

**Stable isotope and geochronological study of the Mawchi Sn-W deposit,  
Myanmar: implications for timing of mineralization and ore genesis**

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## Abstract

Myanmar is endowed with abundant Sn-W mineralization, pre-eminent amongst which is the world-class Mawchi deposit. In the Mawchi area, N-S trending vertical or steeply dipping quartz veins are hosted by both Eocene granite and Carboniferous to Early Permian metasediments. Three stages of ore formation are recognized; (i) tourmaline-cassiterite stage (ii) main ore stage and (iii) sulfide stage. Tourmaline, cassiterite and pyrite-I are early-formed minerals and are representative of the first stage. Their deposition continued together with wolframite, scheelite, molybdenite, arsenopyrite, pyrite-II, fluorite and danalite, which form the second stage. This was followed by the successive deposition of sulfides such as chalcopyrite, sphalerite, galena and Pb-Bi sulfides. A  $^{40}\text{Ar}/^{39}\text{Ar}$  magmatic biotite plateau age of  $41.50 \pm 0.16$  Ma (MSWD = 2.5) is determined for the Mawchi biotite granite which we interpret as the time the granite cooled through the biotite blocking temperature, and is consistent with a previously reported LA-ICP-MS U-Pb zircon concordia age of  $42.72 \pm 0.94$  Ma (MSWD = 2). A molybdenite Re-Os model age of  $42.4 \pm 1.2$  Ma indicates that Sn-W mineralization was synchronous with late Eocene granitic magmatism. Our  $^{40}\text{Ar}/^{39}\text{Ar}$  hydrothermal muscovite plateau ages from the tourmaline granite ( $40.14 \pm 0.14$  Ma; MSWD = 1.48) and quartz vein selvages ( $40.80 \pm 0.12$  Ma; MSWD = 0.47) define the timing of hydrothermal alteration and simultaneous veining that accompanied the late stage of ore forming at Mawchi.

Fluid inclusion microthermometry from cassiterite, scheelite, quartz, and fluorite reveals that ore fluids in the cassiterite-tourmaline stage and main ore stage are characterized by moderate homogenization temperatures ( $T_h = 260^\circ - 345^\circ\text{C}$ ) and salinities (4.5 – 15.7 wt. % NaCl equiv), while the sulfide stage is characterized by lower temperatures ( $T_h = 175^\circ - 260^\circ\text{C}$ ) and moderate to low salinity (2.5 – 13 wt. % NaCl equiv). The mean  $\delta^{34}\text{S}$  of all sulfides is  $2.9 \pm 2.9$  ‰, which is suggestive that the overall system is dominated by magmatic sulfur. The similarity of  $\delta^{34}\text{S}$  values in galena and Pb-Bi sulfides ( $-1.3 - 2.7$  ‰) suggest that sulfur, and inference the Pb and Bi were transported by a common fluid, probably of magmatic origin. The calculated  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  of the hydrothermal fluid associated with cassiterite and tourmaline is 7.3 to 8.4 ‰. Scheelite and quartz deposition is characterized by lighter  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  values (2.1 to 4.9 ‰) indicating that the ore fluid might be mixed with another source of water, perhaps,

meteoric. Hydrogen isotopic compositions ( $\delta D = -51$  to  $-121$  ‰) again indicate that origin of the ore fluid is magmatic, but  $\delta D$  ( $\sim -120$  ‰) is low enough to support the  $\delta^{18}O_{H_2O}$  data that suggests, in part, a meteoric water component to the hydrothermal fluids at Mawchi.

Keywords: Mawchi, Sn-W, stable isotope, geochronology, timing, ore genesis

## 1. Introduction

Myanmar is richly endowed with mineral resources and has a long history of mining for lead-zinc-silver, tin-tungsten, gold-copper and gemstones (Thein, 1991; United Nations, 1996; Gardiner et al., 2015). Tin-tungsten (Sn-W) mineralization is one of the principal mineral resources in the country with 11,000 tonnes of tin concentrate produced in 2014, accounting for 3.72% of total world production (USGS, 2015).

The Southeast Asia tin belt, extending for 2800 km in length and up to 400 km wide extends from Billiton Island, Indonesia, in the south, through Peninsula Malaysia via southern peninsula Thailand to central and northern Thailand, eastern Myanmar and Yunnan Province (China). Collectively, the four granite belts of southeast Asia (Fig. 1a) represent one of the great metallogenic belts of the World – their metallogenic endowment is dominated by Sn-W (Eastern Granite Province, Central Granite Province and Western Granite Province), with significant Cu-Au-Mo porphyry-epithermal mineralization being a feature of the Central Valley Province.

The major tin and tungsten deposits are associated with the granites of Central Granite Province (CGP) and Western Granite Province (WGP). CGP contains the S-type granites of Triassic to Early Jurassic (Ng et al., 2015b) whereas WGP is made up of Cretaceous to Eocene I- and S-type granites (Charusiri et al., 1993; Cobbing et al., 1992; Barley et al., 2003; Mitchell et al., 2012). Mawchi deposit is located in the middle of WGP. The CGP is Indosinian in age and for

more than 40 years has been explained by a late Triassic continent- continent or continent- island arc collision consequent on the closure of Palaeo-Tethys (Mitchell, 1977; Ng et al., 2015a&b). Granites and tin mineralization in the WGP overlap in time the India- Asia collision at ca 60 Ma and ca 45 Ma but the southern end of the WGP is far from the nearest exposures of continental Indian crust in northernmost Myanmar. S-type granites in the WGP might be explicable by slab detachment beneath Myanmar- Thailand in response to the Indian collision, but a more local tectonic control related to subduction of the oceanic Indian plate seems more probable (Gardiner et al., 2015; Jiang et al., 2017).

The WGP is dominated by ilmenite-series rocks (Charusiri et al., 1993; Myint et al., 2017 a & b) of granodioritic to syenogranitic composition (Cobbing et al., 1992; Cobbing, 2011). In Thailand, the Cretaceous to Eocene (88-50Ma) granitoids of both the Sn-W bearing S-type and barren I-type granites intrude Carboniferous to Permian clastic sedimentary rocks at Phuket, Ranong and Pilok (Charusiri et al., 1993). In Myanmar, magnetite-series I-type granitic to dioritic rocks are exposed in some localities (eg. Yebokson, Yinmabin, Nattaung, Mokpalin etc.) (Cobbing et al., 1992; Barley et al., 2003; Mitchell et al., 2012) whereas highly fractionated ilmenite-series biotite and muscovite granites are spatially associated with Sn-W mineralization in the vicinity of Padatchaung, Mawchi, Dawei, Myeik and Yadanabon areas. Available geochronological data indicates that these tin granites crystallized during Cretaceous to Eocene (107 - 42 Ma) (Brook and Snelling, 1976; Myint et al., 2013; Paik and Zaw, 2014; Gardiner et al., 2016; Myint et al.,

2017a&b; Li et al., under submission). In the northern segment of the Sibumasu terrane, the Tengchong-Baoshan granitoid province (TBGP) hosts large- and medium-sized Sn deposits (Wang et al., 2014; Chen et al., 2015). These tin granites of the Yunnan Province link to the granitoids of northeastern Myanmar and emerge as the likely northern continuation of WGP. They are mostly biotite granites with high silica content (73.3 - 76.2 wt.%) and, have metaluminous to weakly peraluminous and high-K calcalkaline compositions. These characters resemble those of the WGP tin granites. Recently LA-MC-ICP-MS U-Pb zircon ages of 53 and 73 Ma have been reported for the Xiaolonghe and Lailishan tin granites, respectively (Chen et al., 2015).

The WGP in Myanmar has over 400 Sn-W occurrences and most of them are associated with granites, granite pegmatites and aplite dykes of the WGP. Sn-W mineralization in the Tin Province is found as cassiterite and/or wolframite-bearing pegmatites and greisen-bordered quartz veins. These are hosted both by the granites, and also by the Slate Belt country rock. Among the Sn-W occurrences, the Mawchi mine is a world class Sn-W deposit with estimated ore reserves of 350,000 tons (Htay et al., 1997). Mawchi is situated 219 km southeast of Naypyitaw and was the world's largest granite-hosted tin-tungsten vein system before World War II. During the period of 1930-1940 the company produced 2000 to 6000 tonnes of ore annually, amounting to 60% of the total production of Myanmar. Approximately 25,000 tonnes of Sn-W ore were produced from 1980-2010 (unpublished Mawchi mine data). Although there is some literature describing the geology of Mawchi mine and surrounding area (Dunn, 1938; Hobson, 1941; Zaw and Thet, 1983;

Myint et al., 2017a), no new data have been published relevant to the ore deposit geology of Mawchi mine in recent years. Here we present and discuss new data associated with the mineralogy, fluid inclusion microthermometry, stable isotope and,  $^{40}\text{Ar}/^{39}\text{Ar}$  and Re-Os geochronology of the Mawchi deposit to constrain the ore genesis and timing of the Sn-W mineralization and implications for Sn-W deposit exploration in the WGP.

## 2. Geological background

### 2.1 Regional geologic setting

As shown in (Fig. 1a) the MMB lies immediately west of the mineralized WGP granites. The MMB also hosts S- type granites but these lack tin mineralization, perhaps because of either a deeper erosion level or source rock different to that which produced the partial melts which resulted in the mineralized granites. The WGP contains Cretaceous to Tertiary granitoid rocks of metaluminous to peraluminous, high K calc-alkaline composition (Cobbing et al., 1992; Mitchell et al., 2012; Myint et al., 2017a&b). Miocene granitic dyke rocks cut the older granitic rocks in some places. The Sn-W mineralization is predominantly associated with granites, granite pegmatites and aplite dykes from the WGP which intruded Carboniferous to Early Permian metasedimentary and sedimentary sequences of the Slate Belt (SB; Bender 1983; Mitchell et al., 2004; Myint et al., 2013; Myint et al., 2017a). The SB comprises the sediments and low grade metasediments of Lebyin, Mawchi and Mergui Groups in Myanmar and Kaeng Krachan Group in

Thailand.

The MMB is a high-grade metamorphic sequence of gneiss, schists and marbles. In the northern part of the belt around Mogok, ruby-bearing phlogopite–diopside–graphite–spinel–bearing marble or calc-silicate and sapphire-bearing post-deformation pegmatitic nepheline syenites are found (Iyer, 1953; Mitchell et al., 2007). The marbles are probably metamorphosed from either the Precambrian (Chhibber, 1934) or Precambrian to Jurassic rocks of the Shan Plateau (Searle and Haq, 1964). Some units of meta-igneous rocks and marbles at and west of Mogok have been correlated with the Silurian and Ordovician units of the Shan Plateau (Thein et al., 1988). Around the Mandalay-Kyaukse-Tharzi areas, the MMB is mostly composed of marbles and calc-silicate rocks with minor schists and gneiss. Some outcrops of low-grade marble contain relict Ordovician to Permian fossils. The fossiliferous rocks and probably the marble and calc-silicates in this segment of the MMB are part of the Paleozoic to Triassic Plateau sequence. Garnet and sillimanite schists with augen gneisses and minor marble continue southwards to the Andaman Sea coast occupying a parallel belt to the WGP and SB.

The Mawchi Sn-W deposit is located in the eastern part of Myanmar. The regional geology is comprised by the NNW-SSE trending juxtaposition of the Mogok Metamorphic Belt (MMB; Searle and Haq, 1964; Mitchell et al., 2007) and metasedimentary and sedimentary rocks of the Mawchi Group (a part of SB) amalgamated with a succession comprising Paleozoic to Triassic shallow marine carbonates and sediments with Jurassic and Cretaceous continental



sediments (Fig. 1b).

Permian Plateau Limestone occurs as NNW-SSE trending intermittent limestone caps on the metasediments and granite striking parallel to shear splays. The prominent rocks of the Mawchi Group are hornfels, mudstones, fine-grained sandstone and metagreywacke. Grayish and reddish colored fine-grained sandstones, argillite and phyllite are common, and grayish and reddish mudstones and claystones occur interbedded between sandstone and argillite in some localities.

## 2.2 Geology of Mawchi Sn-W district

The geology of the area is characterized by the metamorphic rocks of MMB, metasedimentary rocks of Carboniferous to Early Permian Mawchi Group (Myint et al. 2017a), Permian Plateau Limestone and intrusive bodies of biotite granite and tourmaline granite of WGP (Hobson, 1941; Myint et al., 2017a) (Fig. 2a). Metamorphic rocks of the MMB, containing gneisses, granite gneisses and schists, are found in the western part of Mawchi area and their contact with the Mawchi Group appears to be defined by a fault. The Mawchi Group shows extreme lithological variations, including; mudstone, shale, fine-grained siltstone and sandstone, slate, quartzite and metagreywacke. Elongate and small granite bodies of the WGP are dispersed and intrude the metasedimentary rocks and Permian Plateau Limestone along NNW-SSE fracture systems. Most

of these granite bodies are coarse-grained porphyritic biotite granites with accessory epidote and apatite (Hobson, 1941).

In the Mawchi Mine area, a granite pluton is composed of medium- to coarse-grained biotite granite, grading into fine- to coarse-grained tourmaline granite towards the mineralized zone. Pervasive tourmalinization is superimposed on the biotite granite. The biotite granite is mainly composed of quartz, feldspar, biotite with trace amount of tourmaline and contains high silica content (74 - 86 wt. %) with A/CNK [molecular  $\text{Al}_2\text{O}_3 / \text{CaO} + \text{Na}_2\text{O} + \text{K}_2\text{O}$ ] >1. It possesses minor hydrothermal alteration characterized by a trace presence of hydrothermal tourmaline. The tourmaline granite consists of quartz, feldspar, muscovite and tourmaline (Myint et al., 2017a). Biotite is rare in the tourmaline granite. Hydrothermal muscovite-sericite clusters or aggregates are commonly associated with and occasionally replace tourmaline. Spider diagrams for the Mawchi granites show with pronounced negative Ba, Sr, Nb and Ti anomalies with positive Rb, Pb, U anomalies and the granites have low Nb/Ta and Zr/Hf values. These granites were derived from a magma generated by the melting of crustal rocks and extensive magmatic differentiation. Trace element geochemistry and the low  $\epsilon\text{Nd}$  (T) values reveal that distinct crustal sources have been involved in emplacement of the peraluminous granitic rocks and their Sn-W enrichment (Myint et al., 2017a). The LA-ICP-MS U-Pb zircon ages of the biotite and tourmaline granites are  $42.72 \pm 0.94$  Ma (n = 12; MSWD = 2) and  $43.71 \pm 0.39$  Ma (n = 12; MSWD = 1.02), respectively (Myint et al., 2017a). Petrographic, geochemical and geochronological studies

indicate that the tourmaline granite of Mawchi mine area is a hydrothermally altered biotite granite (Myint et al., 2017a).

Several satellite Sn-W prospects surround the Mawchi deposit, comprising; Htawmawkhee, Ketaunggalay, Hteelakhee and Mosakhee. The Htawmawkhee prospect is composed of parallel and inclined small quartz veins that strike NW-SE. These 2 to 14 cm thick veins are hosted by argillite and contain cassiterite, wolframite and tourmaline. Nearly N-S trending sheeted veins are also observed. In the Ketaunggalay prospect, wolframite-bearing quartz and pegmatite veins are hosted by metasandstone of the Mergui Group and strike in a nearly N-S direction. Hteelakhee is 12 km south of the Mawchi mine and is characterized by cassiterite and wolframite bearing quartz veins and sheeted veins hosted by the argillite and sandstone. The Mosakhee prospect is situated near the village of Mosakhee where wolframite and stannite are associated with some sulfides such as galena and sphalerite (Myint, 2015). The quartz veins trend N-S and are hosted by the metasediments.

### **3. Mawchi Sn-W mineralization**

The Mawchi ore body is classified into three types according to vein geometry and mineral assemblage. The principal mineralization is represented by cassiterite-wolframite bearing quartz veins that are hosted by the tourmaline granite and metasedimentary rocks (Fig. 2b&c). These

217 veins are generally confined to a N-S trending fracture system (Fig. 3a), which upwardly taper  
218 forming a minor stockwork style vein network (Fig. 3b). In the vein system close to the limestone,  
219 scheelite-danalite assemblage form a skarn assemblage. Danalite occurs as bands parallel to the  
220 vein wall (Fig. 3c) or as irregular patches in the vein. Post-mineralized veins are horizontal and  
221 gently inclined, cutting through mostly metasedimentary rocks. These veins are, generally, quartz-  
222 tourmaline veins and are barren of ore minerals and are common in the uppermost part of the ore  
223 body (Fig. 3d).

224 Cassiterite and wolframite occur in the marginal zone (Fig. 3e) as well as in the central  
225 zone of the veins, whereas sulfide minerals occur only in the central zone, suggesting that oxide  
226 ore minerals crystallized early and sulfide minerals late in the paragenetic sequence. Fractures are  
227 occasionally observed to host an early formed mineral assemblage of tourmaline – cassiterite  $\pm$   
228 pyrite (Fig. 3f). Massive bands and pockets of cassiterite  $\pm$  tourmaline  $\pm$  muscovite  $\pm$  pyrite are  
229 also common at the vein margins (Fig. 3g).

230 The common gangue minerals in the Mawchi vein system are tourmaline, micas  
231 (muscovite, sericite, gilbertite, and lepidolite), danalite, fluorite, chlorite, clay minerals (kaolinite,  
232 illite), calcite and beryl. Tourmaline, the most common gangue mineral, is dispersed widely, not  
233 only in the veins, but also in the host rocks. Tourmaline also represents the dominant alteration  
234 associated with Sn-W mineralization, with its deposition preceding that of wolframite and other  
235 sulfides in some cases. Other common alteration assemblages are albitization, sericitization,

silicification and kaolinization. Albitization occurs in the highly tourmalinized granite (Myint et al., 2017) whereas sericitization is characterized by the mica selvages along the contact of host rock and quartz veins. The sericitization postdates the tourmalinization as can be seen clearly in the tourmaline granite (Fig. 3h).

Assay data corresponding to the veins from various levels are tabulated in Table 1 and indicate a downward decrease of Sn and WO<sub>3</sub> content. The Sn:WO<sub>3</sub> varies among different veins ranging mostly between 1: 0.7 and 1: 0.5.

## **4. Methodology**

### *4.1 Mineralogical and fluid inclusion study*

The mineralogical study utilized transmitted and reflected light microscopy, and x-ray diffractometry (XRD), Scanning Electron Microscopy (SEM), and Electro Probe Micro Analysis (EPMA). The mineral identification by the XRD method was performed using a Rigaku Ultima IV X-ray Diffractometer at the Economic Geology Laboratory, Kyushu University. Elemental analyses were performed using a Shimadzu Superscan SSX-550, equipped with an energy-dispersive x-ray analyzer (SEM-EDX) at the Advanced Research Center and a JEOL JXA8530F field emission EPMA at the Department of Environmental Changes, Kyushu University. The fluid

inclusion analysis were performed by measuring the final ice melting point (salinity calculations) and heating the fluid inclusion samples (homogenization temperature) using a calibrated Linkam-THMS600 heating-freezing stage.

#### 4.2 Ar-Ar methods

Magmatic biotites from the biotite granite of the Mawchi area were separated with the objective of determining the magmatic cooling age. Hydrothermal muscovites from the tourmaline granite exposed near the mine site and a quartz vein from an underground working in the deposit were selected to determine the timing of alteration that is later than the tourmalinization which is spatially and genetically associated with the Sn-W ore deposition.

Mica samples were irradiated together with the TCR sanidine standard (batch 2; 28.34Ma recalculated by [Renne et al., 1998](#)) at the NRG-Pattern HFR Rodeo facility in the Netherlands, and later analyzed at Lund University, Sweden using a Micromass 5400 Mass Spectrometer with a Faraday and an electron multiplier. The mass spectrometer is equipped with a metal extraction line, containing two SAES C50-ST101 Zr-Al getters and a cold finger cooled to app. -155°C by a Polycold P100 cryogenic refrigeration unit. One or two grains of mica were loaded into a copper planchette consisting of a number of 3 mm holes. Samples were step-heated using a defocused 50W CO<sub>2</sub> laser rastered over the samples to provide even heating of all grains. Sample clean up

times were 5 min (using two hot Zr-Al getters and a cold finger). The laser was rastered over the samples to provide even heating of all grains. The analytical process is automated and runs on a Macintosh OS 10.2 platform with software modified specifically for the laboratory at the University of Lund. Time zero regressions were fitted to data collected from 10 scans over the mass range of 36 to 40. Peak heights and backgrounds were corrected for the mass discrimination, isotopic decay and interfering nucleogenic Ca-, K-, and Cl-derived isotopes. Isotopic production values for the cadmium lined position in the Petten reactor are  $^{36}\text{Ar}/^{37}\text{Ar}(\text{Ca}) = 0.000270$ ,  $^{39}\text{Ar}/^{37}\text{Ar}(\text{Ca}) = 0.000699$ , and  $^{40}\text{Ar}/^{39}\text{Ar}(\text{K}) = 0.00183$ .

Blanks were measured before each sample and after every three-sample step. Blank values were subtracted from the sample signal for all incremental steps. Age plateaus were determined following the criteria of [Dalrymple and Lanphere \(1971\)](#), specifying the presence of at least three contiguous incremental heating steps with statistically indistinguishable ages and constituting more than 50 % of the total  $^{39}\text{Ar}$  released during the experiment. Plateau ages are quoted with uncertainties at the  $2\sigma$  level ([Hermansson et al., 2008](#)). The Taylor Creek Rhyolite (TCR) sanidine standard ( $28.34 \pm 0.16$  Ma; [Renne et al., 1998](#)) was used as flux monitor. J-values were calibrated with a precision of 0.25 %. The uncertainties in the J-values have been propagated into the uncertainties in the plateau ages. However, the uncertainties in the decay constant and in the standards are not propagated.

### 4.3 Re-Os methods

The molybdenite in the Mawchi deposit is fine-grained (0.2 – 1 mm) and rare. The molybdenite is associated with cassiterite-wolframite-sulfides bearing quartz-tourmaline vein, and possesses mica and tourmaline selvages. Molybdenite from one sample was physically (crushed, magnetic and heavy liquid separation, and hand-picked) isolated from a quartz vein to yield a mineral separate of several tens of milligrams.

Rhenium-osmium (Re-Os) analysis was conducted using isotope-dilution negative ionization mass spectrometry at Durham University ([Selby and Creaser, 2001](#)). The molybdenite sample, together with a known amount of tracer solution ( $^{185}\text{Re}$  + normal Os) was digested in a carius tube with 3 mL HCl (11 N) and 6 mL (15 N  $\text{HNO}_3$ ) at 220°C for 24 hours. Osmium from the acid solution was isolated via solvent extraction ( $\text{CHCl}_3$ ), and purified by using micro-distillation. The Re from the Os extracted solution was extracted via NaOH-Acetone extraction and purified by anion chromatography ([Cumming et al., 2012](#)). The isolated Re and Os fractions were loaded onto Ni and Pt wire filaments, respectively, with the isotope compositions determined via Faraday cups of a Thermo Scientific TRITON mass spectrometer. The Re-Os data are blank corrected using values of  $2 \pm 0.03$  ppt for Re,  $0.11 \pm 0.04$  ppt for Os, with an  $^{187}\text{Os}/^{188}\text{Os}$  of  $0.25 \pm 0.02$ . The Model Re-Os age is calculated using the equation  $\ln(^{187}\text{Os}/^{187}\text{Re} + 1)/\lambda$ , using the decay constant ( $\lambda$ ) of  $1.666\text{e}^{-11}\text{a}^{-1}$  ([Smoliar et al., 1996](#)). Uncertainties include all sources of



analytical uncertainty (Re and Os mass spectrometer measurements, blank abundances and isotopic compositions, spike calibrations and reproducibility of standard Re and Os isotopic values), both with and without the uncertainty in the decay constant (0.35%; [Smoliar et al., 1996](#); [Selby et al., 2007](#)). All uncertainties are presented at the  $2\sigma$  absolute level.

#### 4.4 *Stable isotopic analysis*

Stable isotope compositions were analyzed at the Scottish Universities Environmental Research Centre (SUERC), UK. For O-isotopic analyses, mineral separates (cassiterite, wolframite, tourmaline, quartz and scheelite) were analyzed by the laser fluorination procedure, involving total sample reaction with excess  $\text{ClF}_3$  using a  $\text{CO}_2$  laser as a heat source (in excess of  $1500^\circ\text{C}$ ). All combustions resulted in 100% release of  $\text{O}_2$  from the mineral lattice. Liberated  $\text{O}_2$  was then converted to  $\text{CO}_2$  by reaction with hot graphite, then analyzed on-line using a VG SIRA 10 spectrometer. Reproducibility is  $\sim \pm 0.3$  ‰. Results are reported in standard notation ( $\delta^{18}\text{O}$ ) as per mil (‰) deviations from the Standard Mean Ocean Water (V-SMOW) standard.

Pyrite, arsenopyrite, chalcopyrite, sphalerite, galena and Pb-Bi sulfides were selected for sulfur isotope analyses. Pyrite-I and Pb-Bi sulfides were prepared for conventional isotopic analysis by standard heavy liquid (sodium polytungstate) and hand picking techniques. One gram of sample was collected from each although around 5 to 10 mg were utilized for isotopic analysis. Minor contamination by non S-bearing phases was tolerated, and has no isotopic effect on the final

data. Sulfides were analyzed by standard techniques (Robinson and Kusakabe, 1975) in which SO<sub>2</sub> gas was liberated by combusting the sulfides with excess Cu<sub>2</sub>O at 1075°C, *in-vacuo*.

Other sulfides such as pyrite-II, arsenopyrite, chalcopyrite, galena and sphalerite were examined by *in-situ* laser combustion from standard polished blocks. The blocks were inserted into a sample chamber, which was evacuated and subsequently filled with an excess of oxygen gas (Fallick et al., 1992). The spatial resolution is determined by the minimum volume of SO<sub>2</sub> gas required for analysis (0.05–0.10 µmol), with minimum spot sizes of approximately <100 µm. Samples were rastered under the laser beam to combust individual zones extending around 500 by 100 µm. The laser beam was a SPECTRON LASERS 902Q CW Nd:YAG laser (1 W power), operating in TEM<sub>00</sub> mode. The SO<sub>2</sub> gas produced by each laser combustion was purified in a miniaturized glass extraction line, using a CO<sub>2</sub>/acetone slush trap to remove water and a standard *n*-pentane trap to separate SO<sub>2</sub> and trace CO<sub>2</sub> (Kelley and Fallick, 1990). Determination of the sulfur isotope composition of the purified SO<sub>2</sub> gas was carried out on-line by a VG SIRA II gas mass spectrometer.

The international standards employed to calibrate the reference gas were NBS-123 and IAEA-S-3, and SUERC standard CP-1 which gave δ<sup>34</sup>S values of +17.1‰, -31.5‰ and -4.6‰ respectively, with 1σ reproducibility around ±0.2 ‰ in this study. Data are reported in δ<sup>34</sup>S notation as per mil (‰) variations from the Vienna Canyon Diablo Troilite (V-CDT) standard. The laser extraction method results in a sulfur isotope fractionation between the host mineral and the SO<sub>2</sub>

349 gas produced via combustion, which is mineral-specific (Kelley and Fallick, 1990).  
350 Experimentally determined fractionation factors are currently available in the SUERC system to  
351 apply correction to several sulfide/sulfosalt minerals, for example, pyrite (+0.8 ‰) and galena  
352 (+2.4 ‰). Further discussion of the laser-induced fractionation is given in Wagner et al. (2002).

353 For H isotope analysis of fluid inclusion water, pure mineral samples of cassiterite,  
354 wolframite and quartz (around 1 g ) with one dominant fluid inclusion population, were heated to  
355 100°C overnight under high vacuum to release labile volatiles and loaded into outgassed Pt  
356 crucibles. Samples were gradually heated by radiofrequency induction in an evacuated quartz tube  
357 to temperatures in excess of 700°C. For H isotope analysis of tourmaline, pure samples of ~ 800  
358 mg were heated to 150°C overnight under high vacuum to release labile volatiles after loading into  
359 outgassed Pt crucibles. Samples were then gradually heated by radiofrequency induction in an  
360 evacuated quartz tube to temperatures in excess of 1200°C. The released water was reduced to H<sub>2</sub>  
361 in a chromium furnace at 800°C (Donnelly et al., 2001), with the evolved gas measured  
362 quantitatively in an Hg manometer, then collected using a Toepler pump. The gas was subsequently  
363 analyzed on a VG 602D mass spectrometer with a manual Hg, high gas compression inlet system.  
364 Replicate analyses of international water standards (V-SMOW and GISP), and an internal standard  
365 (Lt Std) gave a reproducibility of  $\pm 2$  ‰. Replicate analyses of the international mineral standard  
366 NBS-30 (biotite) also gave reproducibility around  $\pm 2$  ‰. A more conservative error for fluid  
367 inclusion samples would be around  $\pm 5$  ‰.

368

## 369 **5. Mineralogy and vein paragenesis**

370 The paragenetic sequence of ore deposition in the Mawchi vein system is presented in [Fig.](#)  
371 [4](#) and has been divided into three stages.

### 372 *5.1 Tourmaline-cassiterite stage*

373 The onset of ore deposition was initiated by pervasive tourmalinization with tin  
374 mineralization. Early-formed tourmaline in the tourmaline granite is likely the precursor of quartz  
375 vein deposition and successive deposition of late tourmaline and cassiterite took place not only  
376 along the small fractures and vein boundary as masses and patches, but also in the highly altered  
377 tourmalinized granitic rocks. In such case, wolframite is rarely found with tourmaline and  
378 cassiterite, and, these early-formed minerals coexist with first generation of pyrite (pyrite-I) in  
379 some places. Cassiterite is the major ore mineral in this stage and has been identified as  
380 idiomorphic crystals as well as a massive band with tourmaline, muscovite and pyrite-I at the vein  
381 margin. Sometimes, fine tourmaline segregations with the cassiterite can be found as dark-colored  
382 patches in the vein. Pyrite-I crystals are, generally idiomorphic or sub-idiomorphic, up to 0.3 mm  
383 long and usually dispersed in both the granite body and veins showing it formed before wolframite.  
384 The late phase of this stage overlapped with the main ore stage and continued until the end of oxide  
385 ore deposition.

### 386 *5.2 Main ore stage*

The second stage of ore deposition (main ore stage) in the Mawchi vein system appears to have begun with the precipitation of major oxide ore minerals such as cassiterite, wolframite and scheelite with subordinate amount of such sulfide minerals as molybdenite, arsenopyrite and second generation pyrite (pyrite-II) (Fig. 5a). Minute molybdenite flakes commonly occur in the deep level vein system and are affected by kink banding adjacent to the undeformed mineral grains such as pyrite-II and chalcopyrite, and pyrite-II penetrates the cleavage plane of molybdenite (Fig. 5b). Molybdenite occurs as isolated flakes in quartz or in tourmaline or at the border of wolfram and quartz (Dunn, 1938). Cassiterite has been identified as idiomorphic crystals as well as forming a mixed matrix with wolframite, arsenopyrite and pyrite in the vein. Color zoning is common (Fig. 5c). Wolframite is the principal tungsten mineral of the Mawchi mine and it can be found either as large dispersed crystals or intergrown with cassiterite and/or with sulfide minerals. Marginal replacement by early-formed sulfide minerals (Fig. 5d) and scheelite replacement are common (Fig. 5e) in wolframite. Although ferberitic in some veins, wolframites from Mawchi veins are compositionally mostly hubneritic to wolframitic in composition. The first generation wolframite is generally wolframitic and intergrown with cassiterite, with the second generation wolframite, which contemporaneously appears with pyrite-II, is mostly of hubneritic composition.

Two distinct generations of pyrite with several modes of formation are identified. Pyrite-I, occurring in considerable amounts and as tiny cubes, is the earliest sulfide mineral followed by pyrite-II and arsenopyrite. Pyrite-II occurs very widely and shows a cataclastic texture in some

places. Pyrite-II formed simultaneously with or somewhat later than arsenopyrite. In the vein system, pyrite-I cubes are surrounded by the fractured pyrite-II and other later-formed sulfides, such as arsenopyrite (Fig. 5f). Clusters of minute acicular pyrite are also observed in the lower part of veins as void-filling minerals among the other ore and gangue minerals. Scheelite forms mainly as a secondary mineral replacing wolframite and partly as milky white to yellowish white crystals in veins (Fig. 5g) close to the contact of limestone and granite. Scheelite is associated with danalite, wolframite, cassiterite and tourmaline, suggesting a skarn forming event. Some scheelite crystals contain tourmaline (Fig. 5h) and cassiterite as inclusions. Interstitial filling of pyrite-II between the grains of cassiterite and wolframite is common (Fig. 6a). Arsenopyrite, one of the most common and abundant of vein sulfides, frequently occurs as anhedral aggregates and is sometimes intergrown with pyrite-II. Most arsenopyrite formed early in the vein sequence, evidently overlapping wolframite and some early generation pyrite-II.

Tourmaline and muscovite-sericite are still common in the main ore stage and danalite, chlorite, early fluorite are other common gangue minerals. After the main ore and sulfide stages, tourmaline continued to deposit together with quartz. Fluorite, another common gangue mineral, is of two distinct types, which differ in color and mode of occurrence. The first fluorite species is generally bluish to purplish or greenish in color spatially associated with wolframite, cassiterite (Fig. 6b), tourmaline, chlorite and muscovite-sericite. Another species of fluorite is mostly greenish color enriched in yttrium and forms yttrifluorite in some places. It is the product of

hydrothermal activity, which was either early or contemporaneous with the deposition of Nb-Ta-REE minerals of the sulfide stage.

### 5.3 Sulfide stage

The third stage (Sulfide stage) is defined by the successive deposition of sulfide minerals such as chalcopyrite, sphalerite, galena and Pb-Bi sulfides with minor formation of fergusonite, wolframo-ixiolite and other Nb-Ta minerals. Tiny and dispersed fergusonite, wolframoixiolite (Fig. 6c), Nb-bearing rutile and other Nb oxide minerals are common in the lower levels of the Mawchi vein system where the mineralization is hosted by granite. The Nb-Ta minerals are probably the result of exsolution and replacement of scheelite and hubneritic wolframite (Myint et al. in review). Sphalerite occurs not only as a replacement mineral in the interstices and grain boundaries of arsenopyrite and pyrite crystals, but also as inclusions in galena. It usually contains 2.4 – 11.6 wt. % Fe, and encloses blebs and tolls of chalcopyrite and stannite, forming as an exsolution pattern (Fig. 6d). Therefore, sphalerite is contemporaneous with chalcopyrite, but younger than arsenopyrite and pyrite, and older than galena. Chalcopyrite occurs not only as tiny veinlets and replacements (type-I) in arsenopyrite, pyrite and other early vein minerals (Fig. 6e), but also as anhedral grains (type-II) (Fig. 6f). Galena is a late sulfide mineral following sphalerite, replacing the former sulfides and enclosing them as inclusions. Galena can be found as marginal replacement along the borders of chalcopyrite and sphalerite (Fig. 6g), along the fractures of pyrite grains, and as scattered mineral grains in the quartz. The galena and sphalerite are sometimes

veined by cerussite.

The later phase of the sulfide stage is marked by the deposition of bismuth-bearing galena and other bismuth-lead sulfides. These are widespread in the veins and replace early-formed sulfides of pyrite, chalcopyrite and sphalerite (Fig. 6h). The bismuth-bearing galena contains 0.23 - 1.37 wt. % Bi, whereas Bi-Pb sulfides (including cosalite) contain 0.2 - 1.76 wt. % Ag. Pyrrhotite is found as very rare accessory and, hematite, covellite, chalcocite and cerussite occur as supergene minerals.

## 6. Results

### 6.1 Ar-Ar and Re-Os dating

Mica Ar-Ar data is presented in Table 4 and, plateau age and isochron diagrams are shown in Fig. 7. Magmatic biotite from the biotite granite at Mawchi yields a well-defined plateau age of  $41.50 \pm 0.16$  Ma (MSWD = 2.5), with an inverse isochron date of  $42.4 \pm 0.3$  Ma (MSWD = 1.7). The hydrothermal muscovites dated from the tourmaline granite and a quartz vein give plateau ages of  $40.14 \pm 0.14$  Ma (MSWD = 1.48) and  $40.80 \pm 0.12$  Ma (MSWD = 0.47) respectively. The plateau ages are consistent with the inverse isochron dates of  $40.20 \pm 0.3$  Ma (MSWD = 1.7) and  $40.73 \pm 0.17$  Ma (MSWD = 1.5), respectively. The Re-Os abundances and model age of quartz vein-hosted molybdenite are given in Table 5. The concentrations of Re and  $^{187}\text{Os}$  of the Mawchi molybdenite are 1.36 ppm and 0.61 ppb, respectively. The Re-Os data yield a Re-Os model age of



42.43 ± 1.21 [42.43 ± 1.22 including the uncertainty in the decay constant] Ma.

## 6.2 Fluid inclusions microthermometry

Liquid-rich two-phase (liquid and vapor) primary fluid inclusions are dominant in massive cassiterite and in cassiterite, scheelite and quartz from the main ore stage and, quartz and fluorite from sulfide stage. Fluid inclusions vary from being circular through tubular to irregular in shape, and are between 7 and 25µm in size. The homogenization temperatures (Th) in primary fluid inclusions of massive cassiterite from cassiterite-tourmaline stage range from 311° to 343°C (Fig. 8), with final ice melting temperatures of -2.7° to -7.9°C suggesting moderate salinity (4.5 to 11 wt. % NaCl equiv) (Fig. 9). Primary inclusions in vein cassiterite, formed during the main ore stage with tourmaline, homogenized between 312° and 340°C and exhibit ice melting temperatures of -2.5° to -5.6°C, corresponding to salinities of 4.7- 11.9 wt. % NaCl equiv. Fluid inclusions in scheelite from the quartz veins record moderate homogenization temperatures (261° -313°C) and final ice melting temperatures that indicates moderate salinity (6.8 – 15.7 wt % NaCl equiv).

Based on salinity, fluid inclusions of fluorite from sulfide stage can be divided into two groups: low salinity (2.5 – 5 wt. % NaCl equiv) inclusions with low Th (174° - 185°C) and, moderate salinity (10– 13 wt. % NaCl equiv) inclusions with low Th (175° - 220°C). Quartz samples from were also studied, yielding Th of 180° to 290°C, similar to temperatures reported by Zaw and Thet (1983), and low salinity (3.5 – 7.7 wt. % NaCl equiv).

482

### 483 6.3 *Stable isotopic systematics*

#### 484 6.3.1 *Sulfur isotopes*

485 A total of 45 samples of sulfide minerals from all stages of mineral paragenesis were  
486 analyzed to ascertain the source of sulfur in the mineralization of the Mawchi deposit.

487 The  $\delta^{34}\text{S}$  data are listed in Table 2 and plotted in Fig. 10.

488 Pyrite-I from the tourmaline granite and vein samples, which are the representatives of the  
489 tourmaline-cassiterite stage, yield  $\delta^{34}\text{S}$  values of +5 ‰ and +5.6 ‰. Pyrite-II and arsenopyrite  
490 from the main ore stage gave  $\delta^{34}\text{S}$  values of 3.8 to 6.2 ‰ and 2.5 to 3.6 ‰, respectively.

491 Chalcopyrite, sphalerite and galena from the sulfide stage reveal a broader range in  $\delta^{34}\text{S}$   
492 values, with type-II chalcopyrite possessing  $\delta^{34}\text{S}$  values from 0.9 - 1.1 ‰, whereas type-I yields  
493 two populations of  $\delta^{34}\text{S}$  values, 4.5 - 6.2 ‰ and 9.4 - 10.6 ‰. The  $\delta^{34}\text{S}$  values of sphalerite range  
494 between -3.2 and 3.2 ‰. Galena defines a narrow range in  $\delta^{34}\text{S}$  values of -1.3 to 1.8 ‰.

495 The final stage of mineralization, bismuth-bearing galena and Pb-Bi sulfides yield  $\delta^{34}\text{S}$   
496 values of 0.7 - 2.7 ‰, which are quite similar to those of galena. Overall, the entire sulfide set  
497 from Mawchi defines a narrow  $\delta^{34}\text{S}$  range around a mean of  $2.9 \pm 2.9\text{‰}$  ( $1\sigma$ ;  $N=45$ ).

498

#### 499 6.3.2 *Oxygen and hydrogen isotopes*

500 Oxygen and hydrogen isotopes were analyzed to identify the source of ore fluid. The

501  $\delta^{18}\text{O}_{\text{mineral}}$ , calculated  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  and hydrogen isotope data of vein minerals are listed in Table 3. The  
502 calculated  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  data of vein minerals are presented in Fig. 11.

503 The mean homogenization temperature of fluid inclusions in vein hosted main ore stage  
504 cassiterite, 325°C, is used to calculate the  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  values of cassiterite with coexisting tourmaline.  
505 The  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  values calculated for scheelite and quartz from main ore stage are based on their  
506 respective mean fluid inclusion homogenization temperatures of 290° and 280°C. From the  
507 paragenetic sequence, the temperature of formation of wolframite ranges between 325° and 290°C.  
508 Thus 310°C was fixed to calculate the best estimate of  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  values of wolframite. The  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$   
509 values for quartz from main ore stage and sulfide stage are based on their respective mean fluid  
510 inclusion homogenization temperatures of 280° and 260°C, respectively.

511 The  $\delta^{18}\text{O}$  values of cassiterite from the Mawchi veins show a narrow range between 4.6  
512 and 5.7 ‰, with the exception of samples from the stockwork zone (nekhee-1 &-2) that possesses  
513 heavier  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  values of 8.3 and 8.4 ‰. The calculated  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  at 325°C is 7.3 to 11.1 ‰. The  
514  $\delta^{18}\text{O}$  values of tourmalines are distinctly homogeneous (10.4 - 10.8 ‰) and the  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  values  
515 range from 7.8 to 8.2 ‰ at an estimated temperature of 325°C. Wolframite has relatively lighter  
516  $\delta^{18}\text{O}$  values compared to cassiterite and tourmaline ranging from 2 to 4.3 ‰, with calculated  
517  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  values of 3.9 to 6.2 ‰. Scheelite has  $\delta^{18}\text{O}$  of 0.6 to 2.7 ‰, with calculated  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  values  
518 of 2.1 to 4.2 ‰. The  $\delta^{18}\text{O}$  values of quartz samples from various veins range from 11.5 to 13.4 ‰  
519 and their calculated  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  values range from 3 to 4.9 ‰.

Inclusion fluids were extracted to determine the  $\delta D$  value of fluids associated with the formation of cassiterite, wolframite and quartz in the Mawchi veins. The hydrogen isotope compositions of tourmaline were also analyzed, yielding  $\delta D_{\text{tourmaline}}$  values of -112 to -99 ‰. Four cassiterite samples from the main ore stage and one from the tourmaline-cassiterite stage were analyzed and their fluid inclusions gave  $\delta D_{\text{H}_2\text{O}}$  values from -86 to -51 ‰. Three wolframite samples from granite-hosted veins and two samples from metasediment-hosted veins gave  $\delta D_{\text{H}_2\text{O}}$  values of -87 to -61 ‰ and -121 to -118 ‰, respectively. Most of the quartz samples have dominant secondary fluid inclusions and, hence, two granite-hosted veins representing the main ore stage that have mostly primary fluid inclusions were selected for a hydrogen isotope study. They yield  $\delta D_{\text{H}_2\text{O}}$  values of -85.3 and -78.4 ‰.

## **7. Discussion**

### *7.1 Timing of magmatic-hydrothermal activity*

The zircon LA-ICP-MS U-Pb ages of  $42.72 \pm 0.94$  Ma and  $43.71 \pm 0.39$  Ma have been interpreted as the crystallization age of zircons and the emplacement age of the Mawchi pluton (Myint et al., 2017). The magmatic biotite Ar-Ar plateau age of  $41.50 \pm 0.16$  Ma is in agreement with the zircon U-Pb ages, however the nominally younger age of the Ar-Ar biotite age suggest that this date represents the timing of igneous cooling below the blocking temperature ( $\sim 350^\circ\text{C}$ ) for Ar in biotite. The hydrothermal muscovite in the tourmaline granite ( $40.14 \pm 0.14$  Ma) and in

the vein selvage ( $40.80 \pm 0.12$  Ma) yield similar Ar-Ar dates, indicating a broadly contemporaneous formation. The Ar-Ar ages agree with the paragenetic relationships. The biotite  $^{40}\text{Ar}/^{39}\text{Ar}$  plateau age represents cooling during the magmatic event, with the muscovite Ar-Ar plateau ages corresponding to the late stage hydrothermal alteration that formed after tourmalinization (i.e. tourmaline-cassiterite and main ore stages).

The molybdenite Re-Os age of  $42.43 \pm 1.21$  Ma also reveals the timing of mineralization and, it agrees with the U-Pb zircon age and biotite Ar-Ar date suggesting that ore deposition was essentially coeval with magmatic crystallization and fluid exsolution. The Re-Os age represents the main ore stage mineralization and muscovite Ar-Ar age expresses the late stage hydrothermal alteration.

## 7.2 Source of sulfur

The  $\delta^{34}\text{S}$  values of pyrite and arsenopyrite range from 2.9 to 6.2 ‰ and from 2.5 to 5.2 ‰ respectively, indicating a relatively homogeneous source, likely dominated by magmatic sulfur in tourmaline-cassiterite and main ore stages. In the later, sulfide stage, type-II chalcopyrite possesses  $\delta^{34}\text{S}$  values between 4.5 and 10.6 ‰ - suggesting a more reduced state than the main ore stage – compared to type-I chalcopyrite grains that exhibit  $\delta^{34}\text{S}$  values of 0.9 to 1.1‰. Sulfur isotope data for sphalerite shows evidence for a lighter source of S (-3.2 to 3.2 ‰), with galena and Pb-Bi sulfides yielding a narrower range in  $\delta^{34}\text{S}$  values (-1.3 to 2.7‰). The similarity of  $\delta^{34}\text{S}$

values, close to 0 ‰ in galena and Pb-Bi sulfides, may indicate that Pb and Bi were transported by a common fluid of purely magmatic origin. Apart from the heavier sulfur in type- I chalcopyrite, the principal feature of the sulfur isotope data is the systematic decrease from around 5-5.4 ‰ to 0-2 ‰ in  $\delta^{34}\text{S}$  with paragenetic sequence (Fig. 10). The  $\delta^{34}\text{S}$  of pyrite-I (5 - 5.4 ‰) in tourmaline-cassiterite stage also coincides with that for crustal-derived granite (Kohut and Recio, 2002; Cao et al., 2016) suggesting that the sulfur in the tourmaline-cassiterite and main ore stages is mainly crustally-derived. The crustal or heavier sulfur isotopic components appear to become progressively lighter during the progressive deposition of the sulfides in the sulfide mineral paragenesis.

Assuming ranges of depositional temperature of 340° to 260°C for main ore stage, possible  $\delta^{34}\text{S}_{\text{H}_2\text{S}}$  values of pyrite-II are 2.3 to 5.0 ‰ (using the data of Omoto and Rye, 1979), and with the mean of all sulfide  $\delta^{34}\text{S}$  being  $2.9 \pm 2.9$  ‰, suggesting that the overall system is dominated by magmatic sulfur (Omoto and Rye, 1979; Wagner et al., 2005; Wagner et al., 2009). Since sulfide stage chalcopyrite occurs after the main ore stage deposition, a temperature range of 260° to 230° C is assumed for type-I and -II chalcopyrite. According to Li and Liu (2006), the  $\delta^{34}\text{S}_{\text{H}_2\text{S}}$  calculation factor for chalcopyrite is constant for these temperatures. Type-I and -II chalcopyrite have the  $\delta^{34}\text{S}_{\text{H}_2\text{S}}$  values of 0.7 - 0.9 ‰ and 4.3 - 10.4 ‰, respectively. Estimating the precipitation temperature of galena and sphalerite as 200°C, the  $\delta^{34}\text{S}_{\text{H}_2\text{S}}$  values of galena will be 1.6 to 4.7‰ at 200°C, whereas those of sphalerite range from -3.6 to 2.8‰ (using the data of Li and Liu, 2006).

Thus, the source of sulfur for the whole mineralization was uniform and sulfur in the fluid was dominantly H<sub>2</sub>S (Shelton et al., 1986; Shelton et al., 1987). The heavier sulfur in the type-I chalcopyrite implies an extraneous sulfur source, probably derived from the country rocks (Kontak, 1990).

The  $\delta^{34}\text{S}_{\text{H}_2\text{S}}$  values of Panasqueira in Portugal and Pasto Bueno in Peru (-4 to 3 ‰ and -2 to 0 ‰, respectively) indicate a magmatic source (Landis and Rye, 1974; Kelly and Rye, 1979; Campbell, 1987). Except for type-I chalcopyrite, the  $\delta^{34}\text{S}_{\text{H}_2\text{S}}$  values of Mawchi (-3.6 to 5 ‰) are the same as those of Panasqueira and Pasto Bueno. The heavier sulfur at Mawchi ( $\delta^{34}\text{S}_{\text{H}_2\text{S}}$  4.3 to 10.4 ‰) is nearly the same value as San Cristobal ( $\delta^{34}\text{S}_{\text{H}_2\text{S}}$  2 to 9 ‰) of Peru, which also has a major sulfide stage, and magmatic sulfur contaminated with sulfur of sedimentary origin (Campbell, 1987). The source of sulfur for Mawchi is interpreted to be predominantly of magmatic origin, with heavier sulfur in type-I chalcopyrite being probably derived from the country rocks which surround the granite body.

### 7.3 Source of ore fluid

The Sn-W mineralization of Mawchi formed by magmatic-hydrothermal fluids of moderate temperature and low to moderate salinity, very similar to those of Wagone (Myint et al., 2017), Lailishan (Shi et al., 1989 in Cao et al., 2017), Pilok (Linnen and Williams-Jones, 1988), Panasqueira Sn-W deposit, Portugal (Kelly and Rye, 1979), and of the ore forming stage of San

596 Rafael Sn-Cu deposit (Wagner et al., 2009) and Pasto Bueno W base-metal deposit (Landis and  
597 Rye, 1974), both in Peru. The ore fluids of the Lailishan, Pilok, Panasqueira, San Rafael and Pasto  
598 Bueno were interpreted as a mixture of magmatic and meteoric waters. The early barren stages at  
599 San Rafael and Pasto Bueno are characterized by hot ( $>380^{\circ}\text{C}$ ) and hypersaline ( $>34$  wt. % NaCl  
600 equiv) fluid that can be interpreted as purely magmatic. However, fluids of magmatic origin  
601 responsible for the Cleveland tin deposit of Australia (7-14 wt. % NaCl equiv; Collins, 1981) and  
602 the Baiganhu District of China (10 - 14 wt. % NaCl equiv; Gao et al., 2014) are not hypersaline.  
603 Thus, the early stage, moderate salinity, fluids at Mawchi can be considered as essentially  
604 magmatic in origin. By contrast, the later low salinity fluid may have evolved by mixing with  
605 either a meteoric or a metamorphic fluid, which is also supported by  $\delta^{18}\text{O}$  and  $\delta\text{D}$  data.

606 The calculated  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  data of cassiterite (7.3 to 8.4 ‰) and tourmaline (7.8 to 8.2 ‰) are  
607 consistent with a magmatic source of ore fluid (Taylor, 1974, 1979). At the lower temperature  
608 stages,  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  data of wolframite, scheelite and quartz (3.9 to 6.2 ‰, 2.1 to 4.2 ‰ and 3 to 4.9  
609 ‰, respectively) are close to being 5 ‰, lower than cassiterite and tourmaline. These lower  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$   
610 values are again consistent with the view that the ore fluid may have mixed with another source of  
611 water, likely of meteoric origin.

612 The oxygen isotope compositional range in cassiterite is small,  $5 \pm 0.4$  ‰, whereas the  
613 hydrogen isotope composition ranges from -51 to -98 ‰ (Fig. 12). The fluid with distinctive  
614 vertical pattern can have resulted from contamination by a highly evolved meteoric water



(Campbell et al., 1984). The fluid inclusions from main ore stage vein minerals (cassiterite, wolframite, quartz) yield  $\delta D_{H_2O}$  values from -87 to -51 ‰ implying that the origin of ore fluid is magmatic, as suggested by sulfur. However, the calculated  $\delta D_{H_2O}$  value ( $\sim -120$ ‰) of some mineral species (wolframite, tourmaline) is low enough to suggest meteoric contamination in the hydrothermal fluids responsible for the late stage of vein deposition. Although some samples are located in the region of metamorphic water (-65‰ to -20‰ of  $\delta D_{H_2O}$  and +5‰ to +25‰ of  $\delta^{18}O_{H_2O}$ ), the uniform  $\delta^{18}O_{H_2O}$  values in same mineral species suggest minor involvement of metamorphic water in the origin of Mawchi ore fluids.

The ore forming fluids at Mawchi are a mixture of magmatic and meteoric waters, which implies a water-rock exchange model. The pattern of the fluid compositions, nearly constant  $\delta^{18}O_{H_2O}$  with a wide range of  $\delta D_{H_2O}$ , is very similar to the pattern at Pasto Bueno (Landis and Rye, 1974) and San Cristobal, Peru (Campbell et al., 1984; Beuchat et al., 2004) and Panasqueira, Portugal (Kelly and Rye, 1979). This pattern represents an exchange of meteoric waters with the host rocks and the igneous pluton at very low water to rock ratios ( $<0.05$ ) (Campbell et al., 1984; Shelton et al., 1987).

#### *7.4 Ore genesis of Sn-W mineralization in WGP (and TBGP): metallogenic epochs and tectonic settings*

Geochronological data constrain Sn-W mineralization of the granites in the WGP to the

period between the Cretaceous and Eocene. Early Cretaceous (128-121 Ma) granitic rocks in the WGP are represented by I-type magmatism which likely originated from subducted oceanic crust (Barley et al., 2003; Searle et al., 2007) and lack associated with Sn-W mineralization. The oldest tin bearing pegmatite of the WGP has been dated from Mawpalaw Taung at  $106.8 \pm 1.6$  Ma (Paik and Zaw, 2014). Younger Cretaceous tin granites from the WGP have been identified at Xiaolonghe (73 Ma; Chen et al., 2015) and also at Pilok (72 Ma Ar-Ar biotite date; Charusiri et al., 1993). U-Pb zircon data yielded  $61.44 \pm 0.62$  Ma and  $61.38 \pm 0.54$  Ma for the Hermyingyi granite and Wagone granirespectively (Li et al., under review), collectively suggesting that wide-spread emplacement of tin granites in the WGP occurred in the late Cretaceous and Paleocene. Biotite K-Ar dating yielded an Earliest Eocene age of  $55 \pm 1$  Ma for the Padatchaung granite (Brook and Snelling, 1976) and, U-Pb zircon dating yielded for Lailishan,  $52.7 \pm 0.3 - 53.0 \pm 0.4$  Ma (Chen et al., 2015) and for Yadanabon,  $50.3 \pm 0.6$  Ma (Gardiner et al., 2016), respectively. The 42 Ma LA-ICP-MS zircon U-Pb ages coupled with  $41.5 \pm 0.2$  Ma biotite Ar-Ar age of Mawchi granites affirms Late Eocene tin granite emplacement in WGP and is the youngest age for tin mineralization in the province. In the MMB west of the SB, S-type granites range from 72 Ma to 44 Ma and lack significant tin mineralization.

The tin granites of WGP were derived from the partial melting of the crust (Gardiner et al., 2016; Myint et al., 2017a&b) and the emplacement of these granites has been variously related to the Cretaceous–Paleogene subduction- and Himalayan orogeny-related collision (Gardiner et al.,

2015), back-arc extension, or post-collisional or syncollisional extension related granitic magmatism (Chen et al., 2015; Jiang et al., 2017; Myint et al. 2017a; Chen et al., 2018). Interpretation of some WGP granites as A-2 type and derived from partial melts of Proterozoic intrusions is also proposed (Jiang et al., 2017). Triassic to Jurassic CGP tin granites related to the Sibumasu - Sukhothai arc collision form the older Asian tin belt while collisional and post-collisional Cretaceous to Eocene WGP tin bearing granites formed a younger belt before, during and after the India – Asia collision. Assimilation of the crust (eg. SB in WGP; Kontum massif in CGP) led to the production of highly siliceous and highly fractionated tin granites (Chen et al., 2018; Ng et al., 2015a; Myint et al. 2017a&b).

662

## 663 **8. Conclusions**

664 The Mawchi Sn-W polymetallic deposit consists of a suite of Sn-W sulfide veins with  
665 minor skarn and stockworks. Three stages of ore deposition were recognized, namely (1) the  
666 tourmaline-cassiterite stage, (2) the main ore stage and (3) the sulfide stage. Tourmaline,  
667 cassiterite, muscovite and pyrite-I are minerals that formed early in the hydrothermal activity and  
668 their deposition occurred in the granite body. Wolframite, scheelite, arsenopyrite, pyrite-II,  
669 molybdenite, fluorite, chlorite, danalite and calcite formed during the main ore stage together with  
670 the products formed earlier. Sulfide minerals such as chalcopyrite, sphalerite, galena and Pb-Bi  
671 sulfides were deposited during the termination of the mineralization with hydrothermal activity

continuing substantially after deposition of the metals.

Our study provides the first isotopic age determination of Sn-W mineralization in Myanmar. Combining new Ar-Ar geochronological data with previous U-Pb zircon data constrains the magmatic-hydrothermal activity at Mawchi to a short time span that implies successive ore deposition and magmatism. The biotite granite at Mawchi was emplaced and crystallized at 43-42 Ma with coeval magmatic-hydrothermal activity resulting in precipitation of the mineralization paragenesis over a period of around 1-2 million years. Successive hydrothermal activity transformed part of the biotite granite to tourmaline granite by adding tourmaline and muscovite of hydrothermal origin. At the same time, cassiterite deposition together with tourmalinization took place in veins emplaced along the fractures in the granite. Muscovite-sericite Ar-Ar data of vein quartz constrains the late stage of mineralization at Mawchi to around 41 Ma.

The proposed deposit model is based on the stable isotope and fluid inclusion data. Homogenization temperatures range from 175° to 340°C for the Mawchi vein system. This relatively moderate temperature, moderate salinity brine ( $\leq 15$  equiv. wt. % NaCl) ore fluid was originally produced by magmatic water and probably mixed with meteoric water later during circulation through adjacent metasediments. However, the sulfur was relatively uniform throughout the hydrothermal activity, and likely purely magmatic derived except for the sulfur in type-I chalcopyrite which was contaminated with sulfur from country rocks. In conclusion, the hydrothermal fluids of Mawchi were derived from the mixing of magmatic and meteoric waters.

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## Figure captions

**Fig. 1 a** Granitoid belts of Myanmar and its adjacent area showing the occurrences of granites and tectonic framework (modified after Cobbing et al., 1992; Mitchell et al., 2007; Sone and Metcalfe, 2008). **b** Regional geologic setting of the Mawchi area (modified after Bender, 1983; DGSE, 2008)

**Fig. 2 a** Geological map of Mawchi Sn-W district showing the localities of Sn-W mineral systems. **b** Deposit geological map and vertical profile of Mawchi mine (modified after Mawchi mine project map)

**Fig. 3 a** quartz vein cutting the granite body: intensive tourmalinization along the contact between vein and host rock. **b** small sheeted veins intruding the metasediments. **c** danalite(da) boundary vein hosted by the granite(gr). **d** later formed quartz-tourmaline barren vein. **e** vein boundary segregates of wolframite (bladed) and cassiterite. **f** fracture filling of quartz and cassiterite. **g** hand specimen collected from massive band of cassiterite with minor amount of tourmaline and clay minerals. **h** hydrothermal muscovite (mu) in fractures of tourmaline (tu)

**Fig. 4** Paragenetic sequence of Mawchi Sn-W polymetallic mineralization

**Fig. 5** Photomicrographs (**a-f, h**) and hand specimen (**g**) showing the mineral distribution of Mawchi veins. **a** cassiterite (ca) cross cut by wolframite (wf) and, replaced and veined by pyrite-II (py-II). **b** kinked molybdenite (mo) replaced by pyrite-II (py-II): small chalcopyrite blebs (cp) replaced in pyrite. **c** zoned cassiterite crystal. **d** cassiterite veined and replaced by wolframite: wolframite replaced by pyrite-II. **e** cleavage planes of wolframite replaced by scheelite (sh). **f** pyrite-I cubes replaced by arsenopyrite (ap) and pyrite-II which is veined by chalcopyrite (cp). **g** scheelite crystal (milky white) associated with cassiterite and wolframite. **h** tourmaline inclusions in scheelite crystal

**Fig. 6 a** interstitial filling of pyrite-II into wolframite grains. **b** greenish fluorite and cassiterite in the vein. **c** parallel flow of wolframoixiolites. **d** pyrite-II replaced by sphalerite (sp) and

chalcopyrite (cp); sphalerite hosts the blebs of chalcopyrite and stannite (st) forming as exsolution pattern. **e** type-II chalcopyrite (cp) and pyrite-II, as infill veinlets along fractures of arsenopyrite (ap). **f** pyrite-II (py-II) replaced by type-I chalcopyrite (cp). **g** sphalerite (sp) replaced and veined by galena. **h** pyrite-II (py-II) and chalcopyrite (cp) replaced by the Pb-Bi sulfide

**Fig. 7**  $^{40}\text{Ar}/^{39}\text{Ar}$  plateau and inverse isochron ages ( $2\sigma$ ) for micas from Mawchi

**Fig. 8** Histograms of homogenization temperatures of fluid inclusions in vein minerals of the Mawchi mine

**Fig. 9** Salinity variation of fluid inclusions in vein minerals from the Mawchi mine

**Fig. 10** The  $\delta^{34}\text{S}$  values of sulfides from the Mawchi mine. Numbers of analyses in parentheses

**Fig. 11** The calculated  $\delta^{18}\text{O}_{\text{H}_2\text{O}}$  values of ore fluid from the Mawchi mine

**Fig. 12** Plot of  $\delta^{18}\text{O}$ - $\delta\text{D}$  of vein minerals from the Mawchi Sn-W polymetallic deposit (modified from Taylor, 1974)

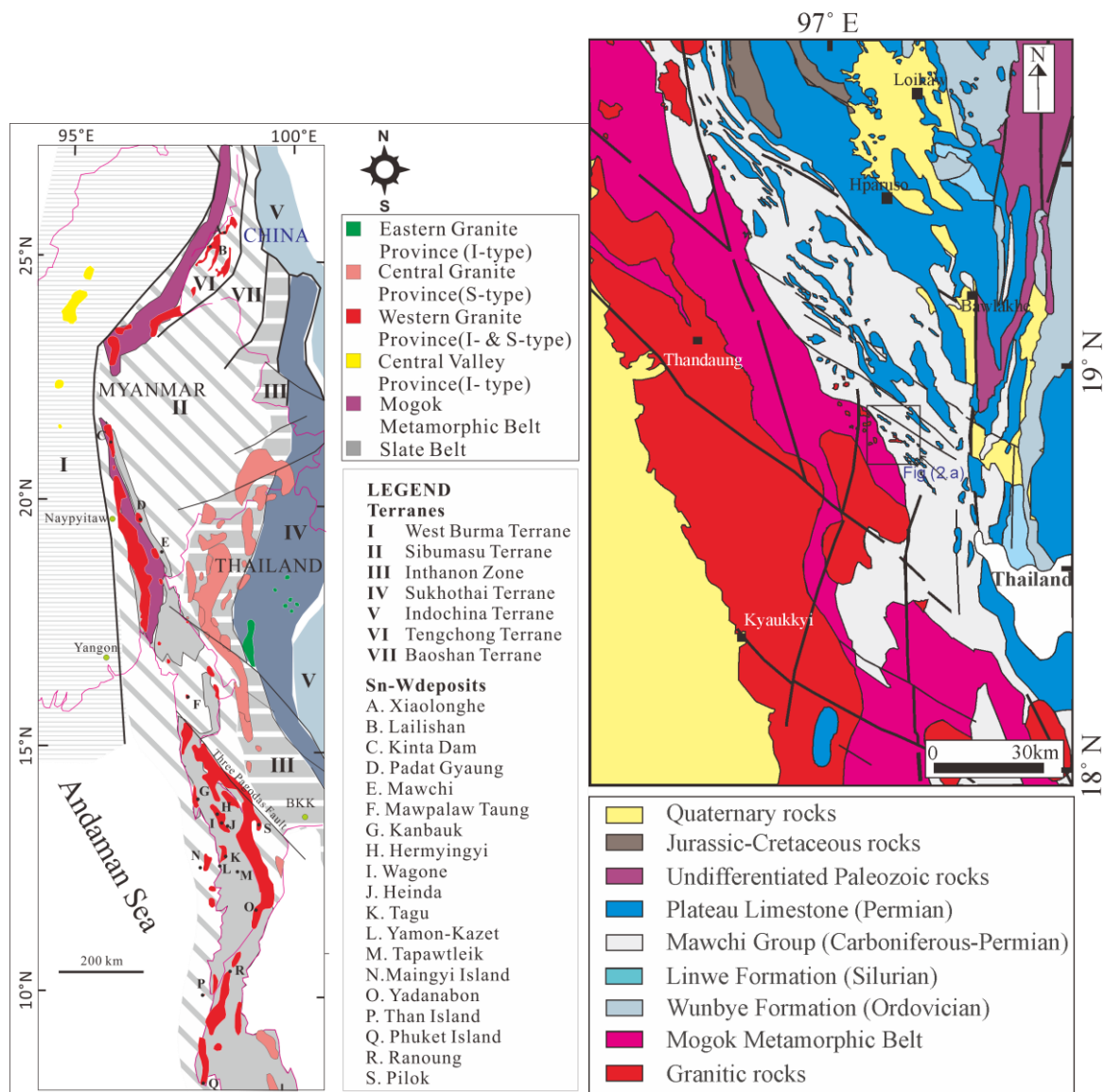


Fig (1)



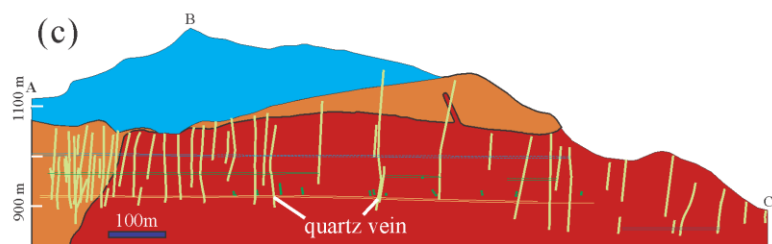
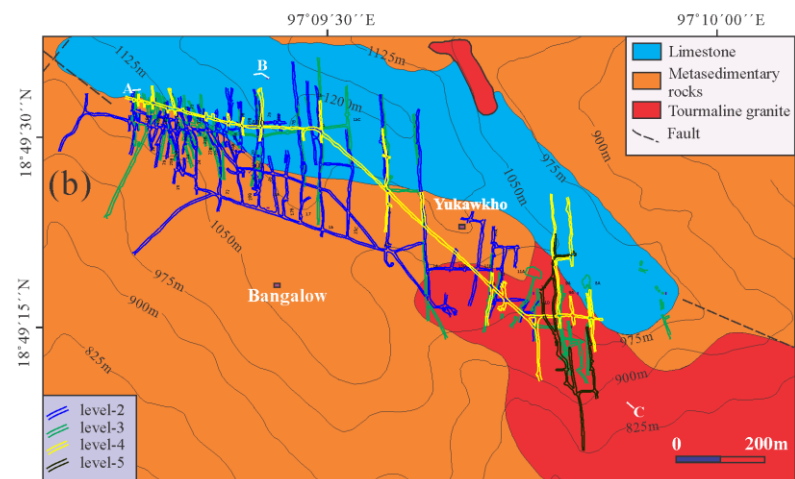
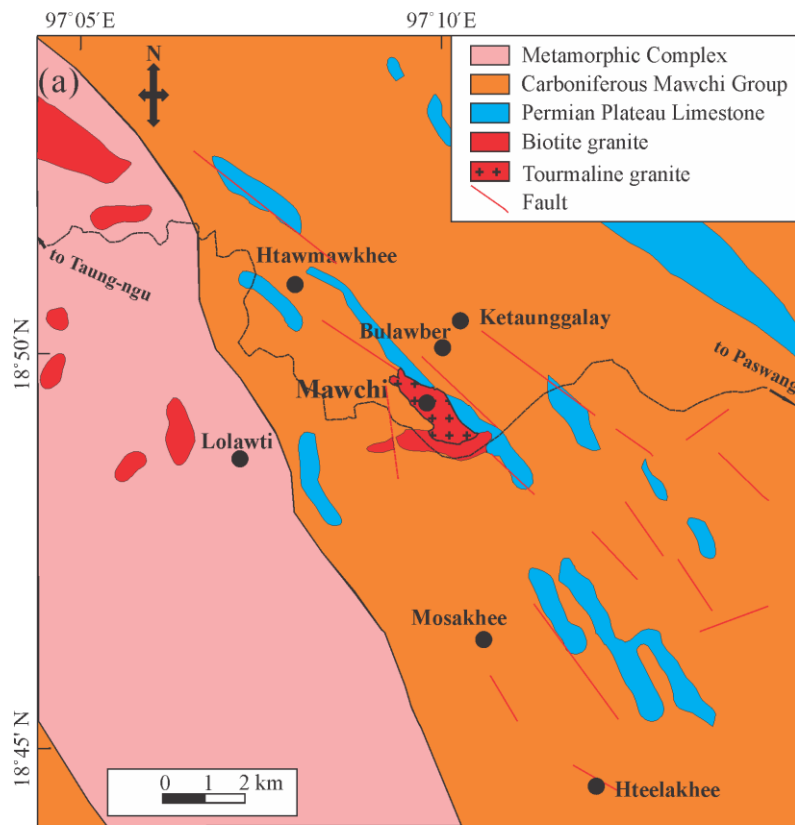


Fig (2)



Fig (3)

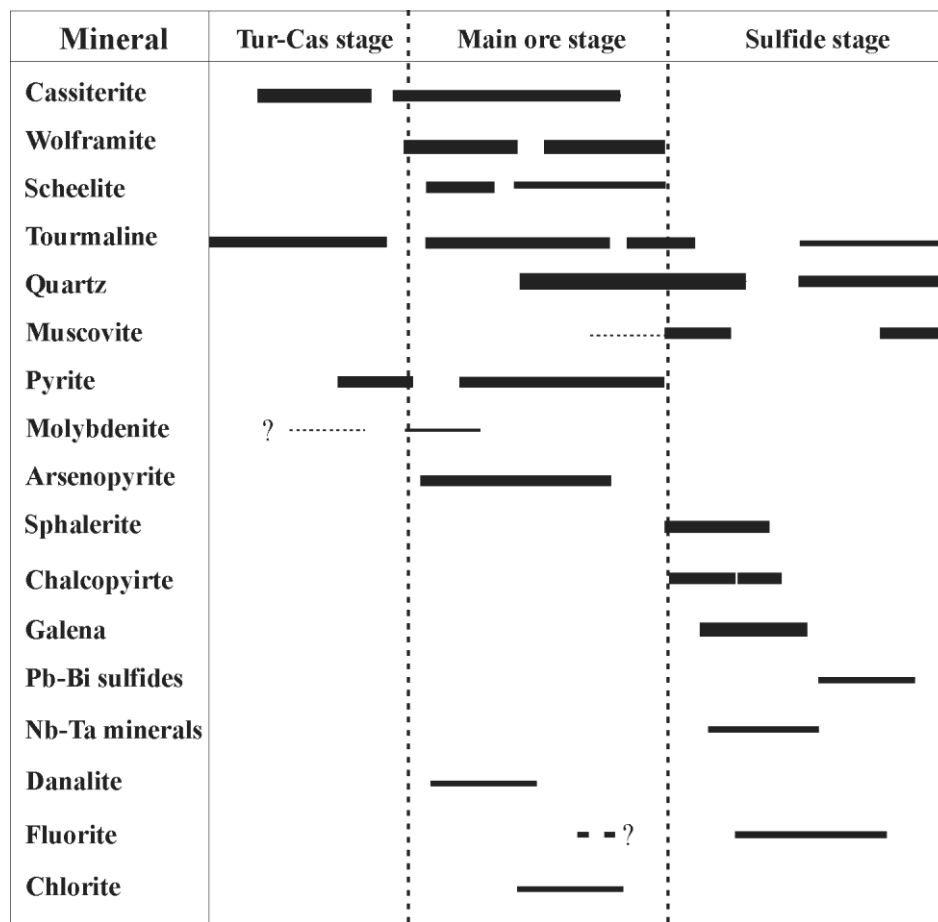


Fig (4)



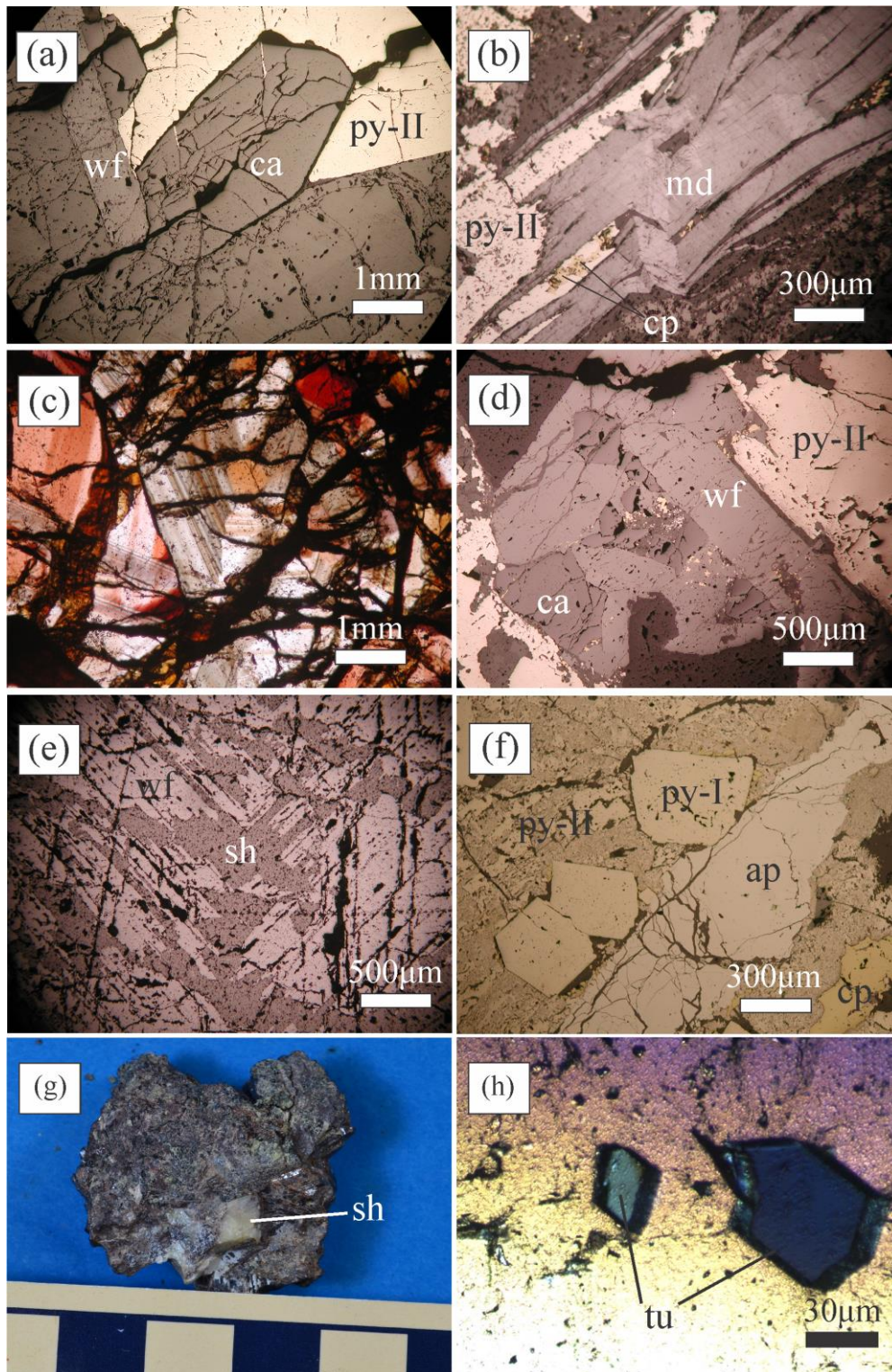


Fig (5)



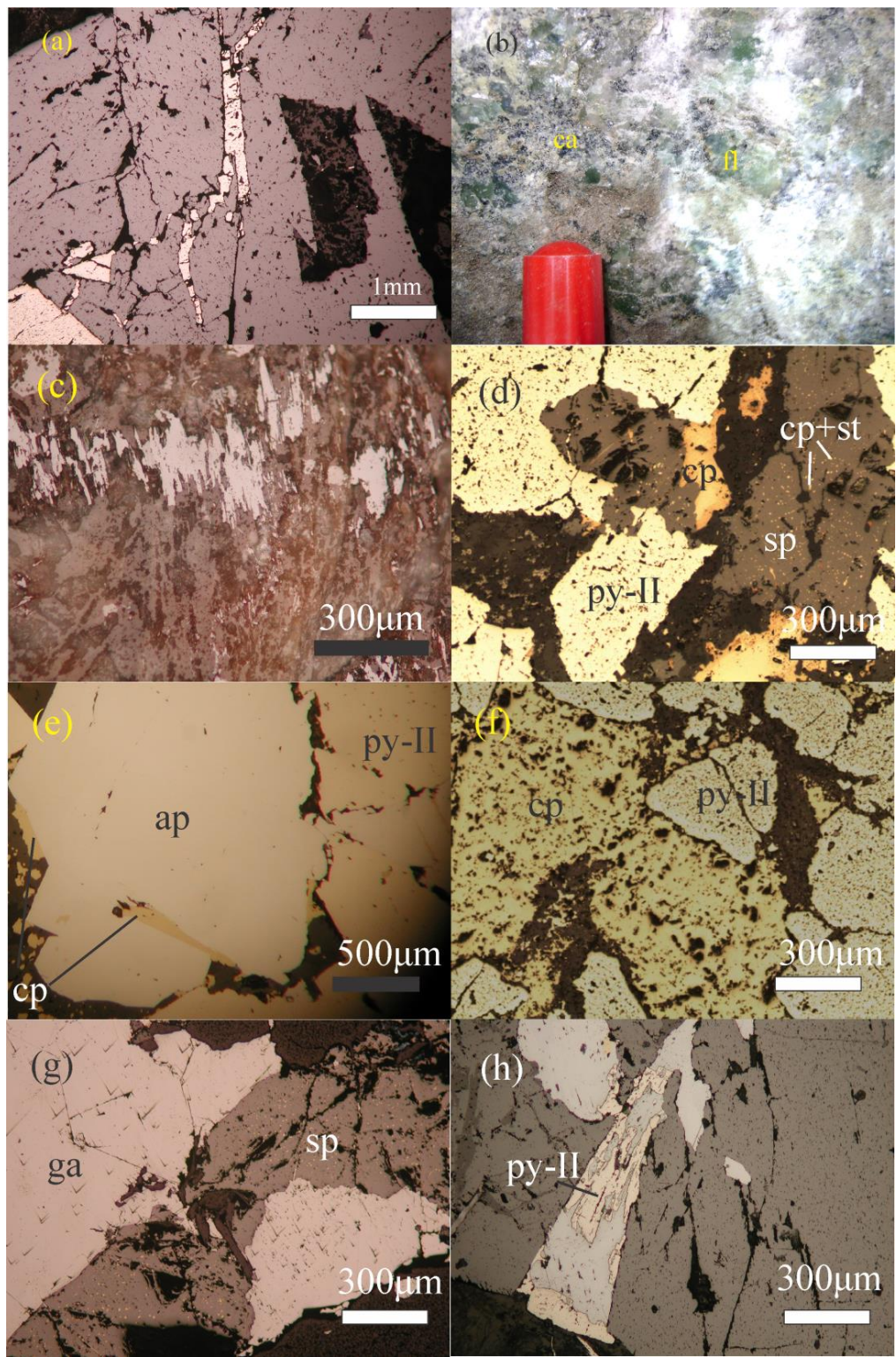


Fig (6)



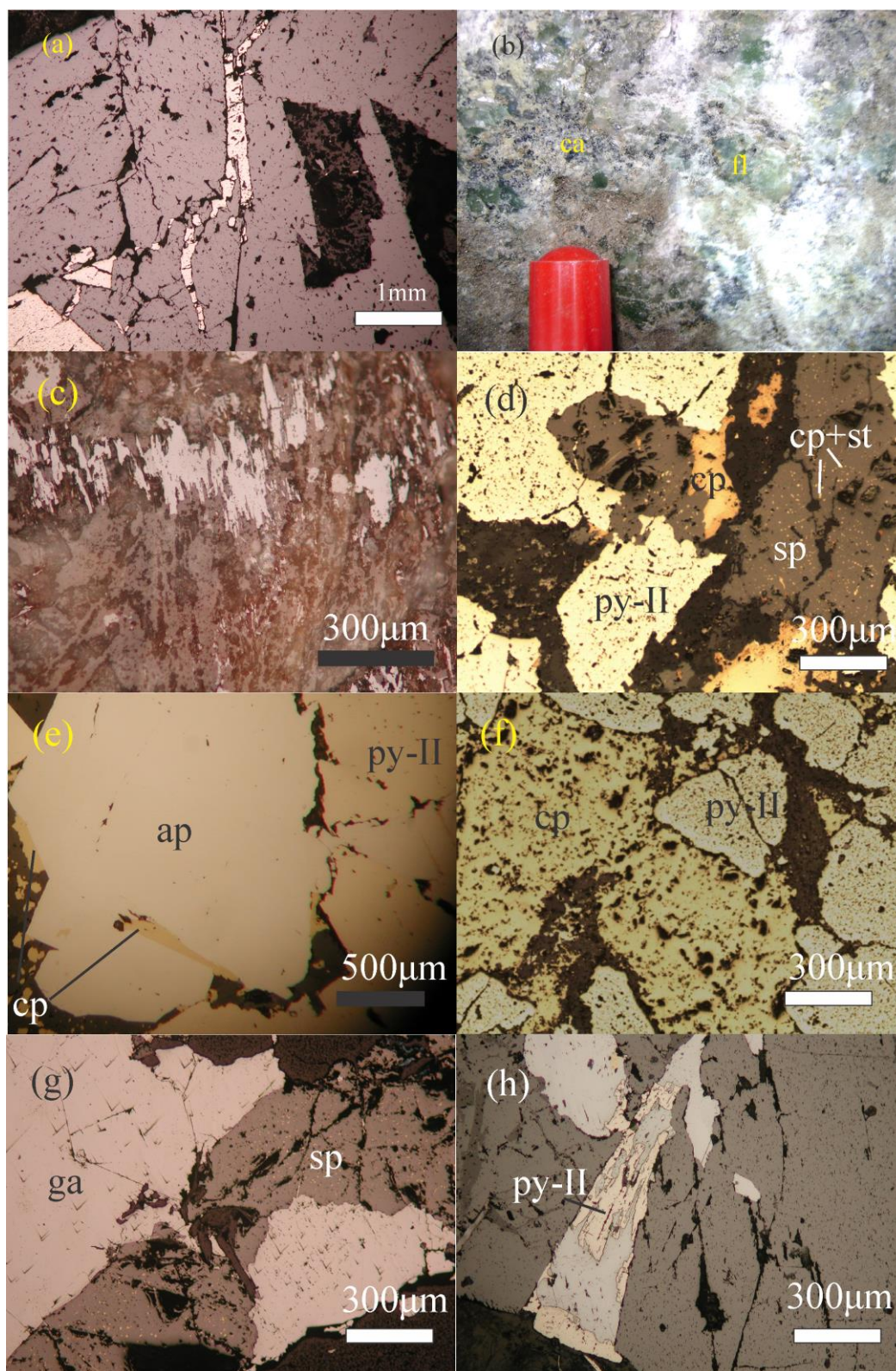
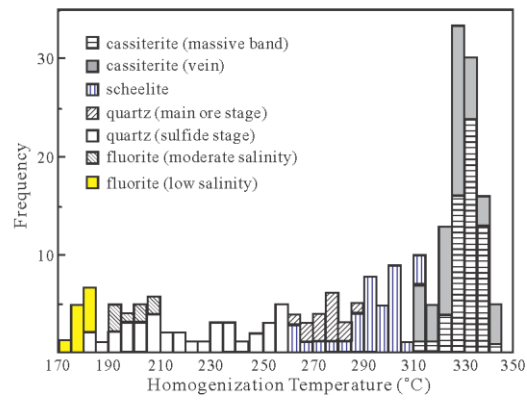
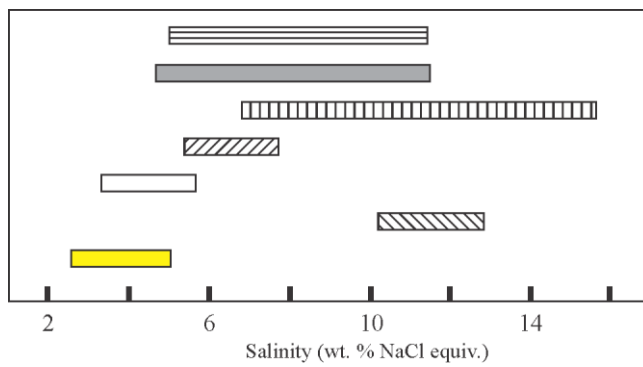


Fig (7)



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956 Fig (8)



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958 Fig (9)

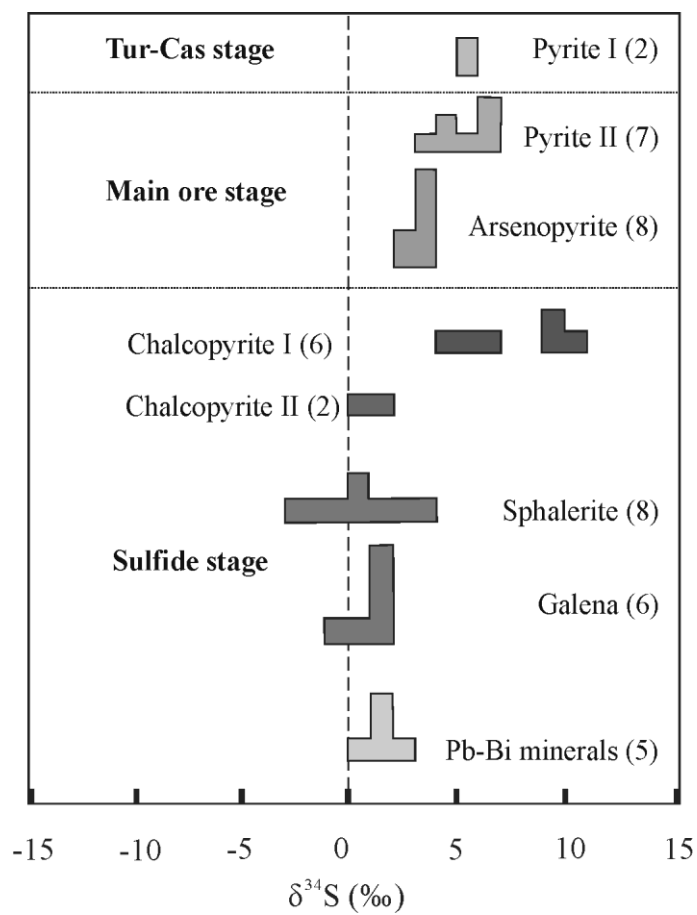


Fig (10)

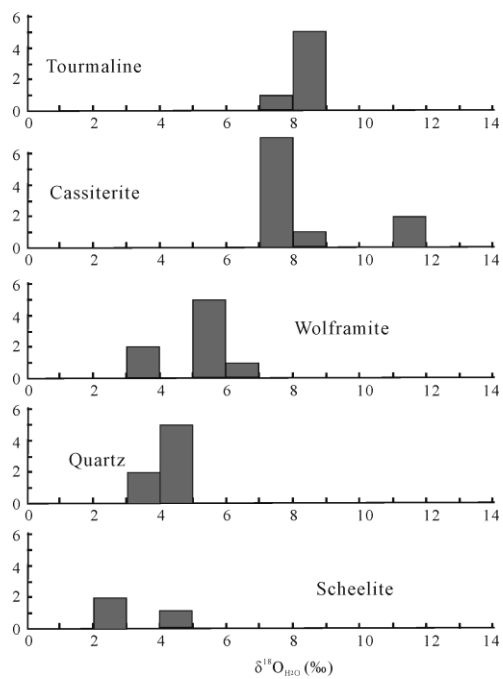
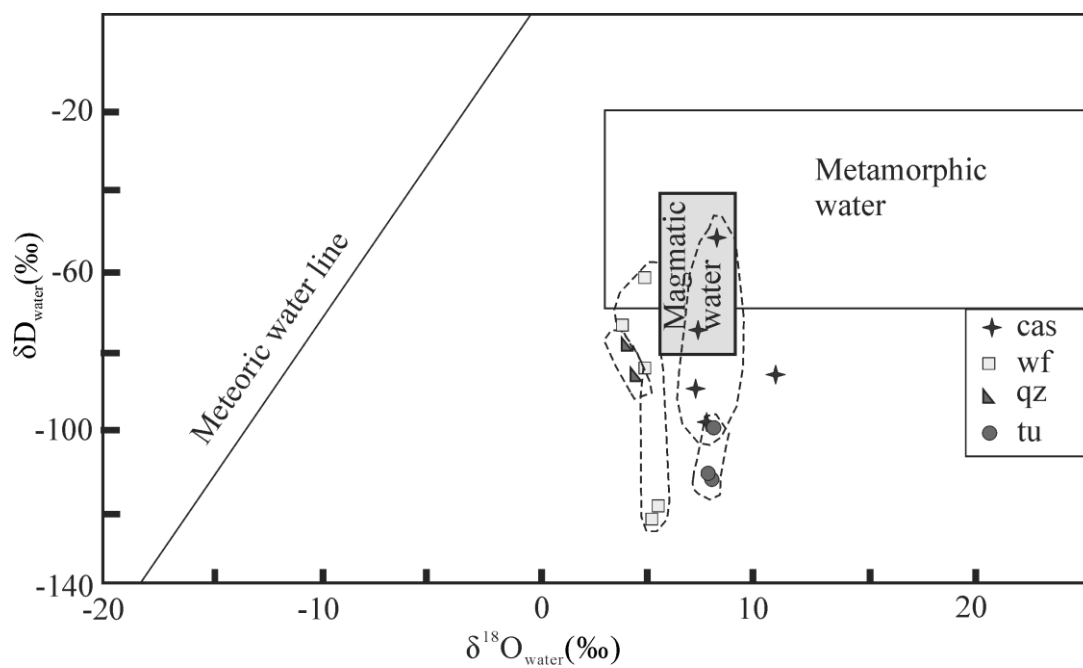


Fig (11)





963

964 Fig (12)