnet grains 81 adjacent to diamond in eclogite, allowing us to investigate the potential effects of diamond-82 forming fluids on garnet oxygen isotope compositions. These data are supplemented by in situ 83

84 major- and trace-element abundance data of garnets and coexisting phases in garnet-rich and

85 clinopyroxene-rich portions of eclogite 09RV09 from the Roberts Victor kimberlite pipe.

86 2. Analytical Methods

87 **2.1 Sample preparation**

88 The studied eclogite xenolith, 09RV09 (total mass ~160 g), is from the Roberts Victor mine, South Africa. The sample was selected on the basis of; 1) its relatively fresh appearance; 2) the 89 presence of diamonds, and; 3) heterogeneously distributed zones with varying modal proportions 90 91 of garnet and clinopyroxene. A 37 g slice derived from this diamondiferous-eclogite (Fig. 2) was used for our study. This carefully examined slice contains zones of varying 92 93 garnet: clinopyroxene modal proportion considered representative of bulk-rock modal variance. To retain textural information during subsequent laser-ablation (LA)-ICP-MS and SIMS 94 analyses, garnets were extracted from a number of distinct zones in this xenolith slice using a 2.5 95 mm diamond-coated steel core drill prior to mounting in the central portion of a 25 mm epoxy 96 97 mount.

98 2.2 In situ major- and trace-element characterisation

All mineral major-, minor-, and trace-element abundances are reported in the

100 supplementary materials along with images of the studied sample regions. Mineral major-

101 element oxide abundances were determined by electron microprobe (EMP) analyses with a five

spectrometer Cameca SX-100 at the University of Alberta. All data were collected with a

- focused (1 μ m) 20 nA beam operating at 15 kV. Counting times for all elements were 20 to 30 s,
- and standard PAP corrections were applied to all analyses using the software of Armstrong
- 105 (1995). Natural and synthetic standards were measured at intervals during each analytical

106 session to assess precision and accuracy. Element concentrations were always within 1 % of 107 accepted values. Detection limits (3σ above background) were typically ≤ 0.03 wt. % for Na₂O, 108 MgO, CaO, NiO, and K₂O, ≤ 0.04 wt. % for Al₂O₃, SiO₂, TiO₂, FeO, and V₂O₃, and ≤ 0.05 wt. % 109 for MnO, P₂O₅, and Cr₂O₃. Additional information pertaining to the EMP methodology is 110 included in the supplementary materials.

111 Garnet trace-element abundances were obtained at the Geological Survey of Canada,

112 Ottawa, using a Photon Machines Analyte 193 nm Ar-F excimer laser coupled to an Agilent

113 7700x quadrupole ICP-MS. Analyses were performed using a 10-16 Hz laser repetition rate at a

photon fluence of 4.4-7.0 Jcm⁻². Data were acquired with a 43 to 69 μ m spot-size for garnets,

and a 52 µm spot for clinopyroxenes. Basaltic glass reference materials, USGS GSD-1G and

116 GSE-1G were utilized as primary standards, with a selection of in-house garnets as secondary

117 reference standards. Further analytical details are included in the supplementary materials.

118 **2.3 Ion-probe oxygen isotope analyses**

All oxygen isotope compositions are reported relative to the Vienna Standard Mean Ocean
Water (VSMOW) standard. Where this can be described as:

121
$$\delta^{18}O_{VSMOW} (\%) = [[({}^{18}O/{}^{16}O)_{sample}/({}^{18}O/{}^{16}O)_{VSMOW}] - 1]*1000].$$

122 For simplicity the VSMOW subscript is omitted and δ^{18} O notation is used in the text herein.

Oxygen isotope compositions of garnets were determined *in situ* with a Cameca IMS-1280 ion microprobe at the Canadian Centre for Isotopic Microanalyses (CCIM) at the University of Alberta following procedures described by Ickert and Stern (2013). Calibration of the matrix correction, as described by Ickert and Stern (2013; cf. Page *et al.*, 2010), was conducted by

analysing a full suite of garnet working standards on a separate grain mount prior to analysing 127 the samples in this study. Seven 2.5 mm diameter cores of eclogite were mounted in a single 25 128 mm epoxy grain mount along with several chips of working standard UAG and secondary 129 130 standard S0088 (a Gore Mountain pyrope-almandine megacryst, and a Jeffrey Mine grossular, respectively). All rock chips and garnets were within 0.5 cm of the centre of the grain mount. 131 Analyses were conducted in a single analytical session, using a 2.5-3.0 nA Cs primary beam with 132 a 15 µm spot diameter. Thirty analyses of UAG were interspersed at regular intervals among 94 133 sample points and seven analyses of S0088 (treated as an 'unknown') were collected during the 134 analytical session for quality assurance. This yielded an S0088 average δ^{18} O-value of 4.13 ± 135 0.09 ‰ (2 σ), n = 7, which coincides with the mean reported in an independent study conducted 136 by Ickert and Stern (2013), indicating that the matrix correction is accurate. Total propagated 137 138 uncertainties (including calibration uncertainty) on each analytical point are ± 0.2 -0.3 ‰ (2 σ). All data are reported in the **supplementary materials**. 139

140 **3. Results**

141 **3.1 Petrographic characteristics**

In 09RV09, garnet-rich, and clinopyroxene-rich zones are unevenly distributed across the ~23 cm² surface area studied (**Fig. 2**). Larger (up to 5 mm in maximum dimension) garnet and clinopyroxene crystals, with well-developed lamellar pyroxene, and garnet exsolution, respectively, account for ~5-10 % of the studied eclogite surface. These large clinopyroxene crystals are turbid in appearance and generally exhibit narrow (<750 µm) sponge-textured rims texturally analogous to those reported for some Bellsbank eclogites (e.g., Taylor and Neal, 1989). The remaining surface area is dominated by rounded interlocking grains (generally 2-3

149	mm in maximum dimension) of garnet and clinopyroxene. Minor interstitial phases include
150	phlogopite (up to 1 mm in maximum dimension), sulphide (generally $< 500 \ \mu m$ maximum
151	dimension), graphite (<650 μ m), diamond (up to 800 μ m in the studied portions of 09RV09), and
152	anhedral clinopyroxene (generally <700 μm). Narrow veins (<150 μm) and cracks (<10 μm) on
153	grain boundaries and penetrating larger crystals contain trapped melt (~60-80 vol. %) with lesser
154	amounts of K-rich phlogopite (\leq 50 µm), clinopyroxene (\leq 20 µm), and small needles of Ti-oxide
155	and sulphide (<1 μ m wide, up to 8 μ m in length). Utilising the textural classification of
156	MacGregor and Carter (1970), the sample is a Group I eclogite. Notably, other accessory
157	mineral phases, for example apatite \pm magnesite \pm monazite \pm kyanite \pm coesite \pm dolomite,
158	which have been observed in other eclogite xenoliths (Sobolev et al., 1994; Snyder et al., 1998)
159	and carbonated high-pressure experimental assemblages (Knoche et al., 1999; Dasgupta et al.,
160	2004) are absent from the studied sample.

2011 Zones rich in garnet have garnet-clinopyroxene ratios of ~90:10 to ~70:30, whereas 162 clinopyroxene-rich zones have garnet-clinopyroxene ratios of ~10:90 to ~30:70. Petrographic 163 study and X-ray tomography (**supplementary materials**) show that abundances and spatial 164 distributions of accessory phlogopite, sulphide, and diamond are not directly correlated with one 165 another, and these do not vary in a systematic manner with garnet and clinopyroxene modal 166 abundances.

167 **3.2 Mineral major- and trace-element abundances**

In situ mineral major- and trace-element abundances of 09RV09 are reported in the
 supplementary materials where reasoning for the use of the geometric form of trace-element
 anomalies throughout this text is discussed. Clinopyroxenes are characterised by Mg# values of

171	86.6-90.0 (where $Mg\# = 100Mg/[Mg+Fe^{total}]$) and jadeite contents that generally range from 13-
172	24 mol. %. Clinopyroxene TiO ₂ abundances range from 0.16-0.29 wt. %, Cr_2O_3 contents vary
173	from <0.05-0.21 wt. %, and K_2O concentrations are generally 0.18-0.29 wt. %, with only two
174	data points at sponge-textured clinopyroxene rims recording <0.04 wt. % K ₂ O. These
175	clinopyroxene compositions are broadly similar to those reported for other diamondiferous and
176	textural Group I eclogites (e.g., Jacob, 2004; Smart et al., 2009).
177	Garnet compositions show small to moderate intra- and inter-grain major-element variation
178	with Mg# of 75.8 to 79.2 and Ca# of 8.2 to 10.4 (where Ca# = $100Ca/[Ca+Mg+Fe^{total}]$), with no
179	consistent trend observed in core to rim traverses. These garnets have low Cr_2O_3 contents (≤ 0.25
180	wt. %), low Cr# values (< 0.70, where Cr# = 100 Cr/[Cr+Al]), combined FeO ^{total} + CaO contents
181	of 13.0-16.6 wt. %, and Na ₂ O contents \leq 0.10 wt. %. This range of garnet major-element
182	compositions (Fig. 3a-b) overlaps that reported for textural Group I and Group II eclogites (\pm
183	diamond; McCandless and Gurney, 1989). Garnet compositions of 09RV09 are broadly
184	consistent with the Group-A garnet major-element classification of Taylor and Neal (1989).
185	Despite significant modal mineral variation, the silicate major-element characteristics of
186	09RV09 do not record the type of systematic mineral compositional variations found to
187	accompany variations in garnet-clinopyroxene proportions in some other modally variable
188	Roberts Victor eclogites. For instance, a diamondiferous eclogite xenolith containing exsolved
189	spinel (< 200 µm; RVSA-71, Ishikawa et al., 2008a-b) and a diamondiferous spinel-free eclogite
190	(HRV-247; Hatton, 1978; O'Reilly and Griffin, 1995; Ishikawa et al. 2008a) showed variations
191	in clinopyroxene jadeite contents, and grossular proportions in garnet that were lower in

192 clinopyroxene-rich zones while the Mg# of both of these phases was lower in garnet-rich zones.

206

193	Chondrite normalised rare-earth-element (REE) abundances of 09RV09 garnets are LREE-
194	depleted relative to HREE ([La/Yb] _N = 0.001-0.019, where N denotes normalisation to CI-
195	chondrite; Fig. 4a). Primitive mantle normalised trace-element abundance patterns (Fig. 4b)
196	display negative Sr- and Ti-anomalies (generally 0.10-0.40 and 0.27-0.47, respectively). In
197	detail, 09RV09 garnets show moderate variations in LREE abundances (e.g., La = 6-70 ppb), Ti,
198	and Hf (88-1386 ppm, and 70-900 ppb, respectively), but little to no variation beyond analytical
199	uncertainties in HREE, Y, Sc, V, and Zr contents (supplementary materials). Abundances of
200	Ti and Hf do not co-vary with La, and La abundances in excess of 20 ppb are restricted to three
201	data points for a single grain distal from zones with the most extreme garnet:clinopyroxene
202	values. The time-resolved ablation signals for these data points do not show resolvable inclusion
203	signals and these La contents are, therefore, considered to represent either a combination of
204	volumetrically minor, finely-disseminated, and evenly distributed LREE-enriched inclusions in

the sampled volume, or higher LREE abundances truly intrinsic to this grain. Notably, all

09RV09 garnet data points are characterised by little to no Eu-anomaly beyond analytical

uncertainties ([Eu/Eu*]_N = 1.0-1.4 and relative errors of 9-21 %, 2σ).

208 In contrast to garnet, all 09RV09 clinopyroxene analyses are LREE-enriched ($[La/Yb]_N =$ 8.52-29.2). These clinopyroxenes lack detectable Eu-anomalies, have positive Sr-anomalies 209 (2.4-4.7), negative Ti-anomalies (0.24-0.42; Fig. 4b), and are generally characterised by positive 210 211 [Zr/Hf]_{PM} values (0.50-0.70) and Ti, V and Sc abundances show little or no variation beyond 212 analytical uncertainties. Detectable variations are present in clinopyroxene LREE, MREE HREE, Y, Sr, Zr, Hf, Nb, and Ta abundances; of these the LREE strongly correlate with one 213 another and HFSE are strongly or very strongly correlated (e.g., supplementary materials). 214 However, comparison of clinopyroxene LREE, Sr, and HFSE abundances shows that these 215

element groups, and Sr, do not co-vary as high degrees of scatter are evident ($R^2 \le 0.3$).

217 Correlations of LREE with MREE, Y, and HREE are moderate to weak, and uncertainties on

218 MREE, HREE, and Y abundances limit confident interpretations of these relationships.

219 Significantly, no strong correlations are evident when abundances of LREE, Sr, Zr, Hf, Nb, and

Ta in clinopyroxene are compared with La contents of texturally associated garnet (not plottedhere).

The range of garnet and clinopyroxene trace-element abundances and inter-element 222 fractionations determined for 09RV09 are broadly analogous to those reported for other Mg-rich 223 eclogites (e.g., Barth et al., 2002; Smart et al., 2009). Gréau et al. (2011) suggested that 224 clinopyroxene trace-element criteria differ between textural Group I and Group II eclogites 225 thereby extending the earlier classification scheme of McCandless and Gurney (1989); 226 comparison with recently recommended discrimination criteria shows that 09RV09 227 228 clinopyroxene Zr abundances are consistent with the Group I classification of Gréau *et al* (2011). On the other hand, Sr and Nd contents of 09RV09 clinopyroxene (472-835 ppm, 7.2-14 ppm, 229 respectively) are higher than other Group I eclogites, and corresponding Ti and Zr abundances 230 231 vary over a range (with associated uncertainties) that overlaps both Group I and Group II characteristics of Gréau et al., (2011). This chemical variability undermines a simple 232 classification scheme based on trace-elements but also emphasises the suitability of sample 233 09RV09 for studying the effects of metasomatic processes on oxygen isotope compositions. 234 Critically, garnets and clinopyroxenes of 09RV09 do not exhibit systematic variations in the 235 trace-element abundances determined for regions with differing garnet:clinopyroxene modal 236 proportions, and abundances and inter-element ratios of trace-elements, which cover a wide-237

range of geochemical properties, do not correlate systematically with major-element

concentrations, Mg#, and/or garnet grossular contents (e.g., Fig. 5a).

240 **3.3 Garnet oxygen-isotope compositions**

In situ oxygen isotope compositions of 94 analytical points in 7 garnets derived from 241 texturally distinct locations in 09RV09 yield δ^{18} O-values of +6.2-6.8 ‰ with no detectable 242 variation outside total analytical uncertainties. The mean, mode, and median of these garnet 243 oxygen isotope compositions are coincident at +6.5 % (Fig. 6a). The garnet δ^{18} O-values show 244 negligible variations, while some major- and trace-element contents vary moderately. However, 245 there is no systematic correlation between garnet oxygen isotope compositions and major-, and 246 trace-element characteristics of 09RV09 garnets and clinopyroxenes at scales ranging from 10's 247 µm to many cm (supplementary materials). 248

The probability of the mean garnet δ^{18} O-value of 09RV09 being equal to that of the mean of 249 the mantle garnet δ^{18} O-distribution is low. Results of t-tests to compare 09RV09 garnet δ^{18} O-250 data with that of Mattey et al., (1994) assuming garnet fractionation factors of <0.5 % for 251 olivine, orthopyroxene, and clinopyroxene, yield a very low probability (p < 0.001) of coincident 252 253 means in all cases (where t-tests include; 1) the Shapiro-Wilk approach that assumes both datasets are derived from normally distributed populations; and, 2) non-parametric Kolmogorov-254 255 Smirnov tests, including the Lilliefors correction, with assumptions of both equal and unequal variances applied during each t-test). In addition, Mann-Whitney Rank sum tests of these data 256 also show that the mean of our garnet δ^{18} O-data is significantly different from garnet-equilibrium 257 values calculated for mantle mineral laser-fluorination data reported by Mattey et al. 1994 (p 258 <0.001). 259

260 **4. Discussion**

Constraining the origin of mantle eclogite xenoliths is of fundamental importance to studies 261 of craton origin and evolution. Isotopic compositions (O, Mg, Pb-Pb, Sm-Nd, Lu-Hf, and Re-262 Os) reported for a number of diamondiferous and non-diamondiferous eclogite xenoliths of the 263 264 Roberts Victor kimberlite suggest that these materials are derived from Archean subducted crust (e.g., Kramers, 1979; McGregor and Manton, 1986; Pearson et al. 1995; Shirey et al., 2001; 265 Jacob et al., 2005; Wang et al., 2012). A recent study of mineral major- and trace-element 266 characteristics and garnet δ^{18} O-compositions of 33 eclogite xenoliths from the Roberts Victor 267 Mine (Gréau et al., 2011), and an investigation of a single texturally complex eclogite (RV07-17; 268 Huang *et al.*, 2014), have questioned the robust nature of garnet δ^{18} O-values as tracers of a 269 crustal precursor for eclogites; these authors suggested that oxygen isotopes are markedly 270 fractionated by mantle metasomatic processes. In particular, Gréau et al. (2011) suggested that 271 garnet δ^{18} O-values are correlated with clinopyroxene incompatible element abundances, arguing 272 that garnet oxygen isotope compositions reflect carbonatite metasomatism. Our detailed 273 investigation of 09RV09 provides an important data-set to evaluate the robustness of garnet 274 oxygen isotope compositions to metasomatic processes. 275

276 4.1 Metasomatic modification of 09RV09

277 Metasomatic modification of 09RV09 is evident in the form of late-stage infiltration along 278 garnet and clinopyroxene grain boundaries (generally <150 μ m wide) and narrow cracks (<10 279 μ m) penetrating coarse-sized (2-3mm) garnets and clinopyoxenes. These narrow features 280 contain trapped melt (~60-80 vol. % of the infiltration zones) with lesser amounts of K-rich 281 phlogopite (<50 μ m), clinopyroxene (<20 μ m), and small needles of Ti-oxide and sulphide (<1

um wide, up to 8 µm in length) and are interpreted as metasomatic in origin. In some cases, 282 needle-like phases (<5µm in length) have nucleated at the grain boundaries of large, pre-existing, 283 garnet and clinopyroxene and have grown outwards into narrow infiltration zones (Fig. 7a). 284 Larger phlogopite grains (up to 500 µm) in 09RV09 generally occur at the point where several 285 narrow (<1 mm wide) infiltration zones connect to one another. Though textural observations 286 alone make it difficult to fully assess the genetic origin of these phases, we consider large 287 phlogopites as crystals formed during metasomatic infiltration experienced by 09RV09; this 288 interpretation is consistent with previous suggestions of a metasomatic origin for phlogopites of 289 other Roberts Victor eclogite xenoliths (e.g., Ongley et al., 1987). The relatively high K₂O 290 contents (generally >0.20 wt. %) of 09RV09 clinopyroxenes are within the range reported for 291 other textural Group I diamondiferous eclogite xenoliths (e.g., McCandless and Gurney, 1989) 292 293 and eclogitic clinopyroxene inclusions in diamond (e.g., Taylor et al., 1998, 2000; Stachel and Harris, 2008 and references therein). Previous studies (e.g., Hatton, 1978; Gréau et al., 2011) 294 have suggested that clinopyroxene K₂O contents above 0.07 wt. % are indicative of metasomatic 295 296 modification associated with diamond formation in mantle environments, and these studies generally show elevated Na₂O contents (>0.09 wt. %) in garnet accompanying the high K 297 concentrations in clinopyroxene (e.g., Hatton, 1978). Further, 09RV09 clinopyroxenes preserve 298 299 notable inter- and intra-grain heterogeneity in the form of LREE, MREE, Sr, Ti and Nb abundances. In contrast, 09RV09 garnets are generally characterised by intra-grain major- and 300 trace-element homogeneity and generally have Na₂O contents <0.09 wt. %. These garnets have 301 higher Ti abundances compared with some - not all - textural Group II eclogites (supplementary 302 materials), but lower Ti contents than mean and median values of eclogitic garnet diamond 303 304 inclusions (e.g., Stachel and Harris, 2008).

The number of diamonds (>100 μ m) observed *in situ* is small (*n* = 7), but those identified occur in regions of intersecting infiltration zones (± adjacent phlogopite). The uneven distribution of 09RV09 diamonds, and their occurrence in interstitial regions, is consistent with observations reported for other diamondiferous eclogite xenoliths (e.g., Anand *et al.*, 2004; Spetsius and Taylor, 2008). However, the clear evidence for metasomatic modification of 09RV09 makes it imperative that potential evidence of the protolith is treated cautiously.

311

4.1.1 Element exchange processes at the grain-scale

The disparate degree of equilibration exhibited by 09RV09 clinopyroxenes and garnets 312 testifies to differing element exchange behaviours in these phases (cf., Burton *et al.*, 1995; 313 Taylor et al., 1996) with respect to the metasomatic history of this sample. An end-member 314 model developed here specifically for 09RV09 references theories of diamond formation and 315 takes account of regional tectonomagmatic events likely to have affected materials in proximity 316 to the Colesburg Lineament, and sampled by the Cretaceous Roberts Victor kimberlite proximal 317 318 to that major intra-cratonic terrane boundary (Fig. 1). Our model involves two metasomatic events influencing a garnet-clinopyroxene protolith after incorporation into the lithospheric 319 320 mantle; the first being ancient and involving metasomatism at great depth by carbon-bearing 321 fluids facilitating diamond formation broadly synchronous with stabilisation of the Kaapvaal Craton >2.5 Ga (see Pearson and Wittig, 2008; Helmstaedt et al., 2010; Shu and Brey, 2015) and 322 potentially concurrent with suturing along the Colesburg Lineament at ~2.9 Ga (Schmitz et al., 323 2004; Shu *et al.*, 2013). This ancient metasomatism is followed by elemental and isotopic 324 equilibration of silicate phases during protracted high-pressure, high-temperature residence in 325 lithospheric mantle. Magnesium may be introduced during ancient metasomatism, and/or a 326 small degree of melt removal may be facilitated by fluid introduction (where melt removal is 327

anticipated to cause only a small shift to lower δ^{18} O-values; Williams *et al.*, 2009). Thus, the 328 relatively magnesian nature of this sample and the homogenous but higher Mg# of garnet in 329 close proximity to diamond (Fig. 3b) may be, at least in part, related to an ancient metasomatic 330 331 event. The second modification event in our model scenario involves a late-stage metasomatic interaction linked to kimberlite arrival and xenolith entrainment at ~124 Ma (Smith et al., 1985) 332 contributing to frozen records of inter-and intra-grain heterogeneity in 09RV09 clinopyroxenes. 333 Assessing the validity of this model requires consideration of the nature of element exchange in 334 eclogitic garnets and omphacitic pyroxenes (supplementary materials). For example, the intra-335 grain homogeneity displayed by 09RV09 garnets may reflect relatively fast element exchange 336 and equilibration of major- and trace-element abundances in garnet during a single metasomatic 337 event when compared with co-existing clinopyroxene. Alternatively, element exchange 338 339 processes in garnet may be orders of magnitude slower than those operating in clinopyroxene 340 leading to the conclusion that the garnets retain robust records of their mantle protolith that are resistant to late-stage small-volume metasomatic modification. 341

Equilibration temperatures calculated for clinopyroxene cores and coexisting garnets in 342 09RV09 are within the range anticipated for cratonic lithospheric mantle materials resident at 343 depths in which diamond is stable. Given this observation, we reason that 09RV09 garnet 344 compositions reflect equilibrated mantle compositions minimally disturbed by late-stage small-345 volume metasomatism. In contrast, trivalent LREE-MREE, tetravalent HFSE, and divalent 346 cations of small ionic radius (e.g, Fe, Mg, Mn) may have diffused relatively rapidly in the outer 347 portions of 09RV09 clinopyroxenes as a result of recent metasomatic disturbance. The observed 348 decoupling between REE and HFSE in 09RV09 clinpyroxenes likely relate to differences in the 349 rate or nature of REE and HFSE element exchange in clinopyroxene, potentially high HFSE 350

blocking temperatures, and/or sequestering of HFSE by volumetrically minor rutile needles
crystallised in corresponding metasomatic infiltration zones. It is likely that all of these factors
contributed during metasomatic modification of 09RV09 clinopyroxenes.

354

4.1.2 Bulk-rock reconstruction and its constraints on 09RV09 metasomatism

The precise nature of fluids that infiltrated 09RV09, related to diamond formation, and 355 kimberlite entrainment, is not well constrained at this time (e.g., speciation, fO_2 , isotopic 356 characteristics). Given the similarities between 09RV09 silicate trace-element characteristics 357 358 and those of silicate diamond inclusion data (e.g., Ireland et al., 1994; Taylor et al., 1996, 2000; 359 Sobolev et al., 1998; Stachel et al., 2004), metasomatic agents that have influenced 09RV09 360 likely resemble the spectrum of compositions reported for diamond fluid inclusions. For these 361 reasons, we model the trace-element composition resulting from mixing between possible protolith compositions and anticipated metasomatic fluids. Modification by kimberlite 362 (generally considered to be CO₂-rich and LREE-enriched; e.g., Becker and Le Roux, 2006; 363 Kjarsgaard et al., 2009) and/or potential LREE-enriched fluids derived from the host kimberlite 364 is possible, but is considered to be volumetrically minor. 365

Bulk-rock reconstructions utilise representative garnet and clinopyroxene trace-element core

367 compositions, trace-element characteristics of altered gabbro (e.g., Hart *et al.*, 1999; Bach *et al.*,

368 2001) previously considered by others as a possible eclogite protolith (e.g., Green and

369 Ringwood, 1967), and trace-element compositions reported for gem-quality diamond (McNeill et

- al., 2009) and fluid inclusions of fibrous diamonds (Klein-BenDavid *et al.*, 2010). Results of
- these calculations indicate that the addition of <<0.03 wt. % of a diamond-forming incompatible-
- 372 element-rich fluid to an oceanic crustal protolith can account for the LREE-enrichment

calculated for the reconstructed bulk-rock compositions of 09RV09 (**Fig. 7b** and **supplementary materials**). The addition of similarly low metasomatic fluid proportions (<<0.05 wt. %) is required if the potential crustal protolith is considered to be derived from a more magnesian (relative to typical gabbro) sheeted dyke complex lacking Eu-anomalies (not shown) and/or an altered basalt of broadly picritic/komatiitic composition (cf., Shirey *et al.*, 2001). Given this evidence for metasomatic modification of 09RV09, we appraise the consequences for our interpretation of the homogeneous garnet δ^{18} O-compositions in this xenolith.

4.2 Oxygen-isotope signatures: metasomatism versus precursor inheritance

In contrast to the prevailing paradigm, it has been suggested that eclogite garnet oxygen 381 isotope compositions in excess of the typical garnet mantle range may reflect secondary 382 overprinting by the passage of carbonatitic melt (e.g., Gréau et al., 2011), and/or could reflect 383 interaction with CO-, OH, and/or CO₂-bearing fluids (e.g., Deines et al., 1991) similar to those 384 reported for diamond inclusions (e.g., Navon et al., 1988; Turner et al., 1990; Izraeli et al., 2001; 385 Klein-BenDavid, 2004, 2007; Tomlinson et al., 2006) and observed in some mantle xenoliths 386 transported by alkali basalts (e.g., Bergman and Dubessy, 1984; Andersen and Neumann, 2001). 387 There are no oxygen isotopic determinations on primary carbonatites erupted in an un-modified 388 389 state from the mantle with which to test this conjecture. Current experimental, empirical, and theoretical partition coefficients combined with fractionation factors reported for basaltic liquids 390 and associated phases at temperatures of ~1000-1300°C (cf., Eiler, 2001 and Chacko et al., 2001) 391 indicate that silicate δ^{18} O-values vary by <0.5 % during the generation and fractional 392 crystallisation of basaltic melts at high-temperatures (1000-1300°C). In addition, pressure effects 393 on isotopic exchange at crustal and upper-mantle conditions are thought to be small due to 394 limited volume changes (<0.005 % for pressure differences of 20-30 kbar; e.g., Clayton et al., 395

396	1975; Polyakov and Kharlashina, 1994). Glass, CO, OH, and CO ₂ species may fractionate δ^{18} O-
397	compositions by detectable amounts (>>0.5 %; Deines et al., 1991), but the effect of possible
398	solutes, and their potential speciation variations, on oxygen isotope fractionation in mantle fluids
399	carrying gaseous molecules is not well constrained at conditions appropriate for mantle
400	environments (e.g., O'Neil, 1986; Bindeman, 2008 and references therein). Zheng (1993)
401	suggested that, under certain circumstances, partial substitution of $[OH]_4^{-4}$ for $[SiO_4]^{-4}$ in
402	grossular molecules could potentially lead to ¹⁸ O-enrichment, and Kohn and Valley (1998)
403	proposed that octahedral cation substitutions may also influence garnet δ^{18} O-values. Oxygen
404	diffuses slowly in garnet even under hydrous conditions (e.g., Lichtenstein and Hoernes, 1992;
405	Cole and Chakraborty, 2001). For these reasons, and considering mass-balance requirements,
406	kinetic processes such as diffusion, and/or solution-precipitation, associated with metasomatic
407	exchange in mantle environments will potentially lead to disequilibrium characteristics in the
408	form of garnet compositional zoning developed during complex multi-stage histories (e.g.,
409	Zhang et al., 2000) anticipated for SCLM residence times up to Gyrs.
410	Given the complex metasomatic history of sample 09RV09 we might expect to see some
411	measureable small-scale variations in oxygen isotope compositions that, for instance, relate to
412	elemental or textural variation. No variation in garnet δ^{18} O-values exists. The data are within

413 measurement uncertainty both within garnet grains and in garnets across the entire xenolith.

414 Similarly, garnets in a coesite-rutile-bearing eclogite from Roberts Victor with abundant veinlets

415 (sample 13-64-136), also lack inter- and intra-grain variation in garnet δ^{18} O-compositions

416 (Russell *et al.*, 2013). These observations contrast to inter-sample garnet δ^{18} O variance reported

- 417 for a texturally complex eclogite (RV07-17; Huang *et al.*, 2014). Moreover, there is no
- 418 correlation at all between clinopyroxene incompatible element abundances such as La and garnet

419	δ^{18} O-values for 09RV09 (Fig. 5b), indicating that incompatible-element enrichment due to
420	metasomatism is unlikely to be the primary control on the oxygen isotopic composition of this
421	eclogite. This result means that a metasomatic origin of the statistically non-robust correlation
422	between La and δ^{18} O-values presented by Gréau <i>et al.</i> (2010; Fig. 5b) is unlikely (p<0.001). No
423	valid mixing curve is evident in the combined data set, especially considering that the Group II
424	eclogites included in the Gréau et al. (2010) are part of a separate group of eclogites from
425	Roberts Victor whose δ^{18} O-values range to above +6 ‰; e.g., Ongley <i>et al.</i> , 1987). This
426	consideration weakens the argument for metasomatic overprinting of garnet δ^{18} O-compositions
427	and hence the existing and new eclogite data plotted on this co-variation diagram, therefore, is
428	not able to provide a unique solution to account for eclogitic garnet δ^{18} O-compositions.
429	Mass balance considerations with respect to garnet oxygen isotope compositions (where
430	oxygen is a major-element) offer a more powerful argument in that to significantly modify the
431	oxygen isotopic composition over 1 ‰ away from the canonical mantle value requires
432	equilibration with substantially larger relative volumes of fluid (or unrealistic δ^{18} O-
433	compositions) than can be accounted for by the degree of trace-element modification in 09RV09.
434	A metasomatic model postulated to drive volumetrically significant ¹⁸ O enrichments (or
435	depletions) in eclogite garnet δ^{18} O-compositions requires that a highly reactive, volatile-rich
436	agent traverse substantial quantities of mantle without equilibrating with the ambient material.
437	Mass balance ("closed system") and Rayleigh ("open system") models (e.g., Fig. 8; Taylor,
438	1977; Criss and Taylor, 1986) place constraints on the degree of fluid-rock interaction required
439	to buffer a fluid with an initial δ^{18} O-value of +7.5 ‰. In these models, we utilised forsterite-
440	calcite and calcite-CO ₂ oxygen isotope fractionation factors (yielding Δ CO ₂ -forsterite = +4.1 ‰;
441	Chiba et al., 1989; Chacko et al., 1991) and a calcite-H ₂ O fractionation factor (giving Δ H ₂ O-

463

442	forsterite = +0.3 ‰; O'Neil et al., 1969; Friedman and O'Neil, 1977). At an assumed
443	temperature of 1100 °C, a closed-system model predicts that 1 g of a fluid, rich in H_2O or CO_2 ,
444	requires 15-20 grams of peridotite (approximated by forsterite = $+5.0$ ‰) to become buffered to
445	a composition within ± 0.5 ‰ of the median peridotitic mantle value. Under open-system
446	conditions, which may provide a more realistic analogue for mantle metasomatism, fluid
447	interaction with substantially less peridotite for a given fluid volume (<1:5 fluid-rock ratio) is
448	required to buffer the fluid δ^{18} O-value to the composition of silicate mantle with which it is
449	interacting. These models show that only minor fluid-rock interaction is required to buffer the
450	oxygen isotope composition of a mantle metasomatic fluid. To generate the 09RV09 garnet
451	δ^{18} O-value of +1 ‰ above the mantle value, and up to +3.5 ‰ observed in other eclogites,
452	mantle pyroxenites and diamond inclusions (MacGregor and Manton, 1986; Pearson et al., 1991;
453	Jacob <i>et al.</i> , 2003; Ickert <i>et al.</i> , 2015) requires not only that very high fluid-rock ratios (≥2:5;
454	Fig. 8) are maintained for a compositionally extreme fluid at the local "sample scale" but that
455	these extreme δ^{18} O-compositions are continuously maintained from the fluid source, throughout
456	its flow at great depth (asthenospheric and/or lithospheric mantle), where the fluid flow regime is
457	likely percolative. Therefore, the probability of a metasomatic fluid with an extreme oxygen
458	isotope composition surviving unmodified during transport through the mantle, itself dominantly
459	peridotitic, and imposing this signature on an eclogite body within the peridotite is very low
460	indeed. This clearly favours the interpretation of 09RV09 garnet δ^{18} O-values as representing a
461	robust tracer of the protolith lithology rather than the product of mantle metasomatism.
462	Evaluating the suggestion that diamond-forming fluids in general may be responsible for

464 peridotite suite garnets included in diamonds analysed by Mattey *et al.* (1994) have a mean δ^{18} O-

generating exotic mantle oxygen isotopic compositions (Gréau et al. 2011) we note that 6

value of +5.3 ‰, identical to typical mantle peridotite. Hence, there is no solid evidence for
appreciable oxygen isotope modification in garnet-bearing mantle materials during
metasomatism by small-volume incompatible-element enriched fluids. Indeed, it is more likely
that δ¹⁸O-compositions of small-degree metasomatic fluids equilibrate with the host rock and,
thus, we reason that mantle eclogites and peridotites impart their oxygen isotope signature on
volumetrically minor and transient fluids during metasomatism and diamond evolution.

471 **4.3 Summary and implications**

To critically appraise the nature of potential metasomatic signatures in mantle eclogite xenoliths, we conducted a multi-technique *in situ* study of a diamondiferous eclogite xenolith with varying garnet:clinopyroxene proportions (09RV09) from the Roberts Victor kimberlite, S. Africa. We provided the first *in situ* measurements of δ^{18} O-values in eclogitic garnet in close proximity to diamond, and retaining textural control, to test theories concerned with metasomatic modification of eclogites during diamond formation, particularly garnet oxygen isotope compositions.

Intra-grain variations in clinopyroxene major-element, LREE-MREE and HFSE contents appear to have resulted from later metasomatism related to kimberlite arrival and xenolith entrainment, yet oxygen isotope compositions in garnet are uniform, within tight analytical uncertainties, across the xenolith in a wide variety of textural settings. SIMS garnet δ^{18} O-values of 6.5 ± 0.2 ‰ are higher than the mean mantle garnet range (4.8-5.5 ‰). There is no covariation of oxygen isotope composition with incompatible element based indicators of metasomatism.

The lack of detectable inter- and intra-grain oxygen isotope variation in 09RV09 garnet. 486 including garnet in close proximity to diamond, indicates that garnet δ^{18} O-compositions are 487 ancient and likely not affect by infiltration of diamond-forming fluids. The intra-sample garnet 488 489 oxygen isotope homogeneity of 09RV09 is of particular interest as available data suggest that intra-sample garnet oxygen isotope homogeneity is likely representative of mantle eclogites in 490 general. Prior laser-fluorination (LF) studies of garnet separates have generally shown highly 491 reproducible eclogite garnet δ^{18} O-compositions within a given sample both at individual 492 laboratories and in inter-laboratory comparison studies (e.g., Rumble et al., 2007). Furthermore, 493 results of other recent *in situ* studies of garnet δ^{18} O-compositions of 52 other eclogite xenoliths 494 have demonstrated intra-sample homogeneity (Russell et al., 2013; Smit et al., 2014; Dongre et 495 al., 2015) with only one known exception; RV07-17 (Huang et al., 2014). Our data, combined 496 497 with the slow time-scales of oxygen isotopic diffusion in the mantle (cf., Russell et al., 2013) and the difficulties in moving metasomatic fluids with exotic oxygen isotopic compositions through 498 the Earth's mantle without buffering their compositions to the mantle δ^{18} O-value support the 499 500 concept that eclogite oxygen isotope compositions largely reflect their crustal precursors.

The oxygen isotope composition of garnets in 09RV09 is significantly different from typical mantle values supporting a crustal origin for its precursor and in line with many other studies of Roberts Victor eclogite xenoliths and eclogitic diamond inclusions (e.g., MacGregor and Manton, 1986; Jacob *et al.*, 2005; Tappert *et al.*, 2005; Schulze *et al.*, 2013; Ickert *et al.*, 2013, 2015), irrespective of their textural groupings.

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511 7. References

- Anand M., Taylor L. A., Misra K. C., Carlson W. D., and Sobolev N. V. (2004). Nature of diamonds in
 Yakutian eclogites: views from eclogite tomography and mineral inclusions in diamonds. *Lithos*,
 77(1), 333-348.
- Andersen T., and Neumann E.R., 2001. Fluid inclusions in mantle xenoliths. *Lithos*, 55(1), 301-320.
- Armstrong, J.T., (1995). A package of correction programs for the quantitative electron microbeam X-ray
 analysis of thick polished materials, thin films, and particles. *Microbeam Anal.*, 4, 177-200.
- Aulbach S., Pearson N.J., O'Reilly S.Y., and Doyle B.J., 2007. Origins of xenolithic eclogites and
 pyroxenites from the Central Slave Craton, Canada. J. Petrol. 48, 1843–1873.
- Bach W., Alt J.C., Niu Y., Humphris S.E., Erzinger J., and Dick H.J.B., 2001. The geochemical
 consequences of late-stage low-grade alteration of lower ocean crust at the SW Indian Ridge: results
 from ODP Hole 735B (Leg 176). *Geochim. Cosmochim. Acta.*, 65, 3267–3287.
- Baker J., Matthews A., Mattey D., Rowley D., and Xue F., 1997. Fluid-rock interactions during ultra-high
 pressure metamorphism, Dabie Shan, China. *Geochim. Cosmochim Acta*, 61(8), 1685-1696.
- Barth M.G., Rudnick R.L., Horn I., McDonough W.F., Spicuzza M.J., Valley J.W., and Haggerty S.E.,
 2001. Geochemistry of xenolithic eclogites from West Africa, part I: a link between low MgO
 eclogites and archean crust formation. *Geochim. Cosmochim. Acta*, 65 (9), 1499–1527.
- Barth M.G., Rudnick R.L., Horn I., McDonough W.F., Spicuzza M.J., Valley J.W., and Haggerty S.E.,
 2002. Geochemistry of xenolithic eclogites from West Africa, part 2; Origins of the high MgO
 eclogites. *Geochim. et Cosmochim. Acta*, 66(24), 4325-4345.
- Beard B.L., Fraracci K.N., Taylor L.A., Snyder G.A., Clayton R.A., Mayeda T.K., and Sobolev N.V.,
 1996. Petrography and geochemistry of eclogites from the Mir kimberlite, Yakutia, Russia. *Contrib. Mineral. Petrol.*, 125, 293-310.
- Becker M., and Le Roex A. P., 2006. Geochemistry of South African on-and off-craton, Group I and
 Group II kimberlites: petrogenesis and source region evolution. *J. Petrol.*, 47(4), 673-703.
- Bergman S.C., and Dubessy J., 1984. CO₂-CO fluid inclusions in a composite peridotite xenolith:
 implications for upper mantle oxygen fugacity. *Contrib. Mineral. Petrol.*, 85(1), 1-13.
- Bindeman I., 2008. Oxygen isotopes in mantle and crustal magmas as revealed by single crystal analysis.
 Rev. Mineral. Geochem., 69(1), 445-478.
- Burton, K. W., Kohn, M. J., Cohen, A. S., and O'Nions, R. K., 1995. The relative diffusion of Pb, Nd, Sr
 and O in garnet. *Earth Planet. Sci. Lett.*, 133(1), 199-211.

- 542 Caporuscio F.A., 1990. Oxygen isotope systematics of eclogite mineral phases from South Africa. *Lithos*,
 543 25, 203-210.
- 544 Carmody L., Barry P.H., Shervais J.W., Kluesner J.W., and Taylor L.A., 2013. Oxygen isotopes in
 545 subducted oceanic crust: A new perspective from Siberian diamondiferous eclogites. Geochem.
 546 Geophys. Geosyst., 14(9), doi: 10.1002/ggge.20220.
- 547 Chacko T., Cole D.R., and Horita J., 2001. Equilibrium oxygen, hydrogen and carbon isotope
 548 fractionation factors applicable to geologic systems. *Rev. Mineral. Geochem.*, 43(1), 1-81.
- Chacko T., Mayeda T. K., Clayton R. N., and Goldsmith J. R., 1991, Oxygen and carbon isotope
 fractionations between CO₂ and calcite. *Geochim. Acta*, 55(10), 2867-2882.
- 551 Chiba H., Chacko T., Clayton R. N., and Goldsmith J. R., 1989, Oxygen isotope fractionations involving
 552 diopside, forsterite, magnetite, and calcite: Application to geothermometry. *Geochim. Cosmochim.*553 *Acta*, 53(11), 2985-2995.
- Clayton R.N., Goldsmith J.R., Karel K.J., Mayeda T K., and Robert C.N., 1975. Limits on the effect of
 pressure on isotopic fractionation. *Geochim. Cosmochim. Acta*, 39(8), 1197-1201.
- Cole D.R., and Chakraborty S., 2001. Rates and mechanisms of isotopic exchange. *Rev. Mineral. Geochem.*, 43(1), 83-223.
- Coleman R. G., Lee D. E., Beatty L. B., and Brannock W. W., 1965. Eclogites and eclogites: their
 differences and similarities. *Geol. Soc. Am. Bullet.*, 76(5), 483-508.
- 560 Criss R. E., and Taylor Jr. H. P., 1986, Meteoric-hydrothermal systems. *Rev. Mineral.*, 16, 373-424.
- Dasgupta R., Hirschmann M. M., and Withers A. C., 2004. Deep global cycling of carbon constrained by
 the solidus of anhydrous, carbonated eclogite under upper mantle conditions. *Earth Planet. Sci. Lett.*,
 227(1), 73-85.
- Deines P., Harris J.W., Robinson, D.N., Gurney, J.J., and Shee, S.R., 1991. Carbon and oxygen isotope
 variations in diamond and graphite eclogites from Orapa, Botswana, and the nitrogen content of their
 diamonds. *Geochim.et Cosmochim. Acta*, 55(2), 515-524.
- Dongre A. N., Jacob D. E., and Stern R. A., 2015. Subduction-related origin of eclogite xenoliths from
 the Wajrakarur kimberlite field, Eastern Dharwar craton, Southern India: Constraints from petrology
 and geochemistry. *Geochim. Cosmochim. Acta*, 166, 165-188.
- Eiler J.M., 2001. Oxygen isotope variations of basaltic lavas and upper mantle rocks. *Rev. Mineral. Geochem.*, 43(1), 319-364.
- Friedman I., and O'Neil, J. R., 1977, Compilation of stable isotope fractionation factors of geochemical interest. In *Data of Geochemistry 6th Ed.*, (Vol. 440). US Govt., Washington, USA.
- Garlick G.D., MacGregor I.D., and Vogel D.E., 1971. Oxygen isotope ratios in eclogites from
 kimberlites. *Science*, 172, 1025–1027.
- Gréau Y., Huang J.-X., Griffin W.L., Renac C., Alard O., and O'Reilly S.Y., 2011. Type I eclogites from
 Roberts Victor kimberlites: Products of extensive mantle metasomatism. *Geochim. Cosmochim. Acta*,
 75, 6927-6954.
- 579 Green D. H., and Ringwood A. E., 1967. An experimental investigation of the gabbro to eclogite
 580 transformation and its petrological applications. Geochimica et Cosmochimica Acta, 31(5), 767-833.

581	Gregory R.T., and Taylor Jr. H.P., 1981. An oxygen isotope profile in a section of Cretaceous oceanic
582	crust, Samail ophiolite, Oman: Evidence for δ^{18} O buffering of the oceans by deep (>5 km) seawater-
583	hydrothermal circulation at mid-ocean ridges. J. Geophys Res, 86(B4), 2737-2755.

- Gurney, J. J., Helmstaedt, H. H., Richardson, S. H., and Shirey, S. B., 2010. Diamonds through time.
 Econ. Geol., 105(3), 689-712.
- Hart S.R., Blusztain J., Dick H.J.B., Meyer P.S., and Muehlenbachs K.,1999. The fingerprint of seawater
 circulation in a 500-meter section of ocean crust gabbros. *Geochim. Cosmochim. Acta.*, 63, 4059–
 4080.
- Hatton C.J., 1978. The geochemistry and origin of xenoliths from the Roberts Victor Mine. *PhD thesis*, *Univ. Cape Town, Cape Town, South Africa.*
- Hatton C.J., and Gurney J.J., 1987. Roberts Victor eclogites and their relation to the mantle. *In: Nixon P.H. (Ed.), Mantle Xenoliths, Wiley, Chichester*, pp. 453-463
- Heaman L.M., Creaser R.A., and Cookenboo H.O., 2002. Extreme enrichment of high field strength
 elements in Jericho eclogite xenoliths: A cryptic record of Paleoproterozoic subduction, partial
 melting, and metasomatism beneath the Slave craton, Canada. *Geology*, *30*(6), 507-510.
- Heaman L.M., Creaser R.A., Cookenboo H.O., and Chacko T., 2006. Multi-stage modification of the
 Northern Slave mantle lithosphere: evidence from zircon-and diamond-bearing eclogite xenoliths
 entrained in Jericho kimberlite, Canada. J. Petrol., 47(4), 821-858.
- Helmstaedt H., and Doig, R., 1975. Eclogite nodules from kimberlite pipes of the Colorado Plateau –
 samples of subducted Franciscan-type oceanic lithosphere. *Phys. Chem. Earth*, 9, 95-111.
- Helmstaedt, H. H., Gurney, J. J., and Richardson, S. H., 2010. Ages of cratonic diamond and lithosphere
 evolution: constraints on Precambrian tectonics and diamond exploration. *Can. Min.*, 48(6), 13851408.
- Huang J.-X., Gréau Y., Griffin W.L., O'Reilly S.Y, and Pearson N.J., 2012. Multi-stage origin of Roberts
 Victor eclogites: Progressive metasomatism and its isotopic effects. *Lithos*, 142-143, 161-181.
- Huang J.-X., Griffin W.L., Gréau Y., Pearson N.J., O'Reilly S.Y., Cliff J., and Martin L., 2014.
 Unmasking xenolithic eclogites: Progressive metasomatism of a key Roberts Victor sample. *Chem. Geol.*, 364, 56-65.
- Ickert R.B., and Stern R.A., 2013. Matrix corrections and error analysis in high-precision SIMS ¹⁸O/¹⁶O
 measurements of Ca-Mg-Fe garnet. *Geostand. Geoanal. Res.*, 37(4), 429-448.
- 611 Ickert R.B., Stachel T., Stern R. A., and Harris, J. W., 2013. Diamond from recycled crustal carbon
 612 documented by coupled δ¹⁸O-δ¹³C measurements of diamonds and their inclusions. *Earth Planet. Sci.* 613 *Lett.*, 364, 85-97.
- Ireland T.R., Rudnick R.L., and Spetsius Z., 1994. Trace elements in diamond inclusions from eclogites
 reveal link to Archean granites. *Earth Planet. Sci. Lett.*, 128(3), 199-213.
- 616 Ishikawa A., Pearson D.G., Maruyama S., Cartigny P., Ketchum R.A., and Gurney J.J., 2008a.
- 617 Compositional layering in a highly diamondiferous eclogite xenolith from the Roberts Victor
- 618 kimberlite, South Africa and its implications for diamond genesis. *IX International Kimberlite Conf.*,
- 619 Abst. No. 9IKC-A-00078.

- 620 Ishikawa A., Pearson D.G., Maruyama S., de Bruin D., and Gurney J.J., 2008b. Compositional variability
- of the Roberts Victor eclogites: evidence for mantle metasomatism involving diamond dissolution. *IX International Kimberlite Conf.*, Abst. No. 9IKC-A-00079.
- Izraeli E.S., Harris J.W., and Navon O., 2001. Brine inclusions in diamonds: a new upper mantle fluid. *Earth Planet. Sci. Lett.*, 187(3), 323-332.
- Jacob D., 2004. Nature and origin of eclogite xenoliths in kimberlites. *Lithos*, 77, 295–316.
- Jacob D., and Jagoutz E., 1994. A diamond-graphite bearing eclogitic xenolith from Roberts Victor
- 627 (South Africa): Indications for petrogenesis from Pb-, Nd-, and Sr-isotopes. In: Meyer H.O.A.,
- Leonardos O.H. (Eds), Kimberlites, Related Rocks, and Mantle Xenoliths. Companhia de Pesquisa de
 Recursos Minerais, Spec. Pub. vol. 1/A, pp. 304-317.
- Jacob D., and Foley S.F., 1999. Evidence for Archean ocean crust with low high field strength element
 signature from diamondiferous eclogite xenoliths. *Lithos*, 48, 317-336.
- Jacob D., Jagoutz E., Lowry D., Mattey D., and Kudrjavtseva G., 1994. Diamondiferous eclogites from
 Siberia: remnants of Archean oceanic crust. *Geochim. Cosmochim. Acta*, 58(23), 5191-5207.
- Jacob D.E., Schmickler B., and Schulze D.J., 2003. Trace element geochemistry of coesite-bearing
 eclogites from the Roberts Victor Kimberlite. *Lithos*, 71, 337-351.
- Jacob D., Bizimis M., and Salters V.J.M., 2005. Lu-Hf and geochemical systematics of recycled ancient
 oceanic crust: evidence from Roberts Victor eclogites. *Contrib. Mineral. Petrol.*, 148(6), 707-720.
- Jagoutz E., Dawson J.B., Hoernes S., Spettel B., and Wanke H., 1984. Anorthositic oceanic crust in the
 Archean Earth. *XV Lunar Planet. Sci. Conf.*, pp. 395-396.
- Kjarsgaard B. A., Pearson D. G., Tappe S., Nowell G. M., and Dowall D. P., 2009. Geochemistry of
 hypabyssal kimberlites from Lac de Gras, Canada: comparisons to a global database and applications
 to the parent magma problem. *Lithos*, 112, 236-248.
- Klein-BenDavid O., Izraeli E.S., Hauri E., and Navon O., 2004. Mantle fluid evolution—a tale of one diamond. *Lithos*, 77(1), 243-253.
- Klein-BenDavid O., Izraeli E.S., Hauri E., and Navon O., 2007. Fluid inclusions in diamonds from the
 Diavik mine, Canada and the evolution of diamond-forming fluids. *Geochim.et Cosmochim. Acta*,
 71(3), 723-744.
- Klein-BenDavid O., Pearson D.G., Nowell G.M., Ottley C., McNeill J.C., and Cartigny P., 2010. Mixed
 fluid sources involved in diamond growth constrained by Sr–Nd–Pb–C–N isotopes and trace elements. *Earth Planet. Sci. Lett.*, 289(1), 123-133.
- Knoche R., Sweeney R. J., and Luth R. W. (1999). Carbonation and decarbonation of eclogites: the role
 of garnet. *Contrib. Mineral. Petrol.*, *135*(4), 332-339.
- Kohn M.J., and Valley J.W., 1998. Effects of cation substitutions in garnet and pyroxene on equilibrium
 oxygen isotope fractionations. J. Meta. Geol., 16(5), 625-639.
- Konrad-Schmolke M., Zack T., O'Brien P.J., and Jacob D.E., 2008. Combined thermodynamic and rare
 earth element modelling of garnet growth during subduction: Examples from ultrahigh-pressure
 eclogite of the Western Gneiss Region, Norway. *Earth Planet. Sci. Lett.*, 272(1), 488-498.
- Kramers J.D., 1979. Lead, uranium, strontium, potassium and rubidium in inclusion-bearing diamonds
 and mantle-derived xenoliths from southern Africa, *Earth Planet. Sci. Lett.*, 42, 58-70.

- MacGregor I.D., and Carter J.L., 1970. The chemistry of clinopyroxenes and garnets of eclogite and peridotite xenoliths from the Roberts Victor Mine, South Africa. *Phys. Earth Planet. Interiors*, 3, 391-397.
 MacGregor I.D., and Manton W.I., 1986. Roberts Victor Eclogites: Ancient Oceanic Crust. *J. Geophys. Res.*, 91(B14), 14,063-14,079.
- Mattey D., Lowry D., and Macpherson C. (1994). Oxygen isotope composition of mantle peridotite.
 Earth Planet. Sci. Lett., *128*(3), 231-241.
- McCandless T.E., and Gurney J.J., 1989. Sodium in garnet and potassium in clinopyroxene; criteria for
 classifying mantle eclogites. *In: Ross J, (Ed) Kimberlites and related rocks. Their mantle/crustal setting, diamonds and diamond exploration*, vol. 2, Blackwell, Carlton, Australia, pp. 827-832.
- McNeill J., Pearson D.G., Klein-BenDavid O., Nowell G.M., Ottley C.J., and Chinn I., 2009. Quantitative
 analysis of trace element concentrations in some gem-quality diamonds. J. Phys. Cond. Matt., 21(36),
 364207.
- Navon O., Hutcheon I.D., Rossman G.R., and Wasserburg G.J., 1988. Mantle-derived fluids in diamond
 micro-inclusions. *Nature*, 355, 784-789.

Neal C.R., and Taylor L.A., 1990. Comment on "Mantle eclogites: evidence of igneous fractionation in
the mantle" by J.R Smyth, F.A. Caporuscio, and T.C. McCormick. *Earth Planet. Sci. Lett.*, 101, 112119.

- Neal C.R., Taylor L.A., Davidson J.P., Holden P., Halliday A.N., Nixon P.H., Paces J.B., Clayton R.N.,
 and Mayeda T.K., 1990. Eclogites with oceanic crustal and mantle signatures from the Bellsbank
 kimberlite, South Africa, part 2: Sr, Nd, and O isotope geochemistry. *Earth Planet. Sci. Lett.*, 99, 362379.
- 682 O'Neil J. R., 1986. Theoretical and experimental aspects of isotopic fractionation. *Rev. Mineral.* 683 *Geochem.*, 16(1), 1-40.
- O'Neil J. R., Clayton R. N., and Mayeda T. K., 1969. Oxygen isotope fractionation in divalent metal
 carbonates. *J. Chem. Phys.*, 51, 5547-5558.
- O'Reilly S.Y., and Griffin W.F., 1995. Trace-element partitioning between garnet and clinopyroxene in
 mantle-derived pyroxenites and eclogites: P-T-X controls. *Chem. Geol.*, 121(1-4), 105-130.
- Ongley J.S., Basu A.R., and Kyser T.K., 1987. Oxygen isotopes in coexisting garnets, clinopyroxenes and
 phlogopites of Roberts Victor eclogites: implications for petrogenesis and mantle metasomatism.
 Earth Planet. Sci. Lett., 83, 80-84.
- Page F.S., Kita N. T., and Valley J.W., 2010. Ion microprobe analyses of oxygen isotopes in garnets of complex chemistry. *Chem. Geol.*, 270, 9-19.
- Pearson D. G., and Wittig N., 2008. Formation of Archaean continental lithosphere and its diamonds: the
 root of the problem. *J.Geol. Soc.*, 165(5), 895-914.
- Pearson D. G., Davies G. R., Nixon P. H., Greenwood P. B., and Mattey D. P. 1991, Oxygen isotope
 evidence for the origin of pyroxenites in the Beni Bousera peridotite massif, North Morocco:
 derivation from subducted oceanic lithosphere. *Earth Planet. Sci. Lett.*, 102(3), 289-301.
- Pearson D. G., Snyder G. A., Shirey S. B., Taylor L. A., Carlson R. W., and Sobolev N. V., 1995.
 Archaean Re–Os age for Siberian eclogites and constraints on Archaean tectonics. *Nature*, *374*(6524),
- 700 711-713.

- Pernet-Fisher J.F., Howarth G.H., Liu Y., Barry P.H., Carmody L., Valley J.W., Bodnar R.J., Spetsius
 Z.V., and Taylor L. A., 2014. Komsomolskaya diamondiferous eclogites: evidence for oceanic crustal
 protoliths. *Contrib. Min. Petro.*, 167(3), 1-17.
- Polyakov V.B., and Kharlashina N.N.,1995. The use of heat capacity data to calculate carbon isotope
 fractionation between graphite, diamond, and carbon dioxide: a new approach. *Geochim. Cosmochim. Acta*, 59(12), 2561-2572.
- Riches A.J.V., Liu Y., Day J.M.D., Spetsius Z.V., and Taylor L.A., 2010, Subducted Oceanic Crust As
 Diamond Hosts Revealed By Garnets Of Mantle Xenoliths From Nyurbinskaya, Siberia, *Lithos*,
 120(3-4), 368-378.
- Rumble D., Miller M. F., Franchi I. A., and Greenwood R. C., 2007. Oxygen three-isotope fractionation
 lines in terrestrial silicate minerals: An inter-laboratory comparison of hydrothermal quartz and
 eclogitic garnet. Geochim. Cosmochim. Acta, 71(14), 3592-3600.
- 713Russell A.K., Kitajima K., Strickland A., Medaris Jr. L.G., Schulze D.J., and Valley J.W., 2013. Eclogite-714facies fluid infiltration: constraints from δ^{18} O zoning in granet. *Contrib. Mineral. Petrol.*, 165, 103-715116.
- Schmidberger S. S., Simonetti A., Heaman L. M., Creaser R. A., and Whiteford S., 2007. Lu–Hf, *in situ*Sr and Pb isotope and trace element systematics for mantle eclogites from the Diavik diamond mine:
 Evidence for Paleoproterozoic subduction beneath the Slave craton, Canada. *Earth Planet. Sci. Lett.*,
 254(1), 55-68.
- Schmitz M. D., Bowring S. A., de Wit M. J., and Gartz V., 2004. Subduction and terrane collision
 stabilize the western Kaapvaal craton tectosphere 2.9 billion years ago. *Earth Planet. Sci Lett.*, 222(2),
 363-376.
- Schulze D.J., Valley J.W., and Spicuzza M.J., 2000. Coesite eclogites from the Roberts Victor kimberlite,
 South Africa. *Lithos*, 54, 23-32.
- Schulze, D. J., Harte, B., Page, F. Z., Valley, J. W., Channer, D. M. D., and Jaques, A. L., 2013.
 Anticorrelation between low δ13C of eclogitic diamonds and high δ18O of their coesite and garnet
 inclusions requires a subduction origin. *Geology*, 41(4), 455-458.
- Smart K.A., Heaman L.M., Chacko T., Simonetti A., Kopylova M., Mah D., and Daniels D., 2009. The
 origin of high-MgO diamond eclogites from the Jericho Kimberlite, Canada. *Earth Planet. Sci. Lett.*,
 284, 527-537.
- Smit K. V., Stachel T., Creaser R. A., Ickert R. B., DuFrane S. A., Stern R. A., and Seller M., 2014.
 Origin of eclogite and pyroxenite xenoliths from the Victor kimberlite, Canada, and implications for
 Superior craton formation. *Geochim. Cosmochim. Acta*, 125, 308-337.
- Snyder G.A., Taylor L.A., Jerde E.A., Clayton R.N., Mayeda T.K., Deines P., Rossman G.R., and
 Sobolev N.V., 1995. Archean mantle heterogeneity and the origin of diamondiferous eclogites,
 Siberia: Evidence from stable isotopes and hydroxyl in garnet. *Amer. Min.*, 80, 799-809.
- Snyder G.A., Taylor L.A., Crozaz G., Halliday A.N., Beard B.L., Sobolev V.N., and Sobolev N.V., 1997.
 The origins of Yakutian eclogite xenoliths. *J. Petrol.*, 38 (1), 85–113.
- Snyder G. A., Taylor L. A., Beard B. L., Crozaz G., Halliday A. N., Sobolev V. N., and Sobolev N. V.,
 1998. Reply to a comment by D. Jacob et al. on 'The Origins of Yakutian Eclogite Xenoliths'. *J. Petrol.*, *39*(8), 1535-1543.

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- Shirey S.B., Carlson R.W., Richardson S.H., Menzies A., Gurney J.J., Pearson D.G., Harris J.W., and Wiechert U., 2001. Archean emplacement of eclogitic components into lithospheric mantle during formation of the Kaapvaal Craton. Geophys. Res. Lett., 28(13), 2509-2512. Shu O., and Brey G. P., 2015. Ancient mantle metasomatism recorded in subcalcic garnet xenocrysts: Temporal links between mantle metasomatism, diamond growth and crustal tectonomagmatism. Earth Planet. Sci. Lett., 418, 27-39. Shu Q., Brey G. P., Gerdes A., and Hoefer H. E., 2014. Mantle eclogites and garnet pyroxenites-the meaning of two-point isochrons, Sm–Nd and Lu–Hf closure temperatures and the cooling of the subcratonic mantle. Earth Planet. Sci. Lett., 389, 143-154. Shu O., Brev G. P., Gerdes A., and Hoefer H. E., 2013. Geochronological and geochemical constraints on the formation and evolution of the mantle underneath the Kaapvaal craton: Lu-Hf and Sm-Nd systematics of subcalcic garnets from highly depleted peridotites. Geochim Cosmochim Acta, 113, 1-20. Sobolev V. N., Taylor L. A., Snyder G. A. and Sobolev N. V., 1994. International Geological Congress, Beijing: University of Beijing, Diamondiferous eclogites from the Udachnaya kimberlite pipe, p. 359. Yakutia. Int. Geol. Rev., 36, 42-64 Sobolev N.V., Snyder G.A., Taylor L.A., Keller R.A., Yefimova E.S., Sobolev V.N., and Shimizu N., 1998. Extreme chemical diversity in the mantle during eclogitic diamond formation: evidence from 35 garnet and 5 pyroxene inclusions in a single diamond. Int. Geol. Rev., 40(7), 567-578. Spetsius Z.V., and Taylor L.A., 2008. Diamonds of Siberia: photographic evidence for their origin. Knoxville, TN: Tranquility Base Press, pp 278. Spetsius Z.V., Taylor L.A., Valley J.W., Deangelis M.T., Spicuzza M., Ivanov A.D., and Banzerak V.I., 2008. Diamondiferous xenoliths from crustal subduction: garnet oxygen isotopes from the Nvurbinskava pipe. Yakutia. Euro. J. Mineral., 20, 375–385. Stachel T., and Harris J.W., 2008. The origin of cratonic diamonds—constraints from mineral inclusions. Ore Geol. Rev., 34(1), 5-32. Stachel T., Aulbach S., Brey G.P., Harris J.W., Leost I., Tappert R., and Viljoen K.S. 2004. The trace element composition of silicate inclusions in diamonds: a review. Lithos, 77(1), 1-19. Tappe S., Smart K. A., Pearson D. G., Steenfelt A., and Simonetti A., 2011. Craton formation in Late Archean subduction zones revealed by first Greenland eclogites. *Geology*, 39(12), 1103-1106. Tappert R., Stachel T., Harris J. W., Muehlenbachs K., Ludwig T., and Brev G. P., 2005. Subducting oceanic crust: the source of deep diamonds. Geology, 33(7), 565-568. Taylor H. P., 1977, Water/rock interactions and the origin of H₂O in granitic batholiths. J. Geol. Soc., 133(6), 509-558. Taylor L.A., and Neal C.R., 1989. Eclogites with oceanic crustal and mantle signatures from the
- Taylor L.A., and Neal C.R., 1989. Eclogites with oceanic crustal and mantle signatures from the
 Bellsbank kimberlite, South Africa, Part I: Mineralogy, petrography, and whole-rock chemistry. *J. Geol.*, 97(5), 551-567.
- Taylor L.A., Snyder G.A., Crozaz G., Sobolev V.N., Yefimova E.S., and Sobolev N.V., 1996. Eclogitic
 inclusions in diamonds: evidence of complex mantle processes over time. *Earth Planet. Sci. Lett.*,
 142(3), 535-551.

- Taylor L. A., Milledge H. J., Bulanova G. P., Snyder G. A., and Keller R. A., 1998. Metasomatic
 eclogitic diamond growth: evidence from multiple diamond inclusions. *Int. Geol. Rev.*, 40(8), 663-676.
- Taylor L.A., Keller R.A., Snyder G.A., Wang W., Carlson W.D., Hauri E.H., McCandless T., Kim K-R.,
 Sobolev N.V., and Bezborodov S.M., 2000. Diamonds and their mineral inclusions, and what they tell
 us: A detailed "pull-apart" of a diamondiferous eclogite. *Int. Geol. Rev.* 42(11), 959-983.
- Tomlinson E.L., Jones A.P., and Harris J.W., 2006. Co-existing fluid and silicate inclusions in mantle
 diamond. *Earth Planet. Sci. Lett.*, 250(3), 581-595.
- Turner G., Burgess R., and Bannon M., 1990. Volatile-rich mantle fluids inferred from inclusions in diamond and mantle xenoliths. *Nature*, 344(6267), 653-655.
- Viljoen K.S., Schulze D.J., and Quadling A.G., 2005. Contrasting Group I and Group II eclogite xenolith
 petrogenesis: Petrological, trace-element and isotopic evidence from eclogite, garnet-websterite and
 alkremite xenoliths in the Kaalvallei Kimberlite, South Africa. *J. Petrol.*, 46(10), 2059-2090.
- Wang S. J., Teng F. Z., Williams H. M., and Li S. G., 2012. Magnesium isotopic variations in cratonic
 eclogites: origins and implications. *Earth Planet. Sci. Lett.*, 359, 219-226.

Williams H. M., Nielsen S. G., Renac C., Griffin W. L., O'Reilly S. Y., McCammon C. A., Pearson N.,
Viljoen F., Alt J.C., and Halliday, A. N., 2009. Fractionation of oxygen and iron isotopes by partial
melting processes: implications for the interpretation of stable isotope signatures in mafic rocks. *Earth Planet. Sci. Lett.*, 283(1), 156-166.

- Zack T., Foley S.F., and Rivers T., 2002. Equilibrium and disequilibrium trace element partitioning in
 hydrous eclogites (Trescolmen, Central Alps). *J. Petrol.*, 43(10), 1947-1974.
- Zhang H.F., Mattey D.P., Grassineau N., Lowry D., Brownless M, Gurney J. J., and Menzies M.A., 2000.
 Recent fluid processes in the Kaapvaal Craton, South Africa: coupled oxygen isotope and trace
 element disequilibrium in polymict peridotites. *Earth Planet. Sci. Lett.*, 176, 57-72.
- Zheng Y.-F., 1993. Calculation of oxygen isotope fractionation in anhydrous silicate minerals. *Geochim. Cosmochim. Acta*, 57(5), 1079-1091.
- Zheng Y. F., Fu B., Gong B., and Li L., 2003. Stable isotope geochemistry of ultrahigh pressure
 metamorphic rocks from the Dabie–Sulu orogen in China: implications for geodynamics and fluid
 regime. *Earth Sci. Rev.*, 62(1), 105-161.

811 FIGURES AND CAPTIONS



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Figure 1: Simplified geological map of south eastern Africa. Location of the Roberts Victor
mine is marked by the red star. This map depicts some of the major structural features within the
Kaapvaal Craton. Image after Schmitz *et al.* (2004).

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820 Figure 2: Hand-specimen (a) and back-scattered electron (BSE) image of 09RV09 (CCIM 821 sample #S1624). Garnet and clinopyroxene modal abundances are heterogeneously distributed 822 at the slice and hand-specimen scale. Sample portions extracted for in situ analyses are 823 delineated by yellow and blue open-circles in (a), and S1624X labels correspond to sub-portion 824 identifiers. The blue circle corresponds to a sample portion in which diamonds was preserved after polishing. Small diamond successfully retained in situ in sub-portion S1624P is intimately 825 826 associated with garnet (b). Phase abbreviations are: Gt = garnet, Pyx = pyroxene, Phlog =827 phlogopite, Sul = sulphide, Di = diamond.





Figure 3: Garnet end-member (a; mol. %) and cation ratios (b). The range of intra- and inter-grain garnet compositions in 09RV09 shown in (a) is enclosed by a black diamond, and these data are compared with the range of garnet compositions reported in previous studies of Roberts Victor eclogites. Data fields in (a) delineate textural Group I and Group II eclogite xenoliths of Roberts Victor studied by Hatton (1978). Lit = literature, and corresponds to data reported by MacGregor and Manton (1986), O'Reilly and Griffin (1995), and Gréau *et al.* (2011). The range of intra- and inter-grain major-element cation values in garnets of 09RV09 is shown in (b), and the standard deviation (SD) was calculated via propagation of typical uncertainties on Ca, Mg, and Fe^{total} (this represents a minimum value as propagated uncertainties on other cations are not included).





Figure 4: Rare-earth-element and extended trace-element abundances of 09RV09 minerals
normalised to the values of CI-Chondrite and primitive mantle reported by McDonough and Sun
(1995). Propagated uncertainties include 2σ precision values determined for each analytical
point. Pm is not measured and is shown as an interpolated space between Nd and Sm (a).



Figure 5: a) Garnet cation compositions and corresponding La/Yb. Propagated uncertainties on 843 La/Yb reflect the 2σ internal precision determined for each analytical point. The typical 844 uncertainty on 100Mg/(Mg+Fe^{total}) represents a minimum value as propagated uncertainties on 845 other cation proportions (e.g., Si) are not included (where SD = standard deviation). b) 846 847 Comparison of Roberts Victor garnet oxygen isotope compositions with La abundances of coexisting clinopyroxenes. Uncertainties on La abundances represent 2σ precision. Linear 848 regressions of the data of Gréau et al. (2010) alone (L1), and incorporating the 09RV09 average 849 (L2), are not strongly correlated ($R^2 \ll 0.7$). The 09RV09 clinopyroxene La abundance data 850 range is depicted by the red bar; inclusion of all 09RV09 clinopyroxene La abundance data 851 (rather than using a single average value) further reduces the correlation coefficient ($R^2 < 0.1$). 852



Figure 6: Histogram of oxygen isotope compositions determined for garnets of 09RV09 using ion probe techniques. Literature data reported for garnet separates of 62 Roberts Victor eclogite xenoliths were sourced from (Garlick *et al.*, 1971; MacGregor and Manton; 1986; Ongley *et al.*, 1987; Caporuscio, 1990; Schulze *et al.*, 2000; Gréau *et al.*, 2011) and the garnet mantle range is after Mattey *et al.*, (1994). A schematic illustration of the range of δ^{18} O-compositions determined for samples of the Samail Ophiolite (b); where δ^{18} O-compositions >+5.9 ‰ are associated with upper sections of oceanic crust altered at low-temperatures (<350 °C; after Gregory and Taylor, 1981).







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Figure 8: Model of the effect of closed- and open-system interaction between CO₂-rich and H₂O-rich metasomatic fluids and 876 peridotite on the fluid δ^{18} O-composition. Details of the model are given in the text. The ordinate of the figure is the difference 877 between the fluid composition and a composition representing a completely rock-dominated system (e.g., where the fluid composition 878 is fixed by the initial isotopic composition of the peridotite and the fractionation factor). The model curves asymptotically approach 879 zero, where any initial ¹⁸O-enrichment or depletion is effectively erased by equilibration with a large enough volume of rock. The 880 grey region marks a ± 0.25 % band around the zero value, reflecting a composition that is effectively indistinguishable from one that is 881 completely rock buffered. Under open-system conditions, and at fluid:rock ratios <1:10, the δ^{18} O-value of the fluid is 882 indistinguishable from the silicate mantle with which it is interacting. 883

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