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Title: Formation of Obsidian Pyroclasts by Sintering of Ash Particles in
the Volcanic Conduit

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Abstract: The ranges in intensity and style of volcanic eruptions, from highly explosive Plinian eruptions to quiescent lava extrusions, depend on the style and efficiency of gas loss from ascending magma. Obsidian pyroclasts - small, glassy pieces of quenched magma found in some volcanic tephra beds - may preserve valuable information about magma degassing in their vesicle textures and volatile contents. Accurate interpretation of their textures and volatiles, however, requires understanding the mechanism of formation of the pyroclasts. Obsidian pyroclasts from the ca. 1325-1350 C.E. North Mono eruption of Mono Craters (CA, USA) were analyzed and found to have H₂O and CO₂ contents indicating that they were formed at pressures in the approximate range 3-40 MPa. Many also contain domains with differing vesicle textures, separated by boundaries containing xenocrystic material, indicating that they are composed of smaller fragments that have sutured together. More than half of the pyroclasts analyzed contained small (~10 µm), highly distorted vesicles, with multi-cusped morphology, interpreted as the remnants of interstitial gas trapped amongst sintered fragments of melt/glass. Rounded vesicles are also common and are interpreted to result from surface tension-driven relaxation of the distorted vesicles. Calculated timescales of sintering and relaxation are consistent with timescales for pyroclast formation indicated by H₂O re-equilibration within the heterogeneous pyroclasts. This sintering model for the origin of obsidian pyroclasts is further supported by the observation that spherical vesicles are found mainly in H₂O-rich pyroclasts, and distorted vesicles mainly in H₂O-poor pyroclasts. We conclude that obsidian pyroclasts generated during the North Mono eruption were formed by cycles of fragmentation, sintering/suturing, and relaxation, over a very wide range of depths within the conduit; we find no evidence to support pumice (foam) collapse as the formation mechanism. Similar textures, and the occurrence of xenolithic material, in obsidian pyroclasts in other eruptions suggest that sintering may be generally responsible for the origin of obsidian pyroclasts. Our conceptual model indicates that volatile contents in obsidian pyroclasts reflect both degassing of bubbly magma and the composition of gas trapped between sintering particles.

Highlights

- Vesicles and volatile contents were measured in Mono Craters obsidian pyroclasts
- Vesicle textures argue for formation of obsidian pyroclasts through ash sintering
- Obsidians made of domains that underwent different degassing and shear histories
- Obsidian pyroclasts formed by cycles of fragmentation, sintering, and relaxation
- No evidence found for obsidian pyroclasts formation by pumice (foam) collapse

1 **Formation of Obsidian Pyroclasts by Sintering of Ash Particles in the**
2 **Volcanic Conduit**

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20

21 **ABSTRACT**

22 The ranges in intensity and style of volcanic eruptions, from highly explosive Plinian
23 eruptions to quiescent lava extrusions, depend on the style and efficiency of gas loss from
24 ascending magma. Obsidian pyroclasts – small, glassy pieces of quenched magma found
25 in some volcanic tephra beds – may preserve valuable information about magma
26 degassing in their vesicle textures and volatile contents. Accurate interpretation of their
27 textures and volatiles, however, requires understanding the mechanism of formation of
28 the pyroclasts. Obsidian pyroclasts from the ca. 1325–1350 C.E. North Mono eruption of
29 Mono Craters (CA, USA) were analyzed and found to have H₂O and CO₂ contents
30 indicating that they were formed at pressures in the approximate range 3–40 MPa. Many
31 also contain domains with differing vesicle textures, separated by boundaries containing
32 xenocrystic material, indicating that they are composed of smaller fragments that have
33 sutured together. More than half of the pyroclasts analyzed contained small (~10 μm),
34 highly distorted vesicles, with multi-cusped morphology, interpreted as the remnants of
35 interstitial gas trapped amongst sintered fragments of melt/glass. Rounded vesicles are
36 also common and are interpreted to result from surface tension-driven relaxation of the
37 distorted vesicles. Calculated timescales of sintering and relaxation are consistent with
38 timescales for pyroclast formation indicated by H₂O re-equilibration within the
39 heterogeneous pyroclasts. This sintering model for the origin of obsidian pyroclasts is
40 further supported by the observation that spherical vesicles are found mainly in H₂O-rich
41 pyroclasts, and distorted vesicles mainly in H₂O-poor pyroclasts. We conclude that
42 obsidian pyroclasts generated during the North Mono eruption were formed by cycles of
43 fragmentation, sintering/suturing, and relaxation, over a very wide range of depths within

44 the conduit; we find no evidence to support pumice (foam) collapse as the formation
45 mechanism. Similar textures, and the occurrence of xenolithic material, in obsidian
46 pyroclasts in other eruptions suggest that sintering may be generally responsible for the
47 origin of obsidian pyroclasts. Our conceptual model indicates that volatile contents in
48 obsidian pyroclasts reflect both degassing of bubbly magma and the composition of gas
49 trapped between sintering particles.

50

51 *Keywords:*

52 obsidian

53 explosive eruption

54 volatiles

55 Mono Craters

56 degassing

57 vesicle

58

59 **1. Introduction**

60

61 The style of silicic volcanic eruptions depends on the behavior of gas that exsolves
62 from magma during ascent (Eichelberger et al., 1986; Jaupart and Allegre, 1991;
63 Gonnermann and Manga, 2003). In the current paradigm, if the gas remains trapped in
64 the bubbly magma (closed–system degassing), then the magma will eventually fragment,
65 driving explosive activity. If, instead, the magma becomes permeable, allowing the gas
66 to escape (open–system degassing), then the magma does not fragment, but collapses to a
67 bubble-poor liquid and effuses as lava (Eichelberger et al., 1986).

68 Evidence for magma degassing (exsolution of magmatic volatiles from the melt) and
69 outgassing (loss of exsolved gas from the magma) is preserved in obsidian pyroclasts –
70 glassy pieces of quenched magma found in some volcanic tephras (e.g., Eichelberger and
71 Westrich, 1981; Taylor et al., 1983; Newman et al., 1988; Rust et al., 2004; Barnes et al.,
72 2014). Variations in concentrations and isotopic compositions of volatiles among
73 obsidian pyroclasts often appear to follow trends expected from closed–system degassing
74 (Taylor et al., 1983; Newman et al., 1988; Barnes et al., 2014). This presents a problem
75 because obsidian pyroclasts are vesicle-poor, yet closed-system degassing implies that
76 gas bubbles remain within the packet of magma in which they form. Various models
77 have been proposed to explain this apparent conundrum, hypothesizing that the closed–
78 system–like volatile signatures arise from disequilibrium bubble growth (Gonnerman and
79 Manga, 2005) or from buffering the magma by CO₂-rich gas that streams through
80 brecciated magma (Rust et al., 2004). Rust et al. (2004) also propose that some obsidian

81 pyroclasts form by ‘annealing’ of glass/melt fragments at the conduit margin, challenging
82 the foam-collapse model for their origin (Eichelberger et al., 1986).

83 In this study, we investigate the origin and evolution of vesicles in obsidian pyroclasts
84 from the most recent eruption of Mono Craters (Sieh and Bursik, 1986) by comparing
85 vesicle textures to the preserved volatile contents. Based on our observations we go
86 further than Rust et al. (2004) and argue that most, if not all, of the obsidian pyroclasts
87 originate from sintering of small glass/melt fragments in the conduit (sintering is used in
88 preference to annealing, following Wadsworth et al., 2014, 2016). Volatile contents in
89 obsidian pyroclasts thus reflect the composition of gas trapped between sintered ash
90 fragments, rather than preserve degassing trends.

91

92 **2. Methods**

93

94 Obsidian pyroclasts were collected from the North Mono tephra bed, dated to the
95 interval 1325–1350 C.E. (Sieh and Bursik, 1986; Bursik, 1993). The eruption began with
96 a series of sub-plinian explosions that spread a thick blanket of tephra to the north and
97 east of Mono Craters (Fig. 1). The eruption shifted to producing pyroclastic surges and
98 flows, and ended by extruding at least five lava domes and flows (coulees) (Sieh and
99 Bursik, 1986). As many as 10 vents were active, and their alignment, plus the extrusion
100 of separate lavas, suggest that the eruption was fed from one or more dikes (Sieh and
101 Bursik, 1986).

102 Samples were collected from two pits dug into the North Mono tephra (Fig. 1). The
103 stratigraphy is the same between pits, and can be matched with that described in Sieh and

104 Bursik (1986); our layers P2, P4, and P6 match their Beds 2, 4, and 6. The relatively
105 lithic- and obsidian-rich nature of P10 suggests it is Bed 7, but it could also be a coarse
106 layer in their Gray Glassy Beds. Bursik (1993) found that Beds 1–6 have ~60–90%
107 pumice and ~1–20% obsidian pyroclasts, with different types of lithics making up the
108 difference. Their Bed 7 contains ~40% obsidian pyroclasts and subordinate amounts of
109 pumice and lithics. Although we did not sieve our tephra layers, those proportions are
110 consistent with our observations.

111 A total of 81 obsidian pyroclasts collected from four tephra layers were analyzed in
112 this study (Fig. 1). All methods are discussed in the Supplemental text; results are
113 presented in Supplemental Tables A.1 to A.7. Vesicle textures were analyzed in 80 of the
114 obsidian pyroclasts. Volatile contents of 54 were measured in this study; volatile contents
115 of the other 27 were reported in Barnes et al. (2014). Together, the measured volatile
116 contents span the entire range of values reported by Newman et al. (1988) and Rust et al.
117 (2004). Vesicularities of 221 pumice and 23 obsidian pyroclasts were also measured.

118

119 **3. Results**

120

121 *3.1. Vesicularities and Vesicle Textures of Pumice Clasts*

122

123 Pumice clasts are gray to white in color, and have vesicularities that range from 20 to
124 78 vol.% (Fig. 2). The distribution of vesicularity in each layer is unimodal, with means
125 between 56 and 68 vol.%. We scanned 14 pumice clasts by High-Resolution X-ray

126 Computed Tomography (HRXCT), and classify them as “highly” vesicular (71–74
127 vol.%), “moderately” vesicular (52–60 vol.%), and “slightly” vesicular (29–44 vol.%).

128 Highly vesicular pumice clasts have elongated (stretched) vesicles, which range
129 widely in size (Fig. 3a). There are planes of kinked (sheared) vesicles that span parts of
130 each clast. Vesicles in most of the moderately vesicular pumice clasts are generally
131 smaller and more uniform in size than those in the highly vesicular samples (Fig. 3b).
132 One moderately vesicular sample, however, has a micro-vesicular interior (Fig. 3c). The
133 boundary between its outer and inner portions is gradational, and so it is not a dense core
134 mantled by more vesicular material. Slightly vesicular pumice lapilli are more diverse
135 texturally. Some are micro-vesicular, like the dense core of the moderately vesicular
136 pumice clast (Fig. 3d). Others consist of highly stretched vesicles, like denser versions of
137 more vesicular clasts (Fig. 3e). Another of these (38 vol.%) is in fact an amalgamation of
138 small vesicular fragments sutured together by crystal-rich ash that contains rock
139 fragments (Fig. 3f). This was the only such pumice found.

140

141 *3.2. Volatile and Crystal Contents of Obsidian Pyroclasts*

142

143 Dissolved concentrations of H₂O and CO₂ in the obsidian pyroclasts range from 0.49
144 to 2.24 wt.% and from 2 to 45 ppm, respectively (Table A.1). These values overlap those
145 reported by Newman et al. (1988) and Rust et al. (2004). Watkins et al. (in press) mapped
146 H₂O and CO₂ across eight of the obsidian clasts and found that concentrations within
147 each varies, on average, by ± 0.3 wt.% H₂O and ± 10 ppm CO₂. These internal variations
148 are typically small compared to the spread between samples (Fig. 4; note that vesicle

149 shapes reported in that figure are discussed later). A wide range in volatile contents is
150 found in each tephra layer, but overall there is a general decrease in H₂O and CO₂
151 contents with increasing stratigraphic height, a trend also found by Newman et al. (1988)
152 and Rust et al. (2004).

153 All obsidian pyroclasts contain ≤ 1 vol.% microphenocrysts and microlites (as distinct
154 from xenocrysts described in Section 3.5). Microphenocrysts include quartz, plagioclase,
155 alkali feldspar, biotite, and hornblende. Microlites (up to a few 10's of microns in size)
156 consist of acicular pyroxene, tabular to skeletal feldspar, and tiny Fe–Ti oxides. Fe–Ti
157 oxide microlites can be disseminated throughout a clast or concentrated in bands. Some
158 clasts contain both concentrated bands and disseminated oxides. Most (60 out of 81)
159 obsidian pyroclasts contain pyroxene and/or feldspar microlites, and all but a few have
160 Fe–Ti oxides. Neither microlite abundance nor type correlates with volatile content in
161 our sample set; however, data in figure 14a of Rust and Cashman (2007) suggest that
162 obsidian pyroclasts in the basal layer of the tephra section – which was not sampled in
163 our study – tend to be richer in microlites, and have lower-than-typical H₂O contents.

164

165 *3.3. Vesicle Textures and Abundances in Obsidian Pyroclasts*

166

167 All but three obsidian pyroclasts contain vesicles (Table A.1). Vesicularities
168 measured in a subset (23 out of 81) of samples range from 0.02 to 8.4 vol.%, although the
169 majority have < 1 vol.% (Fig. 2). No obsidian pyroclast appears more vesicular, and those
170 measured cover the range of vesicle textures (described below); the range of 0.02–8.4
171 vol.% is thus representative of the population. Vesicles are distributed heterogeneously,

172 typically in bands or clusters, rather than being dispersed homogeneously. Vesicle
173 number density (N_V) spans many orders of magnitude (Table A.1). More than half of
174 those measured have $N_V > 10^{6.8}$ vesicles cm^{-3} , with the highest being $10^{8.1}$ cm^{-3} . The
175 lowest measured N_V is $10^{4.3}$ cm^{-3} , although three obsidians lack vesicles entirely.

176 A total of 6958 vesicles were measured in 57 obsidian pyroclasts optically and using
177 HRXCT (Tables A.2 to A.7). Vesicles vary widely in shape, but most can be classified in
178 5 categories, which we call spherical (Fig. 5a, b), rounded ellipsoid (Fig. 5c), sharp-
179 tipped ellipsoid (Fig. 5d), distorted (Fig. 6a–c), and stretched–distorted (Fig. 6d–f).
180 While we group data together based on shape and describe the five classes separately,
181 many pyroclasts contain more than one type of vesicle (Table A.1).

182 Spherical – Spherical vesicles are equant in shape (the three axis lengths are within
183 10% of each other) and have smooth, concave surfaces (Fig. 5a). They occur mainly in
184 H_2O –rich obsidian pyroclasts (Fig. 4). They are ~1 to 120 μm in size (Table A.3), with a
185 median of 10 μm , and tend to be larger at higher H_2O contents. Overall, their N_V ranges
186 between $10^{4.3}$ – $10^{8.0}$ cm^{-3} (Table A.1), and occur both in isolation and in clusters (Fig. 5b).

187 Rounded ellipsoid – Ellipsoidal vesicles have smooth, concave surfaces, like spherical
188 ones, but are elongated by >10% in one direction (Fig. 5c, d). They can be divided into
189 two sub-categories, based on the curvature at the end of the long axis. Rounded
190 ellipsoids have dome-shaped ends that are curved slightly more tightly than other parts of
191 the vesicle (Fig. 5c). Sharp-tipped ellipsoids (discussed in the next subsection) come to
192 relatively narrow points with much tighter curvature (Fig. 5d).

193 Rounded ellipsoid vesicles are found mainly in H_2O –rich obsidian pyroclasts (Fig. 4)
194 and occur in N_V from $10^{4.8}$ to $10^{7.6}$ cm^{-3} (Table A.1). Their long axes range from 4 to

195 1880 μm , with short axes from ~ 1 to $300 \mu\text{m}$ (Table A.4). Intermediate axes are generally
196 closer in length to short axes. Their sizes correlate with H_2O content, and larger vesicles
197 are usually found in smaller N_V . The degree of elongation can be described by a
198 deformation parameter (D), where

199

$$200 \quad D = \frac{l_a - l_b}{l_a + l_b} \quad (1)$$

201

202 with l_a and l_b being the lengths of the long and short axis (Taylor, 1934; Rust and Manga,
203 2002). For rounded ellipsoid vesicles, D ranges from ~ 0.050 (almost spherical) to 0.885
204 (highly elongated), with larger vesicles tending to be more elongated.

205 Sharp-tipped ellipsoid – Sharp-tipped ellipsoid vesicles (Fig. 5d) are rarer than
206 rounded ones, but occur over a wider range of H_2O contents (Fig. 4) and in N_V values
207 typically $< 10^5 \text{ cm}^{-3}$ (Table A.1). Long axes range from ~ 4 to $1070 \mu\text{m}$; short axes, from
208 ~ 1 to $270 \mu\text{m}$ (Table A.5). Their sizes correlate weakly with H_2O content, and their D
209 values range from ~ 0.050 to 0.863 .

210 Distorted – Distorted vesicles are very irregularly shaped, with jagged edges, sharp
211 contortions, convex and concave surfaces, and offshoots (Fig. 6a–c). They are
212 transparent, isotropic, and hollow when exposed, proving they are vesicles. They occur
213 in obsidian pyroclasts across the spectrum of volatile contents, but are more common in
214 those with $\leq 1.5 \text{ wt.}\% \text{ H}_2\text{O}$ (Fig. 4). They tend to be numerous, with N_V values of $10^{5.3-7.6}$
215 cm^{-3} (Table A.1), and most commonly occur in bands that cross the length of an obsidian
216 pyroclast (Fig. 6a) or in isolated clusters (Fig. 6c). The bands often link together in three
217 dimensions and surround regions of less vesicular glass. Characterizing their shapes is

218 difficult because of their irregularity. Roughly, their median size is $4\ \mu\text{m} \times 10\ \mu\text{m}$
219 (Table A.6), but can reach $\sim 120\ \mu\text{m}$ in length, and are usually bladed in shape. Larger
220 ones tend to be wider and have smoother surfaces.

221 Stretched-distorted – “Stretched-distorted” vesicles look similar to distorted ones,
222 except they are significantly elongated in one direction (Fig. 6d, e). Usually, they are
223 stretched in a common direction (Fig. 6d), and are deformed along planes in some
224 instances (Fig. 6f). Like distorted vesicles, these vesicles occur mainly in H₂O-poorer
225 pyroclasts (Fig. 4) and tend to be numerous, mostly $N_V > 10^7\ \text{cm}^{-3}$ (Table A.1). The
226 longest dimension ranges from $\sim 2\text{--}222\ \mu\text{m}$ (median = $11\ \mu\text{m}$), while their shortest
227 dimension is $< 1\text{--}32\ \mu\text{m}$ (median = $2\ \mu\text{m}$) (Table A.7). There is no correlation between
228 their size and volatile contents.

229

230 *3.4. Vesicle associations and assemblages*

231

232 Approximately 40% of obsidian pyroclasts contain more than one type of vesicle
233 (Table A.1). Spherical vesicles occur most commonly with rounded ellipsoid vesicles
234 (e.g., Fig 5c). Stretched–distorted vesicles occur most commonly with sharp–tipped
235 ellipsoid vesicles, but distorted vesicles occur most commonly with stretched–distorted
236 vesicles. On the other hand, rounded ellipsoid vesicles never occur with distorted or
237 stretched–distorted vesicles, and stretched–distorted vesicles never occur with spherical
238 vesicles.

239 Some obsidian pyroclasts consist of domains with different vesicle textures (Fig. 7).
240 Some domains differ in abundances of the same vesicle type, others by degrees of vesicle

241 elongation, and some differ substantially by the alignment of elongated vesicles. About
242 20% of the obsidian pyroclasts have strongly discordant domains. Volatile contents can
243 also differ between domains (e.g., Fig. 8a). It thus appears that a significant fraction of
244 obsidian pyroclasts are composed of smaller fragments that have welded together.

245

246 *3.5. Xenocryst Fragments and Abundances*

247

248 Many obsidian pyroclasts contain distinct bands with numerous pieces of crystals, on
249 order of ~1 to 10 μm in size (Fig. 8). They are identified as crystalline by their relief,
250 birefringence (Fig. 8c), and brightness in reflected light. Their jagged edges suggest that
251 many are broken fragments. The bands can be isolated or grouped together, and often
252 contain large crystals and rock fragments (Fig. 7b). The crystal-rich nature of these
253 bands is in sharp contrast to the nearly holohyaline host rhyolite (Fig. 8c). Bands in four
254 different obsidian pyroclasts were all found to contain quartz, alkali feldspar, plagioclase,
255 biotite, hornblende, and Fe-Ti oxides (typically magnetite); some also contain apatite.

256 The sharp contrast in crystal content, and the common coexistence with rock
257 fragments and anomalously large crystals, indicate that the bands are xenocrystic, and we
258 identify them as the “xenolithic powder” seen by Rust et al. (2004). The crystal
259 assemblage of these bands matches that in many Mono Craters rhyolitic domes and lavas
260 (Kelleher and Cameron, 1990), and so the bands are not small blebs of mafic magma or
261 excavated carbonate lake sediments. Slightly elevated (by ~10 ppm) CO_2 contents are
262 associated with some bands, but not all. This differs from the association of locally
263 highly elevated CO_2 contents with “xenocrystic dust” in obsidian pyroclasts from

264 Newberry volcano (Rust and Cashman, 2007). We do not consider these bands to be
265 tuffisite veins, which can occur in rhyolitic lavas and pyroclasts, because those tend to be
266 cracks filled with ashy material derived from the erupting magma (Stasiuk et al., 1996;
267 Gonnerman and Manga, 2003; Tuffen et al., 2003; Noguchi et al., 2008; Castro et al.,
268 2012; Kolzenburg et al., 2012).

269 Some xenocryst bands cut across entire obsidian pyroclasts; they are usually planar
270 and extend 100's of microns (and farther) in two directions, but are only a few to 10's of
271 microns in the third dimension. Their margins are usually sharp (Fig. 8a), but some are
272 diffuse and appear smeared out. Some bands thin and grade into discontinuous trails of
273 crystal fragments (Fig. 8d). Isolated trails of xenocrysts – sometimes only a few 10's of
274 microns long – are seen in many pyroclasts, including those without distinct bands.
275 Vesicles within xenocryst bands are most commonly distorted or stretched–distorted (Fig.
276 7b, 8b). The bands can be orientated up to $\sim 45^\circ$ to vesicles (Fig. 7b). In many cases, the
277 margins that separate vesicular domains within obsidian pyroclasts are xenocryst bands
278 (Fig. 7b). Together, xenocryst bands and discontinuous trails are found in $\sim 68\%$ of the
279 pyroclasts, across the entire spectrum of H₂O and CO₂ contents.

280

281 **4. Discussion**

282

283 Obsidian pyroclasts form by viscous deformation, and thus any model for their
284 formation requires consideration of their thermal history. Newman et al. (1988) suggest
285 that obsidian pyroclasts represent the “cool glassy margins” of the volcanic conduit.
286 Furthermore, Rust et al. (2004) propose that the chilled margins may cross the glass

287 transition multiple times during their formation. Consequently, obsidian pyroclasts may
288 experience temperatures from magmatic (T_m) to below the glass transition (T_g). To
289 determine an approximate value for T_m , we analyzed the compositions of 10 pairs of
290 magnetite–ilmenite phenocrysts in four of the Mono Craters rhyolite flows, and found
291 temperatures of 744 to 820°C (Ghiorso and Evans, 2008); for convenience we adopt
292 800°C as T_m . Viscous deformation ceases below T_g , which is often taken as the
293 temperature at which melt viscosity (η) = 10^{12} Pa s [a value relevant for rhyolite cooling
294 at a rate of order 1 K min⁻¹ (Gottsmann et al., 2002)]; we use this definition to set the
295 lower temperature bound of interest. For H₂O contents of 0.49 to 2.2 wt.%, T_g ranges
296 from ~600 to ~465°C (Hess and Dingwell, 1996). The thermal range of interest,
297 therefore, is ~200–335°C, depending on H₂O content.

298 Watkins et al. (in press) report heterogeneities in H₂O and CO₂ contents within
299 individual obsidian pyroclasts (e.g., Fig. 8a), and used diffusion modeling to show that
300 the observed heterogeneities would be erased in a matter of hours at T_m , regardless of
301 H₂O content. Diffusion slows at lower temperature (Ni and Zhang, 2008) and at T_g , the
302 maximum time allowed by the observed heterogeneities increases by a factor of 10–100
303 compared with T_m . Watkins et al. (in press) concluded that obsidian pyroclasts are
304 composed of subdomains of melt or sintered ash that were juxtaposed above T_g within a
305 few hours prior to quenching.

306

307 *4.1. Interpretation of Vesicle Shapes in Obsidian Pyroclasts*

308

309 We argue that all vesicle textures can be produced by starting with populations of
310 distorted bubbles, followed by variable degrees and timing of shear and relaxation.
311 Relaxation occurs when $T > T_g$ because bubbles tend to become spherical over time under
312 the action of surface tension (σ) (N m^{-1}). The characteristic timescale over which a non-
313 spherical bubble relaxes towards spherical is the relaxation time (λ_b), which is given by

$$315 \quad \lambda_b = \frac{\eta r}{\sigma} \quad (2)$$

316
317 where r (m) is the equivalent spherical radius of the bubble (Taylor, 1934; Rust and
318 Manga, 2002; Llewellyn and Manga, 2005). If a spherical bubble is deformed by simple
319 shear, pure shear, or a combination of the two, it becomes ellipsoidal and, at steady strain
320 rate ($\dot{\gamma}$) (s^{-1}), the shape, described by D (Equation 1) is a simple function of the capillary
321 number ($Ca = \lambda_b \dot{\gamma}$) (Taylor, 1934; Rust and Manga, 2002). Above a critical Ca (i.e. for
322 high strain rate) ellipsoidal bubbles develop sharp tips. The variation among spherical,
323 rounded ellipsoid, and sharp-tipped ellipsoid vesicles can therefore be explained by
324 varying degrees of shearing, and subsequent relaxation, of initially spherical bubbles.

325 Highly complex shapes like those of the distorted vesicles cannot be produced by
326 variations in shear and relaxation of spherical bubbles (e.g., Rust and Manga, 2002).
327 Moreover, distorted vesicles are generally not aligned (Fig. 6a), which would be expected
328 if they formed by shearing. On the other hand, stretched-distorted vesicles are typically
329 aligned; we thus infer that they formed by shearing of distorted bubbles. While it is not
330 possible to form distorted vesicles by shear deformation of spherical bubbles, distorted
331 bubbles will tend to relax towards spherical over time (Rust and Manga, 2002). Applying

332 shear at different times during the relaxation of distorted bubbles could lead to either
333 stretched–distorted vesicles (if applied relatively early) or ellipsoid vesicles (if applied
334 relatively late).

335

336 *4.2. Evidence for a Fragmental Origin for Obsidian Pyroclasts*

337

338 The occurrence of xenocryst bands and rock fragments in a large majority of obsidian
339 pyroclasts is strong evidence that the obsidians did not form by pumice (foam) collapse
340 (Eichelberger et al., 1986; Westrich and Eichelberger, 1994). This interpretation is
341 supported by the observation that many obsidian pyroclasts are composed of domains that
342 experienced different vesiculation and deformation histories (Figs. 6, 7). In addition, the
343 wide gap in vesicularity between obsidian and pumice (Fig. 2) is hard to reconcile with a
344 foam-collapse origin, which would necessitate that the magma passed through a stage of
345 intermediate vesicularity that is not observed. Finally, recent experiments demonstrate
346 that foam collapse takes much longer than the time available for the formation of the
347 North Mono obsidian pyroclasts. Martel and Iacono-Marziano (2015) decompressed
348 hydrous rhyolite at magmatic temperatures and pressures, causing their samples to
349 vesiculate, and then eventually collapse in the manner proposed by Eichelberger et al.
350 (1986). Their findings imply that at least 10^5 hours are required for foam collapse at the
351 lowest melt viscosity that we expect during formation of the obsidian pyroclasts [$\eta \approx 10^6$
352 Pa s for 2.2 wt.% H₂O at T_m , using the method of Hess and Dingwell (1996)]. More
353 viscous samples would take even longer. This is clearly inconsistent with both the time
354 above T_g inferred from the heterogeneities of H₂O within the pyroclasts (Watkins et al., in

355 press), and the wide gap in vesicularities between pumice and obsidian (Fig. 2). We
356 conclude that some other process besides foam collapse led to the formation of the Mono
357 Craters obsidian pyroclasts.

358 Some obsidian pyroclasts are composed of separate vesicular domains, and the
359 contacts between them consist of xenocrysts and rock fragments (e.g., Figs. 7–8). This
360 shows that some bands of ash-sized (and larger) xenocrysts became incorporated where
361 pieces of obsidian sutured together. Some sutures remained as coherent bands through an
362 obsidian pyroclast (Fig. 8a); other xenocryst bands are short, isolated trails only 10's to
363 100's of microns long. These disconnected bands most likely represent instances where
364 obsidian fragments welded together more thoroughly. In some cases, there is evidence
365 that these welded pyroclasts underwent further deformation and broke apart again, to
366 leave small, isolated fragments of xenocrysts, which then welded together with other
367 fragments. Each vesicular domain seen in Figure 7, for example, contains xenocryst
368 bands orientated obliquely to the outer, xenocryst-lined sutures. This pyroclast thus
369 records multiple cycles of fragmentation and suturing.

370 Distorted vesicles show textures that are also strongly suggestive of a fragmental
371 origin. They are often organized into zones that separate regions of more-or-less vesicle-
372 poor glass (Fig. 6a, b). These textures are consistent with partial sintering of fragments
373 of obsidian and ash-sized glass of various sizes. The jagged and contorted shapes of the
374 vesicles then result from the trapping of gas between the fragments. Indeed, the distorted
375 vesicles seen in Figure 6c have multi-cusped forms that are what one would expect if gas
376 were trapped in the junctions between sintering glass fragments. The preservation of

377 these distorted shapes requires that no significant relaxation of the trapped gas bubbles
378 occurred; relaxation is discussed in detail later in Section 4.4.

379 This interpretation of the distorted vesicles is supported by their relationship to
380 volatile heterogeneities and xenocryst bands. Watkins et al. (in press) found that the
381 poorly vesicular regions in P2B–N (Fig. 6d) have different volatile contents than the
382 bands of distorted vesicles that separate them. In addition, the glassy regions on either
383 side of bands of distorted vesicles in P10–I (Fig. 6a) have different volatile contents.
384 These domains are also separated by xenocryst bands and rock fragments. In general,
385 vesicles that occur within xenocryst bands are most often distorted or stretched–distorted
386 (Figs. 7, 8d), suggesting that xenocryst bands and distorted vesicles are linked, and both
387 mark where ash and obsidian fragments sintered together.

388

389 *4.3. Conceptual Model for the Formation of Obsidian Pyroclasts*

390

391 Rust et al. (2004) proposed that some pyroclastic obsidians form through sintering of
392 ash fragments. Here, we extend that model and argue that it is a general process that
393 forms most, if not all, obsidian pyroclasts generated during the North Mono eruption.
394 Xenocryst bands, which record cycles of fragmentation and sintering, are found in
395 obsidian pyroclasts across the spectrum of measured volatile contents and in every tephra
396 layer sampled, indicating that this process occurs over a wide range of depths in the
397 conduit (Fig. 4), and at all stages of development of vesicular textures.

398 We argue that distorted vesicles are not relicts of collapsed magmatic bubbles, but are
399 remnants of sintering, marking where ash fragments sintered together (Fig. 9). In this

400 model, the spectrum of vesicle textures results from varying degrees of shearing and
401 relaxation that act on starting populations of distorted vesicles formed by sintering. Both
402 sintering and relaxation processes require time (see Section 4.4); consequently, the most
403 likely location for these processes is at the conduit margin, where erupting material may
404 be accreted and held for variable amounts of time, as proposed by Newman et al. (1988).
405 After a packet of ash has begun to weld to form obsidian-like magma, it can be re-
406 incorporated into the erupting mass and ejected to form obsidian pyroclasts with mainly
407 distorted vesicles, or stretched-distorted vesicles if that packet is sheared shortly before
408 (or while) it is re-incorporated (Fig. 9). Stalling for longer times at the margins of the
409 conduit allows vesicles to relax, and thus re-incorporated pieces of this magma will
410 produce obsidian pyroclasts with spherical to ellipsoid vesicles, depending on the degree
411 of shear that acts on the packet (Fig. 9). The sequence of sintering, shearing, and
412 relaxation can be repeated, resulting in pyroclasts with multiple domains of varying
413 vesicular textures (Fig. 7, 8d).

414 Previous models suggest that obsidian forms in localized regions of fragmentation
415 within a conduit that is otherwise filled with continuous magma. Rust et al. (2004)
416 conclude that North Mono obsidian pyroclasts formed at conduit margins, where magma
417 was brecciated by shear. Castro et al. (2014) conclude that obsidian lavas are produced
418 through fragmentation and outgassing of vesicular magma along tuffisite veins. Although
419 our data do not preclude these mechanisms, two lines of evidence support a model in
420 which obsidian pyroclasts are formed in a conduit filled with fragmental material. First,
421 the wide range in volatile contents indicates that obsidian formed over a great range of
422 pressures (Table A.1) and hence depths (likely a few kilometers, depending on the

423 density of the conduit-filling material), and was transported to the surface rapidly enough
424 to preserve relatively steep gradients in volatile contents (Watkins et al., 2012; Watkins et
425 al., in press). It is easier to imagine this happening in a conduit filled with an erupting
426 gas-particle mixture than in one filled with continuous magma. Second, obsidian
427 pyroclasts contain subdomains with different water contents. Although juxtaposition of
428 clasts with different water contents can occur in tuffisite veins (Saubin et al., 2016), again
429 it is easier to imagine this happening in a conduit filled with fast-moving fragmental
430 material.

431 Our conceptualization requires that the environment at the conduit wall varies from
432 depositional to erosional over time and space, suggesting that flow in the conduit is
433 unsteady. It is possible that the sub-Plinian character of the eruption (Bursik, 1993)
434 favors episodic cycling between deposition and erosion. Obsidian pyroclasts with higher
435 volatile contents become less common up-section in the stratigraphy (Table A.1;
436 Newman et al., 1988; Rust and Cashman, 2007; Barnes et al., 2014). It is possible that as
437 the eruption waned (Sieh and Bursik, 1986), the main location for deposition and erosion
438 of sintered material became shallower in the conduit.

439 Finally, we note that very similar vesicle textures are seen in obsidian pyroclasts
440 erupted at Newberry Volcano (Rust and Cashman, 2007). Those pyroclasts also contain
441 abundant veins of xenocrystic material. These features are compatible with our model,
442 and we conclude that the Newberry Volcano samples could have formed by sintering of
443 ash, suggesting that our model provides a general mechanism for generating obsidian
444 pyroclasts.

445

446 4.4. Timescales and Conditions of Formation of Obsidian Pyroclasts

447

448 Our model postulates that the spectrum of vesicular textures results from varying
449 degrees of shearing and relaxation of bubbles that were, originally, distorted gas pockets
450 trapped between sintered fragments (Fig. 9). Both sintering and relaxation require that T
451 $> T_g$, a condition that can exist for only a few hours at T_m across the range of H₂O
452 contents (Watkins et al., in press). A first test of the model is to determine whether this
453 allows sufficient time for bubble relaxation.

454 The characteristic relaxation time for a deformed bubble λ_b is given by equation 2.
455 Surface tension is only weakly controlled by melt composition and T , and so for the
456 limited range in H₂O contents of interest (Fig. 4) it is reasonable to assume that σ is
457 constant, with a value of $\sim 0.2 \text{ N m}^{-1}$ (Walker and Mullins, 1981; Bagdassarov et al.,
458 2000; Gardner and Ketcham, 2011; Gardner et al., 2013). Relaxation time therefore
459 depends mainly on viscosity and bubble radius (Fig. 10). We note that the early stages of
460 relaxation of a distorted bubble will proceed more rapidly because of the tighter radius of
461 curvature, but for a bubble to become spherical, it must have an amount of time available
462 for relaxation that is at least as long as the time calculated using equation 2 with the
463 spherical bubble radius. Given the range in measured H₂O contents (Fig. 4), and allowing
464 T to range from T_m to T_g , viscosity varies from $\sim 10^6$ to 10^{12} Pa s. Bubble radius varies
465 over an approximate range of 1 to 100's of microns. Also shown in Figure 10 is the range
466 in time allowed – bracketed at ca. 1–10 hours at T_m (Watkins et al., in press). This time is
467 determined from modeling of diffusion of H₂O and CO₂, and diffusivity is a weaker
468 function of temperature than is viscosity (Ni and Zhang, 2008); hence, the time allowed

469 increases at $T < T_m$ as indicated by the shift to longer times at lower T . This analysis
470 demonstrates that, at high H₂O content, there is ample time for bubbles to relax at T_m , and
471 even near T_g there is still time for small bubbles to relax (Fig. 10a). At low H₂O contents,
472 however, relaxation can only occur near T_m and only for relatively small bubbles (Fig.
473 10b).

474 A first-order conclusion, based on this analysis, is that distorted vesicles are more
475 likely to be preserved in samples with lower H₂O content because the melt is more
476 viscous and hence, bubbles relax more slowly. A logical corollary is that spherical
477 vesicles should be more common in melts with higher H₂O content. We tested this
478 hypothesis quantitatively using analysis of variance (ANOVA) of the H₂O and CO₂
479 contents of the pyroclasts, by dividing obsidian pyroclasts into three groups: distorted,
480 spheroidal, and stretched. The “distorted” group includes samples with predominantly
481 distorted or stretched-distorted vesicles; the “spheroidal” group includes samples with
482 predominantly spherical, rounded ellipsoid, or sharp-tipped ellipsoid vesicles; and the
483 “stretched” group includes samples with predominantly rounded ellipsoid, sharp-tipped
484 ellipsoid, or stretched-distorted vesicles (thus it repeats some samples from each of the
485 spheroidal and distorted categories). The rationale for these groupings is that distorted
486 vesicles (whether stretched or not) indicate samples in which little bubble relaxation has
487 occurred. Spheroidal vesicles, by contrast, represent samples in which bubbles have
488 relaxed considerably. Samples with stretched vesicles should not differ from their non-
489 stretched counterparts, except that they were sheared during, or shortly before, quench.

490 For H₂O concentrations, ANOVA indicates that the null hypothesis that samples in
491 the spheroidal and distorted categories are drawn from the same distribution can be

492 rejected at the 0.01% level (P-value = 0.000026). Hence, it is highly likely that samples
493 in those two categories came from populations with different H₂O concentrations. By
494 contrast, the null hypothesis that samples in the spheroidal and stretched categories are
495 drawn from the same distribution cannot be rejected at the 1% level (P-value = 0.057).
496 The same is true for the distorted and stretched categories (P-value = 0.040). The
497 findings are the same for CO₂ concentrations, (P-values of 0.00011, 0.18, and 0.030,
498 respectively, for the three null hypotheses). The volatile and vesicle shape data therefore
499 support the hypothesis that spheroidal vesicles are more common in H₂O-rich (low
500 viscosity) melts, whereas distorted vesicles are more common in H₂O-poor (high
501 viscosity) melts (Fig. 4). The data in figure 14b of Rust and Cashman (2007) also
502 support this broad correlation in North Mono obsidian pyroclasts.

503 More generally, our model postulates that distorted vesicles are gas pockets formed
504 by sintering ash particles. Wadsworth et al. (2016) showed that the characteristic
505 sintering timescale (λ_s) for randomly packed, monodisperse spherical particles of silicate
506 melt is also given by equation 2, except with particle radius (L) replacing bubble radius r
507 (i.e., $\lambda_s = \frac{\eta L}{\sigma}$). We do not have a direct measure of L , but can approximate it from the
508 number density of distorted vesicles, such that $L \approx \sqrt[3]{N_V^{-1}}$. N_V ranges from $10^{5.3}$ to $10^{8.1}$
509 cm^{-3} , thus $L \approx 20\text{--}170 \mu\text{m}$. Using the same values for surface tension and viscosity as
510 above, and reading the vertical axis in Fig. 10 as particle radius, we find that there is
511 ample time at high H₂O contents to sinter ash together across almost the entire range of
512 temperature (Fig. 10a). At low H₂O contents, however, sintering is only likely for small
513 ash sizes (Fig. 10b). We note that the sintering timescale presented is likely a maximum,
514 because it assumes no shear or compression, both of which accelerate sintering (Rahaman

515 et al, 1986; Rahaman and De Jonghe, 1990). Regardless, our analysis shows that, to first
516 order, ash sintering can happen over the range of conditions of interest (time, magma
517 temperature, and water content) of the eruption (Fig. 10).

518 Wadsworth et al. (2016) found that sintering monodisperse particles results in a final
519 vesicularity of ~3 vol.%, whilst obsidian pyroclasts from the North Mono tephra typically
520 have vesicularities of <1 vol.% (Table 1). It is likely that the initial population of ash
521 particles is polydisperse, which allows sintering to proceed to lower vesicularities
522 (Wadsworth et al., 2016). In addition, resorption of volatiles from vesicles into the melt,
523 as a result of either cooling (McIntosh et al., 2014) or increased pressure (Watkins et al.,
524 2012), would also lower vesicularity.

525 Our calculations indicate that sintering is favored for ash-sized particles (Fig. 10).
526 This may explain why sintering of large pumice clasts is uncommon (Fig. 3f), even in an
527 eruption that is generating abundant obsidian pyroclasts. Sintering may even require ash
528 to act as the ‘glue’ that facilitates assembling larger clasts that are composed of smaller
529 domains (e.g., Figs. 7, 8c, 8d). Xenolithic material often lines sutures between sub-
530 domains, supporting this hypothesis. The lack of sutured pumice clasts may reflect a
531 lower likelihood of trapping larger clasts at the conduit wall, because they presumably
532 move with greater momentum and they typically make up a smaller proportion of the
533 erupting mixture (e.g., Carey and Sigurdsson, 1982; Alfano et al., 2016).

534

535 *4.5. Implications for Magmatic Degassing*

536

537 The formation of obsidian pyroclasts by sintering of ash implies that the vesicles
538 within them, and associated dissolved volatiles, reflect not only the degassing trends of
539 bubbly magma but also the composition of gas trapped between particles. That volatile
540 record can then be modified by internal diffusive re-equilibration (Watkins et al., in
541 press) and distorted by break-up and re-sintering of disparate fragments (e.g., Fig. 8a).
542 Variations in H₂O and CO₂ contents thus do not reflect either exclusively closed- or
543 exclusively open-system trends (Newman et al., 1988).

544 Dissolved H₂O and CO₂ contents in obsidian pyroclasts are controlled by their
545 solubilities, which are, in turn, controlled largely by pressure (Blank et al., 1993; Holtz et
546 al., 1995; Liu et al., 2005). Accordingly, local volatile concentrations reflect pressures of
547 formation, and suggest that obsidian pyroclasts formed at pressures ranging from ~3 to 43
548 MPa (Fig. 4). Additionally, in our model, dissolved H₂O and CO₂ contents reflect the
549 composition of trapped gas. Volatile contents in each obsidian pyroclast indicate that the
550 trapped gas composition, calculated using the model of Liu et al. (2005), ranges from 65
551 to 96 mol.% H₂O (Fig. 4). Gas composition does not vary systematically with inferred
552 formation pressure; instead, its composition is roughly constant at 87±6 mol.% H₂O,
553 despite the wide range in pressure, and regardless of vesicle type. Obsidian pyroclasts
554 with mainly spherical bubbles record a gas of 87±5 mol.% H₂O, and those with distorted
555 vesicles record a gas of 85±7 mol.% H₂O.

556 Our model generally supports the idea that the gas trapped in obsidian pyroclasts –
557 and therefore the gas in the conduit – was buffered at a roughly constant composition
558 (Rust et al., 2004). If it were not buffered, then it should have become progressively
559 richer in H₂O from continued degassing of the ash (i.e., open-system degassing). In that

560 case, gas trapped at shallow levels (low pressures) would be richer in H₂O, and the
561 outcome would be that those obsidian pyroclasts with mainly distorted vesicles would
562 contain a higher ratio of H₂O to CO₂ than those with mainly spherical vesicles, which is
563 not what is seen (Fig. 9).

564

565 **5. Conclusions**

566

567 Obsidian pyroclasts from the ca. 1325–1350 C.E. eruption of North Mono tephra
568 provide a window into processes that occurred in the conduit during the eruption.
569 Detailed analysis of the vesicle textures show that a range of vesicle shapes exist, from
570 highly distorted and multi-cusped to spherical; vesicles also show variable degrees of
571 shearing. Based on the inferred thermal-temporal environment in which the clasts
572 formed, we show that vesicles of all types can be generated by relaxation and shearing of
573 a starting population of distorted vesicles. We conclude that highly distorted vesicles
574 represent the remnants of the interstitial spaces between sintered fragments of ash, and
575 that all obsidian pyroclasts analyzed formed by sintering and relaxation of ash fragments.
576 We also find that obsidian pyroclasts formed through repeated fragmentation and welding
577 over a wide range of pressures (~3–43 MPa) – and hence a wide range of depths – within
578 the conduit. We conclude that the conduit environment provides multiple opportunities
579 for fragmented magma to stall, weld, and re-fragment, on its way to the surface. By
580 contrast, we are able to eliminate the possibility that the obsidian pyroclasts formed
581 through collapse of a permeable magma foam. The occurrence of similar vesicle textures
582 and xenolithic material in obsidian pyroclasts in other eruptions (Rust and Cashman,

583 2007) suggests that ash sintering along conduit walls may be a general mechanism for
584 generating obsidian pyroclasts.

585

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587

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598

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600

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731

732 **FIGURE CAPTIONS**

733

734 **Figure 1:** Schematic stratigraphic sections for pits A and B, both of which were ~1.5–
735 1.75 m in thickness. Pumice fragments and obsidian pyroclasts were collected from
736 layers labeled “P” in the two pits. Pit A is the same as that in Barnes et al. (2012).
737 Inset map shows the location of the two pits, relative to the positions of the obsidian
738 domes and the cumulative tephra isopachs (in mm) for the ca. 1325–1350 C.E. Mono
739 Craters eruption (modified from Newman et al., 1988).

740

741 **Figure 2:** Vesicularity distributions of pumice lapilli (open bars) and obsidian pyroclasts
742 (solid bars). Tephra layer names mark the median pumice vesicularities. Although
743 only 23 obsidian pyroclasts were measured, these are representative of the range of
744 textures seen. P2B–24 is the one pumice lapilli that consists of smaller vesicular
745 fragments sutured together with crystal-rich ash (See Fig. 3f).

746

747 **Figure 3:** Slices from HRXCT scans of Mono Craters Pumice (white scale bar in each
748 image is 1mm long). **a)** P10B–30, vesicularity = 74 vol.%. Note the relatively large
749 range in vesicle sizes; **b)** P10B–41, vesicularity = 55 vol.%. Vesicle sizes are
750 generally smaller than in more vesicular samples (compare to **a**); **c)** P6B–54,
751 vesicularity = 52 vol.%. Note that this pumice has a denser microvesicular core
752 (marked by arrow) that resembles less vesicular samples (see **d**); **d)** P2B–17,
753 vesicularity = 44 vol.%. Vesicle sizes are significantly smaller than those in more
754 vesicular samples; **e)** P6B–33, vesicularity = 41 vol.%. This pumice consists of highly

755 stretched vesicles, that appear to occur in bands that differ in the degree of stretching;
756 **f**) P2B-24, vesicularity = 38 vol.%. this pumice consists of many smaller vesicular
757 fragments sutured together with crystal-rich ash; some vesicular fragments are
758 outlined to aid the reader. The ash also contains rock fragments (two of which are
759 indicated by arrows).

760

761 **Figure 4:** H₂O–CO₂ concentrations in obsidian pyroclasts, with symbols indicating
762 predominant vesicle type in the obsidian (see legend). The average error associated
763 with the analyses is shown. Equilibrium fluid pressures (gray lines, in MPa) and fluid
764 compositions (dotted lines, in mole fraction of H₂O) are shown, calculated from the
765 model of Liu et al. (2005) and assuming T = 800°C.

766

767 **Figure 5:** Representative photomicrographs of spherical and ellipsoidal vesicle textures:
768 **a**) spherical vesicles in P2-T (scale bar = 100 μm); **b**) clustered spherical vesicles in
769 P4B-C (scale bar = 100 μm). Note that a few larger vesicles appear to have
770 deformed the smaller surrounding vesicles; **c**) rounded ellipsoid vesicles in P2-N
771 (scale bar = 300 μm); **d**) sharp-tipped ellipsoid vesicles in P2-S (scale bar = 500 μm).

772

773 **Figure 6:** Representative photomicrographs of distorted vesicle textures: **a**) and **b**) P10-I
774 (note **b** is located within **a**; scale bar = 250 μm); **c**) P2B-D (scale bar = 100 μm);
775 Representative photomicrographs of stretched-distorted vesicle textures: **d**) P2B-N
776 (scale bar = 200 μm); **e**) P2B-B (scale bar = 200 μm); **f**) P6B-B (scale bar = 300 μm).

777

778 **Figure 7:** Representative photomicrograph of an obsidian pyroclast (P10-B) consisting of
779 **a)** sub-domains of vesicle textures (outlined by curved dashed lines; scale bar = 1000
780 μm), many of which are divided by thin bands of xenocrysts and rock fragment, as
781 shown in **b)**. Note that vesicles in the domains shown in **b)** are orientated nearly 90°
782 to each other and show differing degrees of stretching.

783

784 **Figure 8:** Representative photomicrograph of xenocryst bands in obsidian pyroclasts: **a)**
785 Domains that differ in orientation of highly stretched-distorted vesicles in P4B-D,
786 separated by a relatively sharp margin that contains numerous small xenocryst
787 fragments (scale bar = 500 μm). Volatile contents of each region are shown. Note
788 that vesicles are orientated differently and show different degrees of stretching in the
789 two domains; **b)** plane polarized and **c)** cross polarized light images of a relatively
790 coherent band of xenocrysts and glass with sharp margins in P4B-T (scale bar = 500
791 μm). Note small stretched-distorted vesicles in the pyroclast are orientated $\sim 12^\circ$ from
792 parallel to the band. Small stringers of the band can be seen at its margin, which
793 appear to be smeared off; those stringers are parallel to the orientation of the vesicles.
794 Compare the high xenocryst abundance to the nearly holohyaline matrix; **d)** thin band
795 of small xenocryst fragments in P4B-T (scale bar = 100 μm). This is a continuation
796 of the same band in **a)**, but about 2 mm away.

797

798 **Figure 9:** Schematic model for the formation of obsidian pyroclasts. Ash collects and
799 sinters along the margins of the volcanic conduit, forming distorted vesicles. Given
800 time, vesicles relax to spherical under the stress of surface tension. Time allowed for

801 relaxation increases with increasing viscosity, which increases as temperature
802 decreases from magmatic to the glass transition and as dissolved H₂O content
803 decreases. Pyroclasts that contain spherical vesicles tend to be low viscosity (H₂O–
804 rich), which is a statistically separate population from high viscosity (H₂O–poor)
805 pyroclasts that contain distorted vesicles (ANOVA of H₂O contents based on *n*
806 number of samples). Vesicles are stretched if shear is exerted on the vesicular
807 magma. If exerted soon after sintering, then stretched–distorted vesicles result; if
808 exerted late, then smooth, elongated vesicles result. H₂O contents of pyroclasts with
809 sheared vesicles overlap with the other populations as expected from being those
810 pyroclasts that experienced shear.

811

812 **Figure 10:** Time required for a bubble of radius *r* to relax or ash particle of size *L* to
813 sinter as a function of temperature, for **a)** 2.2 wt.% H₂O and **b)** 0.49 wt.% H₂O; these
814 values are the highest and lowest water contents measured in our samples. The gray
815 band delimits the time available for these processes during the North Mono eruption,
816 based on H₂O diffusion modeling of Watkins et al (in press). Dashed curves are
817 where sintering and relaxation take longer than the time allowed by diffusion
818 modeling. Surface tension (σ) equals 0.2 N m⁻¹ for all times.

819

820

Figure
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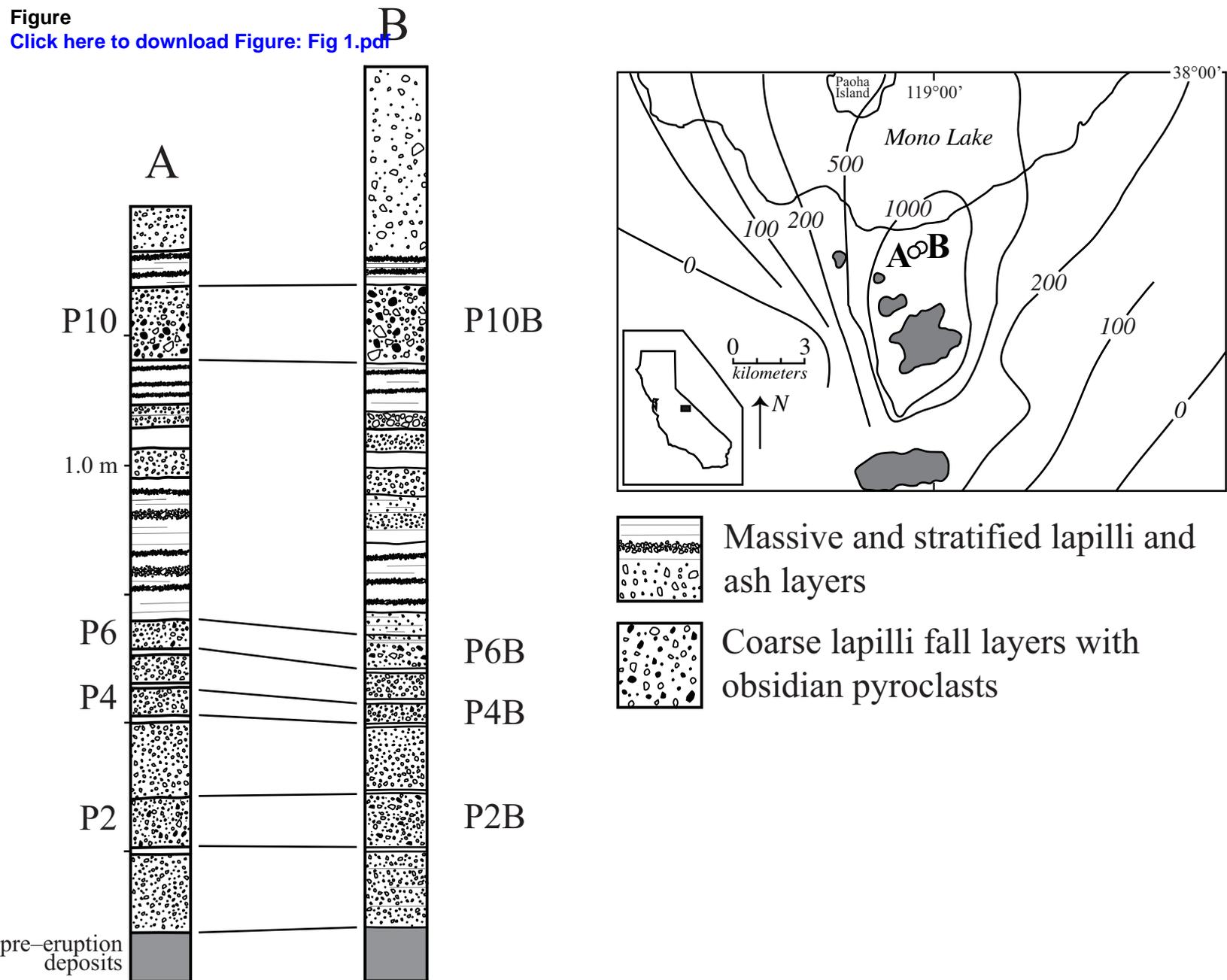
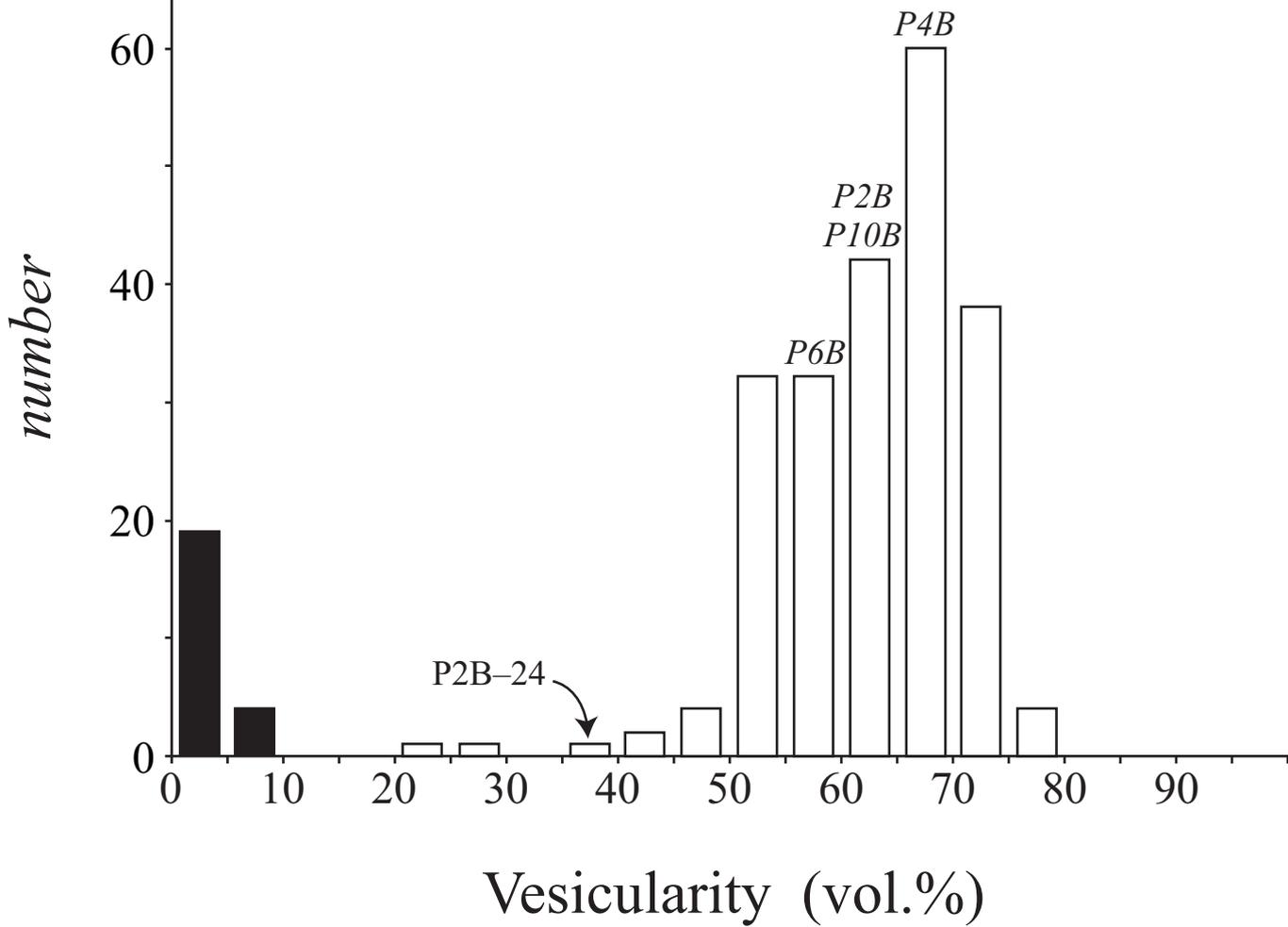


Figure 1



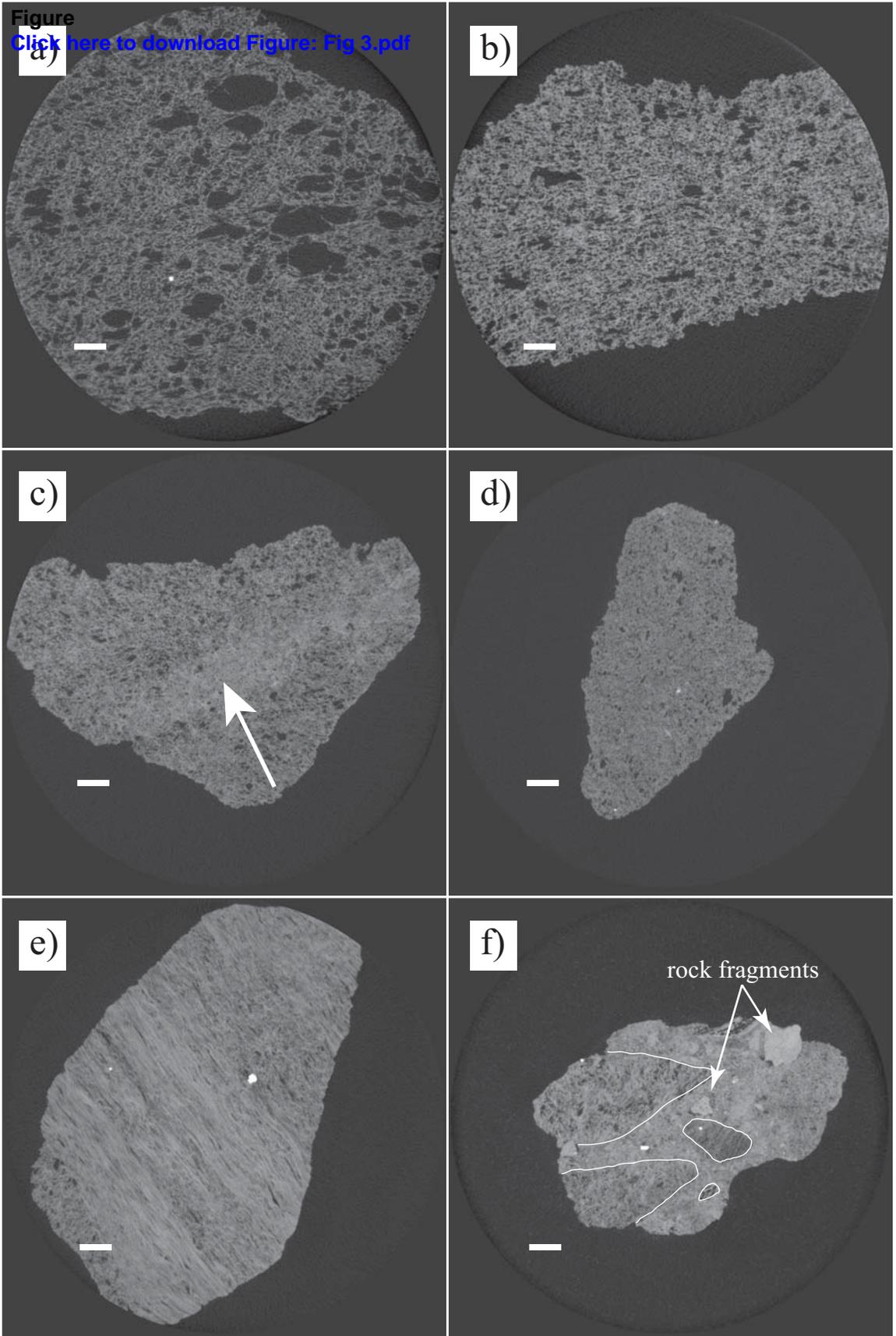


Figure 3

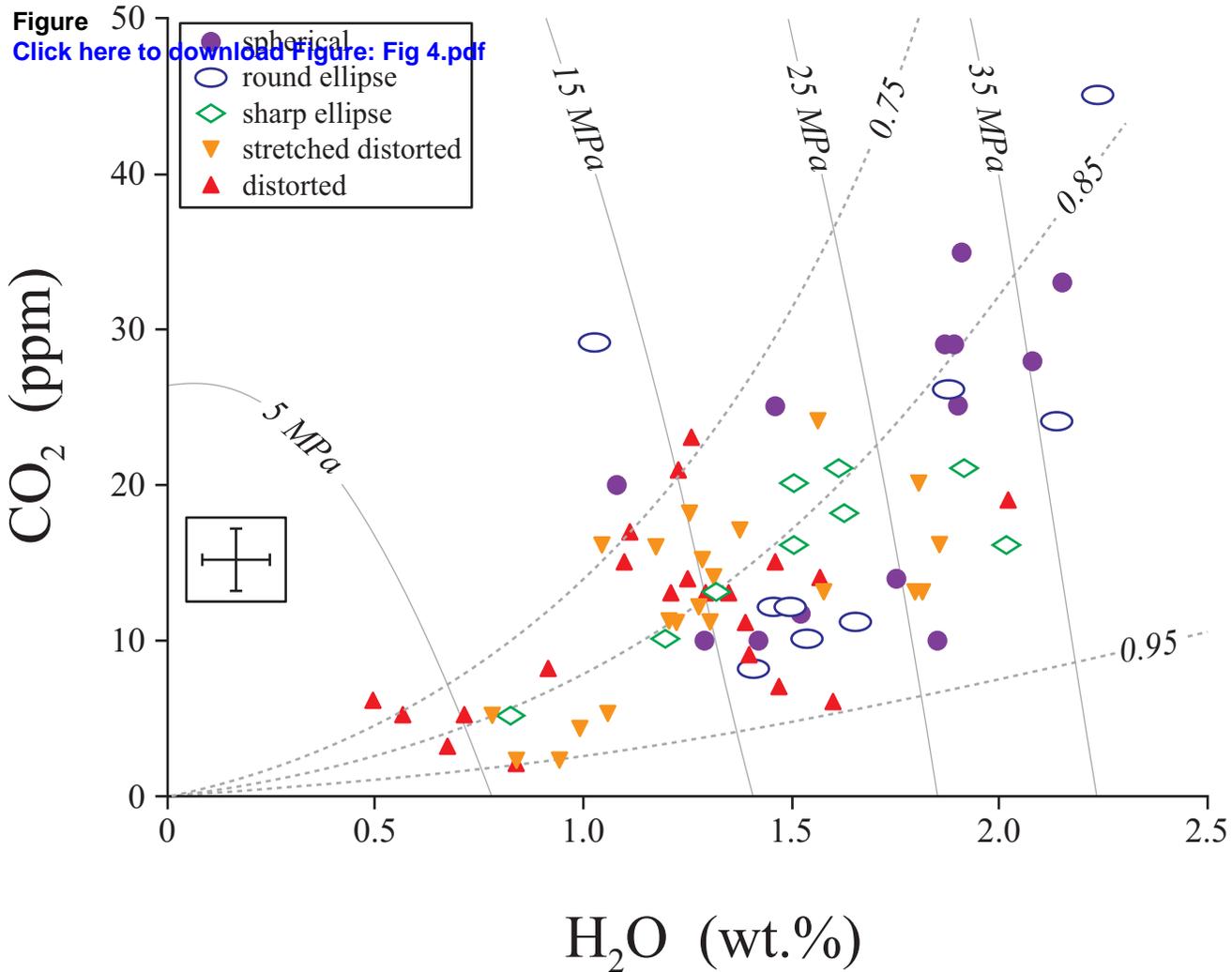
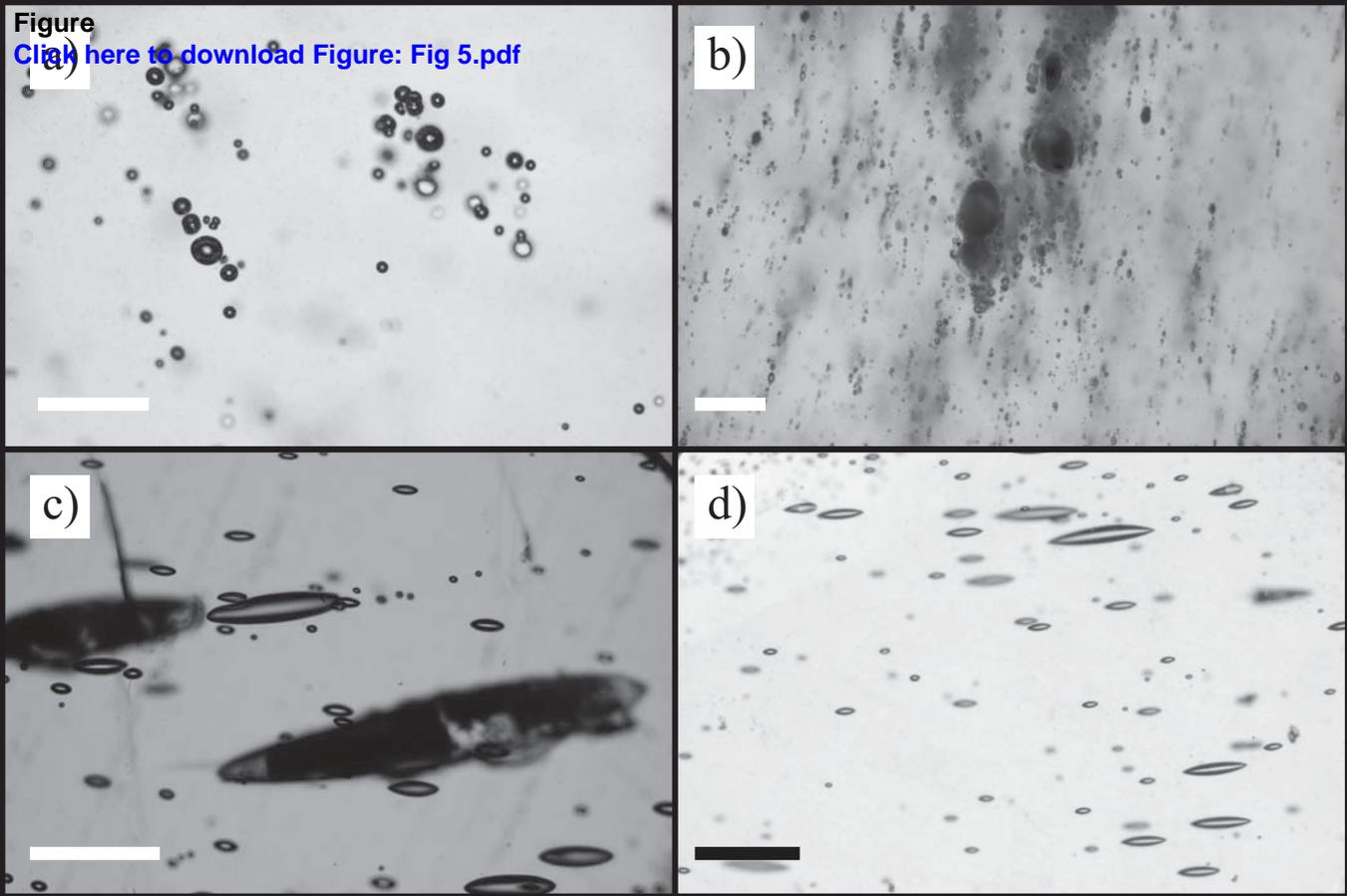


Figure 4



Figure

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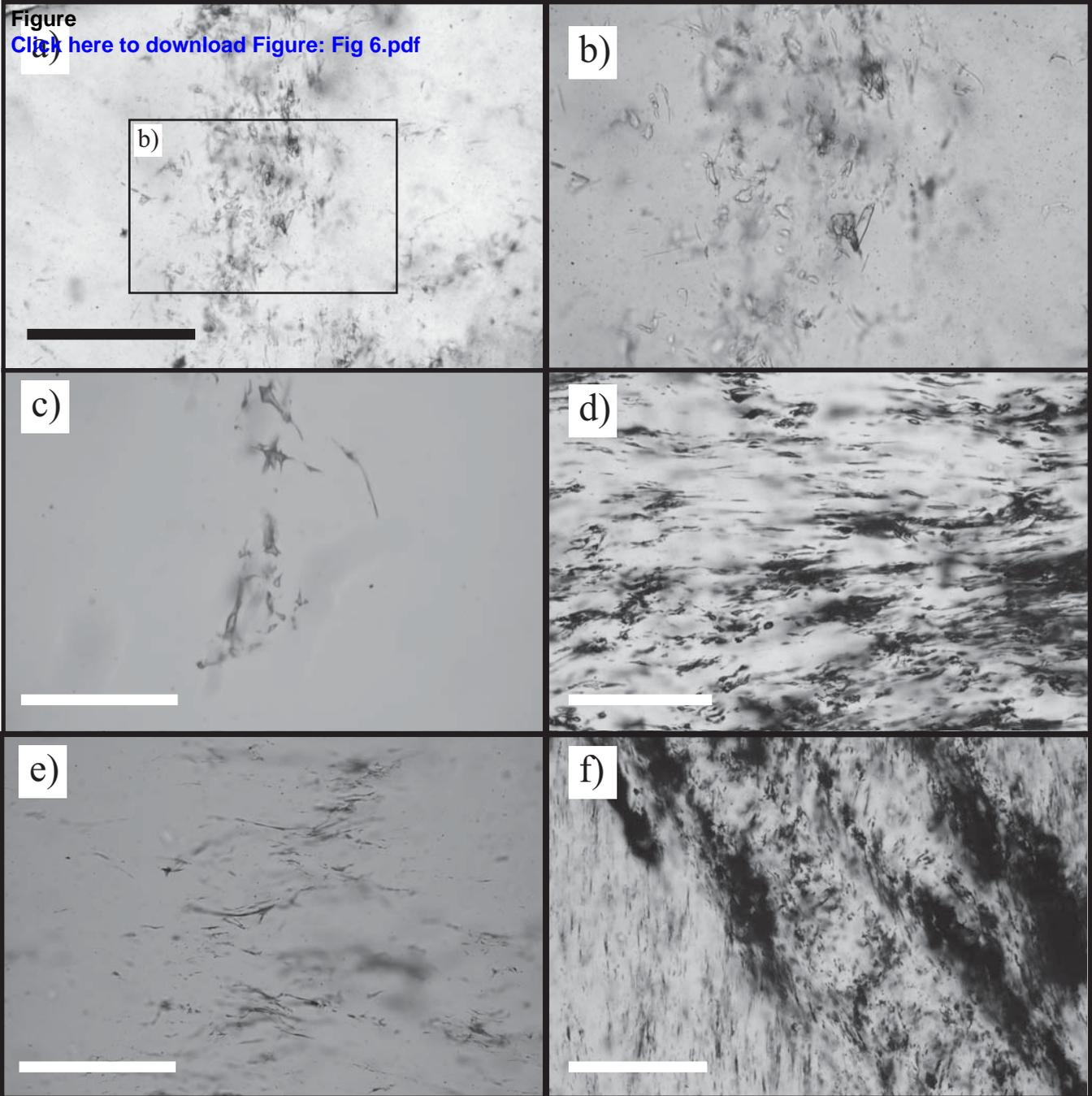


Figure 6

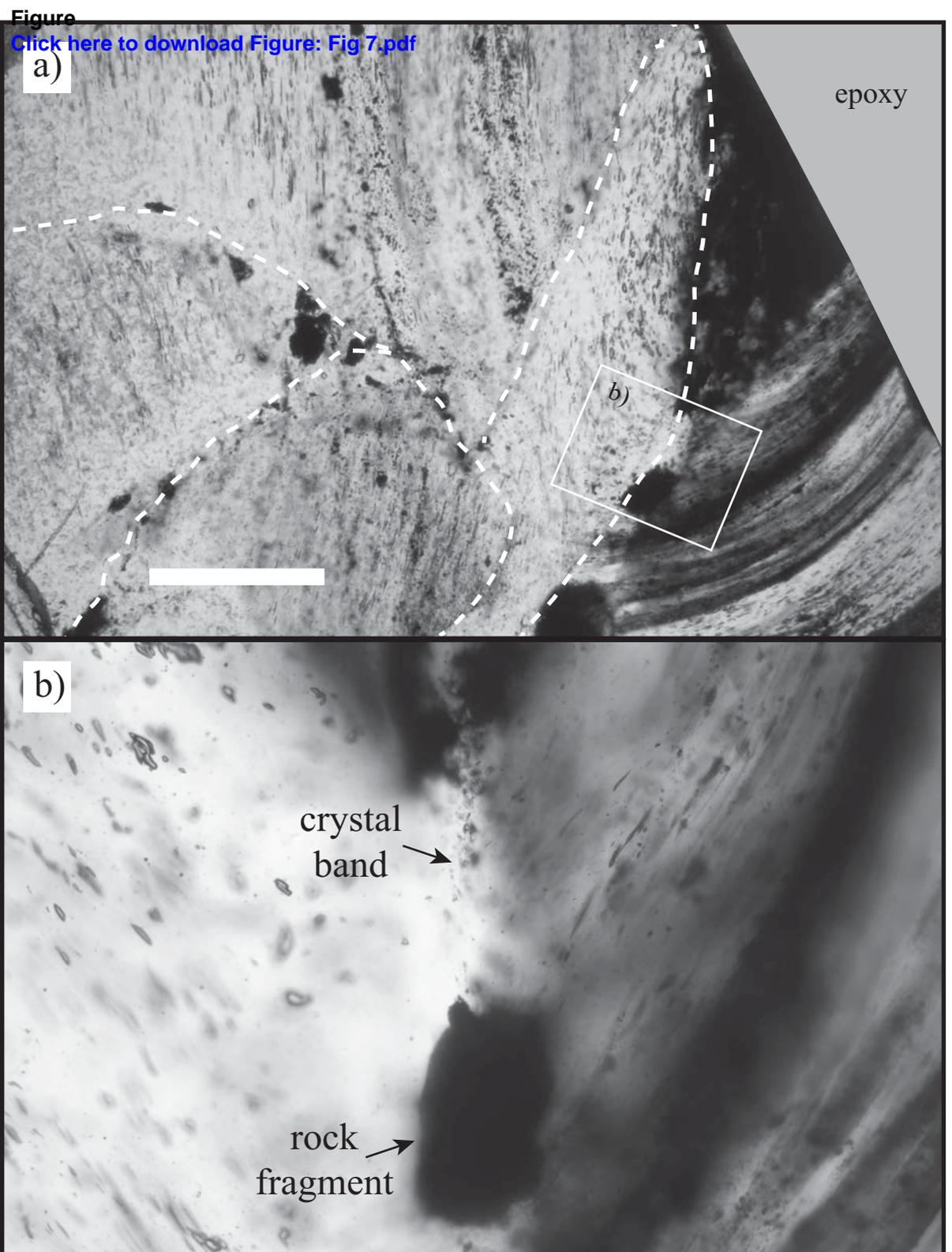


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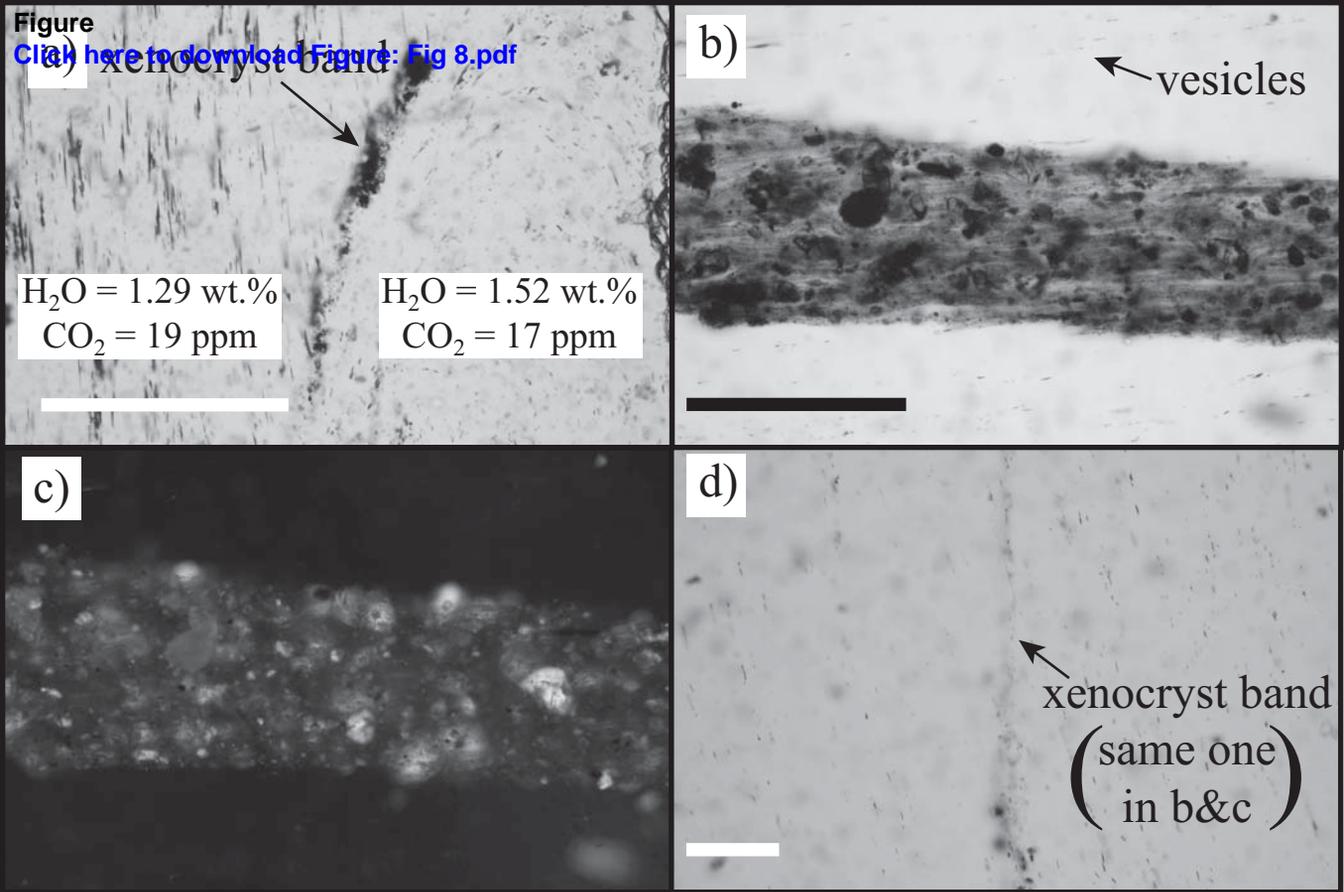


Figure 8

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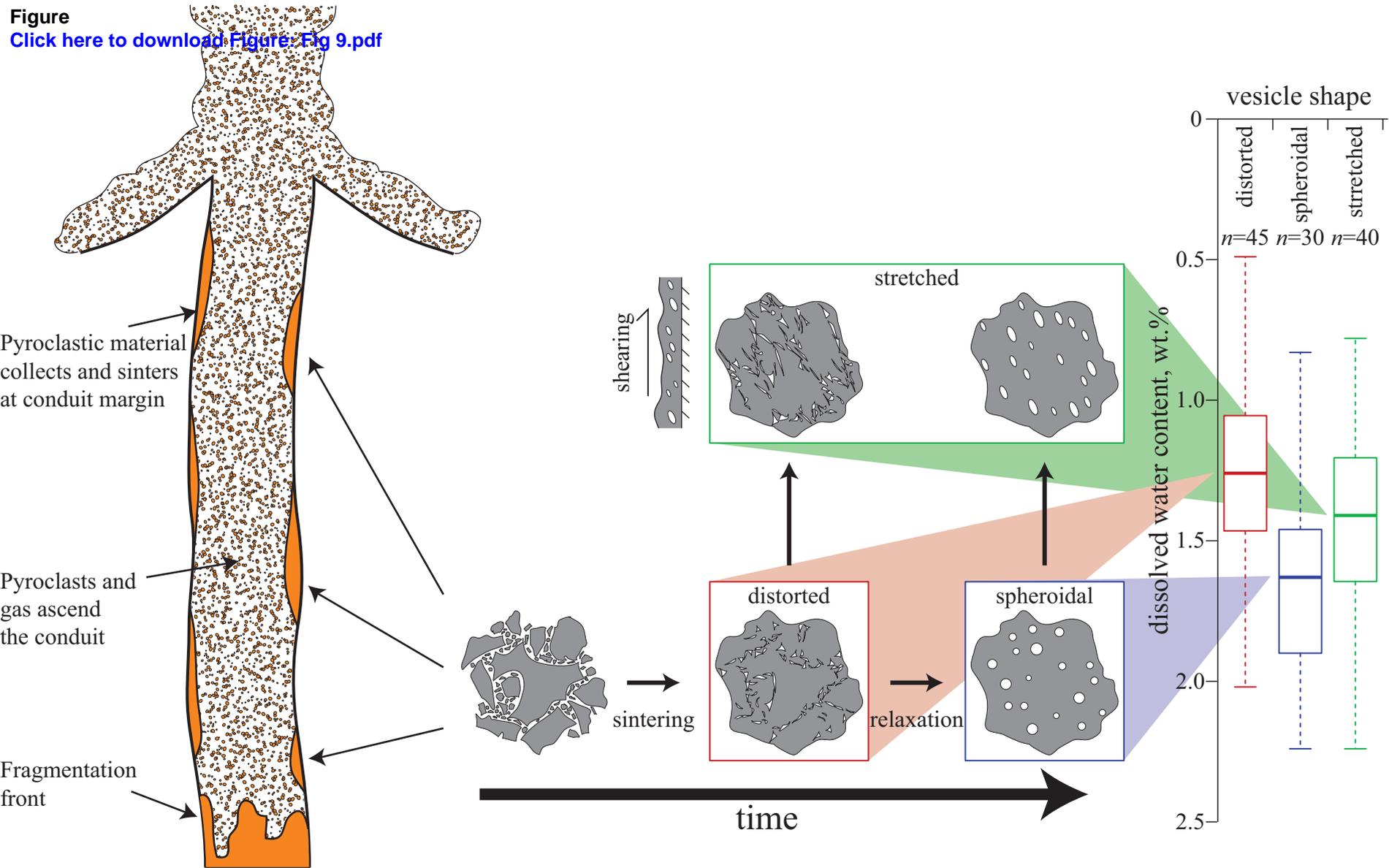


Figure 9

Bubble or particle radius (μm)

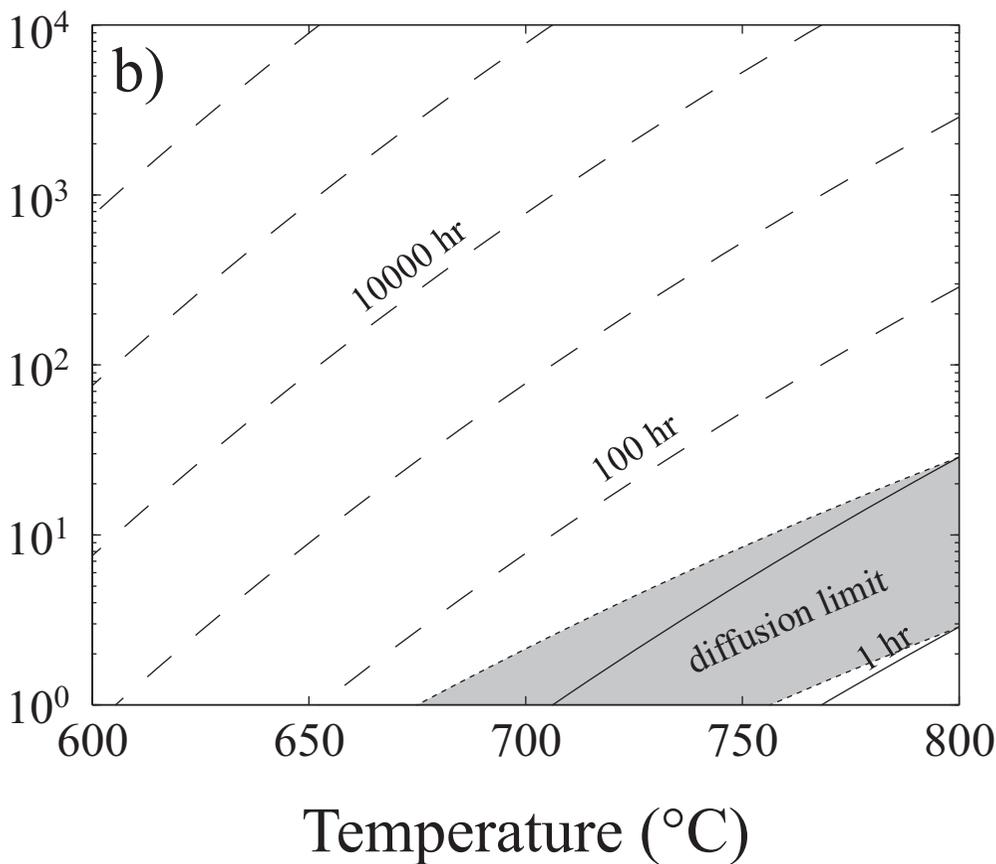
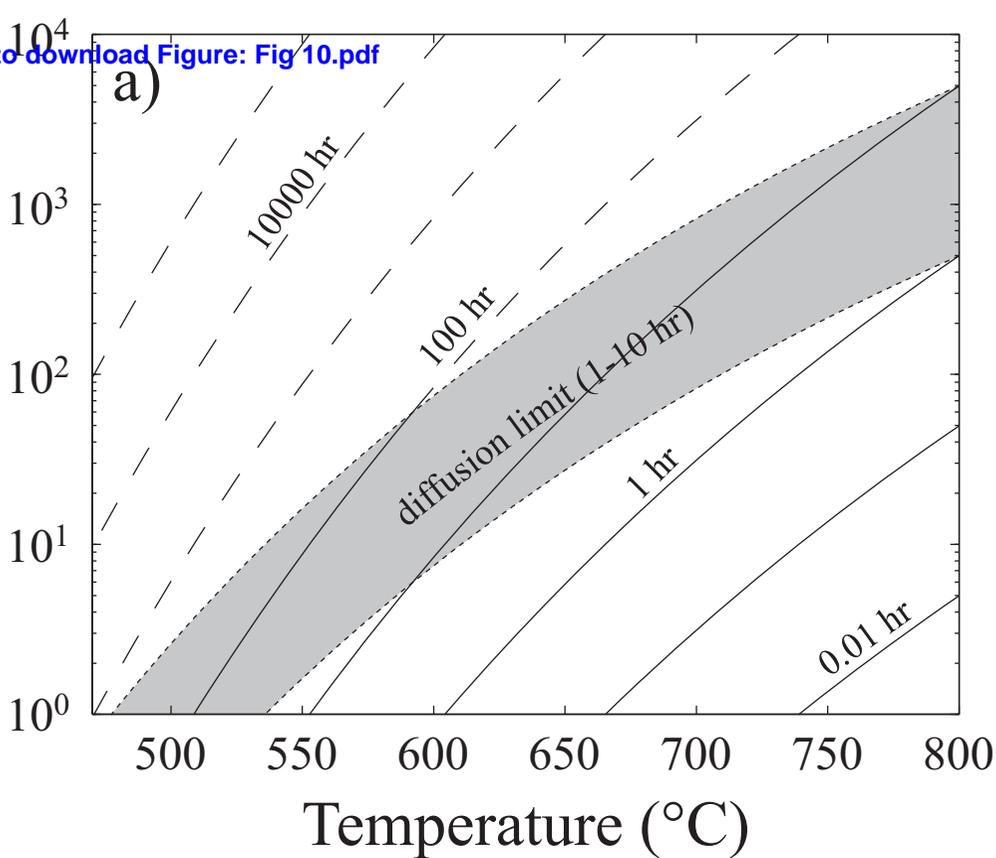


Figure 10