Simulation of three component two phase flow in porous media using method

² of lines

3 Salim Goudarzi*, Simon A. Mathias and Jon G. Gluyas

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5 Received: N/A / Accepted: N/A

Abstract Numerical simulation of compositional flow problems commonly involves the use of 1st or 2nd order Euler time stepping. Method of lines (MOL), using highly accurate and efficient ODE solvers, is an alternative technique which, although frequently applied to the solution of two-phase, two-component flow problems, has generally been overlooked for problems concerning more than two components. This article presents the development of a numerical simulator for 1D, compressible, two-phase, three-component, radially symmetric flow using 10 the method of lines (MOL) and a 3rd order accurate spacial discretization using a weighted essentially non oscil-11 latory (WENO) scheme. The MOL implementation enables application of the MATLAB ODE-Solver, ODE15s, 12 for time integration. Simulation examples are presented in the context of CO₂ injection into a reservoir containing 13 a mixture of CH₄ and H₂O. Following an assumption of constant equilibrium ratios for CO₂ and CH₄, a ternary 14 flash calculator is developed providing closed-form relationships for exact interpolation between equations of 15 state for CO₂-H₂O and CH₄-H₂O binary mixtures. The numerical code is successfully tested and verified for a 16 range of scenarios by comparison to an existing analytical solution. 17

18 **1 Introduction**

¹⁹ Continuum scale simulation of multi-component multi-phase (MCMP) flow in porous media is often used to bet-²⁰ ter understand a range of reservoir pore-space exploitation problems including petroleum production, geothermal ²¹ energy, groundwater management and waste disposal. Generally it is assumed that multiple fluid phases co-exist ²² as volume fractions of an infinitesimal volume. These phases typically include aqueous liquid, non-aqueous liq-²³ uid, gas and solid. Each phase can represent a mixture of multiple components (e.g., H₂O, CO₂ and CH₄). The

^{*}Department of Earth Sciences, Durham University, Durham, UK E-mail: salim.goudarzi@durham.ac.uk

volume fraction of the *j*th phase is mathematically represented as a volume fraction of the pore-space, often referred to as the saturation, S_j [-]. Multiple components are then considered as mass fractions of the various phases. For example, X_{ij} would denote the mass fraction of the *i*th component in the *j*th phase.

Mathematical simulation of such problems involves solving a coupled set of conservation equations for each component (Orr, 2007). Equations of state (EOS) are required to understand how the components partition into different phases. This is particularly important because the mobilities of the components are strongly controlled by their distribution between the present fluid phases. Firstly, component mass fractions, X_{ij} , affect the viscosity of the phases (Centeno et al., 2011). Secondly, the permeability available to each phase is a non-linear function of the phase saturations, S_i (Mathias et al., 2013).

In the absence of diffusion and capillary pressure, this problem is governed by a set of coupled hyperbolic and parabolic transport equations. Hyperbolic equations frequently give rise to the formation of shocks, leading to difficulties with regard to obtaining accurate solutions. Problems associated with one dimensional transport of incompressible fluids, in the absence of capillary pressure and under isothermal conditions, can be solved exactly using the method of characteristics (Orr, 2007). However, even under these restricting conditions, great care must be taken when considering non-zero initial conditions and non-unity boundary conditions (for example, see section 4.3 of Orr (2007)).

Alternative techniques involve the application of approximate methods. The spatial dimension is typically treated using conservative methods such as finite volume (Chen et al., 2006). Alternatively, one can consider the use of finite elements (Chen et al., 2006) or pseudospectral methods (van Reeuwijk et al., 2009). Such spatial schemes give rise to either stability problems or numerical diffusion due to truncation terms associated with the Taylor's expansion, the latter of which can be reduced using flux limiters or their variants (e.g. Mallison et al., 2005).

Handling of the temporal term, which is critical to resolving the non-linear nature of the problem, generally
 revolves around the choice of explicit or implicit treatment. Fully explicit treatment, although easier to implement,
 can run into severe time-step limitations due to the well known CFL (Courant - Friedrichs - Lewy) condition. Fully
 implicit treatment leads to an unconditionally stable solution (as far as time-stepping is concerned), but leads to
 additional numerical diffusion. Furthermore, implementation of the solution is significantly more challenging.

⁵¹ Popular approaches for solving MCMP problems in this context are the so-called semi-implicit methods, the
 ⁵² most common variant of which is referred to as ImPES (Implicit Pressure Explicit Saturation) (Chen et al., 2006)

or ImPEM (Implicit Pressure Explicit Mass). In ImPES, the governing equations are rearranged to identify a transport equation of hyperbolic (or nearly hyperbolic) nature and a pressure equation (of parabolic or elliptic character). The pressure equation is solved implicitly which allows for larger time-steps. The transport equation is solved explicitly, allowing easier implementation and reduced computational memory requirements, hence the semi-implicitness. Both the implicit and explicit time-stepping typically employ simple first-order schemes.

⁵⁸ Multi-step-multi-order time integration algorithms (Shampine et al., 1997) represent an alternative method, ⁵⁹ which treats the temporal term in a more accurate fashion. These techniques maintain a specific time integration ⁶⁰ error while maximizing the time-step size. Moreover, due to the wide availability (e.g. MATLAB or FORTRAN ⁶¹ with NAG) of high quality solvers and simplicity of implementation, there is no need to redevelop the sophisti-⁶² cated solution algorithms. Rather, the so-called method of lines (MOL) approach can be taken. In this case, the ⁶³ partial differential equations (PDE) are discretized in space to form a set of coupled ordinary differential equa-⁶⁴ tions (Wouwer et al., 2005). These can then be solved simultaneously using any ordinary differential equation ⁶⁵ (ODE) solver of choice.

For two-phase immiscible flow, where one of the phases is treated as inviscid, the MCMP problem reduces to 66 a single PDE often referred to as Richards' equation (RE). This equation is commonly solved to better understand 67 hydrological problems associated with unsaturated soils. Numerical studies by Farthing et al. (2003) and Kees 68 and Miller (2001) have shown that applying MOL with higher-order time integration to the solution of RE leads 69 to both improved accuracy and computational efficiency. Indeed, there are many recent articles (Mathias et al., 70 2006, 2008b, Ireson et al., 2009) reporting MOL solutions of RE using the MATLAB ODE solver, ODE15s, 71 which is particularly suitable for stiff systems of ODEs (Shampine et al., 1997). ODE15s has also been found 72 to provide useful solutions to non-Darcian flow problems (Mathias et al., 2008a, 2015, Wen et al., 2009) and 73 two-phase immiscible flow problems (Mathias et al., 2009). 74

Often when dealing with partial differential equations it is useful to distinguish between dependent and independent variables (Stroud and Booth, 2007, p. 122). In this case, time and space are independent variables. All other variables are dependent variables.

⁷⁸ Application of MATLAB ODE solvers to multi-component partially miscible problems has proven more ⁷⁹ challenging. Consider N_c components residing in N_p phases. The problem will be defined by N_c mass conservation ⁸⁰ equations. However, considering the various values of S_j and X_{ij} , it can be understood that there will be at ⁸¹ least $(N_c + 1)N_p$ dependent variables. It is therefore necessary to choose N_c dependent variables to solve for.

When using an ODE solver, the user must construct an ODE function. Within this function, a scalar value 85 of time is provided as an input along with an associated vector of the PDVs. The user must define the ODE 86 function such that it calculates the derivatives of the PDVs with respect to time, which generally involves using a 87 combination of chain rule and product rule differentiation. This results in a need to evaluate the partial derivatives of the bulk fluid mass per unit pore-space, F [ML⁻³], with respect to each component mass fraction, z_i [-] (these 89 terms will be mathematically defined later in the article). For conventional first-order time-stepping, it is arguably 90 acceptable to evaluate these derivatives using first or second order finite differencing. However, given the high 91 accuracy associated with the use of MATLAB's ODE solvers, it is pertinent to obtain these derivatives in exact 92 form wherever possible. 93

There are many detailed works concerning applications of MOL for immiscible two-phase flow and two-94 component two-phase flow problems (e.g. Amaziane et al., 2012, Vohralik et al., 2013, Bourgeat et al., 2013, Mathias et al., 2014). Mallison et al. (2005) present a numerical simulation of an MCMP problem using MOL 96 in conjunction with a 3rd and 4th order Rung-Kutta time integration method. However, Mallison et al. (2005) 97 provides no discussion concerning the casting of equations in terms of PDVs. Indeed, little information is available as to how to obtain exact equations to describe the necessary partial derivatives, $\partial F/\partial z_i$, needed to solve 90 MCMP problems for situations concerning more than two components. In this article, we focus on obtaining 100 such expressions for three-component two-phase problems. These are implemented within a radial flow simula-10' tor using MATLAB. Comparisons are then made, in the context of enhanced gas recovery by CO₂ injection, with 102 an associated analytical solution, previously presented by Hosseini et al. (2012). 103

104 2 Governing equations

¹⁰⁵ Consider three components: CO₂, CH₄ and H₂O, denoted hereafter as i = 1, 2 and 3, respectively. The three ¹⁰⁶ components can partition into a gas phase and an aqueous liquid phases, denoted hereafter as j = 1 and 2, ¹⁰⁷ respectively. A horizontally orientated, homogeneous and isotropic cylindrical reservoir of radius, r_e [L] and ¹⁰⁸ formation thickness, H [L], is invoked. The reservoir is initially filled with a mixture of CH₄ and H₂O. CO₂ is ¹⁰⁹ injected into the center of the reservoir via a fully-completed vertically orientated injection well of radius, r_w dimensional radially symmetric conservation equations:

$$\frac{\partial G_i}{\partial t} = -\frac{1}{r} \frac{\partial (rH_i)}{\partial r}, \quad i = 1, 2, \dots N_c \tag{1}$$

where *t* [T] is time, *r* [L] is radial distance from the injection well, N_c [-] is the number of components considered and G_i [ML⁻³] and H_i [ML⁻²T-1] are the mass of component *i* per unit volume of rock and mass flux of component *i*, respectively, found from:

$$G_i = \phi \sum_{j=1}^{N_p} \rho_j X_{ij} S_j \tag{2}$$

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$$H_i = \sum_{j=1}^{N_p} \rho_j X_{ij} q_j \tag{3}$$

where ϕ [-] is the reservoir porosity, N_p [-] is the number of phases considered, ρ_j [ML⁻³], S_j [-] and q_j [LT⁻¹] are the density, saturation (a volume fraction of the pore-space) and volumetric fluid flux of phase *j*, respectively, and X_{ij} [-] is the mass fraction of component *i* in phase *j*. Note that $\sum_{i=1}^{N_c} X_{ij} = 1$ and $\sum_{j=1}^{N_p} S_j = 1$.

¹¹⁹ The volumetric fluxes are calculated using Darcy's law:

$$q_j = -k \frac{k_{rj}}{\mu_j} \frac{\partial P_j}{\partial r} \tag{4}$$

where k [L²] is the reservoir permeability and k_{rj} [-], μ_j [ML⁻¹T⁻¹] and P_j [ML⁻¹T⁻²] are the relative permeability, dynamic viscosity and pressure of the *j*th phase, respectively.

For two phase flow, without loss of generality, the relative permeability functions are assumed to take the form of power laws (Mathias et al., 2013):

$$k_{r1} = k_{r1_0} \left(\frac{S_1 - S_{1c}}{1 - S_{1c} - S_{2c}} \right)^{n_1}, \quad S_{1c} \le S_1 \le 1 - S_{2c}$$
(5)

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$$k_{r2} = k_{r20} \left(\frac{S_2 - S_{2c}}{1 - S_{1c} - S_{2c}} \right)^{n_2}, \quad S_{2c} \le S_2 \le 1 - S_{1c}$$
(6)

where S_{jc} [-], k_{rj_0} [-] and n_j [-] are the critical saturation, end-point relative permeability and power law exponent for phase *j*, respectively.

¹²⁷ Furthermore, it is assumed that the following density mixing rule applies (Orr, 2007)

$$\rho_j = \left(\sum_{i=1}^{N_c} \frac{X_{ij}}{\rho_{ij}}\right)^{-1} \tag{7}$$

where ρ_{ij} [ML⁻³] is the density of the *i*th component in the *j*th phase.

¹²⁹ **3** Recasting in terms of primary dependent variables

An appropriate choice of primary dependent variables (PDVs) to solve for are the global fluid pressure, P_{131} [ML⁻¹T⁻²], defined in this case by (Chen et al., 2006, p. 342)

$$P = \sum_{j=1}^{N_p} S_j P_j \tag{8}$$

and the bulk mass fraction of each component, z_i [-], defined by

$$z_i = G_i / F \tag{9}$$

where F [ML⁻³] is the bulk fluid mass per unit volume of rock

$$F = \sum_{i=1}^{N_c} G_i = \phi \sum_{j=1}^{N_p} \rho_j S_j$$
(10)

134 Note that $\sum_{i=1}^{N_c} z_i = 1$.

In some previous studies, the mass of each component per unit volume of rock, G_i , have also proven effective as PDVs in this context (Amaziane et al., 2012, Bourgeat et al., 2013). However, an advantage of using P and z_i (for $i = 1, 2, ..., N_c - 1$) as PDVs (as opposed to say G_i) is that z_i are independent of P. For a given volume of fluid mixture, the mass fractions of each component, z_i will not change with pressure. However, the associated mass of each component per volume of rock, G_i , may change with pressure, depending on how the individual component mass densities, ρ_{ij} , vary with pressure. Furthermore, z_i are the variables used in the ternary diagram (discussed later in the article), which determine the equilibrium properties of the three-component fluid mixture.

¹⁴² Differentiating Eq. (9) with respect to time leads to

$$\frac{\partial z_i}{\partial t} = \frac{1}{F} \left(\frac{\partial G_i}{\partial t} - z_i \frac{\partial F}{\partial t} \right) \tag{11}$$

¹⁴³ where, from Eqs. (1) and (10)

$$\frac{\partial F}{\partial t} = -\frac{1}{r} \sum_{i=1}^{N_c} \frac{\partial (rH_i)}{\partial r}$$
(12)

Application of the chain-rule to Eq. (10) and rearranging leads to

$$\frac{\partial P}{\partial t} = \left(\frac{\partial F}{\partial P}\right)^{-1} \left(\frac{\partial F}{\partial t} - \sum_{i=1}^{N_c - 1} \frac{\partial F}{\partial z_i} \frac{\partial z_i}{\partial t}\right)$$
(13)

The main focus of this article is the derivation and application of exact formulae for the relationships defining $\partial F/\partial P$ and $\partial F/\partial z_i$ for $i = 1, 2, ..., N_c - 1$. Note that Eqs. (11) to (13) are important because they directly relate the time-derivatives of the PDVs to the original mass conservation statements.

4 Differentiating the *F* **function**

¹⁴⁹ Considering the identity in Eq. (10), the total derivative of F can be written as

$$dF = F \frac{d\phi}{\phi} + \phi \sum_{j=1}^{N_p} S_j \rho_j \left(\frac{d\rho_j}{\rho_j} + \frac{dS_j}{S_j} \right)$$
(14)

When there are only two phases, $S_2 = 1 - S_1$. From Eqs. (2), (9) and (10), it can then be understood that

$$S_1 = \left[1 - \frac{\rho_1(z_i - X_{i1})}{\rho_2(z_i - X_{i2})}\right]^{-1}$$
(15)

¹⁵¹ which on differentiation leads to

$$\frac{dS_1}{S_1} = S_2 \sum_{j=1}^2 (-1)^j \left[\frac{d\rho_j}{\rho_j} + \left(\frac{dz_i - dX_{ij}}{z_i - X_{ij}} \right) \right]$$
(16)

¹⁵² Invoking Eq. (7), it can also be shown that

$$\frac{d\rho_j}{\rho_j} = \rho_j \sum_{i=1}^{N_c} \frac{X_{ij}}{\rho_{ij}} \left(\frac{d\rho_{ij}}{\rho_{ij}} - \frac{dX_{ij}}{X_{ij}} \right)$$
(17)

It is now assumed that component densities are unaffected by composition such that $\partial \rho_{ij}/\partial z_i = 0$. Additionally noting that

$$\frac{\partial z_i}{\partial z_k} = 0, \quad i \neq k \tag{18}$$

- the remaining challenge is to define the dX_{ij} terms.
- The terms X_{ij} can be further defined by:

$$X_{ij} = \begin{cases} z_i, & S_1 = 0, 1 \\ \\ x_{ij}, & 0 < S_1 < 1 \end{cases}$$
(19)

- where x_{ij} [-] are the equilibrium mass fractions of the *i*th component in the *j*th phase within the two-phase region.
- 158 Then dX_{ij} is given by:

$$\frac{\partial X_{ij}}{\partial z_i} = \begin{cases}
1, & S_1 = 0, 1 \\
\frac{\partial x_{ij}}{\partial z_i}, & 0 < S_1 < 1
\end{cases}$$

$$\frac{\partial X_{ij}}{\partial P} = \begin{cases}
0, & S_1 = 0, 1 \\
\frac{\partial x_{ij}}{\partial P}, & 0 < S_1 < 1
\end{cases}$$
(20)
(21)

It is important to note that for two component two phase problems, x_{ij} only varies with pressure. However, for three component two phase systems, the problem is much more complicated because the equilibrium constants, x_{ij} , are no longer constant with composition (i.e., z_1 and z_2).

EOSs for the binary mixtures of CO₂-H₂O and CH₄-H₂O, are well characterized by relatively simple sets of equations. In this work the binary mixture properties of CO₂-H₂O and CH₄-H₂O were obtained using expressions provided by Spycher et al. (2003) and Duan and Mao (2006), respectively. These can be joined together to form a ternary system by assuming that the equilibrium ratios, K_i [-], defined as

$$K_i = \frac{x_{i1}}{x_{i2}},\tag{22}$$

¹⁶⁶ for components 1 and 2 (but not 3) do not vary with composition.

¹⁶⁷ When K_1 and K_2 are constant, the two-phase region, for a given temperature and pressure, is defined by two ¹⁶⁸ straight lines on a ternary diagram (see figure 1). Consequently, following the work of Juanes (2008), it can be ¹⁶⁹ shown that values of x_{ij} can be related back to their values obtained from the binary mixtures, a_{ij} , by the set of ¹⁷⁰ linear equations:

$$x_{1j} = Aa_{1j}$$
 and $x_{2j} = (1-A)a_{2j}$ (23)

where $a_{1j} = x_{1j}$ when $z_2 = 0$, $a_{2j} = x_{2j}$ when $z_1 = 0$ and A [-] is a weighting parameter that linearly interpolates between the bounding tie-lines that coincide with the z_2 and z_3 axes of the ternary diagram. Writing out Eq. (23) for components 1 and 2 and eliminating A leads to:

$$a_{i1}x_{i2} = a_{i2}x_{i1}$$
 and $x_{2i} = (a_{1i} - x_{1i})a_{2i}/a_{1i}$ (24)

¹⁷⁴ Similarly it can be said that, on a given tie-line in the two-phase region,

$$z_i = Bx_{i1} + (1 - B)x_{i2} \tag{25}$$



Fig. 1 Illustration of a two phase three component system on a ternary diagram. Tie-lines are shown as dashed lines. The barycentric coordinates of points of the ternary diagram correspond to the total composition of the fluid mixture.

where B [-] is a weighting parameter that linearly interpolates between the bounding lines that define the twophase region of the ternary diagram. Writing Eq. (25) out for components 1 and 2 and then eliminating B leads to

$$(z_1 - x_{12})(x_{21} - x_{22}) = (z_2 - x_{22})(x_{11} - x_{12})$$
(26)

Note that the equilibrium ratio for H_2O (component 3) is not necessarily constant and is defined by the

179 relationship

$$K_3 = \frac{1 - x_{11} - x_{21}}{1 - x_{12} - x_{22}} \tag{27}$$

Using Eq. (24) to eliminate the x_{i2} and x_{2j} terms yields the quadratic equation

$$0 = T_2 x_{11}^2 + T_1 x_{11} + T_0 \tag{28}$$

181 where

$$T_0 = (a_{21} - a_{22})a_{11}z_1 \tag{29}$$

$$T_1 = (a_{11} - a_{12})(a_{22} - z_2) - (a_{21} - a_{22})(a_{12} + z_1)$$
(30)

$$T_2 = (a_{12}a_{21} - a_{11}a_{22})/a_{11} \tag{31}$$

182 which has the solutions

$$x_{11} = \frac{-T_1 \pm \sqrt{T_1^2 - 4T_0 T_2}}{2T_2} \tag{32}$$

Eq. (32) gives an explicit expression for x_{11} with respect to z_1 and z_2 . Furthermore, from Eq. (28) we have the total derivative of x_{11} :

$$dx_{11} = -\frac{x_{11}^2 dT_2 + x_{11} dT_1 + dT_0}{2T_2 x_{11} + T_1}$$
(33)

185 where

$$dT_0 = (da_{21} - da_{22})a_{11z_1} + (a_{21} - a_{22})(a_{11}dz_1 + da_{11}z_1)$$
(34)

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$$-(da_{21} - da_{22})(a_{12} + z_1) - (a_{21} - a_{22})(da_{12} + dz_1)$$
(35)

187

$$dT_2 = da_{12}\frac{a_{21}}{a_{11}} + da_{21}\frac{a_{12}}{a_{11}} - da_{11}\frac{a_{12}a_{21}}{a_{11}^2} - da_{22}$$
(36)

which, in conjunction with Eq. (24), provides expressions for all the other derivatives, dx_{ij} . Expressions for dX_{ij}

 $dT_1 = (da_{11} - da_{12})(a_{22} - z_2) + (a_{11} - a_{12})(da_{22} - dz_2)$

can then be obtained from Eqs. (21) and (20).

¹⁹⁰ 5 Additional considerations required for accommodating capillary pressure

¹⁹¹ When capillary pressures can be assumed negligible, the information in this section can be ignored. However, for ¹⁹² non-negligible capillary pressure, the fluid properties x_{ij} and ρ_{ij} should be calculated from the phase pressure, P_j , ¹⁹³ as opposed to the global pressure, P. An equation of state can provide derivatives of these variables with respect ¹⁹⁴ to P_j . But to obtain derivatives with respect to P, the following transformations must be applied:

$$\frac{\partial x_{ij}}{\partial P} = \frac{\partial P_j}{\partial P} \frac{\partial x_{ij}}{\partial P_j}$$
(37)

$$\frac{\partial \rho_{ij}}{\partial P} = \frac{\partial P_j}{\partial P} \frac{\partial \rho_{ij}}{\partial P_j}$$
(38)

¹⁹⁵ For two phase flow systems, Eq. (8) reduces to

$$P_j = P - (-1)^j (1 - S_j) P_c \tag{39}$$

where $P_c = P_1 - P_2$ is the capillary pressure.

¹⁹⁷ Noting that P_c is generally expressed uniquely as a function of S_1 (e.g. van Genuchten, 1980) and that $dS_1 = -dS_2$, differentiating Eq. (39) with respect to P leads to

$$\frac{\partial P_j}{\partial P} = 1 - \left(P_c + (-1)^j (1 - S_j) \frac{\partial P_c}{\partial S_1} \right) \frac{\partial S_1}{\partial P}$$
(40)

Recalling Eq. (16) and that z_i are independent of P it can be said that

$$\frac{1}{S_1}\frac{\partial S_1}{\partial P} = S_2 \sum_{j=1}^2 Y_j \frac{\partial P_j}{\partial P}$$
(41)

200 where

$$Y_j = (-1)^j \left[\frac{1}{\rho_j} \frac{\partial \rho_j}{\partial P_j} - \frac{1}{(z_i - X_{ij})} \frac{\partial X_{ij}}{\partial P_j} \right]$$
(42)

201 from which we obtain

$$\frac{\partial S_1}{\partial P} = \left[\frac{1}{S_1 S_2} + \sum_{j=1}^2 \left(P_c + (-1)^j (1 - S_j) \frac{\partial P_c}{\partial S_1}\right) Y_j\right]^{-1} \sum_{j=1}^2 Y_j \tag{43}$$

202 6 Numerical solution

A weighted essentially non oscillatory (WENO) scheme (Shu, 2009) was used to approximate the flux, H_i , in Eq. (1). For more details on the WENO method and its applications, the reader is referred to (Coralic and Colonius, 2014, Noelle et al., 2007, Zhang and Shu, 2012) and references therein. The WENO scheme implemented here 205 uses a 2nd order stencil to approximate the flux at the mid-points (k + 1/2 and k - 1/2). A central difference 207 approximation is then used to evaluate the derivative of the flux at grid point k, which is 3rd order accurate in 208 space (Shu, 2009):

$$\left(\frac{\partial G_i}{\partial t}\right)_k = -\frac{1}{r_k} \left[\frac{r_{k+\frac{1}{2}}(H_i)_{k+\frac{1}{2}} - r_{k-\frac{1}{2}}(H_i)_{k-\frac{1}{2}}}{r_{k+\frac{1}{2}} - r_{k-\frac{1}{2}}} \right]$$
(44)

The pressure derivative contained within the $(H_i)_{k+\frac{1}{2}}$ term was approximated using the following central difference expression:

$$\left(\frac{\partial P}{\partial r}\right)_{k+\frac{1}{2}} = \frac{P_{k+1} - P_k}{r_{k+1} - r_k} \tag{45}$$

The Jacobian matrix is a matrix describing the dependencies of each equation in the system of equations on every other equation. In this problem we are solving for three PDVs, which by virtue of the discretised conservation equations, are mutually dependent on one another. Conventional second-order finite difference approximations of second-order derivatives at a grid point, k, will depend on three points, namely k - 1, k and k + 1. However, when using the WENO scheme described above, additional information from k - 2 and k + 2 is required, leading to a penta-diagonal structure for the Jacobian matrix. For systems of equations with more than one PDV, a so-called block-diagonal structure is formed.

In MATLAB, it is possible to provide the ODE solver with the Jacobian pattern as a sparse matrix of zeros and ones to indicate where the Jacobian matrix is nonzero. This tells the solver to only evaluate the sparse system and not the full matrix. In stiff problems with non-constant Jacobian matrices, specifying the Jacobian pattern a priori can lead to significant savings in computation time. A pictorial representation of the Jacobian pattern for the problem discussed in this article is presented in Fig. 2.

Spatial grids were locally refined near the well-bore and gradually coarsened away from the injection well to ensure simulation stability. Simulation time steps are refined automatically using MATLAB's ODE15s adaptive time stepping scheme. ODE15s is particularly suitable for stiff problems where the governing equations include a combination of terms that lead to rapid variation in the solution and terms that vary slowly. For the problem



Fig. 2 Block-penta-diagonal Jacobian pattern for the sparse system for n=10 grid points and 3 PDVs (nz:number of nonzero elements).

described in this article, compositions propagate very slowly thorough the reservoir compared to the pressure

²²⁸ waves, which move much more rapidly.

ODE15s is a multi-order multi-step solver. The scalar relative error tolerance and the absolute error tolerance are set to MATLABs default values, 10^{-3} and 10^{-6} , respectively. For more information about how the solver is implemented, readers are referred to Shampine et al. (1997, 2001).

232 7 Model verification

- 233 Results from the numerical solution (implemented in MATLAB) were compared against results from the an-
- alytical solution by Hosseini et al. (2012). Both models assume a fully completed well at the center of a 1D
- radially symmetric flow field of radial extent, r_e . Capillary pressure, molecular diffusion and gravity effects were
- ²³⁶ neglected. Initial and boundary conditions were applied as follows:

Parameter (unit)	Deep	Shallow
P_0 , initial pressure (MPa)	32	10
T , temperature (o C)	124	40
M_0 , injection rate (kg/s)	16.21	16.21
r_w , well radius (m)	0.25	0.25
r_e , reservoir radius (km)	13.7	13.7
φ, porosity (-)	0.2	0.2
k, permeability (m ²)	10^{-13}	10^{-13}
<i>H</i> . formation thickness (m)	50	50

Table 1 Model parameters used for verification of the numerical simulation.

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Parameter	
S_{1c} , critical gas saturation	0.10
S_{2c} , residual aqueous phase saturation	0.50
k_{r10} , gas end-point relative permeability	0.30
k_{r20} , liquid end-point relative permeability	0.512
$n_1 = n_2$, power-law exponents	3.0

Table 2 Relative permeability model parameters used in Eqs. (5) and (6).

Δ

$$\begin{aligned} & r_{w} \leq r \leq r_{e}, t = 0 \\ S_{1} = S_{10}, & r_{w} \leq r \leq r_{e}, t = 0 \\ P = P_{0}, & r_{w} \leq r \leq r_{e}, t = 0 \\ H_{1} = M_{0}/(2\pi r_{w}H), r = r_{w}, & t > 0 \\ H_{2} = 0, & r = r_{w}, & t > 0 \\ H_{3} = 0, & r = r_{e}, & t > 0 \\ H_{1} = 0, & r = r_{e}, & t > 0 \\ H_{1} = 0, & r = r_{e}, & t > 0 \\ H_{2} = 0, & r = r_{e}, & t > 0 \\ H_{3} = 0, & r = r_{e}, & t > 0 \\ H_{3} = 0, & r = r_{e}, & t > 0 \end{aligned}$$

· •

where S_{10} [-] is the initial gas saturation and the other parameters are as specified in tables 1 and 2.

Fig. 3 shows gas saturation profiles for CO_2 injection into deep (Figs. 3a and 3b) and shallow (Figs. 3c and 3d) reservoirs with 10% residual CH_4 , during a simulation period of 1,000 days. The solid lines and dashed lines represent results from the analytical solution and numerical solution, respectively.

The high gas saturation around the injection well, often referred to as a dry-out zone (Mathias et al., 2011), is due to the vaporization of the residual water saturation by the injected CO_2 . A CH₄ bank with about 22% gas saturation in front of the CO₂ plume develops. The length of the CH₄ bank increases with time. Correspondence between the analytical solution and the numerical solution is very good for both gas phase saturation and pressure



10²

r (m)

10⁴

10

10⁰

Fig. 3 a) Comparison of gas saturation profiles between the analytical and numerical simulation of CO₂ injection into a deep reservoir with initial gas saturation, $S_{10} = 0.1$, after 10, 100 and 1,000 days. b) Corresponding pressure profiles for deep reservoir. c) Gas saturation profiles for a shallow reservoir. d) Corresponding pressure profiles for of a shallow reservoir. A total of 300 nodes were used for the numerical simulation.



Solid: Analytical Solution

Dashed: Numerical Simulation



²⁴⁵ buildup. The numerical solution is seen to accurately locate the associated shock fronts while considering the ²⁴⁶ partial miscibility of both CO₂ and CH₄ in H₂O.

For the shallow reservoir, just ahead of the CH₄ bank, it can be seen that the numerical solution predicts a slightly lower gas saturation as compared to the analytical solution. This is due to the fact that the analytical solution assumes constant fluid properties and hence is not capturing volume change effects due to pressure change. This discrepancy is not noticeable for the deep reservoir because the gas compressibility is a lot lower at 34 MPa.

The possibility that the above mentioned discrepancy was due to insufficient mesh refinement was investigated by comparing simulation results using a 600 point grid and a 900 point grid. The comparison study confirmed that the discrepancy was not due to numerical error.

Similar simulations but with different initial gas saturations are compared in Fig. 4. It is found that the extent of the dry-out region is insensitive to the initial gas saturation. The extent of the dry-out region is smaller for the shallow reservoir and the volume of the gas plume is larger. The reduced dry-out region, in this case, is due to the reduced evaporation that occurs at cooler temperatures. The increased gas volume is due to the reduced gas density that occurs at lower pressures. Again it can be seen that the numerical solution is able to accurately predict the analytical results of Hosseini et al. (2012).

261 7.1 Numerical solution performance

The performance of the MOL numerical solution was explored further through a grid convergence study. The shallow reservoir scenario, depicted in figures 3c and d, was repeated using different numbers of grid cells. Numerical solution performance was quantified by calculating the mean absolute normalized error (MANE) between results (from each grid cell) from the numerical solution and those from the analytical solution for a given time. Figure 5 shows a plot of MANE for gas saturation and pressure for the different times previously presented in figures 3c and d. Grid convergence can be seen to have been achieved at around 300 grid cells.

The converged MANE for pressure is around 0.02%. However, the converged MANE for gas saturation is quite high at around 0.25%. This is due to conceptual differences between the numerical solution and the analytical solution. Recall that the analytical solution assumes constant fluid properties whereas the numerical solution is allowing for variations of fluid properties with pressure and composition, as discussed above. Hosseini et al. (2012) was able to achieve a very similar level of accuracy for a very similar set of simulations using the al. (2012)). GEM uses an adaptive-implicit solver as described by Colins et al. (1992).

In terms of computation time, the 300 grid cell numerical solution using our new MOL approach was able to simulate 1000 days of gas injection in around 5 minutes using an Intel Xeon CPU E5-2630 2.30 GHz (2 processors). For comparison, the GEM simulations undertaken by Hosseini et al. (2012) took around 2 hours on an Intel Xeon CPU E5-2687 3.10 GHz (2 processors) (Hosseini, 2015, Personal Communications). Clearly, our new MOL solution has the potential to offer significant computation time savings in this context.



Fig. 5 Mean absolute normalized error (MANE) between the numerical and analytical pressure and gas saturation values for different times, plotted against the number of grid cells used for the shallow-reservoir case previously depicted in figures 3c and d.

280 8 Effect of initial gas saturation

- As was shown in the previous section, numerical simulations of CO₂ injection into a reservoir containing CH₄
- predict the accumulation of a CH₄ bank at the head of the CO₂ plume (Oldenburg and Doughty, 2011, Battistelli
- and Marcolini, 2009, Taggart, 2010).
- The system in discussion can be differentiated into three regions (Hosseini et al., 2012). These regions,
- starting from the injection point and moving outward, are:
- $_{286}$ 1. a single-phase, dry-out region around the well-bore filled with pure CO₂.
- $_{287}$ 2. a two-phase, two-component system containing CO₂ and H₂O.
- $_{288}$ 3. a two-phase, two-component system containing CH₄ and H₂O.

- Within the two-phase mixture, each phase propagates at a rate according to its mobility. The mobility of each
- ²⁹⁰ phase varies from one region to another due to associated compositional changes. As a consequence, a trailing
- shock forms at the contact between regions (1) and (2) and a leading shock forms at the contact between regions
- ²⁹² (2) and (3).



Fig. 6 Schematic diagram illustrating the three-region system associated with CO₂ injection into a reservoir initially containing CH₄ and H₂O. J:Injection, T:Trailing Shock, L:Leading Shock, G:Gas Bank, I:Initial.

The development of the CH_4 bank ahead of the CO_2 has been explained as follows (Taggart, 2010, Oldenburg 293 and Doughty, 2011, Hosseini et al., 2012). As CO_2 is injected, it partitions into the gas phase and the aqueous 294 phase. The initially dissolved CH_4 exsolves immediately, and is then pushed ahead of the growing CO₂ plume 295 leading to the development of a CH₄ bank (Oldenburg and Doughty, 2011). Mathematically, the system is con-296 strained to constantly enter and leave the two phase region along the tie-lines representing the injection and initial 297 compositions, therefore the leading CH₄ bank is free from injected gas, CO₂ (Taggart, 2010). 298 Intuitively, it is expected that the amount of CH₄ initially present should affect the methane bank satura-299 tion; the more the initial CH_4 saturation, the higher the bank saturation. However, numerical simulation of CO_2 300 injection for different initial gas saturations (everything else being the same), show that the bank saturation is in-30 dependent of the initial CH₄ saturation (see Figs. 4 and 7). In fact, Hosseini et al. (2012) showed mathematically 302 that the CH₄ bank saturation is independent of the initial gas saturation. 303 This can be further explained using the principles of fractional flow theory (Pope, 1980, Orr, 2007). Because 304 of the differences in phase viscosities in the two phase region (i.e. between mixtures of CO₂-H₂O and CH₄-305 H₂O), flow occurs on different fractional flow curves in the two phase region (Fig. 7). Figs. 7a, c and e show the 306 fractional flow curves (plots of H_i/ρ_{i1} against G_i/ρ_{i1}) for CO₂ and CH₄ along with the locations of the shock 307 fronts for different initial gas saturations. 308

The partial derivative $\partial H_i/\partial G_i$ represents the wave velocity of the system. The wave velocities of the shock fronts are found from the gradients of straight lines that link the two conditions on either side of the shock. Fractional flow theory dictates that valid solutions should satisfy both the velocity constraint and the so-called entropy constraint (Orr, 2007). The velocity constraint implies that wave velocity should always decrease with increasing distance from the injection boundary. The entropy constraint implies that the shock wave velocity should be equal to the gradient of the fractional flow curve immediately upstream of the shock.

³¹⁵ Due to the zero initial condition for G_1 , the only valid path on the CO₂ fractional flow curve is a tangent ³¹⁶ (i.e., (0,0) to L). On the other hand, velocities must be equal at the contact between a pair of different fluids ³¹⁷ (Pope , 1980) (i.e. points G and L in Fig. 6). This means that the gas bank saturation (point G) is dictated by the ³¹⁸ intersection of the tangent to the CO₂ curve with the CH₄ curve.

Figs. 7b, d and f show the corresponding saturation profiles for different initial gas saturations. The level of saturation at point G is always determined by the tangent from (0,0) to point L. Physically, this implies that the bank saturation is only dependent on how fast the injected gas propagates. The solid and dashed lines in Figs. 7b, d and f are from the analytical solution and numerical solution, respectively. There is an excellent correspondence between the two. The analytical solution was developed on the basis of the fractional theory described above. The numerical solution therefore further confirms the finding of Hosseini et al. (2012), that the CH₄ bank saturation is independent of the initial gas saturation.

However, if the initial gas saturation is sufficiently high it may not be possible to build a solution that travels from the initial condition (point I in figure 6) to the intersection point of the tangent line with the CH₄ fractional flow curve (point G in figure 6). As a consequence, a CH₄ bank will no longer be present. Therefore, it should be noted that the existence of the CH₄ bank is dependent on the initial gas saturation (LaForce and Johns, 2010). Nevertheless, providing the CH₄ bank exists, the CH₄ bank saturation is independent of the initial gas saturation.

9 Summary and conclusions

A numerical simulator was developed for 1D, compressible, two-phase, three-component, radially symmetric flow using the method of lines (MOL) and a 3rd order accurate spatial discretization using weighted essentially non oscillatory (WENO) scheme. The MOL implementation enabled application of the state of the art MATLAB ODE-Solver, ODE15s, for time integration. Pressure and overall component mass fraction (z_i) were selected as the primary dependent variables (PDV). The sparsity of the system of equations was taken advantage of by



Fig. 7 Illustration of CH₄ bank saturation independence of initial CH₄ mass fraction for the deep reservoir scenario after 1000 days of injection (with parameters as set in Table 1): a) and b) $S_{1_I} < S_{1_G}$, c) and d) $S_{1_I} = S_{1_G}$, e) and f) $S_{1_I} > S_{1_G}$.

- provision of a penta-diagonal Jacobian pattern for the ODE solver. Simulation examples were developed in the context of CO_2 into a reservoir containing a mixture of CH_4 and H_2O .
- Following an assumption of constant equilibrium ratios (K_i) for CO₂ and CH₄, it was possible to derive a ternary flash calculator by deriving closed-form relationships for exact interpolation between equations of state for CO₂-H₂O and CH₄-H₂O binary mixtures. This helped ensure improved stability and mass conservation and led to reduced computational requirements associated with multi-dimensional lookup tables. The numerical code was successfully tested and verified for a range of scenarios by comparison to an existing analytical solution.

344 10 Aknowledgements

This work was funded by Centrica plc and a NERC Oil and Gas Catalyst award (NE/L008076/1). The authors are also grateful for the useful comments made by several anonymous reviewers. These have led to a significantly improved manuscript.

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