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1 Tracing weathering regimes using the **lithium** isotope
2 composition of detrital sediments

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9 **ABSTRACT**

10 Lithium (Li) isotopes are a promising tracer of chemical weathering processes for
11 both modern and ancient times. In order to improve the use of Li isotopes in the
12 sedimentary record, here we calibrate the relationship between weathering intensity and
13 detrital Li isotope composition ($\delta^7\text{Li}$) using the fine fraction of modern large river
14 sediments. By using independent estimates for sediment provenance to calculate the Li
15 isotope signature of the rock from which the sediments derive through weathering, we
16 show that source rock variability (in particular the relative contribution of sedimentary vs.
17 igneous rocks) must be corrected for before using Li isotopes as a weathering proxy. We
18 also show that for rivers draining mountain ranges, the contribution of particles derived
19 from sedimentary rocks to river sediments is correlated to their Li/Al ratio, making it
20 possible to use Li contents to estimate the average source rock composition. Once
21 corrected for bedrock variability, the Li isotope signature defines a negative relationship
22 with the weathering intensity (ratio between silicate weathering rate and total denudation

rate) with highest Li isotope fractionation for the highest weathering intensity.

Altogether, we propose a set of new relationships between weathering, erosion, provenance and Li isotopes that can be used to quantify present-day and paleo-weathering using detrital sediment.

INTRODUCTION

The surface of Earth evolves through the combined action of chemical weathering reactions and physical denudation. The former process partly dissolves the continental crust, exports cations to the ocean, and consumes atmospheric CO₂ from the atmosphere, whereas physical denudation exports solid products to the ocean. Hence, over long time scales, silicate weathering is one of the major processes regulating the amount of atmospheric CO₂ and therefore controlling the evolution of climate (Berner et al., 1983). Despite this recognition, it is still poorly known how silicate weathering has changed in the past and what are the parameters controlling weathering rates and intensities over long timescales. The emerging Li isotope proxy has a great potential for tracing silicate weathering because Li is a soluble element predominantly hosted in silicate rocks and its isotopes are fractionated during water-rock interactions (Huh et al., 2001). Although the relation between the river dissolved Li isotope composition ($\delta^7\text{Li}$) and weathering is still a matter of debate (e.g., Dellinger et al., 2015; Huh et al., 2001; Pogge von Strandmann and Henderson, 2015), this has served as a basis to investigate the $\delta^7\text{Li}$ of ancient marine carbonates (e.g., Misra and Froelich, 2012; Pogge von Strandmann et al., 2013) in order to unravel global past weathering conditions (Bouchez et al., 2013; Li and West, 2014; Misra and Froelich, 2012).

45 Complementary to this approach, the characterization of weathering conditions
46 with Li isotopes can also be achieved with the study of detrital sediments. However,
47 several difficulties arise for using sediments $\delta^7\text{Li}$ to trace weathering and erosion in
48 modern systems. First, detrital sediments not only contain weathered products but also
49 unweathered minerals deriving from the physical erosion of the bedrock (Dellinger et al.,
50 2014). Secondly, the $\delta^7\text{Li}$ of river sediments is quite variable depending on the sediment
51 grain size (Dellinger et al., 2014). Finally, as on average 80% of the total Li in the
52 weathering system is contained in the solid load and only 20% in the dissolved load
53 (Misra and Froelich, 2012), terrigenous sediment $\delta^7\text{Li}$ is usually close to that of source
54 rock (Bouchez et al., 2013), such that the source rock $\delta^7\text{Li}$ has a large imprint on the
55 composition of the detrital material (Dellinger et al., 2014). In this contribution, we use
56 the available data from the literature to investigate how the $\delta^7\text{Li}$ of medium to large river-
57 sediments relates to the present-day weathering regimes.

58 **PRESENT-DAY RIVER SEDIMENTS DATASET**

59 The data set used in this study includes some of the largest Earth river systems:
60 the Amazon, Mackenzie, Ganges and Brahmaputra (Dellinger et al., 2015, 2014), Congo
61 (Henchiri et al., 2016) and Changjiang (also known as the Yangtze River; Wang et al.,
62 2015). It is complemented by new data from the Stikine and Skeena River from the
63 western Canadian cordillera, two large rivers draining predominantly volcanic rocks
64 (Gaillardet et al., 2003). All these rivers drain a surface larger than 19,000 km², and cover
65 a wide range of lithology, climate, and weathering regimes such that this data set can be
66 considered as globally representative. In the following, we characterize the modern
67 weathering regimes by using the weathering intensity parameter “(W/D)”, which is

68 independent of the drainage area and the discharge, and defined as the ratio between
69 chemical silicate weathering rates W (calculated as the flux of river dissolved material
70 sourced from silicate rocks only) and total denudation rate D , thereby corresponding to
71 the proportion of weathering material transported in the dissolved form by the river.

72 Within the set of rivers from this study, we distinguish two groups: (1) rivers
73 draining high elevation mountain ranges (the Andes, Himalayas, Western Canadian
74 Cordillera) characterized by moderate to high erosion rates ($>100 \text{ t km}^{-2} \text{ yr}^{-1}$), low
75 weathering intensity ($W/D < 0.1$), and a kinetically limited weathering regime (West et
76 al., 2005); and (2) rivers draining lowlands with deep weathered soils, low erosion rates,
77 and high weathering intensity ($W/D > 0.1$) corresponding to the supply-limited
78 weathering regime (West et al., 2005).

79 Overall, for rivers draining mountain ranges, the Li concentration of river
80 sediments ranges from 10 to $102 \mu\text{g g}^{-1}$ (ppm) and the $\delta^7\text{Li}$ of the sediments
81 ($\delta^7\text{Li}_{\text{suspended_load}}$, precision better than ‰) from -4‰ to $+5\text{‰}$ (Fig. 1). The rivers draining
82 lowlands have on average lower $\delta^7\text{Li}$ (-6.5‰ to -3‰) and Li concentration compared to
83 rivers draining mountain ranges. Interestingly, the $\delta^7\text{Li}$ is negatively correlated to the Li
84 concentrations in large rivers draining mountain ranges, whereas no relationship is
85 observed for rivers draining the lowlands (Fig. 1).

86 REMOVING THE NON-WEATHERING EFFECTS

87 In large rivers draining mountain ranges, the $\delta^7\text{Li}$ and Li concentration of river
88 sediments are strongly dependent upon grain size and lithological source (Dellinger et al.,
89 2014). In order to extract the weathering signal contained in river sediments, it is

necessary to remove the effect of both grain size and lithological variability on the Li isotope composition.

Grain Size Effects

Fine sediments have generally lower $\delta^7\text{Li}$ and higher Li concentration than coarse sediments (Fig. 1). This is because they contain a higher proportion of particles sourced from sedimentary rocks and a higher proportion of weathering products sourced from modern soils that carry the fingerprint of the modern weathering regime (Dellinger et al., 2014). Therefore, similarly to element ratios (e.g., Lupker et al., 2013), large changes of the $\delta^7\text{Li}$ in sedimentary archives can simply arise from change of the sediment grain size and lead to false interpretation in terms of shift in weathering conditions. In order to mitigate this grain size effect, in the present study we use for each river the finest sediment sample available (size < 63 μm), which for most cases has been collected at or near the river channel surface. However, even these fine sediments contain a non-negligible fraction of unweathered material that needs to be taken into account for the interpretation of their $\delta^7\text{Li}$ data.

The Influence of Lithology

The Li isotope signature that can be measured in weathered detrital material depends upon (1) the initial $\delta^7\text{Li}$ of river sediments, prior to chemical weathering (i.e., the Li isotope signature of its source rock material, hereafter referred to as “ $\delta^7\text{Li}_{\text{source_material}}$ ”), which is a lithological effect and (2) the extent of fractionation between this initial signature and the neoformed weathering material, which in turn depends on isotope fractionation factors and on Li partitioning between secondary minerals and the dissolved load (Bouchez et al., 2013). The weathering signal is contained only in the second effect.

113 This is why in the following, we consider the difference in isotope composition between
114 the fine suspended sediment and the original bedrock, $\Delta^7\text{Li}_{\text{fine-source}} = \delta^7\text{Li}_{\text{fine suspended_load-}}$
115 $\delta^7\text{Li}_{\text{source_material}}$.

116 The “ $\delta^7\text{Li}_{\text{source_material}}$ ” corresponds to the weighted average of the $\delta^7\text{Li}$ of the
117 various erosion sources: igneous, metamorphic and sedimentary rocks (hereafter referred
118 to as “shale” because shales are by far the most important type of sedimentary rock
119 contributing to the Li budget; Holland, 1984) that have distinct average Li concentration
120 and isotope composition (Dellinger et al., 2014; Sauzéat et al., 2015; Tomascak et al.,
121 2016). Indeed, the Li isotope variability of the present-day upper continental crust
122 (UCC), as characterized by using inherently integrative samples like large river sand
123 (Dellinger et al., 2014) or desert loess (Sauzéat et al., 2015), corresponds mostly to a
124 mixture between a Li-rich, low- $\delta^7\text{Li}$, shale end member ($\delta^7\text{Li}$ of $-0.5 \pm 1\text{‰}$ and Li/Al of
125 $1.0 \pm 0.1 \times 10^{-3}$) and a Li-poor, high- $\delta^7\text{Li}$, igneous end member ($\delta^7\text{Li}$ of $3.5 \pm 1.5\text{‰}$,
126 Li/Al of $0.30 \pm 0.12 \times 10^{-3}$). This mixing trend can be used to determine the Li isotope
127 composition of river sediments prior to weathering ($\delta^7\text{Li}_{\text{source_material}}$) provided that the
128 relative contribution of the different source materials to the sediments is known.

129 Quantification of the provenance of sediments can be achieved using Sr-Nd
130 isotopes and/or insoluble trace element ratios, provided that the composition of the major
131 bedrock drained by each river basin is known and that they have contrasted signatures
132 (e.g., Allègre et al., 1996). Here, we use independent tracers of provenance (Nd, Sr
133 isotopes and trace elements) in order to quantify the erosion sources of the fine sediments
134 and determine the $\delta^7\text{Li}_{\text{source_material}}$ for fine sediments from each of the large rivers. We
135 note that the procedure of choice to characterize average source rock is specific to each

136 large river basin. In the Amazon, Congo, and Ganges-Brahmaputra basins, sedimentary,
137 igneous, and metamorphic rocks have distinct Nd isotope composition making it possible
138 to quantify precisely the provenance of fine sediments using Nd isotopes (see the GSA
139 Data Repository¹). For the Mackenzie River, we use trace elements (Cr, Ti, Cs) ratios
140 (see the Data Repository). We emphasize that this calculation provides information on
141 source rocks for fine sediments only, which can be different for other grain sizes
142 (Bouchez et al., 2011a; Dellinger et al., 2014; Garçon and Chauvel, 2014).

143 However, in some cases like that of the Changjiang basin, neither Sr-Nd isotopes
144 nor trace elements can be used to assess the provenance of sediments because the
145 signatures of the main rock sources are under-constrained or not significantly different
146 (Chetelat et al., 2013). An alternative method is to identify another tracer that correlates
147 with an independent measure of the proportion of shale-derived particles (i.e., the amount
148 of sedimentary recycling) quantified across the rest of our river set. Among all the
149 chemical elements tested, the best correlation with the proportion of shale-derived
150 particles is obtained for the Li/Al ratio (Fig. 2). Li concentrations are normalized to Al to
151 remove dilution effect by quartz or other Li-free phases. This good relationship stems
152 mostly from the fact that (1) shales and igneous rocks have significantly distinct Li
153 contents, and (2) that chemical weathering has a minor impact on sediment Li content for
154 a majority of samples. However, we note that sediments from some lowland rivers have
155 lost a significant fraction of Li because of their higher weathering intensity (Fig. 3).
156 Using the relationship between Li and the proportion of sediments derived from erosion
157 of shales calibrated on rivers draining mountain ranges, we infer the proportion of shales

contributing to the fine sediment load of the Changjiang basin river, which makes our set of $\delta^7\text{Li}_{\text{source_material}}$ estimates complete (see Data Repository).

Using the calculated proportion of shales and igneous rocks with the $\delta^7\text{Li}$ of the rock end members, we estimate the Li isotope composition of the rock source ($\delta^7\text{Li}_{\text{source_material}}$) for each fine river sediment sample (see the Data Repository). We note that the calibration of the Li provenance tracer in detrital sediments might also serve to determine the provenance and the $\delta^7\text{Li}_{\text{source_material}}$ of ancient detrital sedimentary archives for which the potential bedrock sources are usually not constrained.

RELATION BETWEEN THE LITHIUM ISOTOPE SIGNATURE AND WEATHERING INTENSITY

In this section we evaluate the relation between sediment $\delta^7\text{Li}$, Li weathering index and the weathering intensity previously defined as the W/D ratio. The Li weathering index of the detrital sediment can be expressed as $(\text{Li}/\text{Al})_{\text{fine_sediments}} / (\text{Li}/\text{Al})_{\text{source_material}}$, where $(\text{Li}/\text{Al})_{\text{source_material}}$ corresponds here to the (Li/Al) ratio the suspended sediments would have in the absence of any silicate weathering reactions. A value of 1 means no loss or gain of Li by the solids during weathering while a value lower than 1 implies loss of Li during weathering. The “ $(\text{Li}/\text{Al})_{\text{source_material}}$ ” values are determined using the rock end member compositions defined in the previous section and the proportion of shales calculated above. The Li weathering index values are negatively correlated to the weathering intensity (Fig. 2.b) with most of the rivers draining mountain ranges having a Li weathering index higher than 0.8 (except for the Solimões and Pastaza rivers), whereas rivers draining lowlands have Li weathering index values between 0.9 and 0.4. The higher Li depletion in sediments transported by lowland rivers is compatible

181 with the higher proportion of dissolved Li transported by these rivers (see the Data
182 Repository). This shows that significant Li depletion in sediments occurs only for very
183 intense weathering conditions (Fig. 2).

184 The relationships between Li isotope signature in modern fine river sediments
185 corrected for the source rock signature ($\Delta^7\text{Li}_{\text{fine-source}}$) can be examined as a function of
186 the weathering regime. A significant negative relationship is observed between the
187 weathering intensity (W/D) and $\Delta^7\text{Li}_{\text{fine-source}}$ (Fig. 3a, $r^2 = 0.75$). This correlation is
188 stronger than that between $\delta^7\text{Li}_{\text{fine-sediments}}$ and W/D (Fig. 3b, $r^2 = 0.48$), showing that
189 source rock composition indeed introduces some scatter in this relationship. Sediments
190 from rivers characterized by high W/D have the lowest $\Delta^7\text{Li}_{\text{fine-source}}$ whereas rivers
191 having low W/D have $\Delta^7\text{Li}_{\text{fine-source}}$ close to zero. The $\delta^7\text{Li}_{\text{sed}}$ and $\Delta^7\text{Li}_{\text{fine-source}}$ also
192 correlate well with more traditional weathering proxies such as CIA ($r^2 = 0.76$, not
193 shown) or K/Th ratio ($r^2 = 0.62$, not shown). We observe no relationship between $\Delta^7\text{Li}_{\text{fine-}}$
194 source and chemical silicate weathering rates.

195 To explain this relationship, we suggest that at high denudation rate (and hence
196 low W/D), incorporation of Li into secondary minerals is hindered by short mineral
197 residence times in soils making the dissolved $\delta^7\text{Li}$ close to $\delta^7\text{Li}_{\text{source_material}}$ (Bouchez et
198 al., 2013; Dellinger et al., 2015; Wang et al., 2015). Therefore, the small amount of
199 secondary minerals formed should have a low $\delta^7\text{Li}$, which could be expected to be
200 reflected in the $\delta^7\text{Li}$ of fine sediments (Dosseto et al., 2015). We suggest that this is not
201 observed because at very low weathering intensity ($\text{W/D} < 0.01$), typical of high-erosion
202 settings, the sediments carried by the rivers contain a high proportion of primary minerals
203 ($\Delta^7\text{Li}_{\text{sed-rock}} = 0$) compared to secondary minerals formed during the modern weathering

cycle, even in the fine sediments (Dellinger et al., 2014, 2015). At high weathering intensity ($W/D > 0.1$) such as in lowland areas, secondary minerals formed during the present-day weathering cycle and characterized by low $\delta^7\text{Li}_{\text{fine_sediments}}$, compose the majority of the sediments and therefore control the bulk sediment Li isotope signature. The contribution of unweathered debris to river fine sediment load is likely to be more prominent for rivers draining fine-grained sedimentary rocks (which is the case of large, lithologically mixed rivers), as these lithologies are highly erodible and produce fine sediments.

IMPLICATIONS AND CONCLUSIONS

These results have important implications regarding the use of Li isotopes in clastic sediments to trace weathering regimes both in the present-day and in the past. First, the $\delta^7\text{Li}$ of modern river sediments is primarily related to weathering intensity (that is, in the modern Earth, mostly a function of the erosion rate; Dixon and von Blanckenburg, 2012) and not to the silicate weathering rate as inferred by other studies (e.g., Dosseto et al., 2015).

Second, the $\delta^7\text{Li}$ of fine river sediments is sensitive only to large changes of the weathering intensity (i.e., by an order of magnitude; Fig. 3a.). In particular, a $\Delta^7\text{Li}_{\text{fine-source}}$ value between -1% and -3.5% can correspond to a large range of possible weathering intensity comprised between 0.005 and 0.08. Indeed, in high-erosion rate settings, the contribution of soil-derived weathered particles is minor compared to that of debris of unweathered bedrock mobilized by mass wasting processes, even in the finest river sediments fraction (Dixon and von Blanckenburg, 2012; Dellinger et al., 2014). However, the $\delta^7\text{Li}$ of clastic sediments can clearly discriminate between high weathering intensity

settings (typical of tropical lowland areas like the Congo River basin) and intermediate to low weathering intensity regimes.

The third important implication is related to the use of Li isotopes in detrital sediments as a proxy for palaeo-weathering regimes. Tracing ancient weathering processes using conventional geochemical tracers in detrital sediments (*e.g.* chemical index of alteration CIA, or chemical ratios combining a soluble and an insoluble element, such as Na/Sm, U/Th or K/Al) can be challenging because these proxies are also influenced by sediment provenance, grain size effects or post-depositional alteration (Bouchez et al., 2011a; Carpentier et al., 2013; Li and Yang, 2010; Lupker et al., 2013). Hence, some recent studies have used Li isotopes in fluvial alluvial deposits (Dosseto et al., 2015), marine detrital sediments (Bastian et al., 2015) or ancient siliciclastic rocks (Li et al., 2016) to determine palaeo-weathering conditions. In comparison to marine carbonates, clastic sediments record the fingerprint of local weathering conditions and have the advantage of not being influenced by biological fractionation (“vital effects”) observed during carbonate precipitation (Vigier et al., 2015).

The results from this study can be used as a framework for interpreting the detrital $\delta^7\text{Li}_{\text{fine_sediments}}$ and Li/Al ratio in sediment archives. It demonstrates the importance of correcting for grain size and lithological variability in order to interpret the detrital Li isotope composition in terms of weathering intensity. It also shows that the $\Delta^7\text{Li}_{\text{fine-source}}$ value is a function of the weathering intensity only for W/D values higher than 0.05. Overall, the combined use of Li/Al, $\delta^7\text{Li}$ and other more traditional weathering proxies (like CIA) has great potential for quantifying both sedimentary recycling and weathering intensity in sediment archives.

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350 **FIGURE CAPTIONS**

351 Figure 1. Li isotope composition as a function of Li concentration for large river
352 sediments. Gray color scale applies only to rivers draining mountain ranges (circles in
353 figure) and corresponds to the logarithm of the D_{90} values (which is the grain size below
354 which 90% of the grains are found in a given sample). For lowland rivers, grain size was
355 not measured, but all lowlands sediment are surface samples, which are generally fine-
356 grained (Filizola, 2003).

357

358 Figure 2. A: The Li/Al ratio in fine sediments as a function of the proportion of shales in
359 the source of river sediments, as calculated using Nd isotopes or trace element ratios (see
360 the Data Repository [see footnote 1]). Black dotted line corresponds to the mixing
361 between shales and igneous rocks. Gray dotted lines labeled with numbers correspond to
362 the Li weathering index values $(\text{Li}/\text{Al})_{\text{sediment_load}} / (\text{Li}/\text{Al})_{\text{source_material}}$ for various

363 proportion of shales. The best fit of the data ($R^2 = 0.82$), excluding the lowland rivers,
364 corresponds to a Li weathering index value of 0.9. The average igneous rock and shale
365 end members as given in Dellinger et al. (2014) and Sauzéat et al. (2015) are also
366 represented. B: Li weathering index $(\text{Li}/\text{Al})_{\text{sediment_load}} / (\text{Li}/\text{Al})_{\text{source_material}}$ in the sediments
367 as a function of the weathering intensity.

368

369 Figure 3. A: Li isotope composition of fine large river sediments corrected from the
370 source rock ($\Delta^7\text{Li}_{\text{fine-source}}$). B: Li isotope composition of large river fine sediments
371 ($\delta^7\text{Li}_{\text{sediment_load}}$), as a function of the weathering intensity (W/D).

372

373 ¹GSA Data Repository item 2017xxx, methods, supplementing text, Figures DR1–DR6
374 and Tables DR1 and DR2, is available online at
375 <http://www.geosociety.org/datarepository/2017/> or on request from
376 editing@geosociety.org.

377

Figure 1

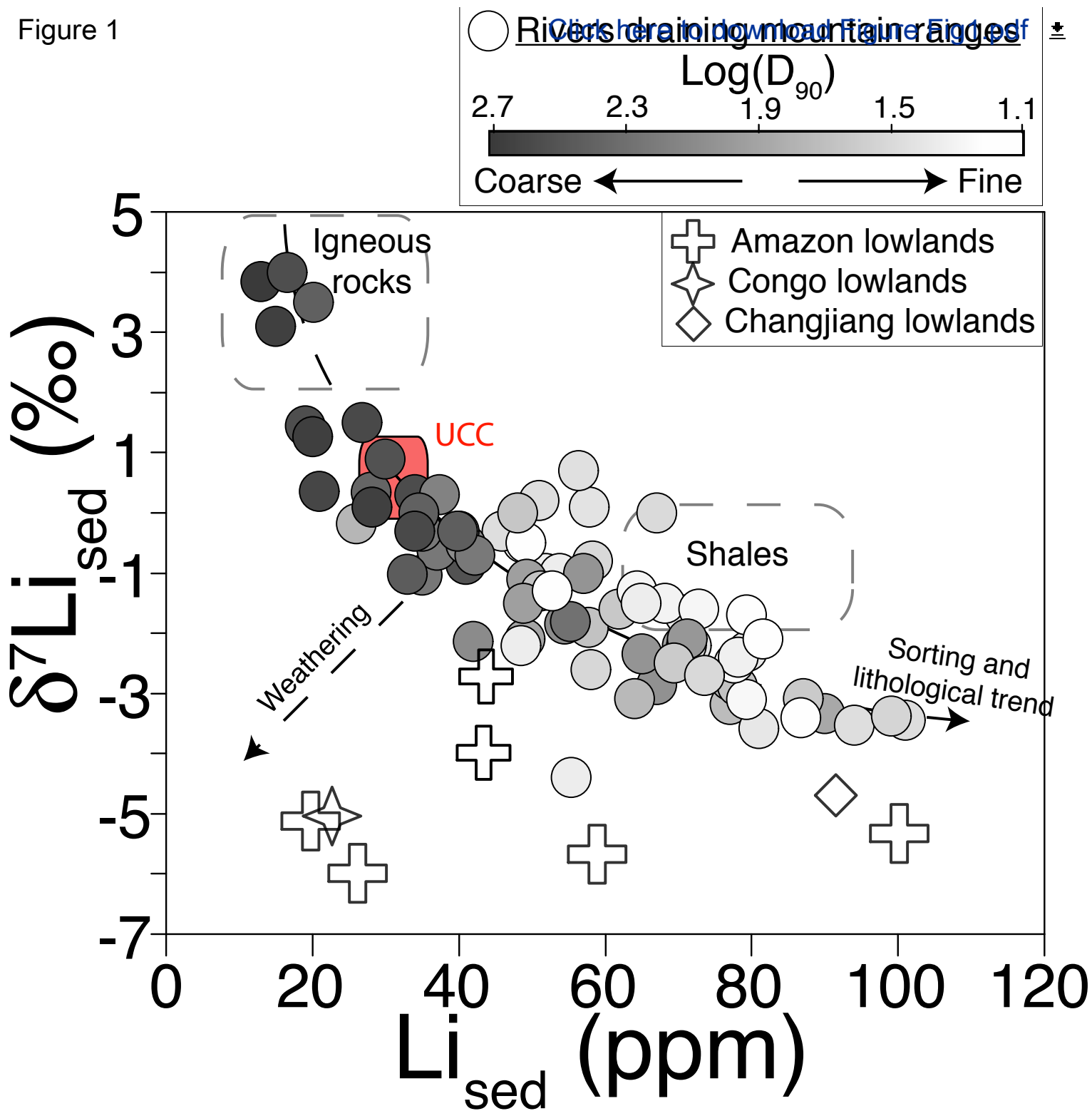


Figure 2

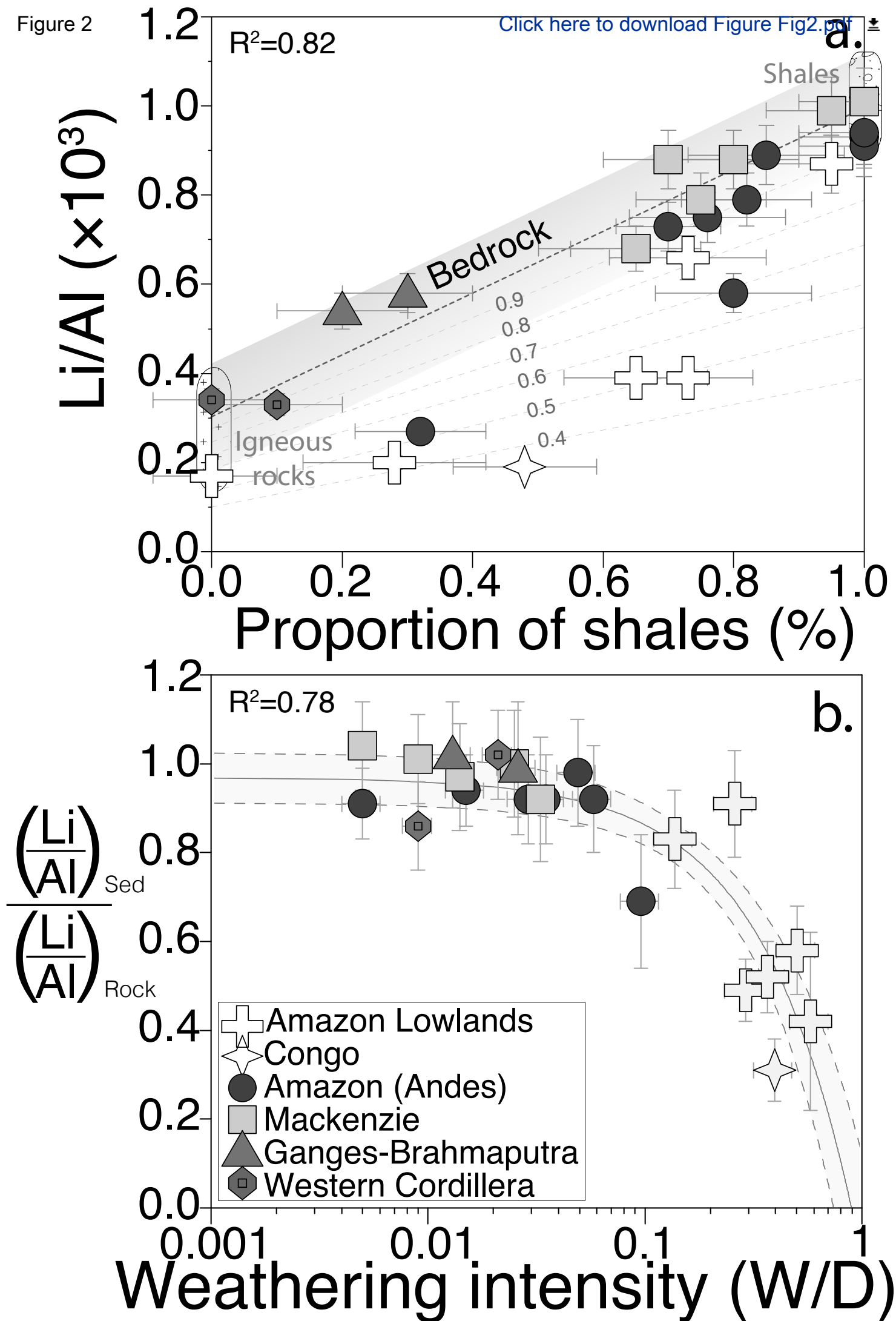
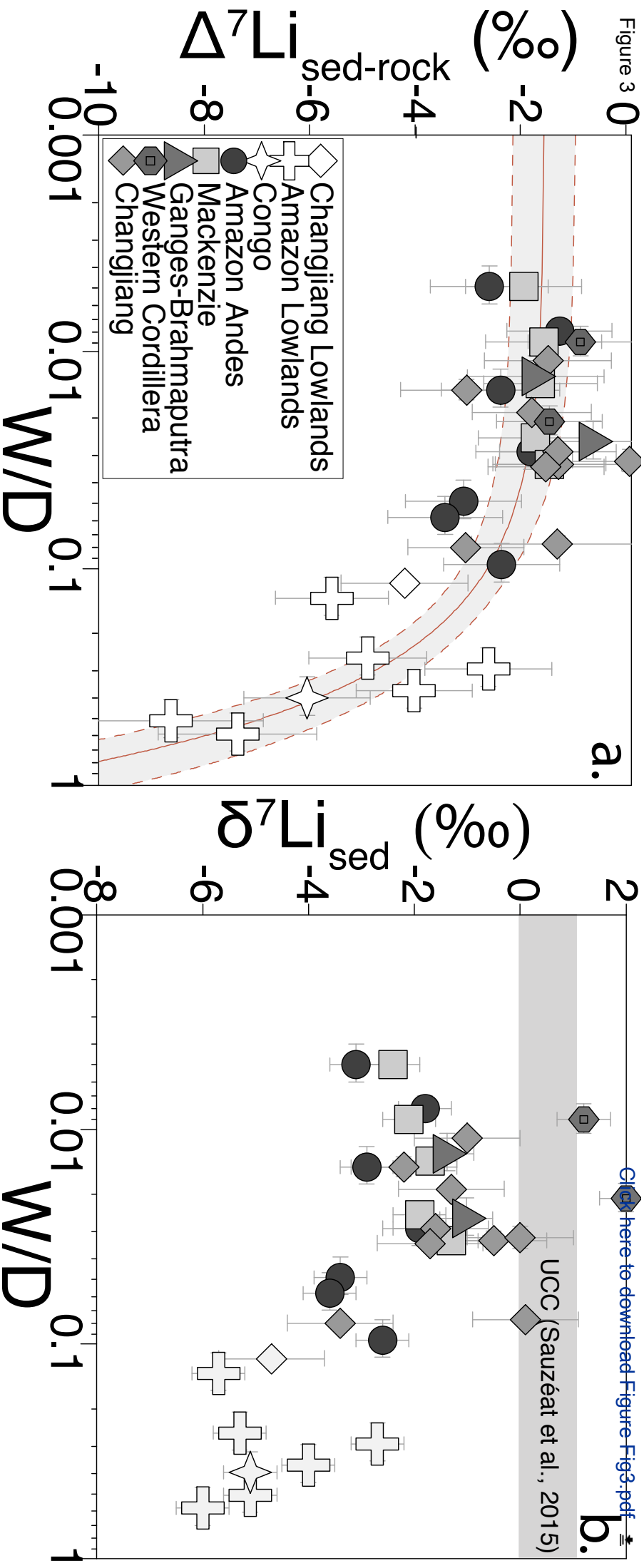


Figure 3



Tracing weathering regimes using the Li isotope composition of detrital sediments

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GSA Data Repository

Contents

Supplemental information

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Figure DR1, DR2, DR3, DR4, DR5 and DR6

Table DR1 and DR2

References

1. Analytical methods

Here, we briefly describe the analytical methods used for measurements of the Li and Nd isotope composition of sediments.

Li isotopes: the digestion, column chemistry and Li isotope measurement procedure for the sediments has been described in Dellinger et al., (2014). About 20 mg of sediment were digested in HF-HNO₃ at 100°C during 24 to 48 hours. Li was then separated from the matrix by ion-exchange chromatography using a method modified from James and Palmer (2000) and described in Dellinger et al. (2014). The lithium isotope composition was measured using a MC-ICP-MS Neptune (Thermo Scientific, Bremen) at IPGP (Paris). Details on the analytical procedure are available in Dellinger et al. (2014). Accuracy and reproducibility of the isotopic measurements were checked through repeated analyses of the basalt reference materials JB-2 yielded $\delta^7\text{Li} = +4.47 \pm 0.53$ ($\pm 2\sigma$, $n = 30$ separations and 15 digestions).

Nd isotopes: Nd isotope composition was measured in the same aliquot used for Li isotopes. Nd was separated from the rest of the matrix by ion-exchange chromatography using the TRU-spec and Ln-spec resins and measured by MC ICP-MS Neptune in IPGP following the method described in Cögeš et al., (2015).

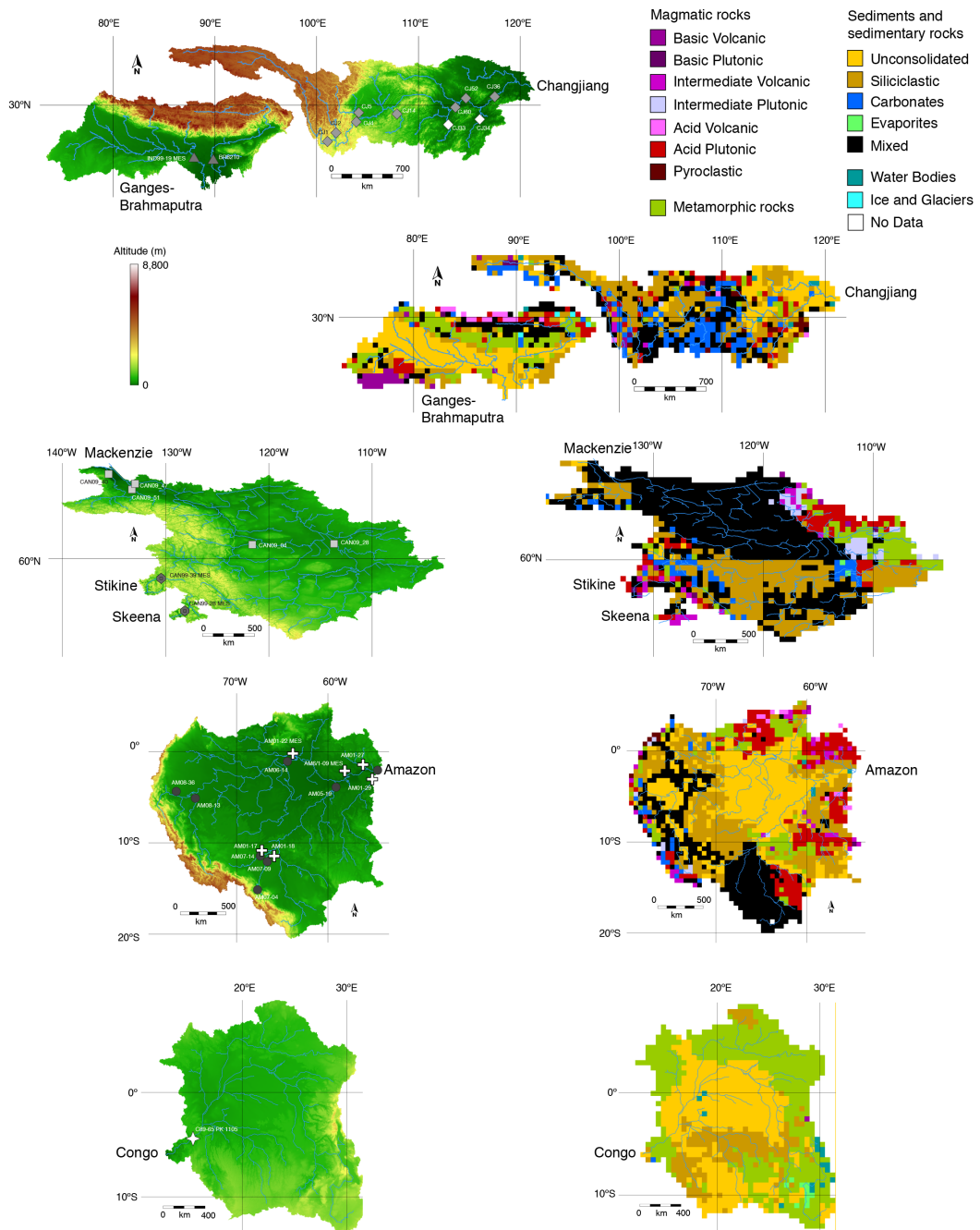


Figure DR1: Maps of the river basins (elevation on the left, geology on the right) from this study, with the locations of the samples. The Geological maps are from Hartmann and Moosdorf (2013). The Changjiang samples are from Wang et al., (2015)

2. Provenance of fine river sediments

Here, we describe how the relative proportion of river sediment derived from the erosion of shales, igneous and high-grade metamorphic rocks is calculated for each river basin. The location of the samples and the geological map of each river basin are shown in Fig. (DR1).

Amazon River: In the Amazon basin, the major rock types are sedimentary rocks in the Andes and foreland/lowlands, igneous rocks (andesites in the northern part of the Amazonian Andes and granitoids in their southern part) and granitic and metamorphic rocks in the Brazilian and Guyana shields. The Nd isotope composition (expressed here as $\epsilon_{Nd}(0)$, the ratio in parts per 10^4 (units) between the measured $^{143}\text{Nd}/^{144}\text{Nd}$ of the sample relative to the present-day value (0.512638) of the $^{143}\text{Nd}/^{144}\text{Nd}$ of the CHondritic Uniform Reservoir (CHUR)) is a reliable tracer of the provenance of river sediments in the Amazon basin because the major rock types have distinct ϵ_{Nd} values (Allègre et al., 1996; Roddaz et al., 2005; Viers et al., 2008). In the following we consider that river sediments in the Amazon Basin result from a binary mixture between two end members, such that the following mixing equation can be written:

$$\epsilon_{Nd}(0)_{sed} = \epsilon_{Nd}(0)_{sha} \times \gamma_{sha} + \epsilon_{Nd}(0)_{Ign} \times (1 - \gamma_{sha})$$

With $\epsilon_{Nd}(0)_{sed}$, $\epsilon_{Nd}(0)_{sha}$ and $\epsilon_{Nd}(0)_{Ign}$, respectively the Nd isotope composition of river sediment, shale and igneous rock. The proportion of shales-derived particles (“ γ_{sha} ” in mass %) can be calculated as:

$$\gamma_{sha}^{Nd} = \gamma_{sha} \times \frac{[Nd]_{sha}}{[Nd]_{sed}}$$

As the average Nd concentration of shale and igneous rocks is not significantly different (Condie, 1993), we consider that $\gamma_{sha} = \gamma_{sha}^{Nd}$. As explained below, the composition of the igneous end member ($\epsilon_{Nd}(0)_{Ign}$) varies between the different sub-basins, while the composition of the shale end member ($\epsilon_{Nd}(0)_{sha}$) is considered homogenous throughout the Amazon River basin.

The Beni River drains exclusively shales from the Bolivian Andes and we use the $\epsilon_{Nd}(0)$ of its sediments to estimate the average $\epsilon_{Nd}(0)$ value of Andean sedimentary rocks. The $\epsilon_{Nd}(0)$ of Beni sediments ranges between -12.2 and -12.9 , in good agreement with the range of $\epsilon_{Nd}(0)$ values measured in Andean sedimentary rocks (-10 to -13 ; Pinto, 2003; Viers et al., 2008) and Neogene South Amazonian foreland basin sediments from Roddaz et al., (2005). Similarly, the

Yata river that drains exclusively the lowlands Tertiary sedimentary rocks (originated from the erosion of the region of the Andes) have a $\epsilon_{Nd}(0)$ of -11.6 , in good agreement with the $\epsilon_{Nd}(0)$ value measured in Beni river sediments. Therefore, we assume a $\epsilon_{Nd}(0)_{sha}$ of -12.0 ± 1.0 for the shale end member. The same value of -12.0 was also assumed for the metasedimentary end-member of the Madre de dios river basin by Basu et al., (1990). In addition, we consider that this value is also representative of the shales from the northern Andes. Indeed, the $\epsilon_{Nd}(0)$ of Solimões sediments defines a trend with the Al concentration (grain size proxy) that tends toward a similar shale $\epsilon_{Nd}(0)$ value (Fig. DR2) and Neogene North Amazonian foreland basin sediments from Roddaz et al., (2005) also plot on the same trend.

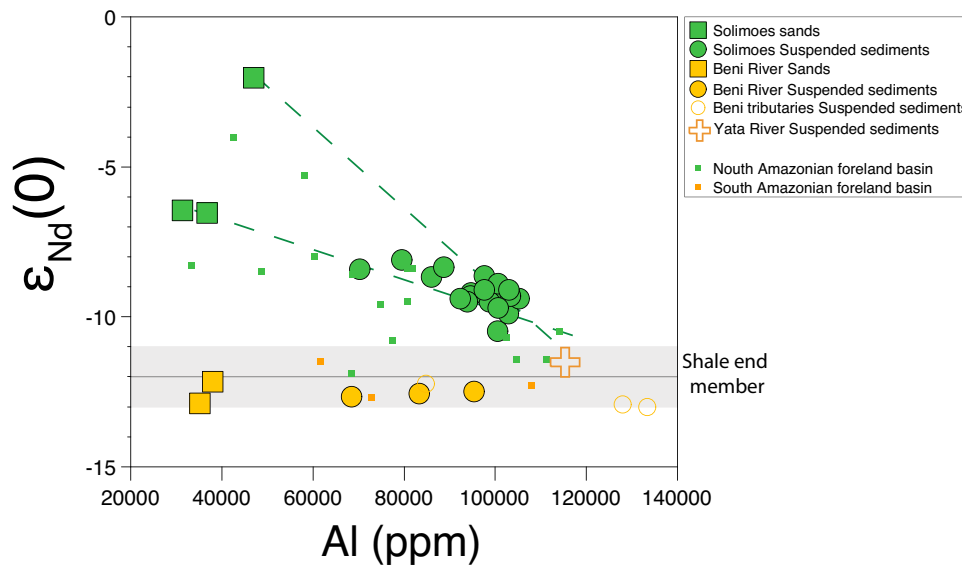


Figure DR2: $\epsilon_{Nd}(0)$ as a function of the Al concentration in sediments. Solimões and Beni data are from Bouchez et al., (2011a), Dellinger et al., (2015) and Viers et al., (2008). Data for foreland basin sediments are from Roddaz et al., (2005). The grey band corresponds to the estimated $\epsilon_{Nd}(0)$ of the mean shale end-member in the Amazon basin.

Regarding the igneous rock end member, the Nd isotope composition measured in Andean Neogene granitoid rocks from the Sierra Blanca in Peru have $\epsilon_{Nd}(0)$ generally between -3 to 0 (Petford and Atherton, 1996). Pemian-Triassic and Carboniferous granitoids in southern Peru have a median $\epsilon_{Nd}(0)$ of -4.7 ± 0.6 (Miskovic, 2008). Accordingly, bed sediments from the

Solimões River draining the Peruvian and Ecuadorian Andes, and for which the major and trace element signature reflect an igneous rock provenance (Bouchez et al., 2011a; Dellinger et al., 2014), have $\epsilon_{Nd}(0)$ ranging from -6.5 to -2.0 . However, volcanic rocks in Ecuador (mostly andesites) have a higher $\epsilon_{Nd}(0)$ ranging between $+2$ and $+6$ with (Bryant et al., 2006), and bed sediments from the Pastaza River derived from the erosion of andesite in Ecuador has a $\epsilon_{Nd}(0)$ of $+1.9$ (Dellinger et al., 2015a). This high $\epsilon_{Nd}(0)$ indicates that Pastaza sediments predominantly derive from the erosion of more recent (Quaternary) andesite. In the following, we use a $\epsilon_{Nd}(0)_{Ign}$ of -3.0 ± 2.0 for the igneous end member of all the Andean-fed rivers, except for the Pastaza River for which we use a $\epsilon_{Nd}(0)_{Ign}$ value of $+2.0 \pm 1.0$.

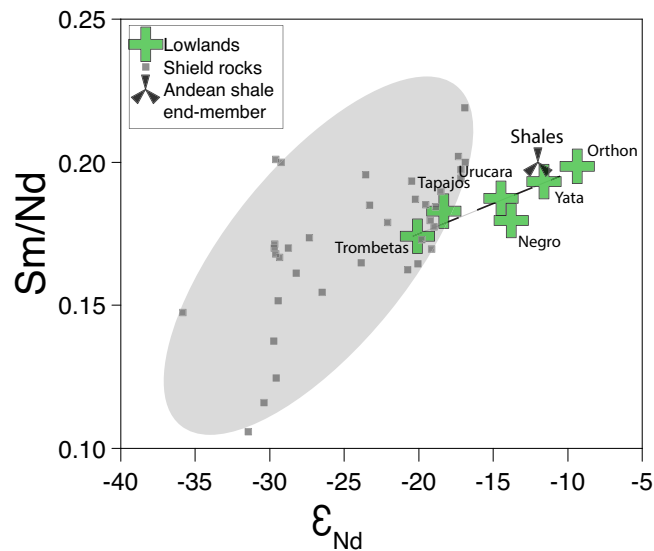


Fig. DR3: Sm/Nd as a function of the ϵ_{Nd} of lowland river sediments. Shield rock data are from Santos et al., (2000). The Tapajós and Pastaza rivers have a $\epsilon_{Nd}(0)$ and Sm/Nd close to the signature of shield rocks whereas Urucara and Negro river sediments contain shale-derived sediments. The shale end-member corresponds to the average composition of Beni river sediments.

Regarding the shield regions, the range of $\epsilon_{Nd}(0)$ measured in rock samples from the Rio Negro, Central Amazon and Tapajós-Parima provinces is very large (-20 to -30) and positively correlated to the Sm/Nd ratio as a result of magmatic differentiation (Allègre et al., 1996; Basu et al., 1990; Hoorn et al., 2009; Roddaz et al., 2005; Santos et al., 2000). The Sm/Nd ratio of rivers

draining the shield range between 0.15 to 0.20, and for a given Sm/Nd ratio, lowlands rivers generally have higher $\epsilon_{\text{Nd}}(0)$ than cratonic rocks (Fig. DR3) and plot on a line trending towards the shale end member (defined as the mean Sm/Nd and $\epsilon_{\text{Nd}}(0)$ of Beni River sediments). Only the Tapajós and Trombetas river sediments have a Nd isotope composition similar to that of the cratonic rocks in the Amazon basin whereas the Rio Negro and Urucara rivers have a $\epsilon_{\text{Nd}}(0)$ intermediate between cratonic rocks and Andean-derived shales, consistently with the findings of Allègre et al., (1996). We consider therefore the Trombetas River composition as representative of the mean Nd isotope composition of the shield end member ($\epsilon_{\text{Nd}}(0)_{\text{Ign}} = -20.0 \pm 2.0$).

Congo River: The lithology of the Congo Basin is mainly composed of Archean and Proterozoic TTG along the basin border, and of sedimentary rocks in the center. To our knowledge, there is no available Nd isotope data on Congo rocks in the literature. However, Allègre et al. (1996) report $\epsilon_{\text{Nd}}(0)$ values for the major Congo tributaries. The $\epsilon_{\text{Nd}}(0)$ range from -12.9 for the Lobaye River to -18.2 for the Likouala. We also measured the $\epsilon_{\text{Nd}}(0)$ on a lateritic soil profile sampled 100 km downstream from Bangui (Henchiri et al., 2016) and developed on TTG rocks. The Nd isotope values range from -18.1 to -18.9 (See table DR2) and is similar to the value of the Likouala.

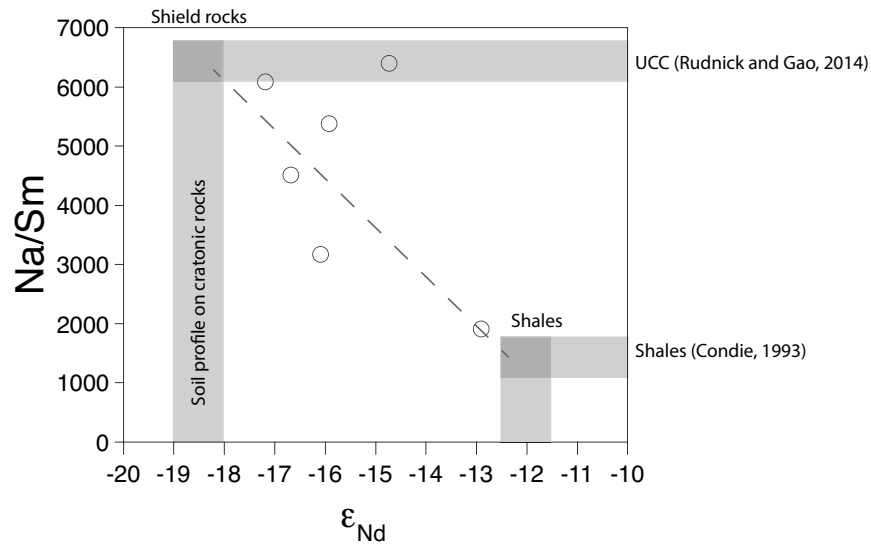


Fig. DR4: Crustal Na/Sm for Congo rivers inferred by Gaillardet et al. (1995) as a function of the sediment ϵ_{Nd} . The Nd isotope composition of the cratonic end member is estimated to be -18.5 ± 0.5 based on the above trend, the Na/Sm of UCC (Rudnick and Gao, 2014) and the Nd

isotope composition of a soil profile developed on cratonic rocks in Bangui (this study). The Nd isotope composition of the shales end-member is estimated to be -12 ± 0.5 by projecting the typical Na/Sm value of global Phanerozoic shales (from Condie, 1993) on the mixing trend. This value is also compatible with the Nd isotope composition of the Lobaye River.

Interestingly, the $\epsilon_{Nd}(0)$ of Congo tributaries sediments is negatively correlated to the Na/Sm ratio of the Congo tributaries continental crust (sediments + dissolved) inferred by Gaillardet et al., (1995) (Fig. DR4). This crustal Na/Sm ratio can be used to distinguish TTG (high Na/Sm) from shales (low Na/Sm resulting from Na depletion during ancient weathering processes). The Likouala has a TTG-like Na/Sm signature whereas the Lobaye has a signature close to that of shales. Altogether, the range of variation of $\epsilon_{Nd}(0)$ in Congo sediments can be explained at first order as resulting from a mixture between a TTG igneous end member of high $\epsilon_{Nd}(0)_{Ign}$ around -19 and a metasedimentary end member of low $\epsilon_{Nd}(0)_{sha}$ (around -12), as for the Amazon. Using eqs. (1) and (2), we calculate that the proportion of shale-derived particles in the Congo River sediments is $40 \pm 10\%$. We note that the correlation between $\epsilon_{Nd}(0)$ and Na/Sm is scattered and therefore that this method for determining the provenance of Congo River sediments gives only a first-order estimate of the proportion of shale-derived particles.

Ganges-Brahmaputra rivers: In the Ganges-Brahmaputra system, the source of sediments has been characterized using Nd and Sr isotopes by previous studies (Galy and France-Lanord, 2001; Lupker et al., 2013; Singh et al., 2008; Singh and France-Lanord, 2002). The Ganges sediments are mostly derived from the erosion of the high-grade metamorphic rocks from the High Himalaya Crystalline (HHC, about 70%) and the rest of the sediments from the low-grade metasedimentary rocks of the Lesser Himalayas (LH, 30%). For the Brahmaputra, Singh and France-Lanord (2002) estimate that in average, between 50 and 70% of the sediments are derived from the erosion of the HHC, about 10% from the LH and the rest from the erosion of the Trans-Himalayan Batholith (igneous rocks). However, the provenance of the Brahmaputra sediments can be quite variable depending upon the depth and the sampling date (Lupker et al., 2013). As the Nd and Sr isotope composition of the Brahmaputra sediment sample from this study (BR8210) has been measured by Lupker et al., (2013), we can use these data along with the estimate of the Himalayan rock end-members (Galy and France-Lanord, 2001; Singh et al., 2008; Singh and France-Lanord, 2002) to determine the provenance for this sample. We calculate that

for this Brahmaputra sample, the proportion of LH, and therefore of shales-derived sediments, is $20\pm 10\%$.

Mackenzie River: The upper crust drained by the main Mackenzie tributaries is composed by a mixture between shales and igneous rocks having granodioritic to granitic composition (Driver et al., 2000; Reeder et al., 1972). The Nd and Sr isotope composition has not been measured on sediments from the Mackenzie tributaries. However, a large dataset of shale chemical composition from the Liard River basin is available in Ross and Bustin, (2009). Dellinger, (2013) shows that some insoluble elements (Cr, Ti, Eu, Cs) have significantly distinct concentration between shales and granitic rocks. It is therefore possible to use these element contents to calculate the proportion of shale-derived and igneous rock-derived particles in the fine sediments of the Mackenzie rivers (Fig. DR.5). The Peel River that drain almost exclusively shales have sediments having chemical signature similar to the mean composition of shales from Ross and Bustin (2009). Other rivers have lower Al-normalized concentrations showing that they contain a larger, yet small fraction of igneous-derived material compared to the Peel. We estimate the composition of the shale end member using the data from Ross and Bustin (2009), Garziona et al., (1997) and Morrow, (1999). For the shale end member, we estimate that $Ti/Al = 0.057\pm 0.002$ (median value $\pm 2\sigma/\sqrt{n}$, $n=101$), $Cr/Al = 1.54\pm 0.12\times 10^{-3}$ ($n=106$) and $Cs/Al = 0.10\pm 0.01\times 10^{-3}$ ($n=50$). For the granitic end member, we use data from Driver et al., (2000) and Heffernan, (2004): $Ti/Al = 0.030\pm 0.005$ ($n=60$), $Cr/Al = 0.13\pm 0.29\times 10^{-3}$ ($n=23$) and median $Cs/Al = 0.050\pm 0.005\times 10^{-3}$ ($n=50$). We calculate that the Mackenzie tributaries contain between 70 and 100% of shale-derived sediments.

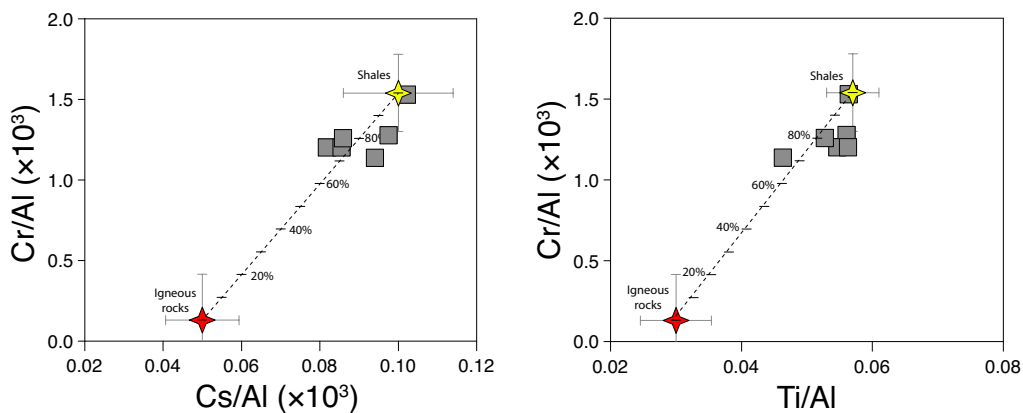


Fig. DR5: Cr/Al as a function of the Cs/Al and Ti/Al ratio for Mackenzie River fine sediments. The igneous and shales end members are also represented (as discussed in the section 2 of the data repository).

Western Cordillera: Most of the bedrock of the Canadian western cordillera rivers is formed of volcanic rocks, with a minor contribution of shales. The Sr isotope composition of both dissolved and suspended material for the Skeena and Stikine rivers is unradiogenic (<0.707 ; Gaillardet et al., 2003) so we consider that the proportion of shales is lower than $10 \pm 10\%$.

Changjiang: For the Changjiang, we determined the proportion of shales by using the Li concentration in sediments and the relation between the Li concentration and proportion of shales defined by other rivers (see section “The influence of lithology” in the main text). We note that samples CJ33 and CJ34 correspond to rivers draining lowland areas (Wang et al., 2015). Hence, the provenance method based on Li might not be applicable to these rivers. However, their weathering intensity is quite low compared to lowland rivers of the Amazon. In addition, the Li content of the sediments is higher than 85 ppm indicating that even if they have lost a significant proportion of Li, their main source is the erosion of shales (in agreement with the lithology in these two river basin; Wang et al., 2015). Based on the uncertainty on the trend between Li concentration and proportion of shales defined by other rivers, the estimated uncertainty on the proportion of shale-derived particles for Changjiang samples is $\pm 15\%$.

3. Calculation of the $\delta^7\text{Li}_{\text{rock}}$ and $(\text{Li}/\text{Al})_{\text{rock}}$

To calculate the Li isotope composition of the bedrock, $\delta^7\text{Li}_{\text{rock}}$ (which is the $\delta^7\text{Li}$ that the sediments would have in absence of chemical weathering) and $(\text{Li}/\text{Al})_{\text{rock}}$ values, we can use the proportion of shales calculated before (γ_{sha}) in combination with the mixing equation in Dellinger et al., (2014):

$$\left(\frac{\text{Li}}{\text{Al}}\right)_{\text{rock}} = \left(\frac{\text{Li}}{\text{Al}}\right)_{\text{sha}} \times \gamma_{\text{sha}} + \left(\frac{\text{Li}}{\text{Al}}\right)_{\text{Ign}} \times (1 - \gamma_{\text{sha}})$$

and

$$(\delta^7\text{Li})_{\text{rock}} = (\delta^7\text{Li})_{\text{sha}} \times \left(\frac{(\text{Li}/\text{Al})_{\text{sha}}}{(\text{Li}/\text{Al})_{\text{sed}}} \right) \times \gamma_{\text{sha}} + (\delta^7\text{Li})_{\text{Ign}} \times \left(\frac{(\text{Li}/\text{Al})_{\text{Ign}}}{(\text{Li}/\text{Al})_{\text{sed}}} \right) \times (1 - \gamma_{\text{sha}})$$

With the shale end member composition being $\delta^7\text{Li}$ of $-0.5 \pm 1\text{‰}$ and Li/Al of $1.0 \pm 0.1 \times 10^{-3}$ and the igneous end member $\delta^7\text{Li}$ of $+3.5 \pm 1.5\text{‰}$, Li/Al of $0.30 \pm 0.12 \times 10^{-3}$ (Dellinger et al., 2014; Sauzéat et al., 2015). For the Congo River, Henchiri et al., (2016) show that the TTG of the Congo basin have possibly high $\delta^7\text{Li}$ ($+6$ to $+7\text{‰}$) due to their Archean age. Therefore, we use a higher $\delta^7\text{Li}_{\text{ign}}$ end member of $+5 \pm 2\text{‰}$ for the Congo River.

For the Ganges-Brahmaputra (G-B) system, we note that G-B sands, which are derived mostly from the erosion of high-grade metasedimentary rocks, have a distinct composition ($\delta^7\text{Li}$ of $0 \pm 1\text{‰}$, Li/Al of $0.40 \pm 0.10 \times 10^{-3}$; Dellinger et al., 2014) relative to igneous and sedimentary rocks. Hence, for the Ganges-Brahmaputra, we consider a mixing between a shale end member and a high-grade metamorphic end member having a $\delta^7\text{Li}$ of $0 \pm 1\text{‰}$ and a Li/Al ratio of $0.40 \pm 0.10 \times 10^{-3}$ (Dellinger et al., 2014).

4. Comparison between the Li depletion in sediments and the fraction of Li transported into the dissolved load

As Li is a soluble element, Li concentration in river sediments is sensitive to weathering. In particular, for a given proportion of shales, lowland rivers have a lower Li/Al ratio than rivers draining mountain ranges (Fig. 2). Yet, this Li-depletion in lowland rivers could be explained by i) the loss of Li by weathering processes and/or ii) an incorrect estimation of the bedrock (*i.e.* the proportion of shales and/or the composition of the rock end members). Following the approach of Gaillardet et al., (1999), we can test these two hypotheses using a steady state assumption. At steady state, the fraction of Li transported as dissolved should be equal to the Li depletion (Li weathering index $(\text{Li}/\text{Al})_{\text{sed}}/(\text{Li}/\text{Al})_{\text{rock}}$, see main text) in the sediments. The proportion of Li riverine export occurring as dissolved (w^{Li} as defined by Bouchez et al., 2013) can be calculated using the Li concentration in the dissolved and suspended load and the concentration of suspended sediment transported by the rivers.

$$w^{Li} = \frac{1}{1 + \frac{[Li]_{sed}}{[Li]_{diss}} \times [SPM]}$$

With $[Li]_{sed}$, $[Li]_{diss}$ and $[SPM]$ being respectively the Li concentrations in sediment, and dissolved load, and the suspended sediment concentration. The comparison between $(1-w^{Li})$ and the weathering index (Fig. DR6) shows that at first order, there is a relatively good agreement between the amount of Li depletion in the sediments and the fraction of dissolved Li export, within uncertainties.

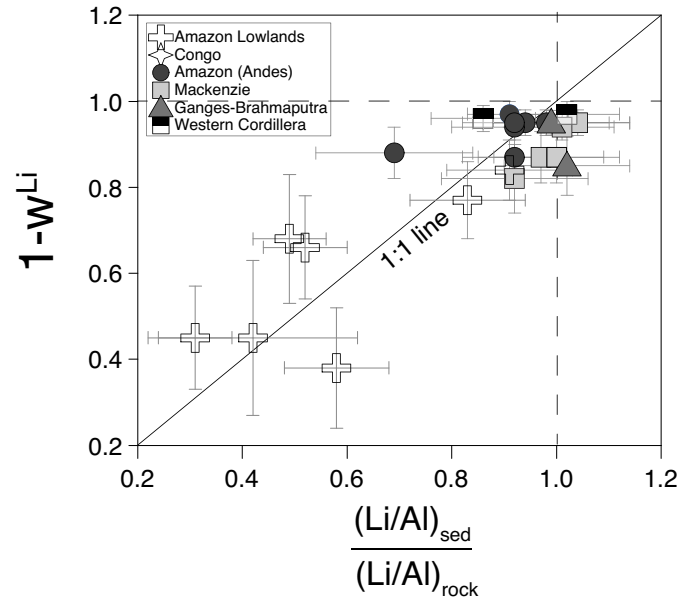


Figure DR6: Proportion of solid Li transported by the river ($1-w^{Li}$) as a function of the Li weathering index calculated as $(Li/Al)_{sed} / (Li/Al)_{rock}$. The 1:1 line corresponds to the line for which $(1-w^{Li})$ is equal to $(Li/Al)_{sed} / (Li/Al)_{rock}$, meaning that the fraction of Li dissolved export is compatible with that of the solid export, a requisite for steady state denudation.

This calculation does not take into account the change in Li concentration in the sediments with depth. Indeed, in large rivers draining mountain ranges, the grain size distribution of the suspended sediment load is generally bimodal with a fine mode composed of clay and silt particles and a coarse mode composed of sands (Bouchez et al., 2011a; Dellinger, 2013; Lupker et al., 2012). The fine sediments ($< 63 \mu m$) are relatively homogeneously distributed in the water

column while coarse sediments (size > 63 μm) concentration strongly increases from the surface toward the bottom of the river channel (Bouchez et al., 2011b; Galy et al., 2007; Lupker et al., 2011). As coarse sediments are mostly composed of quartz, the Li concentration in sediments decreases toward the bottom of the channel because of dilution by increasing quartz amount. This implies that the $(1-w^{\text{Li}})$ values calculated from surface sediments are underestimated. However, as shown by Dellinger et al., (2015), the difference is relatively small (less than 0.05 units of w^{Li} values). In addition, we note that in lowland rivers suspended sediments collected at any depth contain a negligible amount of coarse (> 63 μm) particles (Meade, 1985). Therefore, the conclusions about the agreement between $(1-w^{\text{Li}})$ values and $(\text{Li}/\text{Al})_{\text{sed}} / (\text{Li}/\text{Al})_{\text{rock}}$ stands whether these values are depth-integrated or calculated with surface sediments only.

Sample name	Basin	River	Location	Longitude	Latitude	References for Li isotope data	Surface	Discharge	D ₅₀	Al	Li	¹⁴³ Nd/ ¹⁴⁴ Nd	ε _{Nd(0)}	δ ⁷ Li	Li/Al (×10 ⁻³)	CIA	[Li] _{diss}	TDS _{Si}
							(km ²)	(m ³ /s)	(μm)	(μg/g)	(μg/g)		(‰)	(g/g)		(mmol/L)	(mg/L)	
Amazon Lowlands																		
AM01-17	Amazon	Orthon	Riberalta	66°02'11.0" W	10°48'39.9" S	Dellinger et al., (2014, 2015)	33485	475		89835	59	0.512158	-9.37	-5.7	0.66	82	0.31	19.6
AM01-18	Amazon	Yata	Mouth	65°35'00.2" W	10°58'44.8" S	Dellinger et al., (2014, 2015)	20770	222		115390	100	0.512043	-11.61	-5.3	0.87	86	0.08	10.4
AM01-27	Amazon	Trombetas	Mouth	N.D.	N.D.	Dellinger et al., (2014, 2015)	250000	6200		115312	20	0.511609	-20.08	-5.1	0.17	92	0.037	7.8
AM01-22 MES	Amazon	Negro	Paricatuba	N.D.	N.D.	Dellinger et al., (2014, 2015)	691000	29000		113016	44	0.511930	-13.81	-2.7	0.39	87	0.031	4.3
AM6/1-09 MES	Amazon	Urucara	Mouth	N.D.	N.D.	Dellinger et al., (2014, 2015)	150000	3000		110691	43	0.511895	-14.50	-4.0	0.39	87	0.049	8.7
AM01-29	Amazon	Tapajos	Mouth	N.D.	N.D.	Dellinger et al., (2014, 2015)	490000	13500		129350	27	0.511698	-18.33	-6.0	0.20	93	0.038	11.5
Congo																		
C89-63 PK 1105	Congo	Congo	Mouth	N.D.	N.D.	Henchiri et al., (2016), this study	3700000	38026		29	0.511796	-16.43	-6.2			0.10	15.0	
C89-65 PK 1105	Congo	Congo	Mouth	N.D.	N.D.	Henchiri et al., (2016), this study	3700000	38026		117769	23	0.511833	-15.70	-5.1	0.19	92	0.10	15.0
Amazon Andes																		
AM01_08	Amazon	Alto Beni	Mouth	67°40'18" W	15°07'24" S	Dellinger et al., (2014, 2015)	21197	750		84800	79	0.512011	-12.24	-1.8	0.93	74	1.62	26.4
AM07_04	Amazon	Beni	Rurrenabaque	67°32'05.2" W	14°28'26.8" S	Dellinger et al., (2014, 2015)	69980	2153	44	95289	87	0.511998	-12.49	-3.1	0.91	76	1.28	17.4
AM07_09	Amazon	Beni	Riberalta	66°05'53.8" W	10°58'44.8" S	Dellinger et al., (2014, 2015)	114379	3772	50	83319	78	0.511994	-12.56	-2.9	0.94	76	0.57	14.9
AM05_19	Amazon	Madeira	Foz Madeira	58°48'30.1" W	03°27'00.9" S	Dellinger et al., (2014, 2015)	1325000	31250	36	111822	99	0.512097	-10.55	-3.4	0.89	78	0.18	12.8
AM06_65	Amazon	Amazonas	Obidos	55°30'18.6" W	01°56'05.3" S	Dellinger et al., (2014, 2015)	4618750	182200	27	107370	81	0.512141	-9.70	-3.6	0.75	78	0.14	11.4
AM08_13	Amazon	Ucayali	Janaro Herrera	73°40'28" W	04°54'28" S	Dellinger et al., (2014, 2015)	352593	12690	50	85684	58	0.512158	-9.37	-1.9	0.73	68	0.66	15.6
AM06_14	Amazon	Solimoes	Manacapuru	60°33'11.0" W	03°19'41.0" S	Dellinger et al., (2014, 2015)	2148000	103000	35	100535	58	0.512101	-10.48	-2.6	0.58	76	0.14	13.1
AM08_36	Amazon	Pastaza	Embouchure	N.D.	N.D.	Dellinger et al., (2014, 2015)	39000	2770	61	97031	26	0.512512	-2.45	-0.2	0.27	68	0.14	24.2
AM07_14	Amazon	Madre de Dios	Riberalta	66°05'54" W	10°57'46" S	Dellinger et al., (2014, 2015)	124231	5602	35	94447	75	0.512106	-10.38		0.79	77	0.22	14.6
Mackenzie																		
CAN09_51	Mackenzie	Red Arctic	Tsiigehtchic	133°46'44.3" W	67°25'34.2" N	Dellinger et al., (2014)	18600	159	23	79022	78.1	0.512066	-11.16	-2.4	0.99	77	0.89	8.0
CAN09_04	Mackenzie	Liard	Fort_Simpson	121°17'6.2" W	61°50'8.6" N	Dellinger et al., (2014)	275000	2450	49	64955	51.2	0.511904	-14.31	-1.7	0.79	73	0.63	8.7
CAN09_47	Mackenzie	Mackenzie	Tsiigehtchic	133°43'23.8" W	67°27'30.1" N	Dellinger et al., (2014)	1680000	8990		76477	67.5	0.512013	-12.19	-1.9	0.88	76	0.57	9.5
CAN10_32	Mackenzie	Middle Channel	Inuvik	134°44'49.8" W	68°24'33.3" N	Dellinger et al., (2014)	1805200		14	90608	79.3	0.511979	-12.86	-1.7	0.88	79		
CAN09_40	Mackenzie	Peel	Fort_McPherson	134°52'8.6" W	68°19'56.6" N	Dellinger et al., (2014)	70600	686	18	80981	81.5	0.512062	-11.23	-2.1	1.01	77	0.71	8.8
CAN09_28	Mackenzie	Slave	Fort_Smith	111°53'25.7" W	60°0'59.5" N	Dellinger et al., (2014)	616400	3200	17	77069	52.8	0.512014	-12.18	-1.3	0.68	76	0.60	12.0
Changjiang																		
CJ1	Changjiang	Jinshajiang	Panzhihua	101°37.551' E	26°34.503' N	Wang et al., (2015)	285556	1309	24	80546	48.5		-2.2	0.60	69	2.07	18.6	
CJ4	Changjiang	Jinshajiang	Yibin	104°33.385' E	28°42.164' N	Wang et al., (2015)	485000	8239	22	82252	48.7		-0.5	0.59	73	0.97	18.5	
CJ14	Changjiang	Changjiang	Chongqing	N.D.	N.D.	Wang et al., (2015)	867000	10945	24	84970	53.8		-1.0	0.63	71	0.78	13.4	
CJ60	Changjiang	Changjiang	Yueyang	113°18.739' E	29°38.079' N	Wang et al., (2015)	1250000	15844			76.6		-1.3			0.37	14.2	
CJ52	Changjiang	Changjiang	Wuhan	114°29.876' E	30°40.747' N	Wang et al., (2015)	1488000	22340	45	83606	61.9		-1.6	0.74	72	0.44	16.9	
CJ36	Changjiang	Changjiang	Datong	117°38.324' E	30°46.085' N	Wang et al., (2015)	1705000	28266	26	98071	70.9		-1.7	0.72	76	0.27	16.7	
CJ2	Changjiang	Yalongjiang	Panzhihua	101°49.200' E	26°40.127' N	Wang et al., (2015)	128444	1857		161917	41.3		0.1	0.26	63	0.68	17.0	
CJ5	Changjiang	Minjiang	Yibin	104°33.721' E	28°48.779' N	Wang et al., (2015)	133000	2732	33	83956	67.0		0.0	0.80	67	0.45	18.3	
CJ33	Changjiang	Xiangjiang	Changsha	112°56.795' E	28°07.364' N	Wang et al., (2015)	94700	2405	15	108398	86.7		-3.4	0.80	82	0.16	13.3	
CJ34	Changjiang	Ganjiang	Nanchang	115°52.053' E	28°40.206' N	Wang et al., (2015)	80900	2104		103092	88.6		-4.7	0.86	80	0.25	22.4	
Ganges-Brahmaputra																		
IND99-19 MES	Ganges	Ganga	Harding bridge	89°1'28.739" E	24°3'10.44" N	Dellinger et al., (2014)	1050000	15622		94322	55.0	0.511725	-17.81	-1.02	0.58		0.49	29.8
BR8210	Brahmaputra	Brahmaputra	Jamuna bg.	N.D.	N.D.	Dellinger et al., (2014)	580000	16161		87063	47	0.511897	-14.45	-1.38	0.54	66	1.08	11.8
Western cordillera																		
CAN99-28 MES	Skeena	Skeena	Kitwanga	N.D.	N.D.	This study	42200	962		83647	28.7			2.0	0.34	64	0.05	7.9
CAN99-39 MES	Stikine	Stikine	Iskut	N.D.	N.D.	This study	29300	755		75071	24.5			1.2	0.33	63	0.10	7.2

Table DR1: Data for all the clastic river sediment samples from this study. The “references” column indicate the source of Li isotope and concentration data used for this study. CIA values have been calculated using similar formula as in McLennan, (1993). Silicate weathering rates (calculated as $Q \times ([Na]_{sil} + [K]_{sil} + [Ca]_{sil} + [Mg]_{sil})$, with “[X]_{sil}” the silicate-derived dissolved concentration of the element X and "Q" the discharge) and erosion rates are from multiple sources (Armijos et al., 2013; Dellinger et al., 2015b; Gaillardet et al., 1997; Guyot et al., 1996; Moquet et al., 2011; Wittmann et al., 2010). [Li]_{diss} corresponds to the dissolved Li concentration, TDS_{sil} to the total dissolved load deriving from silicates weathering (in mg/L). Sediment concentration corresponds to the mean measured suspended sediment concentration for each river (in mg/L).

Sample name	River	Weathering Intensity (W/D)	Proportion of shales	δ^{Li}_{rock}	$\Delta^{Li}_{sed-rock}$	$1-w^{Li}$	$(Li/Al)_{sed} / (Li/Al)_{rock}$	References
(‰)								
(‰)								
Amazon Lowlands								
AM01-17	Orthon	0.137	0.73 ± 0.12	-0.10 ± 0.89	-5.58 ± 1.07	0.77 ± 0.09	0.83 ± 0.11	Dellinger et al., (2014, 2015)
AM01-18	Yata	0.259	0.95 ± 0.14	-0.36 ± 0.93	-4.90 ± 1.12	0.84 ± 0.07	0.91 ± 0.12	Dellinger et al., (2014, 2015)
AM01-27	Trombetas	0.504	0.00 ± 0.10	3.50 ± 1.50	-8.63 ± 1.75	0.38 ± 0.14	0.58 ± 0.10	Dellinger et al., (2014, 2015)
AM01-22 MES	Negro	0.291	0.73 ± 0.10	-0.11 ± 0.90	-2.61 ± 1.20	0.68 ± 0.15	0.49 ± 0.07	Dellinger et al., (2014, 2015)
AM6/1-09 MES	Urucara	0.367	0.65 ± 0.11	0.04 ± 0.90	-4.02 ± 1.10	0.66 ± 0.12	0.52 ± 0.08	Dellinger et al., (2014, 2015)
AM01-29	Tapajos	0.581	0.28 ± 0.14	1.37 ± 1.11	-7.37 ± 1.50	0.45 ± 0.18	0.42 ± 0.20	Dellinger et al., (2014, 2015)
Congo								
C89-63 PK 1105	Congo	0.396	0.35 ± 0.10			0.51 ± 0.13		Henchiri et al., (2016), this study
C89-65 PK 1105	Congo	0.396	0.48 ± 0.11	1.00 ± 1.11	-6.05 ± 1.20	0.45 ± 0.12	0.31 ± 0.07	Henchiri et al., (2016), this study
Amazon Andes								
AM07_04	Beni	0.005	1.00 ± 0.10	-0.50 ± 1.00	-2.59 ± 1.12	0.97 ± 0.03	0.91 ± 0.08	Dellinger et al., (2014, 2015)
AM07_09	Beni	0.015	1.00 ± 0.10	-0.50 ± 1.00	-2.38 ± 1.12	0.95 ± 0.03	0.94 ± 0.08	Dellinger et al., (2014, 2015)
AM05_19	Madeira	0.049	0.85 ± 0.12	-0.30 ± 1.00	-3.08 ± 1.10	0.95 ± 0.03	0.98 ± 0.12	Dellinger et al., (2014, 2015)
AM06_65	Amazonas	0.058	0.76 ± 0.12	-0.2 ± 1.00	-3.43 ± 1.09	0.94 ± 0.03	0.92 ± 0.12	Dellinger et al., (2014, 2015)
AM08_13	Ucayali	0.029	0.70 ± 0.08	-0.10 ± 0.89	-1.80 ± 1.05	0.87 ± 0.06	0.92 ± 0.10	Dellinger et al., (2014, 2015)
AM06_14	Solimoes	0.096	0.80 ± 0.12	-0.24 ± 0.96	-2.36 ± 1.10	0.88 ± 0.06	0.69 ± 0.15	Dellinger et al., (2014, 2015)
AM08_36	Pastaza		0.32 ± 0.10	1.10 ± 0.93	-1.28 ± 1.14			Dellinger et al., (2014, 2015)
AM07_14	Madre de Dios	0.035	0.82 ± 0.10			0.95 ± 0.03	0.92 ± 0.10	Dellinger et al., (2014, 2015)
Mackenzie								
CAN09_51	Red Arctic	0.003	0.95 ± 0.10	-0.47 ± 0.91	-1.94 ± 1.10	0.95 ± 0.03	1.04 ± 0.10	Dellinger et al., (2014)
CAN09_04	Liard	0.007	0.75 ± 0.10	-0.04 ± 0.91	-1.62 ± 1.08	0.87 ± 0.06	0.97 ± 0.12	Dellinger et al., (2014)
CAN09_47	Mackenzie	0.016	0.80 ± 0.10	-0.20 ± 0.91	-1.72 ± 1.08	0.87 ± 0.06	1.00 ± 0.12	Dellinger et al., (2014)
CAN10_32	Middle Channel		0.70 ± 0.10	-0.04 ± 0.91	-1.65 ± 1.05		1.10 ± 0.12	Dellinger et al., (2014)
CAN09_40	Peel	0.006	1.00 ± 0.10	-0.50 ± 0.91	-1.56 ± 1.10	0.94 ± 0.03	1.01 ± 0.10	Dellinger et al., (2014)
CAN09_28	Slave	0.020	0.65 ± 0.10	0.15 ± 0.91	-1.45 ± 1.07	0.82 ± 0.08	0.92 ± 0.14	Dellinger et al., (2014)
Changjiang								
CJ1	Jinshajiang	0.008	0.50 ± 0.15	0.82 ± 1.13	-3.02 ± 1.26	0.80 ± 0.09		Wang et al., (2015)
CJ4	Jinshajiang	0.016	0.50 ± 0.15	0.78 ± 1.16	-1.28 ± 1.20	0.79 ± 0.09		Wang et al., (2015)
CJ14	Changjiang	0.003	0.58 ± 0.15	0.48 ± 1.03	-1.48 ± 1.20	0.92 ± 0.04		Wang et al., (2015)
CJ60	Changjiang	0.006	0.84 ± 0.15	0.49 ± 1.02	-1.79 ± 1.13	0.96 ± 0.03		Wang et al., (2015)
CJ52	Changjiang	0.012	0.68 ± 0.15	-0.3 ± 0.99	-1.29 ± 1.11	0.92 ± 0.04		Wang et al., (2015)
CJ36	Changjiang	0.010	0.78 ± 0.15	-0.2 ± 0.99	-1.52 ± 1.10	0.95 ± 0.03		Wang et al., (2015)
CJ2	Yalongjiang	0.032	0.38 ± 0.15	1.40 ± 1.20	-1.30 ± 1.63	0.64 ± 0.12		Wang et al., (2015)
CJ5	Minjiang	0.019	0.74 ± 0.15	-0.1 ± 0.97	0.07 ± 1.09	0.92 ± 0.04		Wang et al., (2015)
CJ33	Xiangjiang	0.014	0.92 ± 0.15	-0.4 ± 0.93	-3.04 ± 1.10	0.92 ± 0.04		Wang et al., (2015)
CJ34	Ganjiang	0.023	1.00 ± 0.15	-0.50 ± 1.00	-4.20 ± 1.20	0.90 ± 0.06		Wang et al., (2015)
Ganges-Brahmaputra								
IND99-19 MES	Ganga	0.026	0.30 ± 0.10	-0.40 ± 0.95	-0.62 ± 1.00	0.95 ± 0.03	0.99 ± 0.15	Dellinger et al., (2014)
BR8210	Brahmaputra	0.013	0.20 ± 0.10	0.33 ± 1.10	-1.71 ± 1.30	0.85 ± 0.07	1.02 ± 0.12	Dellinger et al., (2014)
Western cordillera								
CAN99-28 MES	Skeena	0.016	0.00 ± 0.10	3.50 ± 1.10	-1.45 ± 1.00	0.97 ± 0.03	1.02 ± 0.10	This study
CAN99-39 MES	Stikine	0.007	0.10 ± 0.10	2.06 ± 1.20	-0.86 ± 1.00	0.96 ± 0.03	0.86 ± 0.10	This study

Table DR1 (next): The parameter “ $1-w^{Li}$ ” corresponds to the proportion of Li transported in the solid form by the river. The parameter “ $(Li/Al)_{sed} / (Li/Al)_{rock}$ ” is the ratio between the Li/Al of the sediments and the Li/Al of the rock.

Sample name	ϵ_{Nd}	$^{143}Nd / ^{144}Nd$
Betou 1A	-18.30	0.511700
Betou 1B	-18.13	0.511709
Betou 1C	-18.16	0.511707
Betou 2A	-18.44	0.511693
Betou 2B	-18.92	0.511668

Table DR2: Nd isotope composition of the Soil profile samples from the Congo river basin. The $\epsilon_{Nd}(0)$ are calculated by normalizing to the CHUR value (0.512638). For details about the samples, see Henchiri et al., (2016).

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Sample name	Basin	River	Location	Longitude	Latitude	References for Li isotope data	Surface (km ²)	Discharge (m ³ /s)	D ₉₀ (µm)	AI (µg/g)	LI (µg/g)	¹³ Nd/ ¹⁴⁴ Nd ε _{ND} (0)	δ ⁷ Li (‰)	L/LI (×10 ⁻³)	CIA (mmol/L)
Amazon Lowlands															
AM01-17	Amazon	Orthon	Fiberalta	66°02'1.0" W	10°48'39.9" S	Dellinger et al., (2014, 2015)	33485	475		89835	59	0.512158	-9.37	-5.7	0.66
AM01-18	Amazon	Yata	Mouth	65°35'00.2" W	10°58'44.8" S	Dellinger et al., (2014, 2015)	20770	222		115390	100	0.512043	-11.61	-5.3	0.87
AM01-27	Amazon	Tionbetas	Mouth	N.D.	N.D.	Dellinger et al., (2014, 2015)	250000	6200		115312	20	0.511609	-20.08	-5.1	0.17
AM01-22 MES	Amazon	Negro	Paricatuba	N.D.	N.D.	Dellinger et al., (2014, 2015)	691000	29000		113016	44	0.511930	-13.81	-2.7	0.39
AM01-09 MES	Amazon	Uucara	Mouth	N.D.	N.D.	Dellinger et al., (2014, 2015)	150000	3000		110691	43	0.511895	-14.50	-4.0	0.39
AM01-29	Amazon	Tapajos	Mouth	N.D.	N.D.	Dellinger et al., (2014, 2015)	490000	13500		129350	27	0.511698	-18.33	-6.0	0.20
Congo															
C89-63 PK 1105	Congo	Congo	Mouth	N.D.	N.D.	Henchiri et al., (2016), this study	3700000	38026		29		0.511796	-16.43	-6.2	0.10
C89-65 PK 1105	Congo	Congo	Mouth	N.D.	N.D.	Henchiri et al., (2016), this study	3700000	38026		117769	23	0.511833	-15.70	-5.1	0.19
Amazon Andes															
AM01_08	Amazon	Alto Beni	Mouth	67°40'18" W	15°07'24" S	Dellinger et al., (2014, 2015)	21197	750		84800	79	0.512011	-12.24	-1.8	0.93
AM07_04	Amazon	Beni	Furenabaque	67°32'05.2" W	14°28'26.8" S	Dellinger et al., (2014, 2015)	69980	2153	44	95289	87	0.511998	-12.49	-3.1	0.91
AM07_09	Amazon	Beni	Fiberalta	66°05'53.8" W	10°58'44.8" S	Dellinger et al., (2014, 2015)	114379	3772	50	83319	78	0.511994	-12.56	-2.9	0.94
AM05_19	Amazon	Madeira	Foz Madeira	58°48'30.1" W	03°27'00.9" S	Dellinger et al., (2014, 2015)	1325000	31250	36	111822	99	0.512097	-10.55	-3.4	0.89
AM06_65	Amazon	Amazonas	Obidos	55°30'18.6" W	01°56'05.3" S	Dellinger et al., (2014, 2015)	4618750	182200	27	107370	81	0.512141	-9.70	-3.6	0.75
AM06_13	Amazon	Ucayali	Jenaro Herrera	73°40'28" W	04°54'28" S	Dellinger et al., (2014, 2015)	352593	12090	50	85664	58	0.512158	-9.37	-1.9	0.73
AM06_14	Amazon	Solimoes	Manacapuru	60°33'11.0" W	03°19'41.0" S	Dellinger et al., (2014, 2015)	2148000	103000	35	100535	58	0.512101	-10.48	-2.6	0.58
AM08_36	Amazon	Pastaza	Embochure	N.D.	N.D.	Dellinger et al., (2014, 2015)	39000	2770	61	97031	26	0.512512	-2.45	-0.2	0.27
AM07_14	Amazon	Madre de Dios	Fiberalta	66°05'54" W	10°57'46" S	Dellinger et al., (2014, 2015)	124231	5602	35	94447	75	0.512106	-10.38		0.79
Mackenzie															
CAN09_51	Mackenzie	Fed Arctic	Tsigilchic	133°46'44.3" W	67°25'34.2" N	Dellinger et al., (2014)	18600	159	23	79022	78.1	0.512066	-11.16	-2.4	0.99
CAN09_04	Mackenzie	Liard	Fort_Simpson	121°17'6.2" W	61°50'8.6" N	Dellinger et al., (2014)	275000	2450	49	64955	51.2	0.511904	-14.31	-1.7	0.79
CAN09_47	Mackenzie	Mackenzie	Tsigilchic	133°43'23.8" W	67°27'30.1" N	Dellinger et al., (2014)	1680000	8990		76477	67.5	0.512013	-12.19	-1.9	0.88
CAN10_32	Mackenzie	Middle Channel	Inuvik	134°44'9.8" W	68°24'33.3" N	Dellinger et al., (2014)	1805200		14	90608	79.3	0.511979	-12.86	-1.7	0.88
CAN09_40	Mackenzie	Peel	Fort_McPherson	134°52'8.6" W	68°19'56.6" N	Dellinger et al., (2014)	70600	686	18	80981	81.5	0.512062	-11.23	-2.1	1.01
CAN09_28	Mackenzie	Slave	Fort_Smith	111°53'25.7" W	60°0'59.5" N	Dellinger et al., (2014)	616400	3200	17	77069	52.8	0.512014	-12.18	-1.3	0.68
Changjiang															
CJ1	Changjiang	Jinshajiang	Panzhuhua	101°37'55.1" E	26°34'50.3" N	Wang et al., (2015)	285556	1309	24	80546	48.5			-2.2	0.60
CJ4	Changjiang	Yibin		104°33'3.85" E	28°42'1.64" N	Wang et al., (2015)	485000	8239	22	82252	48.7			-0.5	0.59
CJ14	Changjiang	Chongqing		N.D.	N.D.	Wang et al., (2015)	867000	10945	24	84970	53.8			-1.0	0.63
CJ60	Changjiang	Yueyang		113°18'7.39" E	29°38'07.9" N	Wang et al., (2015)	1250000	15844			76.6			-1.3	0.37
CJ62	Changjiang	Changjiang	Wuhan	114°29'8.76" E	30°40'7.47" N	Wang et al., (2015)	1488000	22340	45	83606	61.9			-1.6	0.74
CJ36	Changjiang	Changjiang	Dalong	117°38'3.24" E	30°46'0.85" N	Wang et al., (2015)	1705000	28286	26	98071	70.9			-1.7	0.72
CJ2	Changjiang	Changjiang	Panzhuhua	101°49'20.0" E	26°40'1.27" N	Wang et al., (2015)	128444	1857		161917	41.3			0.1	0.26
CJ5	Changjiang	Minjiang	Yibin	104°33'7.21" E	28°48'17.7" N	Wang et al., (2015)	133000	2782	33	83956	67.0			0.0	0.80
CJ33	Changjiang	Xiangsha	Changsha	112°56'7.95" E	28°07'3.64" N	Wang et al., (2015)	94700	2405	15	108398	86.7			-3.4	0.82
CJ34	Changjiang	Ganjiang	Nanchang	115°52'0.83" E	28°40'2.06" N	Wang et al., (2015)	80900	2104		103092	88.6			-4.7	0.86
Ganges-Brahmaputra															
IND99-19 MES	Ganges	Ganga	Harding bridge	89°1'28.739" E	24°3'10.44" N	Dellinger et al., (2014)	1050000	15622		94322	55.0	0.511725	-17.81	-1.02	0.58
BR8210	Brahmaputra	Brahmaputra	Jamuna bg.	N.D.	N.D.	Dellinger et al., (2014)	580000	16161		87063	47	0.511897	-14.45	-1.38	0.54
Western cordillera															
CAN9-28 MES	Sikeena	Sikeena	Kiwanga	N.D.	N.D.	This study	42200	962		83647	28.7			2.0	0.34
CAN9-39 MES	Sikeena	Sikeena	Iskut	N.D.	N.D.	This study	29300	755		75071	24.5			1.2	0.33

TDS _{SI}	Sample name	River	Cation silicate Weatherin g rate	Total silicate Weatherin rate (W)	Sediment concentrat ion	Erosion rate (E)	Denudation rate (D)	Weathering Intensity (W/D)	Proportion of shales	$\delta^{18}\text{O}_{\text{rock}}$	$\Delta^{17}\text{O}_{\text{ker/rock}}$	$1-\text{w}^{\text{Li}}$	(L/A) _{sed} / (L/A) _{rock}
(mg/L)			(t/km ² /yr)	(t/km ² /yr)	(mg/L)	(t/km ² /yr)	(t/km ² /yr)			(‰)	(‰)		
Amazon Lowlands													
19.6	AM01-17	Orthon	2.8	8.8	123	55	64	0.137	0.73 ± 0.12	-0.10 ± 0.89	-5.58 ± 1.07	0.77 ± 0.09	0.83 ± 0.11
10.4	AM01-18	Yata	0.8	3.5	30	10	14	0.259	0.95 ± 0.14	-0.36 ± 0.93	-4.90 ± 1.12	0.84 ± 0.07	0.91 ± 0.12
7.8	AM01-27	Trombetas		6.1	8	6	12	0.504	0.00 ± 0.10	3.50 ± 1.50	-8.63 ± 1.75	0.38 ± 0.14	0.58 ± 0.10
4.3	AM01-22 MES	Negro		5.7	10	14	20	0.291	0.73 ± 0.10	-0.11 ± 0.90	-2.61 ± 1.20	0.68 ± 0.15	0.49 ± 0.07
8.7	AM6/1-09 MES	Urucara		5.5	15	10	15	0.367	0.65 ± 0.11	0.04 ± 0.90	-4.02 ± 1.10	0.66 ± 0.12	0.52 ± 0.08
11.5	AM01-29	Tapajos		10.0	8	7	17	0.581	0.28 ± 0.14	1.37 ± 1.11	-7.37 ± 1.50	0.45 ± 0.18	0.42 ± 0.20
Congo													
15.0	C89-63 PK 1105	Congo		5.3	25	8.1	13	0.396	0.35 ± 0.10			0.51 ± 0.13	
15.0	C89-65 PK 1105	Congo		5.3	25	8.1	13	0.396	0.48 ± 0.11	1.00 ± 1.11	-6.05 ± 1.20	0.45 ± 0.12	0.31 ± 0.07
Amazon Andes													
26.4	AM01_08	Alto Beni	20.5	29.5	3403	3800	3829	0.008	1.00 ± 0.10	-0.50 ± 1.00	-1.26 ± 1.00		0.91 ± 0.08
17.4	AM07_04	Beni	8.9	16.9	3234	3140	3157	0.005	1.00 ± 0.10	-0.50 ± 1.00	-2.59 ± 1.12	0.97 ± 0.03	0.91 ± 0.08
14.9	AM07_09	Beni	5.8	15.5	980	1020	1036	0.015	1.00 ± 0.10	-0.50 ± 1.00	-2.38 ± 1.12	0.95 ± 0.03	0.94 ± 0.08
12.8	AM05_19	Madeira	3.2	9.5	248	184	194	0.049	0.85 ± 0.12	-0.30 ± 1.00	-3.08 ± 1.10	0.95 ± 0.03	0.98 ± 0.12
11.4	AM06_65	Amazonas		14.2	186	231	245	0.058	0.78 ± 0.12	-0.15 ± 1.00	-3.43 ± 1.09	0.94 ± 0.03	0.92 ± 0.12
15.6	AM08_13	Ucayali	8.5	16.9	527	570	587	0.029	0.70 ± 0.08	-0.10 ± 0.89	-1.80 ± 1.05	0.87 ± 0.06	0.92 ± 0.10
13.1	AM06_14	Solimoes		19.8	124	188	207	0.096	0.80 ± 0.12	-0.24 ± 0.96	-2.36 ± 1.10	0.88 ± 0.06	0.69 ± 0.15
24.2	AM08_36	Pastaza	31.7	54.3					0.32 ± 0.10	1.10 ± 0.93	-1.28 ± 1.14		0.92 ± 0.10
14.6	AM07_14	Madre de Dios	7.5	20.7	401	570	591	0.035	0.82 ± 0.10			0.95 ± 0.03	
Mackenzie													
8.0	CAN09_51	Red Arctic	1.37	2.2	1454	392	395	0.005	0.95 ± 0.10	-0.47 ± 0.91	-1.94 ± 1.10	0.95 ± 0.03	1.04 ± 0.10
8.7	CAN09_04	Liard	1.19	2.4	595	167	170	0.014	0.75 ± 0.10	-0.04 ± 0.91	-1.62 ± 1.08	0.87 ± 0.06	0.97 ± 0.12
9.5	CAN09_47	Mackenzie	1.03	1.6	377	64	65	0.025	0.80 ± 0.10	-0.20 ± 0.91	-1.72 ± 1.08	0.87 ± 0.06	1.00 ± 0.12
	CAN10_32	Middle Channel							0.70 ± 0.10	-0.04 ± 0.91	-1.65 ± 1.05		1.10 ± 0.12
8.8	CAN09_40	Peel	1.73	2.7	961	295	297	0.009	1.00 ± 0.10	-0.50 ± 0.91	-1.56 ± 1.10	0.94 ± 0.03	1.01 ± 0.10
12.0	CAN09_28	Slave	1.14	2.0	348	57	59	0.033	0.65 ± 0.10	0.15 ± 0.91	-1.45 ± 1.07	0.82 ± 0.08	0.92 ± 0.14
Changliang													
18.6	CJ1	Jinshajiang	1.4	2.7	1211	175	178	0.015	0.50 ± 0.15	0.82 ± 1.13	-3.02 ± 1.26	0.80 ± 0.088	
18.5	CJ4	Jinshajiang	4.8	9.9	538	289	299	0.033	0.50 ± 0.15	0.78 ± 1.16	-1.28 ± 1.20	0.79 ± 0.088	
13.4	CJ14	Changliang	1.7	5.4	1226	488	494	0.011	0.58 ± 0.15	0.48 ± 1.03	-1.48 ± 1.20	0.92 ± 0.04	
14.2	CJ60	Changliang	1.8	5.7	740	296	302	0.019	0.84 ± 0.15	0.49 ± 1.02	-1.79 ± 1.13	0.96 ± 0.03	
16.9	CJ52	Changliang	3.3	8.0	522	271	279	0.029	0.68 ± 0.15	-0.31 ± 0.99	-1.29 ± 1.11	0.92 ± 0.04	
16.7	CJ36	Changliang	2.4	8.7	475	249	257	0.034	0.78 ± 0.15	-0.18 ± 0.99	-1.52 ± 1.10	0.95 ± 0.03	
17.0	CJ2	Yalongliang	3.1	7.8	205	93	101	0.077	0.38 ± 0.15	1.40 ± 1.20	-1.30 ± 1.63	0.64 ± 0.117	
18.3	CJ5	Mingliang	7.1	11.9	552	358	370	0.032	0.74 ± 0.15	-0.07 ± 0.97	0.07 ± 1.09	0.92 ± 0.04	
13.3	CJ33	Xiangliang	1.7	10.6	153	122	133	0.080	0.92 ± 0.15	-0.36 ± 0.93	-3.04 ± 1.10	0.92 ± 0.04	
22.4	CJ34	Ganliang	3.2	18.4	169	138	157	0.117	1.00 ± 0.15	-0.50 ± 1.00	-4.20 ± 1.20	0.90 ± 0.055	
Ganges-Brahmaputra													
29.8	IND99-19 MES	Ganga		14	1129	530	544	0.026	0.30 ± 0.10	-0.40 ± 0.95	-0.62 ± 1.00	0.95 ± 0.029	0.99 ± 0.15
11.8	BR8210	Brahmaputra		10	921	810	820	0.013	0.20 ± 0.10	0.33 ± 1.10	-1.71 ± 1.30	0.85 ± 0.069	1.02 ± 0.12
Western cordillera													
7.9	CAN99-28 MES	Skeena	4.2	5.7	362	261	266	0.021	0.00 ± 0.10	3.50 ± 1.10	-1.45 ± 1.00	0.97 ± 0.03	1.02 ± 0.10
7.2	CAN99-39 MES	Silkine	4.1	5.9	755	614	620	0.009	0.10 ± 0.10	2.06 ± 1.20	-0.86 ± 1.00	0.96 ± 0.03	0.86 ± 0.10