

1 **Better Together: Water Treatment Residual and Poor-Quality Compost**

2 **Improves Sandy Soil Fertility**

3 C.E. Clarke^{1*}, W. Stone^{2,3}, A.G. Hardie¹, J.N. Quinton⁴, L. Blake⁵ and K.L. Johnson⁵

4 ¹ Department of Soil Science, Stellenbosch University, Stellenbosch, South Africa

5 ² Water Institute, Stellenbosch University, Stellenbosch, South Africa

6 ³ Department of Microbiology, Stellenbosch University, Stellenbosch, South Africa

7 ⁴ Lancaster Environment Centre, Lancaster University, Lancaster LA1 4YQ, UK

8 ⁵ Department of Engineering, Durham University, Durham, DH1 3LE, UK

9 * Corresponding Author C.E. Clarke cdowding@sun.ac.za

10

11 **Abstract**

12 Water treatment residual (WTR) is an under-utilized clean water industry byproduct, generally
13 disposed to landfill. This study assesses the benefits and risks of ferric-WTR as a soil
14 amendment or co-amendment for plant growth in a nutrient-poor sandy soil. A 12-week pot
15 trial tested the efficacy of WTR and a locally available, low-quality, municipal compost as
16 single (1, 5, 12.5% dry mass) and co-amended treatments (1:1 WTR:compost ratio, at 2%, 10%
17 and 25%) on wheat growth in a sandy soil. The low total N content of the compost and low
18 WTR P and K contents resulted in significantly lower (up to 50% lower; $p < 0.05$) plant biomass
19 in single amendments compared to the control, while the highest co-amendment produced
20 significantly higher plant biomass (33% higher; $p < 0.05$) than the control. This positive co-
21 amendment effect on plant growth is attributed to balanced nutrient provision, with P and K
22 from the compost and N from the WTR. Foliar micronutrient and Al levels showed no toxic
23 accumulation, and co-amended foliar Mn levels increased from near deficient (20 mg/kg) to
24 sufficient (50 mg/kg). Total WTR metals were well below maximum land application
25 concentrations (USDA). Trace element bioavailability remained the same (Ni, Cu, Hg) or

26 significantly decreased (B, Al, Cr, Mn, Fe, Zn, As, Cd; $p < 0.05$) during the pot trial. These
27 results suggest, within this context, that WTR is a safe soil improvement technology and can
28 be combined with poor quality local composts to improve yields in sandy soil.

29 Keywords: Fe-WTR, Waste Recycling, drinking water purification, Arenosol

30

31 **1. Introduction**

32 Water treatment residual (WTR) is a global byproduct of drinking water treatment which
33 purifies raw water to produce drinking water for municipalities. Basibuyuk and Kalat (2004)
34 reported that several million tons of WTR are produced in Europe every year, with production
35 estimated to double within the next decade. In Africa, WTR production is also set to increase
36 due to an ~~increasing~~ growing population requiring increasing access to clean drinking water.
37 WTR is most commonly disposed in landfill, both globally (Basta et al., 2000) and within
38 South Africa (Herselman, 2013). Alternative uses of this waste byproduct are of global interest
39 to water companies, many of which are looking towards zero waste strategies to reduce costs
40 and contribute to the United Nations Sustainable Development Goals (SDG 12, Responsible
41 Production and Consumption; UN, 2016).

42

43 WTR consists of flocculating agents (ferric and aluminium oxyhydroxides), de-watering agents
44 (polyelectrolytes), activated carbon and flocculated material from the catchment dams,
45 including clay particles, microbes and dissolved organic matter (Matilainen et al., 2010). Given
46 the soil-like composition of WTR, land application is an important potential disposal option.
47 The implications of land application have been well researched (Ippolito et al., 2011). One of
48 the major problems encountered with land application of WTR is the high P-fixation capacity
49 of the Fe and Al-oxyhydroxides (Elliot and Dempsey, 1991; Ippolito et al., 2003; Norris and
50 Titshall, 2012). Addition of WTR to soils results in yield loss and P-deficiency symptoms in

51 maize (Rengasamy et al., 1980), lettuce (Elliott and Singer, 1988) and sorghum-Sudan grass
52 (Heil and Barbarick, 1989). Another problematic factor is the high concentrations of
53 bioavailable Al and Mn in WTR (Ippolito et al., 2011; Novak et al., 2007; Titshall and Hughes,
54 2005), which may result in phytotoxic conditions.

55

56 Compost is commonly used to improve both chemical (fertility and phytotoxicity) and physical
57 (aggregation and water holding capacity) properties of soils. It is well-established that compost
58 addition can reduce the P sorption capacity of Al and Fe oxides in soils (Havlin et al., 2005),
59 yet the use of WTR and compost as a co-amendment is not well-researched. Hsu and Hseu
60 (2011) looked at the co-addition of a good quality (C:N ratio = 20, total N = 3.9%) compost
61 with Al-WTR. In contrast to the above-mentioned studies, they observed an increase in the
62 growth of Bahia grass with Al-WTR added as a single amendment. Co-aAddition of the
63 compost improved growth but not significantly. Compost also increased plant available P in
64 co-amended treatments, although plant tissue P was not significantly affected. In many small-
65 scale farming systems in Africa, compost quality is often poor, with high C:N ratios and
66 typically low total N contents (Vanlauwe and Giller, 2006). Our research findings showcase
67 the first use of a ferric-WTR and poor quality compost co-amendment as a cost-effective soil
68 improvement technology to improve crop productivity through balanced nutrient provision, in
69 sandy soils from Southern Africa.

70

71 Sandy soils are ubiquitous throughout Africa, where despite their low fertility and low water
72 holding capacity, they support crop production in small-scale dryland systems. Dryland
73 farming in sandy soil has a high risk of crop failure due to crop susceptibility to water stress,
74 which is exacerbated in nutrient-deprived plants (Steynberg et al., 1989). Infertile soils affect
75 both plant growth and human nutrition. For example, communities solely subsisting on crops

76 grown in sandy soils in Maputoland, South Africa, had elevated incidences of dwarfism and
77 endemic osteoarthritis due to nutrient deficiencies (Ceruti et al., 2003).

78

79 The Cape Flats region, just outside Cape Town, has nutrient poor, sandy soils of aeolian origin.
80 The area is predominantly occupied by low-income communities and hosts the largest informal
81 settlement in the Western Cape (Statistics South Africa, 2016). Residential urban agriculture
82 is uncommon, mainly due to lack of space, but also due to the nutrient poor soils and restricted
83 access to irrigation water. However, in a community where unemployment levels are over
84 30% (Western Cape Government, 2017), backyard vegetable gardens can provide fresh
85 produce to supplement the common maize staple. Thus, any improvement to the soils in terms
86 of increased water holding capacity and nutrient provision, could stimulate backyard
87 gardening, impacting community health and wellbeing. The Faure raw water treatment works
88 is the main supplier of ~~drinking-potable~~ water to the City of Cape Town, producing
89 approximately 14 000 tons Fe-WTR per year (personal communication, City of Cape Town
90 Municipality, 2018), and lies physically close to the Cape Flats region. Currently Faure WTR
91 is transported approximately 50 km to a local landfill site. Therefore, if the Faure WTR could
92 be used to improve the Cape Flats soil it would be beneficial for both the municipality and the
93 local inhabitants. In this study we focus on the safety and plant response to WTR amendments
94 and compare the effect of WTR and a typical low quality compost added separately and as a
95 co-amendment on plant yield, bioavailable metals and plant nutrient levels in typical sandy
96 Cape flats soil.

97 **2. Materials and methods**

98 **2.1 Sample collection and characterization**

99 Water Treatment Residual was obtained from Faure water treatment works, outside Cape
100 Town. The main storage dam, and the only reservoir supplying Faure at the time of sampling,

101 is the Theewaterskloof dam. The plant uses $\text{Fe}_2(\text{SO}_4)_3$, ~~lime~~ CaCO_3 , a chemical coagulant
102 (Praestol 2540, a copolymer of acrylamide and sodium acrylate) and varying amounts of
103 activated charcoal for odour control (Titshall and Hughes, 2005). The resulting WTR is a
104 mixture of ferric hydroxides, reservoir sediments, flocculated organic acids, coagulant and
105 activated carbon. Samples of WTR were collected on three dates - 28 February, 9 May and 15
106 May 2017. During this period the Western Cape was experiencing a severe drought, and
107 turbidity and odour levels were elevated due to the increased microbial blooms. This increased
108 coagulant and activated carbon use during water purification. The three individual samples
109 were air-dried (30°C, 1 month) before being crushed to pass through a 2 mm sieve. The three
110 individual samples were chemically analysed to assess elemental variation, before being
111 thoroughly combined for re-analysis and subsequent application in incubations, chemical
112 analyses, and plant trials.

113

114 The commercially available compost used in this study is made from municipal green waste
115 (chipped garden refuse) and was used and analysed without sieving. The total C and N content
116 of the compost was analysed on a milled subsample.

117

118 The sandy soil was collected from a fallow field outside Brackenfell (Western Cape). The
119 Quartzipsamment soils of this region are typical acid variants of the Cape Flats sands. These
120 sands are windblown marine deposits, that have been leached of all carbonates, have an
121 inherently low nutrient status and are mildly acidic (Schloms et al., 1983). The top 30 cm of
122 soil was collected, air-dried and passed through a 2 mm sieve before analysis. Details of the
123 basic characterization methods and statistical analysis are provided in the Supporting
124 Information.

125

2.1.1 Trace element content and availability

126 Trace elements (TE) were measured in i) aqua regia (USEPA method 3015a), and ii) NH_4NO_3
127 (representing bioavailable fraction) following the DIN 19730 procedure (Herselman, 2013).
128 Extracts, prepared in triplicate, were analysed for metals using ICP-MS with an Agilent 8800
129 QQQ ICP-MS.

130 **2.2 Pre-Trial Incubation Analyses**

131 Incubation profiles of pH, EC, Mn and P were assessed, to inform application rates. ~~Four~~ Six
132 application levels (0, 10, 25, 50, 75 and 100%) of (a) WTR and (b) a 1:1 WTR-Compost
133 mixture were added on a dry weight % basis to the soil. Each air-dried sample (50 g) was wet
134 to field water capacity, covered in parafilm to prevent moisture loss and incubated at room
135 temperature ($\pm 25^\circ\text{C}$) in duplicate for two weeks. Samples were regularly weighed to confirm
136 moisture retention. Samples were analyzed post-incubation for pH, EC, Mn and P as described
137 in Supplementary Materials-.

138 **2.3 Pot Trials**

139 Pot trials were set up to assess the impact of increasing application rates of WTR, compost and
140 the WTR-Compost (WTR-Comp) co-amendment on wheat growth and elemental
141 accumulation in nutrient-poor sandy soils. The application rates used were 0 (control), 1, 5
142 and 12.5% (w/w) for the single compost or WTR treatments and 0, 2, 10 and 25% (w/w) for
143 the 1:1 WTR-Comp co-amendment. All treatments were prepared in triplicate. Pots (5L) were
144 packed to a bulk density of 1500 kg/m^3 . Six wheat seeds (*Triticum aestivum L.*) per pot were
145 planted and thinned to 3 plants per pot after germination. Pots were weighed and watered twice
146 a week, maintaining field water capacity. Greenhouse pot placement was randomized and
147 randomly re-organized twice during the 3-month trial. Pots were fertilized using the wheat
148 recommendation of the Fertilizer Society of South Africa (FSSA, 2007) for Western Cape
149 sandy soils (N = 130, P = 50, K = 75, Ca = 40, Mg = 13 and S = 40 kg/ha). The 500 mL fertilizer
150 concentrate was added as three applications over the 3 month trial period.

151 **2.4 Post-Trial Analyses**

152 After 3 months of growth, the pot trial was terminated. The above-ground plant material was
153 harvested by cutting the plant at soil level. Roots were weighed after soil material was removed.
154 Plant material was oven-dried (60°C) overnight and weighed per pot. Total macro- and
155 micronutrients of the dried above-ground plant material were determined using the Kjeldahl
156 method (N), and acid digestion and ICP-MS (P, Ca, Mg, K, Na, Fe, B, Zn, Mn, Cu and Al;
157 Elsenberg Plant Laboratory).

158

159 Soil from the pots was sieved (2 mm) to remove roots ~~hairs~~ and air-dried. The NH_4NO_3
160 extractable metals (see Section 2.1), were measured on the pre- and post-trial soil mixtures.

161

162 **3 Results and Discussion**

163 **3.1 WTR, Compost and Soil Characterization**

164 The properties of the sandy soil, WTR and compost are given in Supplemental Table S1. The
165 soil is mildly acid ($\text{pH}_{\text{water}} = 6.5$), with very low EC ($64 \mu\text{S}/\text{cm}$), total C (0.6%) and total N
166 (0.04%). The P level in the soil (52 mg/kg) is above the 33 mg/kg recommended for most crops
167 (Mehlich, 1985). Bray II K levels in the soil are extremely low (9 mg/kg), falling well below
168 the recommended 50 mg/kg for winter wheat production (FSSA, 2007). The WTR has a neutral
169 pH in water (7.8) and low EC ($319 \mu\text{S}/\text{cm}$). The total C is 17%, which includes flocculated
170 dissolved organic C and the added activated carbon. The total N content of the WTR is 0.35%,
171 which is in the typical range for South African WTRs (0.02 – 0.52%), but lower than reported
172 for Faure WTR in 2005 (0.52%; Titshall and Hughes, (2005)). Thus, the severe drought had
173 not significantly increased the total N content of the WTR. The mineral N content (165 mg/kg)
174 of the WTR falls within the range of typical WTRs in South Africa (Titshall and Hughes, 2005)
175 and those reviewed by Ippolito et al. (2011). The Mehlich III P concentration in the WTR is

176 within the lower region of the range reported by Dayton and Basta (2001), between 1.6 and
177 54.4 mg/kg.

178

179 The compost has a slightly alkaline pH_{water} (7.8), very high EC (5410 $\mu\text{S}/\text{cm}$) and a relatively
180 low total C content (9.6%) for a compost. Despite an acceptable C:N ratio (25), the total N
181 content of the compost (0.38%) falls well below the 1% threshold recommended in composts
182 intended for fertilizer use (Barker, 1997). The mineral N content (7 mg/kg) of the compost is
183 also very low, falling short of that required to support crop growth (50-200 mg/kg; (Mulvaney,
184 1996)). On the other hand, the compost has ample plant available K and P (145 and 2944
185 mg/kg, respectively).

186 The aqua regia metal concentrations of the three Faure WTR samples collected at different
187 times are shown in Supplemental Table S2. Iron is the dominant metal (14-19%), with
188 substantial Al concentrations (5.3- 7.7%). Manganese is variable (0.05 – 0.29%) but lower than
189 the values reported by Titshall and Hughes (2005) for Faure WTR in 2005 (0.7 and 1.8%). The
190 source of Mn in the Faure WTR is anticipated to be from impurities in the ferric sulphate or
191 lime used during the water treatment process (Titshall and Hughes, 2005). The lower Mn values
192 measured in this study suggests that purer sources or lower quantities of these additives are
193 currently being used. With the exception of Mn, Zn and Ni, which were higher in summer
194 (February), the trace elements in the WTR do not differ substantially between sampling dates.
195 The metal concentrations of all samples are well below both the United States Environmental
196 Protection Agency (USEPA, 2000) and the more conservative South African guidelines
197 (Herselman, 2013) for the maximum allowable limits for land application.

198

199 The bioavailable metals (NH_4NO_3 extract) for the soil, composite WTR and compost are given
200 in Table 1. Prior to WTR land application in South Africa, receiving soils must be analysed for

201 bioavailable metals to assess the soils' suitability for receiving waste (Herselman, 2013). The
202 Cape Flats sand has metal concentrations far below the maximum limit permitted for soils that
203 will receive WTR (Herselman, 2013). The pure WTR had slightly elevated bioavailable Mn
204 concentrations (17 mg/kg) however, there are no plant micronutrient thresholds for NH_4NO_3
205 extracts, so Mehlich III extracts of the soil, compost and WTR were conducted. The Mn
206 concentrations in the Mehlich III extracts were 2.2 (± 0.2), 22.9 (± 0.7) and 124.0 (± 2.6) mg/kg
207 for the soil, compost and WTR, respectively. The available Mn in the soil is well below the
208 critical minimum level required for crop growth (10 mg/kg; (Sims and Johnson, 1991) and Mn
209 deficiencies could be expected. There are no clear guidelines for phytotoxic Mn levels in soils,
210 but application of the WTR in the Cape Flats sand up to rates of 10% (w/w) would bring the
211 Mn concentrations close to the minimum critical level. The compost contained a very high
212 bioavailable As concentration (141 $\mu\text{g}/\text{kg}$), which may be due to pesticide residues in
213 municipal green waste or inclusion of treated wood in the composted material (Adriano, 2001).
214 This compost was selected for its low C and N content. The elevated As was an unexpected
215 property of the widely used compost and although it adds an interesting aspect to the study, the
216 emphasis is on metals in the WTR, rather than metals in an inherently variable compost stream.
217 ~~Despite this high As level, the compost was still used in the trial as it represents the most widely~~
218 ~~available compost material, used by local organic farmers and backyard gardeners (Gibozi,~~
219 ~~2018).~~

220 Table 1 Bioavailable trace element concentrations ($\mu\text{g}/\text{kg}$) in the pot trial materials, together with threshold
 221 limits for metal concentrations in the soil where WTR will be applied (Herselman, 2013)

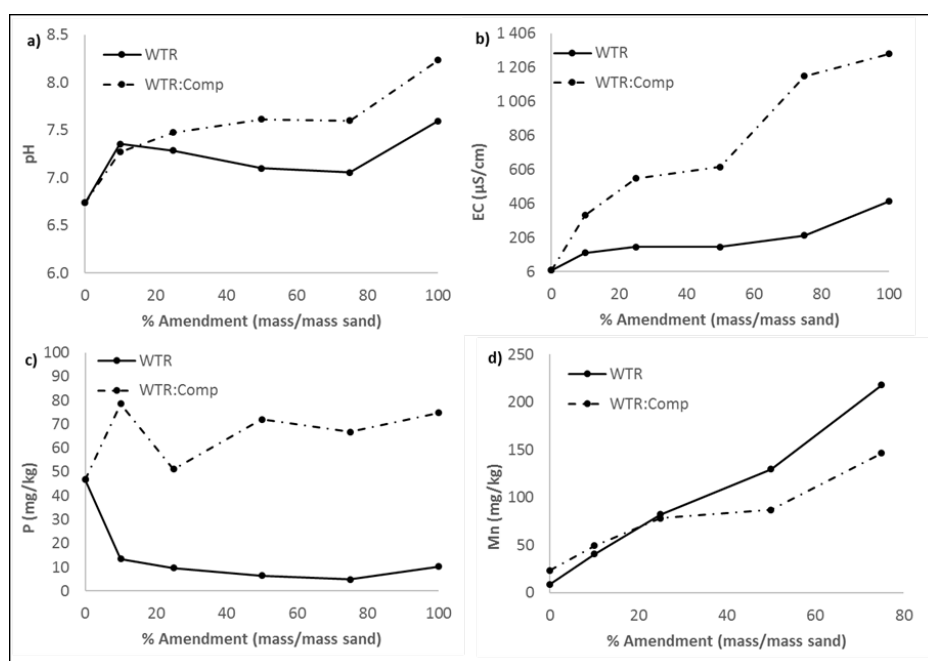
Element	Receiving soil limit	Soil	WTR	Compost
B		31.5	188.7	659.0
Al		208.7	60.3	2473.1
Mn		194	17000	343
Fe		126.8	130.8	1534.2
Ni	1200	3.3	94.7	19.5
Cu	1200	9.8	363.6	113.9
Zn	5000	57.6	100.0	96.3
As	14	1.7	30.1	141.3
Cd	100	0.2	0.5	0.3
Hg	7	0.03	<0.05	0.06
Pb	3500	1.0	1.4	5.1

222

223 3.2 Pre-Trial incubation studies

224 Prior to the pot trial design, 14-day incubations were performed at field water capacity with i)
 225 WTR and ii) a 1:1 WTR-Compost co-amendment added to the sandy soil, at 4-6 application
 226 rates between 0 and 100% (dry w/w). The results of the incubation studies (Figure 1) provide
 227 insights into the effects extreme loadings of WTR and WTR-Compost co-amendments might
 228 have on important soil parameters. Both WTR and the co-amendment increased pH (Figure
 229 1a), which would benefit acid soils, although increasing the pH above 7.5 is undesirable as it
 230 can result in trace element deficiencies (Havlin et al., 2005). The higher pH readings in the
 231 incubation studies, compared to the initial characterization (Supplemental Table S1) is assigned
 232 to longer equilibrium times during the incubation. At higher loadings the 1:1 WTR-Compost
 233 co-amendment exceeded $500 \mu\text{S}/\text{cm}$ (Figure 1b) which is considered the critical EC level (in a
 234 1:5 water extract) where plant growth is affected negatively (Sonmez et al., 2008). The compost
 235 is likely to be the main contributor to salinity with an $\text{EC} > 5000 \mu\text{S}/\text{cm}$ (Supplemental Table
 236 S1). To keep EC within tolerable levels, the 1:1 WTR-Compost co-amendment loadings should
 237 be below 25%. The high P-sorption potential of the WTR is evident from the incubations
 238 (Figure 1c) and increases with WTR application rate in the single amendment. However,

239 compost co-addition increases plant-available P suggesting that the organic matter might
 240 alleviate this limitation to a degree. Bioavailable Mn concentrations increase linearly with
 241 increasing loading rates (Figure 1d). These incubation results suggest that maximal application
 242 rates should be kept below 25% WTR to prevent phytotoxic Mn conditions developing in the
 243 soil. Based on these incubation studies the maximum WTR application rate was set at 12.5%
 244 and the WTR-Comp co-amendment was set at 25%.

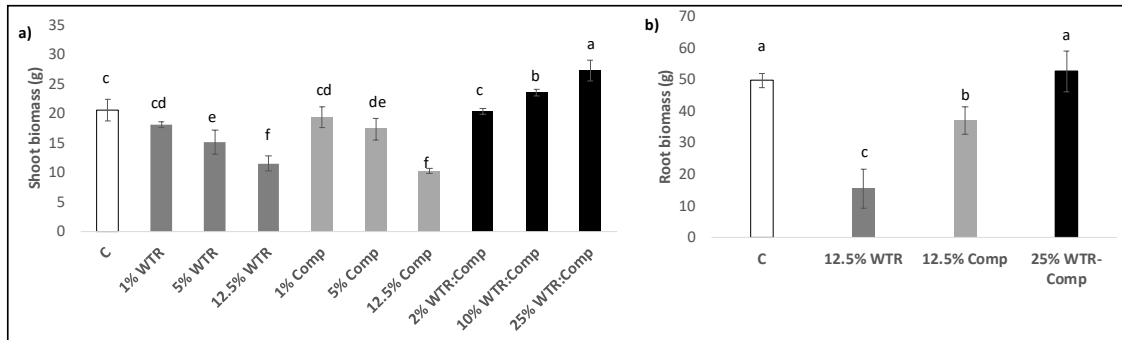


245
 246 Figure 1 Pre-trial incubations, investigating the effect of increasing application rates of WTR and a 1:1 WTR-
 247 Compost (WTR:Comp) co-amendment on (a) pH, (b) EC (c) P (Mehlich III) and d) Mn (Mehlich III). Average
 248 of duplicate incubations shown. **Results repeatable**

249 3.3 Pot Trial: Post-Harvest Plant Physiology and Chemistry

250 The above- and below-ground biomass of the treatments are shown in Figure 2a and b,
 251 respectively. The WTR-Comp co-amendment resulted in significantly higher (up to 33%;
 252 $p < 0.05$) above-ground biomass than the control at the two highest application rates (10 and
 253 25% WTR-Comp). The individual compost and WTR treatments had a significant negative
 254 effect on above-ground biomass (up to 50% lower), with biomass concomitantly decreasing
 255 with increasing amendment rates. The below-ground biomass for the highest amendment

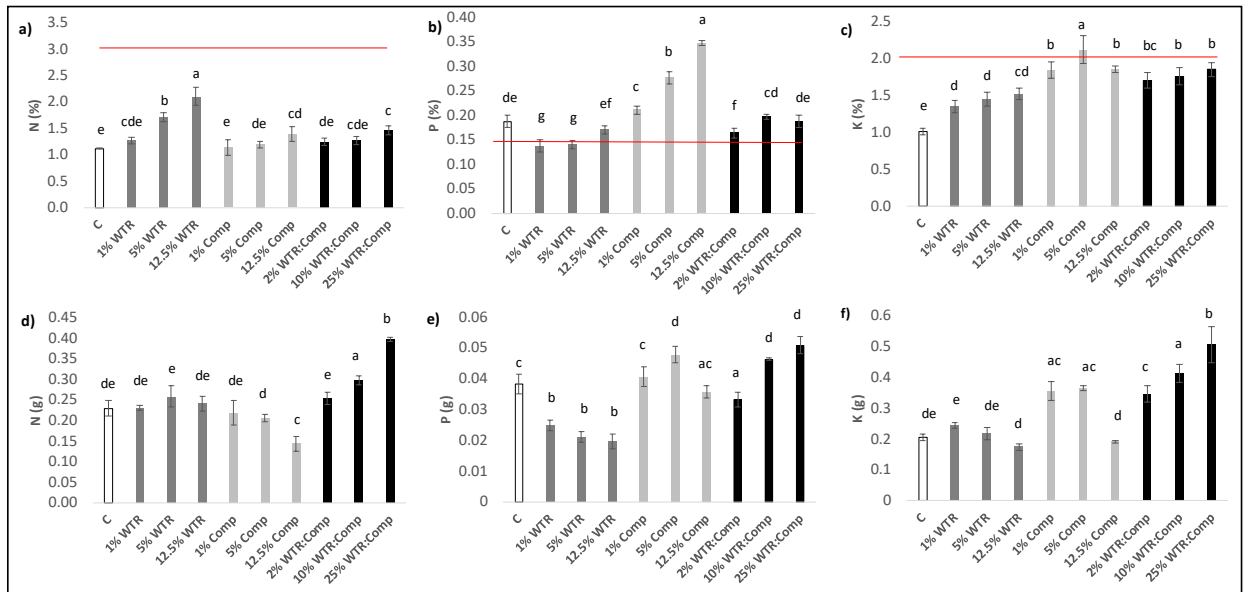
256 loadings showed a similar pattern, significantly lower root biomass in the single amendments
 257 than the control, while the co-amended treatment showed no significant difference to the
 258 control.



259
 260 Figure 2 The effect of single WTR and compost amendments, and co-amendments (WTR:Comp), on plant
 261 growth parameters, (a) total above-ground biomass and (b) root biomass. Bars that do not differ significantly
 262 ($p < 0.05$) contain the same letter.

263 At the end of the trial, plants in all treatments except for the 12.5% WTR started to show N –
 264 deficiency symptoms through older leaf yellowing and senescence despite fertilizer
 265 application. Plants in the 12.5% WTR treatment did not show deficiency symptoms, most likely
 266 due to the fact that this treatment was significantly stunted (Figure 2a) and thus utilized less of
 267 the applied N, confirmed by the leaf N-levels (Figure 3a). Although plants from all treatments
 268 were well below the critical N-level (3%) for wheat (Plank and Donohue, 2000), the 12.5%
 269 WTR treatment had the highest N weight percent, followed by the 5% WTR treatment. The
 270 highest co-amendment (25% WTR-Comp) showed significantly higher (30%; $p < 0.05$) leaf N-
 271 levels than the control, despite these plants being 33% larger. The compost amended treatments
 272 all showed similar leaf N-levels to the control, although plants in the higher loadings were
 273 severely stunted (Figure 2a).

274



275

276 Figure 3 Foliar macronutrient contents of harvested wheat plants as a weight percentage a)-c) and as absolute
 277 accumulation in grams d)-f). Critical macronutrient levels for wheat (Plank and Donohue, 2000) shown by red
 278 lines. Bars that do not differ significantly ($p < 0.05$) contain the same letter.

279 The leaf P-levels showed the opposite trend to the N-levels, with the two lowest single WTR
 280 amendments having significantly lower leaf P-levels than the control while all single
 281 amendment compost treatments had significantly higher P-levels than the control (Figure 3b).
 282 The two highest co-amended treatments did not show a significant difference to the control in
 283 terms of P content. All treatments were above the 0.15% critical level for P in wheat (Plank
 284 and Donohue, 2000), except for the two lowest WTR treatments. The slightly higher P content
 285 of the 12.5% WTR treatment is attributed to smaller plant size. Potassium levels are generally
 286 below the critical level of 2% (Plank and Donohue, 2000) but all treatments significantly
 287 increased the K level compared to the control (Figure 3c).

288

289 The poor plant response to the compost is not surprising, considering the low total and mineral
 290 N content of this material (Supplemental Table S1). The fact that the compost treatments
 291 performed worse than the control suggests that N-immobilisation is taking place in these
 292 treatments. This is also illustrated by the total grams of N taken up by the plants (Figure 3d),

293 which shows the plants in the compost treatment assimilated the lowest amount of nitrogen
294 into their leaves. In contrast, the two highest co-amendments took up significantly more N
295 than the control or the single WTR treatments. The same trend is observed with the absolute
296 amount of P in the leaves (Figure 3e). While the single compost treatments showed the highest
297 weight % P (Figure 3b), the co-amendment treatments showed higher absolute P-levels,
298 because the biomass of these plants was greater. This was also true for K accumulation (Figure
299 3f).

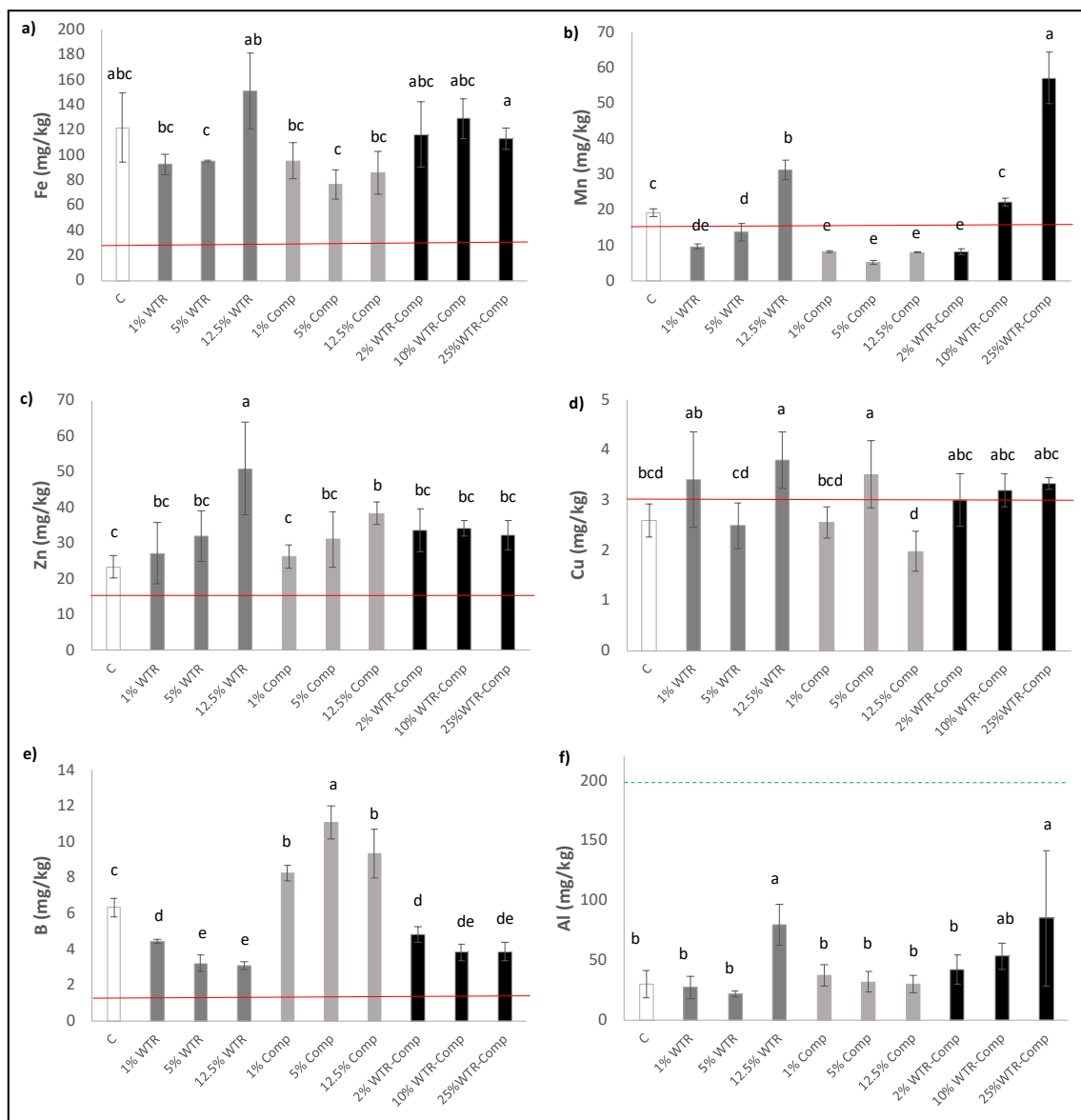
300

301 When interpreting these growth response results in light of the nutrient contents in the compost
302 and WTR it is clear that both amendments are providing different macronutrients, with the
303 WTR adding mineral N while the compost contributes P and K. Although total provision of
304 nutrients by the co-amendment is likely to be the main cause of improved growth, there is also
305 the potential for the organic matter from the compost to sorb to the WTR surface and prevent
306 the fixation of added P to the oxide surfaces (Havlin et al., 2005).

307

308 The foliar micronutrient and Al levels of the wheat plants are given in Figure 4. Foliar Mn in
309 the control is at the lowest critical limit for wheat growth (Figure 4b). Addition of compost
310 with WTR at 25% had the largest effect on foliar Mn, raising the concentration to sufficiency
311 levels (20-150 mg/kg). This increase was significantly greater than addition of WTR alone,
312 indicating a synergistic effect on plant uptake of Mn in the co-amendment. Possible reasons
313 for this synergy include lowering of the redox potential in the soil and addition of Mn-
314 associated microbiomes, which may aid in Mn mobilization in the rhizosphere (Rengel, 2015).
315 Manganese is often flagged as a possible problematic metal in WTR (Novak et al., 2007;
316 Titshall and Hughes, 2005). The incubation experiments also indicate that Mn phytotoxicity
317 might be an issue at higher loadings (Figure 1d). The foliar analysis shows that even at the

318 highest levels of WTR application (12.5%), the foliar Mn concentrations were only at sufficient
 319 levels and far below the toxicity threshold (380 mg/kg) for small grains (Keisling et al., 1984).
 320 Thus, for nutrient poor soils, such as the Cape Flats sands, WTR-Compost co-amendments
 321 could constitute an important source of Mn plant nutrition although careful monitoring would
 322 be required if repeated WTR additions were made to such a sandy soil.-



323
 324 Figure 4 Foliar micronutrient and Al concentrations provided with critical values for wheat production (Plank
 325 and Donohue, 2000) and Al toxicity threshold (Pais and Benton Jones, 1997). Bars that do not differ significantly
 326 ($p < 0.05$) contain the same letter.

327 Aluminium constitutes up to 7.7% of the WTR used in this study, thus Al toxicity in plants was
 328 considered a potential risk when applying the material to an acid soil. Only treatments with the
 329 highest loading of WTR (12.5% WTR and 25% WTR-Comp) showed a significant increase in
 330 foliar concentrations and these were well below (less than half) the Al toxicity level for crops
 331 (Pais and Benton Jones, 1997).

332 3.4 Pot Trial: Bioavailable trace elements

333 Bioavailable TE were measured before and after the pot trial on selected treatments (Table 2).
 334 Before the trial B, Mn, Fe, Ni, Cu and As concentrations significantly increased ($p < 0.05$) in
 335 the 25% WTR-Comp treatment while Al, Zn and Cd concentrations significantly decreased (p
 336 < 0.05) compared to the control. WTR and WTR-Comp treatments significantly decreased Pb,
 337 while compost on its own significantly increased Pb ($p < 0.05$). The increase in TE
 338 bioavailability before the trial is attributed to the higher TE content of the amendments (Table
 339 1), while the decrease of Al, Zn and Cd is most likely due to an increased pH in the soil system
 340 (Figure 1a).

341 Table 2 Trace element concentrations ($\mu\text{g}/\text{kg}$) in 1M NH_4NO_3 extracts of selected soil treatments analysed
 342 before and after the wheat pot trial.

Element	Receiving soil limit ^a	Soil Screening Values ^b (mg/kg)	Control		12.5% Compost		12.5% WTR		25% WTR+Comp	
			Before	After	Before	After	Before	After	Before	After
B			31.5	32.5	215.3	176.7	26.0	26.6	152.7*	100.8
Al			208.7	191.6	146.4*	111.6	28.3	23.9	67.5*	44.3
Mn			194.3	173.8	283.2*	101.9	3931.6*	404.8	2292.6*	473.9
Fe			126.8	110.3	371.8*	309.2	52.7	32.5	252.2*	133.2
Ni	1200	91	3.3	3.4	5.8*	4.7	8.4	7.6	9.9	9.4
Cu	1200	200	9.8	20.1	60.2	57.9	29.5	29.6	37.2	34.3
Zn	5000	3700	57.6	51.3	44.6*	33.0	19.4*	10.2	23.2*	13.6
As	14	5.8	1.7	1.9	29.3	17.4	4.8*	1.7	11.7*	4.1
Cd	100	7.5	0.16	0.18	0.13*	0.08	0.08	0.04	0.09	0.06
Hg	7	1	0.03	0.03	0.04	0.05	0.02	0.01	0.03	0.02
Pb	3500	20	1.02	1.29	1.89*	1.24	0.11	0.11	0.14	0.13

343 ^a. According to (Herselman, 2013) ^b South African Soil screening values for the protection of water sources using a dilution
344 factor of 20 (DEA, 2010)

345 * marks significance between before and after concentrations at a 95% confidence limit

346 When adding a waste to a soil, it is important to consider any mobilizing effects plant growth
347 might have on the bioavailability of metals. The TE either showed no change or significantly
348 decreased in post-trial bioavailability (Table 2). For all the compost- and WTR-treated soils,
349 extractable Mn concentrations were significantly lower after the pot trial. Importantly,
350 phytotoxic Al was not mobilized and either showed little change or decreased during the trial.
351 Plant available As levels were elevated in the compost (Table 1) and for the 12.5% compost
352 treatment levels were beyond the threshold for soils to receive additional WTR (Herselman,
353 2013), both before and after the pot trial. Pre-trial As concentrations (11.7 µg/kg) in the 25%
354 WTR-Comp were significantly lower ($p < 0.05$) than in the pre-trial 12.5% compost treatment
355 (29.3 µg/kg). This is attributed to the capacity of WTR to strongly chemisorb As (McCann et
356 al., 2018; Sarkar et al., 2007) and suggests WTR addition to an As-rich compost could reduce
357 bioavailable As content.

358

359 With the exception of As in the compost treatment, the bioavailable TE measured after the trial
360 were substantially below the maximum extractable threshold for receiving soils (Table 2). ~~This~~
361 ~~means multiple additions of WTR, even at very high loading rates (375 tons WTR/ha), would~~
362 ~~be possible on these sandy soils (Herselman, 2013).~~ In addition, all TE concentrations are far
363 below the soil screening guidelines for the protection of water sources (Table 2) thus the risks
364 of trace metal contamination of ground- and surface water, even at very high WTR application
365 rates, appears low. The maximum rates applied in this trial are unrealistically high (375 tons
366 WTR + 375 tons compost/ha), but indicate multiple applications of WTR at lower rates would
367 keep TE levels within guideline levels, however further work must establish responsible

368 application rates. In addition, elevated As in the compost, highlights the importance of
369 screening the metal content of compost used as a co-amendment.

370 **3.5 Implications for WTR-Compost co-amendments**

371 In African small-scale farming systems, organic residues are often available but are of poor
372 quality with high C:N ratio and/or low total N (Vanlauwe and Giller, 2006). The compost used
373 in this study was of extremely poor quality, with low total C and N contents, high salinity and
374 unacceptably high As levels. Addition of WTR to this compost provided mineral N, increased
375 certain deficient trace elements and decreased the bioavailable As content, creating a more
376 favorable growth medium than compost on its own. The compost, in turn, provided K and
377 countered or reduced P-sorption tendencies of the WTR.

378
379 The mildly acidic sandy soils used in this study are ubiquitous in Africa (Jones et al., 2013)
380 and communities relying solely on these soils for food are at greater risk of malnutrition due to
381 insufficient soil micronutrients (Ceruti et al., 2003). Our results suggest that WTR-Compost
382 co-amendments are a viable option to improve crop productivity where the two materials are
383 abundantly available and within the context of considering transport costs versus economic and
384 social benefits of improved soil function.

385
386 The potential risks associated with land application of wastes are contamination of soil and
387 groundwater resources (Pritchard et al., 2010). Sandy soils lack clays and sesquioxides, which
388 sequester contaminants and often buffer the soil and underlying groundwater against
389 contamination. Such soils, especially the acid variants, are considered high risk for land
390 application of wastes. ~~are poorly buffered and thus are highly susceptible to contamination~~
391 and transfer of contaminants to groundwater sources (Pritchard et al., 2010), therefore the soils
392 used in this study are 'high risk' for land application of wastes. The results obtained here

393 suggest that even at extreme loadings (375 ton WTR/ha), contamination risks from heavy
394 metals are low, although these need to be verified under field conditions using multiple WTR
395 applications.
396 Wheat was used in this study as an indicator crop, in subsistence agriculture leafy greens are
397 frequently grown to supplement the maize staple. Leaf nutrition and metal uptake in edible
398 leaves needs to be determined in assessing the safety of WTR land application. In addition,
399 there are other potential toxicity risks of using WTR in agriculture land application, which are
400 seldom addressed ~~in land application studies~~. ~~Such risks~~ These include microbial contamination
401 from polluted water sources, phyto-uptake and toxicity of micropollutants (pharmaceuticals,
402 pesticides, plasticides, etc.), as well as the toxicity of the chemical additives used in coagulation
403 and flocculation. All of these risks should be investigated before large-scale land applications
404 of WTR are permitted on ~~such~~ susceptible sandy soils.

405 **Acknowledgements**

406 We are grateful to the City of Cape Town's Water & Sanitation Department and the staff at the
407 Scientific Services Branch and Faure Water Treatment Works, particularly Allen Blackenberg,
408 for generous assistance in obtaining the waste material. Opinions expressed and conclusions
409 arrived at, are those of the authors and are not necessarily to be attributed to the City of Cape
410 Town. This work was funded by N8 pump priming fund awards administered by both Durham
411 and Lancaster Universities.

412

413 **Supplementary Material** includes description of basic characterization methods, statistical
414 methods and characterization data for materials used

415 **References**

416 Adriano D.C. (2001) Trace elements in Terrestrial Environments Springer, New York, NY.
417 Barker A.V. (1997) Composition and Uses of Compost, Agricultural Uses of By-Products and
418 Wastes, American Chemical Society. pp. 140-162.

- 419 Basibuyuk M., Kalat D.G. (2004) The use of waterworks sludge for the treatment of
 420 vegetable oil refinery industry wastewater. *Environmental Technology*, 25:373-380.
- 421 Ceruti P., Pooley J., Fey M.V. (2003) Soil nutrient deficiencies (Mseleni Joint Disease) and
 422 dwarfism in Maputaland, South Africa, in: H. C. W. Skinner and A. R. Berger (Eds.), *In*
 423 *Geology and Health: Closing the Gap*, Oxford University Press, New York. pp. 151-
 424 154.
- 425 Dayton E.A., Basta N.T. (2001) Characterization of drinking water treatment residuals for use
 426 as a soil substitute. *Water Environ Res* 73:52-7.
- 427 DEA. (2010) Framework for the management of contaminated land. , May 2010.
- 428 Elliot H.A., Dempsey B.A. (1991) Agronomic effects of land application of water treatment
 429 sludges. *J. Am. Water Works Assoc.* 84:126.
- 430 Elliott H.A., Singer L.M. (1988) Effect of water treatment sludge on growth and elemental
 431 composition of tomato (*Lycopersicon esculentum*) shoots. *Communications in Soil*
 432 *Science and Plant Analysis* 19:345-354. DOI: 10.1080/00103628809367943.
- 433 FSSA. (2007) FSSA fertilizer handbook, Fertilizer Society of South Africa.
- 434 Gibozi T.K.S. (2018) Evaluation of a peri-urban smallholder farmers' soil amendment
 435 practices on soil quality and crop growth, yield and quality, Department of Soil
 436 Science, MSc Thesis. Stellenbosch University.
- 437 Havlin J.L., Beaton J.D., Tisdale S.L., Nelson W.L. (2005) *Soil fertility and Fertilizers: an*
 438 *introduction to nutrient management*. 7th ed. Pearson Prentice Hall, Upper Saddle
 439 River, N.J.
- 440 Heil D.M., Barbarick K.A. (1989) Water Treatment Sludge Influence on the Growth of
 441 Sorghum-Sudangrass. *Journal of Environmental Quality* 18:292-298. DOI:
 442 10.2134/jeq1989.00472425001800030008x.
- 443 Herselman J.E. (2013) Guideline for the utilisation and disposal of water treatment residue.
 444 Water Research commission No. 559/13.
- 445 Hsu W.M., Hseu Z.Y. (2011) Rehabilitation of a Sandy Soil With Aluminum-Water Treatment
 446 Residual. *Soil Sci* 176:691-698.
- 447 Ippolito J.A., Barbarick K.A., Elliott H.A. (2011) Drinking water treatment residuals: a review
 448 of recent uses. *J Environ Qual* 40:1-12.
- 449 Ippolito J.A., Barbarick K.A., Heil D.M., Chandler J.P., Redente E.F. (2003) Phosphorus
 450 retention mechanisms of a water treatment residual. *J Environ Qual* 32:1857-64.
- 451 Jones A., Breuning-Madsen H., Brossard M., Dampha A., Deckers J., Dewitte O., Gallali T.,
 452 Hallett S., Jones R., Kilasara M., Le Roux P., Michéli E., Montanarella L., Spaargaren
 453 O., Thiombiano L., Van Ranst E., Yemefack M., Zougmore R. (2013) *Soil Atlas of*
 454 *Africa*. European Commission, Publications Office of the European Union,
 455 Luxembourg. 176 pp. ISBN 978-92-79-26715-.
- 456 Keisling T.C., Thompson L.F., Slabaugh W.R. (1984) Visual symptoms and tissue manganese
 457 concentrations associated with manganese toxicity in wheat. *Communications in Soil*
 458 *Science and Plant Analysis* 15:537-540. DOI: 10.1080/00103628409367495.
- 459 Matilainen A., Vepsäläinen M., Sillanpää M. (2010) Natural organic matter removal by
 460 coagulation during drinking water treatment: a review. *Advances in colloid and*
 461 *interface science* 159:189-197.
- 462 McCann C.M., Peacock C.L., Hudson-Edwards K.A., Shrimpton T., Gray N.D., Johnson K.L.
 463 (2018) In situ arsenic oxidation and sorption by a Fe-Mn binary oxide waste in soil. *J*
 464 *Hazard Mater* 342:724-731. DOI: 10.1016/j.jhazmat.2017.08.066.

- 465 Mehlich A. (1985) Mehlich 3 soil test extractant: A modification of Mehlich 2 extractant.
 466 Commun. Soil Sci. Plant Anal. 15:1409.
- 467 Mulvaney R.L. (1996) Inorganic Nitrogen in: D. L. Sparks (Ed.), *In Methods of Soil Analysis*,
 468 Soil Sci. Soc. Am., Madison, Wis.
- 469 Norris N., Titshall L.W. (2012) The distribution of inherent phosphorus in fifteen water
 470 treatment residues from South Africa. *Water SA* 38:715-720.
- 471 Novak J.M., Szogi A.A., Watts D.W., Busscher W.J. (2007) Water treatment residuals
 472 amended soils release Mn, Na, S, AND C. *Soil. Sci.* 172:992-1000. DOI:
 473 10.1097/ss.0b013e3181586b9a.
- 474 Pais I., Benton Jones J. (1997) *The Handbook of Trace Elements* St Lucie Boca Raton, Florida.
- 475 Plank C.O., Donohue S.J. (2000) Small Grain-Barley, Oats, Rye, Wheat, in: C. R. Cambell (Ed.),
 476 Reference sufficiency ranges for plant analysis in the southern region of the United
 477 States, Southern Cooperative Series Bulletin no. 394.
- 478 Pritchard D.L., Penney N., McLaughlin M.J., Rigby H., Schwarz K. (2010) Land application of
 479 sewage sludge (biosolids) in Australia: risks to the environment and food crops.
 480 *Water Sci Technol* 62:48-57. DOI: 10.2166/wst.2010.274.
- 481 Rengasamy P., Oades J.M., Hancock T.W. (1980) Improvement of soil structure and plant
 482 growth by addition of alum sludge. *Communications in Soil Science and Plant*
 483 *Analysis* 11:533-545. DOI: 10.1080/00103628009367061.
- 484 Rengel Z. (2015) Availability of Mn, Zn and Fe in the rhizosphere. *Journal of soil science and*
 485 *plant nutrition* 15:397-409.
- 486 Sarkar D., Makris K.C., Vandanapu V., Datta R. (2007) Arsenic immobilization in soils
 487 amended with drinking-water treatment residuals. *Environmental Pollution* 146:414-
 488 419. DOI: <https://doi.org/10.1016/j.envpol.2006.06.035>.
- 489 Schloms B., Ellis F., Lambrechts J.J.N. (1983) Soils of the coastal platform, in: H. Deacon, et
 490 al. (Eds.), *Fynbos palaeoecology: A preliminary synthesis*, South African National
 491 Scientific Programmes Report No 75, South African National Scientific Programmes
 492 Report No 75.
- 493 Sims J.T., Johnson C.V. (1991) Micronutrient soil tests, in: J. J. Mortvedt, et al. (Eds.),
 494 *Micronutrients in Agriculture*, Soil Science Society of America, Madison, WI.
- 495 Sonmez S., Buyuktas D., Okturen F., Citak S. (2008) Assessment of different soil to water
 496 ratios (1:1, 1:2.5, 1:5) in soil salinity studies. *Geoderma* 144:361-369. DOI:
 497 <https://doi.org/10.1016/j.geoderma.2007.12.005>.
- 498 Statistics South Africa. (2016) Provincial profile: Western Cape, Statistics South Africa.
 499 Report 03-01-07.
- 500 Steynberg R.E., Nel P.C., Hammes P.S. (1989) Drought sensitivity of maize (*Zea mays* L.) in
 501 relation to soil fertility and water stress during different growth stages. *South African*
 502 *Journal of Plant and Soil* 6:83-85. DOI: 10.1080/02571862.1989.10634487.
- 503 Titshall L.W., Hughes J.C. (2005) Characterisation of some South African water treatment
 504 residues and implications for land application. *Water SA* 31:299-307.
- 505 USEPA. (2000) Maximum concentration permitted for Land Application: Biosolids
 506 Technology fact sheet Land Application of Biosolids.
- 507 Vanlauwe B., Giller K.E. (2006) Popular myths around soil fertility management in sub-
 508 Saharan Africa. *Agriculture, Ecosystems & Environment* 116:34-46. DOI:
 509 <https://doi.org/10.1016/j.agee.2006.03.016>.
- 510 Western Cape Government. (2017) City of Cape Town: Socio-Economic Profile, Western
 511 Cape Government.

512 **Tables**

513 Table 1 Bioavailable trace element concentrations ($\mu\text{g}/\text{kg}$) in the pot trial materials, together
 514 with threshold limits for metal concentrations in the soil where WTR will be applied
 515 (Herselman, 2013)

Element	Receiving soil limit	Soil	WTR	Compost
B		31.5	188.7	659.0
Al		208.7	60.3	2473.1
Mn		194	17000	343
Fe		126.8	130.8	1534.2
Ni	1200	3.3	94.7	19.5
Cu	1200	9.8	363.6	113.9
Zn	5000	57.6	100.0	96.3
As	14	1.7	30.1	141.3
Cd	100	0.2	0.5	0.3
Hg	7	0.03	<0.05	0.06
Pb	3500	1.0	1.4	5.1

516

517 Table 2 Trace element concentrations ($\mu\text{g}/\text{kg}$) in 1M NH_4NO_3 extracts of selected soil
 518 treatments analysed before and after the wheat pot trial.

Element	Receiving soil limit ^a	Soil Screening Values ^b (mg/kg)	Control		12.5% Compost		12.5% WTR		25% WTR+Comp	
			Before	After	Before	After	Before	After	Before	After
B			31.5	32.5	215.3	176.7	26.0	26.6	152.7*	100.8
Al			208.7	191.6	146.4*	111.6	28.3	23.9	67.5*	44.3
Mn			194.3	173.8	283.2*	101.9	3931.6*	404.8	2292.6*	473.9
Fe			126.8	110.3	371.8*	309.2	52.7	32.5	252.2*	133.2
Ni	1200	91	3.3	3.4	5.8*	4.7	8.4	7.6	9.9	9.4
Cu	1200	200	9.8	20.1	60.2	57.9	29.5	29.6	37.2	34.3
Zn	5000	3700	57.6	51.3	44.6*	33.0	19.4*	10.2	23.2*	13.6
As	14	5.8	1.7	1.9	29.3	17.4	4.8*	1.7	11.7*	4.1
Cd	100	7.5	0.16	0.18	0.13*	0.08	0.08	0.04	0.09	0.06
Hg	7	1	0.03	0.03	0.04	0.05	0.02	0.01	0.03	0.02
Pb	3500	20	1.02	1.29	1.89*	1.24	0.11	0.11	0.14	0.13

519 ^a. According to (Herselman, 2013) ^b South African Soil screening values for the protection of water sources using a dilution
 520 factor of 20 (DEA, 2010)

521 * marks significance between before and after concentrations at a 95% confidence limit

522